Synthesis, X-ray characterization, DFT calculations and Hirshfeld surface analysis of Mⁿ⁺ ions (n = 2,3; M = Ni, Cd, Mn, Co and Cu): The role of secondary bonding and steric effects in complexes based on thiosemicarbazone

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Compound	[Ni(L1)(N ₃)]	[Ni(L2)(N ₃)]	$[Cu(L1)(\mu-Cl)]_2[Cu(L1)(Cl)]_2$	$[Cu(L3)(\mu-N_3)]_2$
Empirical formula	C ₁₃ H ₁₁ N ₇ NiS	C ₁₄ H ₁₃ N ₇ NiS	$C_{52}H_{44}Cl_4Cu_4N_{16}S_4$	$C_{28}H_{22}Cu_2N_{14}S_2$
Formula weight	356.06	370.08	1417.23	745.80
Temperature / K	293	293	150	100
Wavelength / Å	0.71073	0.71073	0.71073	0.71073
Crystal system	Triclinic	Monoclinic	Monoclinic	Orthorhombic
Space group	Pl	C2/c	$P2_1/n$	Pbca
Unit cell dimensions				
<i>a</i> / Å	6.812(3)	11.0428(8)	13.2776(5)	13.7877(7)
b / Å	8.063(3)	12.3443(5)	15.4640(6)	12.9538(6)
<i>c</i> / Å	13.349(5)	22.9690(11)	13.4326(5)	15.6337(8)
α / °	97.10(3)	—	-	—
β/°	96.91(3)	94.927(5)	102.185(1)	—
γ/°	99.02(3)	—	-	—
Volume / Å ⁻³	711.2(5)	3119.5(3)	2695.91(18)	2790.4(2)
Ζ	2	8	2	4
Calc. density / Mg/m ³	1.663	1.576	1.746	1.775
Absorp. coefc. / mm ⁻¹	1.517	1.387	1.966	1.725
<i>F</i> (000)	364	1520	1432	1512
Crystal size	$0.19 \times 0.15 \times 0.06$	$0.15 \times 0.10 \times 0.05$	$0.10 \times 0.06 \times 0.04$	$0.31 \times 0.15 \times 0.13$
θ range / °	2.82-28.88	3.76-27.00	2.03–27.14	2.52–29.13
Limiting indices / h,k,l	-9/9, -10/10, -17/17	-13/14, -15/15, -29/29	-17/16, -18/19, -15/17	-15/18, -14/17, -21/20
Refl. collect/unique (R _{int})	15636/3455 (0.0264)	11052/3395 (0.0342)	29992/5950 (0.0560)	31742/3759 (0.0325)
Completeness θ / °	97.8/28.00	99.4/27.00	99.6/27.14	100.0/29.13
Absorp. correct.	Analytical	Multi-scan	Multi-scan	Multi-scan
Max. /min. transm.	0.9145/0.7614	0.934/0.819	0.7455/0.6845	0.7459/0.6131
Data / parameters	3455/199	3395/213	5950/361	3759/209
Goodness-of-fit on F ²	1.037	0.914	1.005	1.082
Final <i>R</i> indices	$R_1 = 0.0294,$	$R_1 = 0.0277,$	$R_1 = 0.0334$	$R_1 = 0.0226,$
	$wR_2 = 0.0625$	$wR_2 = 0.0601$	$wR_2 = 0.0673$	$wR_2 = 0.0689$
<i>R</i> indices (all data)	$R_1 = 0.0441,$	$R_1 = 0.0551$	$R_1 = 0.0599$	$R_1 = 0.0341,$
	$wR_2 = 0.0674$	$wR_2 = 0.0848$	$wR_2 = 0.0754$	$wR_2 = 0.0733$
Largest dif. peak/hole	0.242/-0.231	0.283/-0.239	0.467/-0.355	0.839/-0.318

