

Supplementary information

Single-crystal structure determination from microcrystalline powders (~5 μm) by an orientation attachment mountable on an in-house X-ray diffractometer

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Materials and methods

Preparation of an L-alanine microcrystal suspension

An as-received sample of L-alanine crystals (Wako Pure Chemical Industries, Ltd.) was pulverized using a mortar and pestle and passed consecutively through 125-, 75-, 45-, and 20-mesh sieves. Then, the powders remaining on the 20-mesh sieve (powder A) and the powders passed through (powder B) were collected. Powder A was dispersed in UV light-curable monomer (Kyoritsu Chemical XVL-14, viscosity 12 ± 2 Pa s) to prepare 20 wt% microcrystal suspensions (suspension A). Powder B was also dispersed in the same monomer and allowed to stand for a few weeks to make larger microcrystals sink. Then, the upper part of the suspension was taken (suspension B). The sizes of microcrystals present in each of suspensions A and B determined by microscope were 20–45 μm and <5 μm , respectively. In this study, the UV light-curable monomer was used only for a suspending medium because it formed a stable well-dispersed suspension of microcrystals; photopolymerization was not performed. Each suspension was poured into a glass capillary (2.5 mm ϕ) and placed on a magnetic unit (a Kyoto University KU model $\chi 10-3$) comprising a rotating unit and a pair of neodymium magnets (~ 1 T in the center). The capillary was rotated by a stepping motor controlled by a computer. The shutter was a metal disk with a diameter of 44 mm made of aluminum and copper plates with a thickness of 2 and 1 mm,

respectively, rotated by a stepping motor in synchronization with the rotation of the capillary. The shutter contained two slits with a width of $\Delta\beta = 10^\circ$ separated by 180° .

***In situ* X-ray measurements of a MOMS**

The orientation attachment (magnetic unit and shutter) was mounted on a Rigaku R-AXIS RAPID diffractometer (collimator size = 0.8 mm ϕ , graphite-monochromated Mo K α radiation, crystal-to-detector distance = 127.40 mm) where the goniometers for χ and ϕ angles were removed. The rotation speed ω of the glass capillary was switched between $\omega_s = 0.93$ rpm and $\omega_q = 3.0$ rpm every 90° . The X-ray measurement began 3 h after the start of nonuniform sample rotation, when 3D alignment of the microcrystals was achieved.

Data were collected at room temperature to a maximum 2θ value of 54.8° . The angle α was altered from 0° to 200° and 340° to 360° for suspension A, and from 0° to 90° and 230° to 360° for suspension B in increments of 10° by changing the starting position of the slits in the shutter before each measurement, respectively. The measurement time for suspension A was 3000 s for the ranges between $0-90^\circ$ and $180-200^\circ$ and 930 s for $90-180^\circ$ and 340° to 360° to ensure that the net X-ray irradiated time per unit degree was 8.3 s for every measurement. The measurement time for suspension B was 8000 s for the ranges between $0-90^\circ$ and $230-270^\circ$ and 2480 s for $270-360^\circ$ to ensure that the net X-ray irradiated time per unit degree was 22.2 s for every measurement. Each measurement time was calculated on the basis of the rotation speeds ω_s and ω_q used. In both measurements, a total of 22 diffraction images were recorded corresponding to the number of different starting positions of the slits. A measurement without the shutter was also performed for suspension A.

Single-crystal analyses

Data were processed using RAPID AUTO software and the diffraction spots were indexed. The crystal structure was solved by direct methods^[1] and then expanded using Fourier techniques and refined. All calculations were performed using the CrystalStructure^[2] crystallographic software package except for refinement, which was performed using SHELXL-2013.^[3] The calculation was performed without prior knowledge of the L-alanine crystal. The diffraction spots located beyond $2\theta = 50.9^\circ$ were not used because this area was shadowed by the magnetic unit.

References

- S1. M. C. Burla, R. Caliendo, M. Camalli, B. Carrozzini, G. L. Cascarano, C. Giacovazzo, M. Mallamo, A. Mazzone, G. Polidori and R. Spagna, *J. Appl. Crystallogr.*, 2012, **45**, 357.
- S2. CrystalStructure 4.0: Crystal Structure Analysis Package, Rigaku Corporation, Inc.: Tokyo, Japan, 2000.
- S3. Sheldrick, G. M. *Acta Crystallogr. C*. 2015, **71**, 3.