 $\textbf{Table S1. Crystal data and structure refinement for [Ni(L1)(N_3)] (1), [Ni(L2)(N_3)] (2), [Cu(L1)(\mu-Cl)]_2[Cu(L1)Cl]_2 (3) and [Cu(L3)(\mu-N_3)]_2 (4). Crystal data and structure refinement for [Ni(L1)(N_3)] (1), [Ni(L2)(N_3)] (2), [Cu(L1)(\mu-Cl)]_2[Cu(L1)Cl]_2 (3) and [Cu(L3)(\mu-N_3)]_2 (4). Crystal data and structure refinement for [Ni(L1)(N_3)] (1), [Ni(L2)(N_3)] (2), [Cu(L1)(\mu-Cl)]_2[Cu(L1)Cl]_2 (3) and [Cu(L3)(\mu-N_3)]_2 (4).$

Compound	[Mn(L1) ₂]·EtOH	[Cd(L1) ₂]·MeOH	$[Co(L2)_2]\cdot N_3$
Empirical formula	$C_{28}H_{28}MnN_8OS_2$	$C_{27}H_{26}CdN_8OS_2$	$C_{28}H_{26}CoN_{11}OS_2$
Formula weight	611.64	655.08	639.65
Temperature / K	100	100	273
Wavelength / Å	0.71073	0.71073	0.71073
Crystal system	Monoclinic	Monoclinic	Monoclinic
Space group	C2/c	C2/c	$P2_{1}/c$
Unit cell dimensions			
<i>a</i> / Å	14.5132(9)	14.0111(11)	9.1405(4)
b / Å	18.7859(9)	19.8804(119	24.3129(12)
c / Å	11.3833(6)	10.7121(8)	15.1947(6)
α/°	-	_	_
β/°	111.460(2)	108.500(3)	120.285(4)
γ/°	-	_	_
Volume / Å ⁻³	2888.4(3)	2829.6(3)	2915.9(2)
Ζ	4	4	4
Calc. density / Mg/m ³	1.407	1.538	1.457
Absorp. coefc. / mm ⁻¹	0.639	0.956	0.772
F(000)	1268	1328	1320
Crystal size	$0.25 \times 0.20 \times 0.18$	$0.14 \times 0.12 \times 0.10$	$0.25 \times 0.24 \times 0.22$
θ range / °	2.90–26.42	2.34–26.43	2.28-28.24
Limiting indices / h,k,l	-18/18, -23/23, -14/14	-17/17, -24/23, -13/13	-12/11, -32/29, -20/20
Refl. collect/unique (R _{int})	20205/2960 (0.0244)	19776/2907 (0.0258)	11180/6673 (0.0564)
Completeness θ / \circ	99.8/25.00	99.4/25.24	99.7/25.00
Absorp. correct.	Multi-scan	Multi-scan	Multi-scan
Max. /min. transm.	0.8937/0.8566	0.9106/0.8780	0.8485/0.8304
Data / parameters	2960/201	2907/182	6673/381
Goodness-of-fit on F^2	1.085	1.048	0.895
Final <i>R</i> indices	$R_1 = 0.0365,$	$R_1 = 0.0486,$	$R_1 = 0.0462,$
	$wR_2 = 0.1025$	$wR_2 = 0.1272$	$wR_2 = 0.1056$
<i>R</i> indices (all data)	$R_1 = 0.0404,$	$R_1 = 0.0534,$	$R_1 = 0.1332,$
	$wR_2 = 0.1053$	$wR_2 = 0.1323$	$wR_2 = 0.1339$
Largest dif. peak/hole	0.876/-0.516	2.801/-1.084	0.417/-310

$\textbf{Table S1 (cont)}. Crystal data and structure refinement for [Mn(L1)_2] \cdot EtOH (\textbf{5}), [Cd(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot EtOH (\textbf{5}), [Cd(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot EtOH (\textbf{5}), [Cd(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot EtOH (\textbf{5}), [Cd(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot EtOH (\textbf{5}), [Cd(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot EtOH (\textbf{5}), [Cd(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot EtOH (\textbf{5}), [Cd(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot EtOH (\textbf{5}), [Cd(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot EtOH (\textbf{5}), [Cd(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot EtOH (\textbf{5}), [Cd(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot EtOH (\textbf{5}), [Cd(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot EtOH (\textbf{5}), [Cd(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot N_3 (\textbf{7}). \\ \textbf{Crystal data and structure refinement for [Mn(L1)_2] \cdot MeOH (\textbf{6}) and [Co(L2)_2] \cdot MeOH (\textbf{6})$

Compound	[Ni(L1)(N ₃)]	$[Ni(L2)(N_3)]$	[Cu(L1)(Cl)]	$[Cu(L1)(\mu-Cl)]_2$	$[Cu(L3)(\mu-N_3)]_2$
M-N _{azm}	1.8459(15)	1.850(2)	1.969(2)	1.967(2)	1.9596(14)
M-X	1.8773(17)	1.882(2)	2.2135(8)	2.2535(7)	2.3761(15)
M-N _{py}	1.9248(17)	1.932(2)	2.018(2)	2.019(2)	2.0379(14)
M-S	2.1507(10)	2.1480(8)	2.2502(8)	2.2628(8)	2.0210(14) ^b
$M-X_p^a$				2.811397(12)	1.9600(14)
N _{azm} -M-X	176.50(7)	175.72(11)	177.49(7)	173.59 (7)	171.17(6)
N _{azm} -M- N _{py}	83.52(7)	83.42(9)	80.93(9)	80.82(9)	79.35(6)
N _{azm} -M-S	86.61(6)	86.70(7)	83.43(7)	83.08(7)	79.22(5) ^b
X-M-N _{py}	93.75(7)	94.12(10)	97.32(7)	87.65(7)	93.86(5)
X-M-S	96.14(6)	95.74(8)	98.36(3)	97.51(3)	96.29(5) ^b
N _{py} -M-S	170.10(5)	170.13(7)	164.29(7)	162.15(7)	156.25(6) ^b
N _{azm} -M-X _p				91.580(1)	90.04(5)
N _{py} -M-X _p				89.865(1)	95.62(6)
X-M-X _p				94.643(1)	82.97(6)
S-M-X _p				98.253(1)	106.88(6) ^b

 $\textbf{Table S2}. Selected bond lengths [Å] and bond angles [°] for [Ni(L1)(N_3)] (1), [Ni(L2)(N_3)] (2), [Cu(L1)(\mu-Cl)]_2[Cu(L1)Cl]_2 (3) and [Cu(L3)(\mu-N_3)]_2 (4).$

* N_{azm} = azomethine atom, N_{py} = pyridine atom,; ^{a)} bridging ligand atom (Cl in **3** or N in **4**); ^{b)}N1 instead of S

Compound	[Mn(L1) ₂]·EtOH	[Cd(L1) ₂]·MeOH	$[Co(L2)_2] N_3$
M-N _{azm}	2.2427(17)	2.342(3)	1.880(3)
M-N _{py}	2.2696(17)	2.406(3)	1.969(3)
M-S	2.5092(6)	2.5675(12)	2.2156(10)
M-N _{azm} ^{(a}	2.2427(17)	2.342(3)	1.888(3)
M-N _{py} ^{(a}	2.2696(17)	2.406(3)	1.955(3)
M-S ^(a)	2.5092(6)	2.5675(12)	2.2274(10)
N _{azm} -M- N _{py}	72.88(6)	69.89(11)	82.21(11)
N _{azm} -M-S	75.40(4)	74.57(8)	86.07(8)
N _{azm} -M-N _{azm} ^{(a}	160.37(9)	157.96(16)	179.46(11)
N _{azm} -M-N _{pv} ^{(a}	92.72(6)	93.39(11)	97.88(12)
N _{azm} -M-S ^(a)	118.58(4)	120.66(8)	93.68(8)
N _{py} -M-S	148.21(4)	144.45(9)	168.02(9)
N _{py} -M-N _{azm} ^{(a}	92.72(6)	93.39(11)	97.62(12)
N _{py} -M-N _{py} ^{(a}	87.27(8)	83.75(15)	90.18(11)
N _{py} -M- S ^{(a}	96.37(4)	97.81(8)	91.14(8)
S-M-N _{azm} ^{(a}	118.58(4)	120.65(8)	94.11(8)
S-M-N _{py} ^{(a}	96.37(4)	97.81(8)	88.95(8)
S-M-S ^(a)	96.80(3)	100.78(6)	92.10(4)
N _{azm} ^{(a} -M-N _{py} ^{(a}	72.88(6)	69.89(11)	82.64(12)
N _{azm} ^{(a} -M-S ^(a)	75.40(4)	74.57(8)	85.81(9)
$N_{pv}^{(a)}-M-S^{(a)}$	148.21(4)	144.45(9)	168.44(9)

Table S2 (cont.). Selected bond lengths [Å] and bond angles [°] for $[Mn(L1)_2]$ ·EtOH (5), $[Cd(L1)_2]$ ·MeOH (6) and $[Co(L2)_2] N_3$ (7).

^{a)}In **5** and **6**, symmetry transformation used to generate equivalent atoms: -x, y, -z+1/2. In **7**, second ligand molecule.

Compound	D-H····A*	D–H	H···A	D····A	∠DHA
$[Ni(L1)(N_3)]$	N_{ta} -H···N _{az} ^t	0.86	2.19	3.041(3)	173.4
t: -x+1, -y, -z+1				× /	
$[Ni(L2)(N_3)]$	N_{ta} -H \cdots N_{az} ^t	0.84	2.23	3.058(4)	167.0
	C7-H7B····S1ª	0.96	2.86	3.735(3)	152.0
t: -x+1, -y, -z+1; a: x-1/2, y+1/2, z					
$[Cu(L1)(\mu-Cl)]_2[Cu(L1)Cl]_2$	N_{ta} -H \cdots S1 ^a	0.88	2.77	3.570(2)	151.2
	N_{ta} -H \cdots S2 ^b	0.88	2.63	3.508(2)	172.3
	C2-H2…Cl2 ^b	0.95	2.82	3.579(3)	137.0
	С2-Н2…Ѕ2Ҍ	0.95	2.93	3.780(1)	149.0
	C15-H15…Cl1 ^a	0.95	2.92	3.613(1)	131.3
a: x, y, z+1; b: x, y, z-1;					
$[Cu(L3)(\mu-N_3)]_2$	C14-H14B…S1 ^a	0.98	2.99	3.871(1)	149.7
a: -x, y+1/2, -z+1/2					
$[Mn(L1)_2]$ ·EtOH	N_{ta} -H \cdots S2 ^a	0.86	2.60	3.368(2)	149.4
	$C4-H4A\cdots S2^{b}$	0.95	2.95	3.625(4)	129.5
	C14-H14B····N _{hz} ^c	0.99	2.67	3.615(12)	158.6
a: -x, -y+1, -z; b: x+1/2, -y+1/2, z+1/2; c: x, y, z+	1			~ /	
$[Cd(L1)_2]$ ·MeOH	N_{ta} -H···S1 ^a	0.88	2.82	3.472(4)	132.0
	C6-H6A…O10 ^b	0.95	2.60	3.485(9)	154.8
	C13-H13A…S1c	0.95	2.99	3.884(5)	158.0
	C1-H1A····O10 ^d	0.95	2.57	3.497(9)	164.2
a: -x+1, -y+1, -z+1; b:-x+1, y, -z+3/2; c: x, -y+1,	z-1/2; d: -x+1/2, -y+1/2, -z+2			× /	
$[Co(L2)_2]\cdot N_3$	N _{ta} -H4A····N _{az}	0.86	2.06	2.915(5)	171.8
	N_{ta} -H8 $\cdots N_{az}^{t}$	0.86	2.18	2.950(6)	149.0
t: x. $-v+1/2$, $z+1/2$	· · · · · · · · · · · · · · · · · · ·				

Table S3. Hydrogen bond parameters [Å, °] for $[Ni(L1)(N_3)]$ (1), $[Ni(L2)(N_3)]$ (2), $[Cu(L1)(\mu-Cl)]_2 [Cu(L1)Cl]_2$ (3) and $[Cu(L3)(\mu-N_3)]_2$ (4), $[Mn(L1)_2]$ ·EtOH (5), $[Cd(L1)_2]$ ·MeOH (6) and $[Co(L2)_2]N_3$ (7).

* N_{ta} = thioamide atom, N_{az}^{t} = azide terminal atom, N_{hz} = hydrazine atom,

Compound	$\pi \dots \pi$ intoractions		~
Compound	π π interactions $C_{\alpha}(1) \cdots C_{\alpha}(1)^{a}$	2608(2)	u 0
	$Cg(1) = Cg(1)^{a}$	3.008(2)	0
	$Cg(1) = Cg(2)^{a}$	3.570(2)	0 24
$[Ni(L1)(N_3)]$	$Cg(1) \cdots Cg(2)^{*}$	3.535(2)	0.34
	$Cg(1)\cdots Cg(2)^{o}$	3.6/1(2)	0.34
	Ring-metal interactions	$Cg(1) \cdots Ni(J)$	p
	$Cg(1) \cdots Ni(1)^{a}$	3.574	18.34
	$Cg(1)\cdots Ni(1)^{o}$	3.994	33.51
Cg(1): ring (N11/S3/C3/N2/N1); $Cg(2)$: ring (N11/N1/C11/C a = 1-x, 1-y, 1-z; b = 2-x, 1-y, 1-z	C12/N113)		
	$\pi^{\dots}\pi$ interactions	Cg(I)····Cg(J)	α
	$Cg(1)\cdots Cg(1)^{a}$	3.576(1)	0
	$Cg(1)\cdots Cg(2)^{a}$	3.735(1)	0.42
	$Cg(1)\cdots Cg(3)^{b}$	3.624(2)	0.44
$[Ni(L2)(N_3)]$	$Cg(2)\cdots Cg(2)^{b}$	3.578(2)	0
	$Cg(2)\cdots Cg(3)^{b}$	3.651(2)	0.08
	Ring-metal interactions	Cg(I)···Ni(J)	β
	Cg(1)···Ni(1) ^a	3.775	23.02
	Cg(3)···Ni(1) ^b	3.863	27.60
Cg(1): ring (Ni1/S1/C8/N3/N2); Cg(1): ring (Ni1/N1/C5/C a = 1/2 - r / 1/2 - v / 1 - r = 1 - r / 1 - v / 1 - r	6/N2); Cg(3): ring (N1/C1/C2/C3/C4/C5)		
a 1/2 x, 1/2 y, 1 2, 0 1 x, 1 y, 1 2	$\pi \cdots \pi$ interactions	Cg(I)····Cg(J)	α
	$Cg(3)\cdots Cg(5)^{a}$	3.554(2)	3.51
	$Cg(4)\cdots Cg(5)^{a}$	3.550(2)	3.33
	$Cg(4)\cdots Cg(9)^{b}$	3.813(2)	4.62
	$Cg(5)\cdots Cg(7)^{c}$	3.323(2)	7.68
	$Cg(6)\cdots Cg(5)^{c}$	3.775(1)	6.00
$[Cu(LI)(\mu-Cl)]_2[Cu(LI)Cl]_2$	$Cg(7)\cdots Cg(9)^d$	3.619(2)	1.19
	$Cg(8)\cdots Cg(9)^d$	3.739(2)	1.45
	Ring-metal interactions	Cg(I)····Cu(J)	В
	Cg(5)···Cu(1) ^c	3.688	33.96
	Cg(5)····Cu(2) ^e	3.943	31.90

Table S4. Intermolecular interaction parameters (Å, °)

	Cg(9)···Cu(1) ^f	3.767	27.61
Cg(3): ring (Cu2/N7/C21/C22/N8); Cg(4): ring (N8/			
Cg(6): ring (Cu1/S1/C7/N2/N3); Cg(7): ring (Cu1/N	[3/C8/C9/N4); Cg(8): ring (N4/C9/C10/C11/C12/C13);		
Cg(9): ring (C1/C2/C3/C4/C5/C6)			
a = 3/2-x, -1/2+y, 3/2-z; b = 2-x, -y, 1-z; c = 2-x, 1-y	v, 1-z; d = 3/2-x, 1/2+y, 1/2-z; e = 3/2-x, 1/2+y, 3/2-z; t = 3/2-x, -1/2+y, 1/2-z		
	$\pi^{\dots}\pi$ interactions	$Cg(I)\cdots Cg(J)$	α
$[C_{11}(I_2)(u, N_1)]$	$Cg(4)$ ···· $Cg(5)^a$	3.822(1)	18.59
$[Cu(LS)(\mu - N_3)]_2$	Ring-metal interactions	Cg(I)····Cu(J)	β
	Cg(6)···Cu(1) ^a	3.700	42.98
Cg(4): ring (S1/C1/N1/C8/C13); Cg(5): ring (N4/C3 a = $1/2$ -x, $-1/2$ +y, z	/C4/C5/C6/C7); Cg(6): ring (C8/C9/C10/C11/C12/C13)		

 $Cg(I) \cdots Cg(J)$: Distance between ring centroids; α : Dihedral angle between planes I and J; β : Angle vector and normal to plane I.

Table S4 (cont.). Intermolecular interaction parameters (Å, °)
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Compound	$\pi \cdots \pi$ interactions	Cg(I)···Cg(J)	α
	$Cg(2)\cdots Cg(5)^{a}$	3.769(2)	3.44
	$Cg(4)\cdots Cg(5)^{b}$	3.769(2)	3.44
	$Cg(5)\cdots Cg(5)^{a}$	3.784(1)	0
$[Mn(L1)_2] \cdot EtOH$	$C-H\cdots\pi$ -ring interactions	Cg(I)····H	γ
	$C1-H1A\cdots Cg(2)^{c}$	2.95	17.19
	$C1-H1A\cdots Cg(3)$	2.95	17.19
	$C3-H3A\cdots Cg(6)^d$	2.82	18.12
Cg(2): ring (Mn1/N1/C5/C6/N2); C Cg(5): ring (N1/C1/C2/C3/C4/C5); a = 1/2-x, 1/2-y, 1-z; b = -1/2+x, 1/	g(3): ring (Mn1/S2a/C7a/N3a/N2a); Cg(4): ring (Mn1/N1a/C Cg(6): ring (C8/C9/C10/C11/C12/C13) '2-y, -1/2+z; c = -x, y, 1/2-z; d = 1/2-x, -1/2+y, 1/2-z	C5a/C6a/N2a);	
	$\pi^{\dots}\pi$ interactions	Cg(I)···Cg(J)	α
$[C_{J}(I_{1})] M_{2}OU$	$Cg(2)\cdots Cg(5)^{a}$	3.785(2)	2.73
$[Ca(L1)_2]$ -MeOH	$Cg(4)\cdots Cg(5)^{b}$	3.785(2)	2.73
	$Cg(5)\cdots Cg(5)^{a}$	3.647(2)	0
Cg(2): ring (Cd1/N1/C5/C6/N2); Cg a = 3/2-x, 1/2-y, 2-z; b = -1/2+x, 1/	g(4): ring (Cd1/N1a/C5a/C6a/N2a); Cg(5): ring (N1/C1/C2/C '2-y, -1/2+z	C3/C4/C5)	
	<i>C-H</i> ··· π -ring interactions	Сд(І)…Н	γ
	C1-H1···Cg(2)	2.94	25.39
$[Co(L2)_2] \cdot N_3$	C1-H1Cg(4)	2.85	20.51
	C15-H15···Cg(1)	2.87	21.70
	C15-H15····Cg(3)	2.89	22.06
Cg(1): ring (Co1/S1/C8/N3/N2); Cg Cg(4): ring (Co1/N5/C19/C20/N6)	g(2): ring (Co1/S2/C22/N7/N6); Cg(3): ring (Co1/N1/C5/C6/	N2);	
$Cg(1) \cdots Cg(J)$: Distance betw	een ring centrolas; α : Dinedral angle between p	anes I and J, p: Angle vector and normal to plane I.	



Fig. S1. Molecular pairing for the compounds 1 (right) and 2 (left) showing the intermolecular hydrogen bonding.



Fig. S2. Crystal packing diagram for the compound **1** showing the intermolecular π - π stacking.



Fig. S3. Crystal packing diagram for the compound 2 showing the π - π stacking interactions.



Fig. S4. Crystal packing diagram for compound 3 showing the intermolecular interactions and the supramolecular synthons.



Fig. S5. Crystal packing diagram for compound 4 showing: a) no classical hydrogen bonds, b) π - π stacking interactions and c) intermolecular interactions between the phenyl ring and the copper atom.



Fig. S6. Crystal packing of the molecules in 5: a) detail of the chains viewed along the *c* axis, including supramolecular heterosynthons of $R_2^2(8)$ motif and b) view along *a* axis showing the π - π stacking interactions and detail of interactions between the pair of pyridine rings.



Fig. S7. Crystal packing of the molecules in 6: a) detail of the chains viewed along the *c* axis, including supramolecular heterosynthons of $R_2^2(8)$ motif and b) view along *a* axis showing the π - π stacking interactions and detail of interactions between the pair of pyridine rings.



Fig. S8. Crystal packing of 7: a) details of anion-cation-anion hydrogen bonds interactions,
b) details of cation-anion-cation hydrogen bonds interactions, c) view of infinite
one-dimensional chains along the c axis and d) details on the C-H…chelate ring interactions.



Fig. S9. Hirshfeld surface of 1 (a) and 2 (b) mapped with shape index function. Areas marker with ovals indicate stacking interactions



Fig. S10. Decomposed fingerprint plots of 1: a) $H \cdots H$, b) $H \cdots C$, c) $H \cdots N$ and d) $C \cdots C$.



Fig. S11. Decomposed fingerprint plots of **2**: a) $H \cdots H$, b) $H \cdots C$, c) $H \cdots N$ and d) $H \cdots S$.



Fig. S12. Hirshfeld surface of **3** mapped with shape index function. Areas marked with ovals indicate stacking interactions.



Fig. S13. Decomposed fingerprint plots for **3**: a) $H \cdots H$, b) $H \cdots S$, c) $H \cdots Cl$ and d) $C \cdots C$.



Fig. S14. Hirshfeld surface of **4** mapped with d_{norm} function.



Fig. S15. Decomposed fingerprint plots of **4**: a) $H \cdots H$ and b) $H \cdots C$.



Fig. S16. Hirshfeld surface of **5** mapped with shape index function. Ovals indicate stacking interactions



Fig. S17. Decomposed fingerprint plots of **5**: a) $H \cdots H$, b) $H \cdots C$, c) $H \cdots N$ and d) $H \cdots S$.



Fig. S18. Hirshfeld surface of **6** mapped with shape index function. Ovals indicate stacking interactions.



Fig. S19. Decomposed fingerprint plots of **6**: a) $H \cdots H$, b) $H \cdots C$, c) $H \cdots S$ and d) $H \cdots O$.



Fig. S20. Hirshfeld surface of 7 mapped with shape index function. Ovals indicate stacking interactions, while arrows indicate C-H $\cdots\pi$ interactions



Fig. S21. Decomposed fingerprint plots of 7: a) $H \cdots H$, b) $H \cdots C$ and c) $H \cdots N$