

Electronic Supporting Information (ESI)

Correlating the melting point alteration with the supramolecular structure in Aripiprazole drug cocrystals.

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Table S1. Important vibrational modes observed in aripiprazole and its cocrystals. All peak positions are referred in cm^{-1} unit scale.

Solid form	C=O stretch	N-H stretch	O-H stretch	C-H stretch	C=C stretch
Aripiprazole	1679	3471 3322	-	3196	1625
Aripiprazole-Resorcinol, I	1653	3298	3584 , 3487 3000-3500 (broad)	3198	1626
Aripiprazole-Catechol, II	1673	3316	3633, 3528 3000-3500 (broad)	3198	1626
Aripiprazole-Hydroquinone, III	1663	3319	3000-3500 (broad)	3197	1626
Aripiprazole-Pyrogallol, IV	1652	3295	3659, 3555 3000-3500 (broad)	3197	1627
Aripiprazole-Phloroglucinol, V	1652	3289	3646, 3575, 3491 , 3000-3500 (broad)	3198	1626

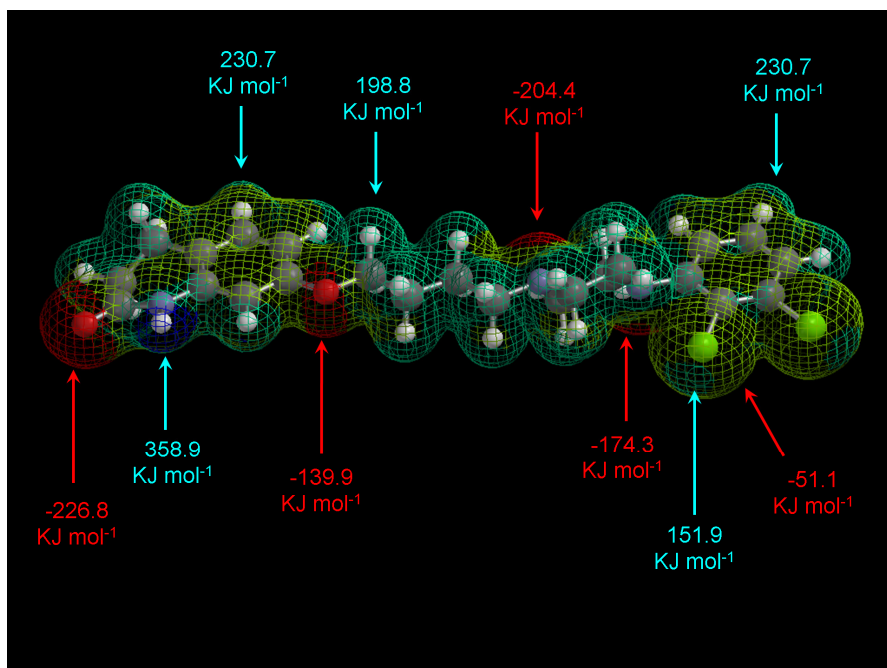


Fig.S1 Electrostatic surface potential (ESP) map on the energy minimized structure of aripiprazole. The red regions indicate electron rich and blue region indicates electron deficient and the colors from blue to green forms the continuum. The chlorine atoms play a dual role of both donor and acceptor due to the anisotropy of the electron distribution on the chlorine atom with $-51.1 \text{ KJ mol}^{-1}$ at the periphery and $+151.9 \text{ KJ mol}^{-1}$ at the centre of the halogen atom.

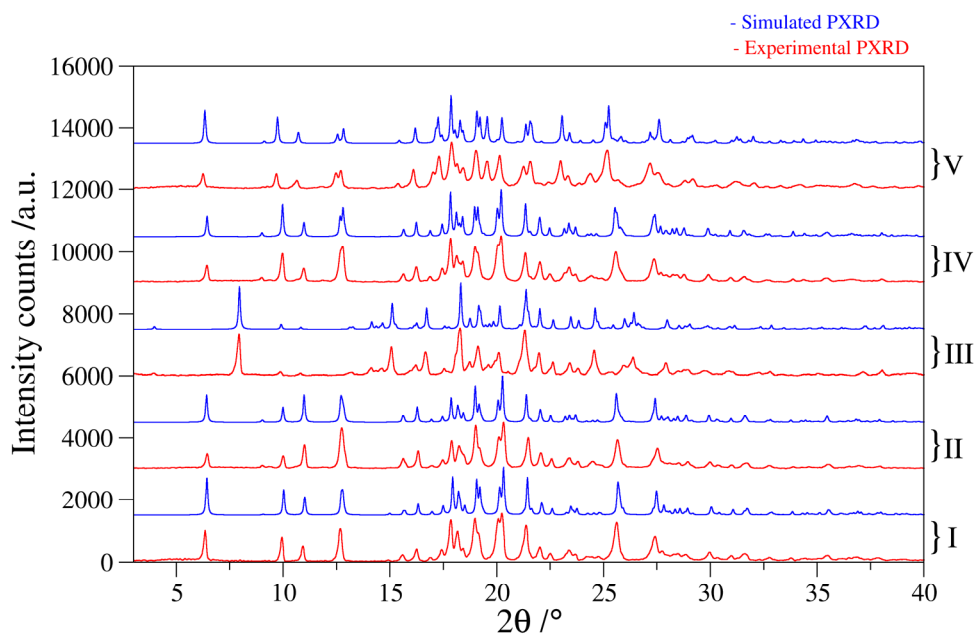
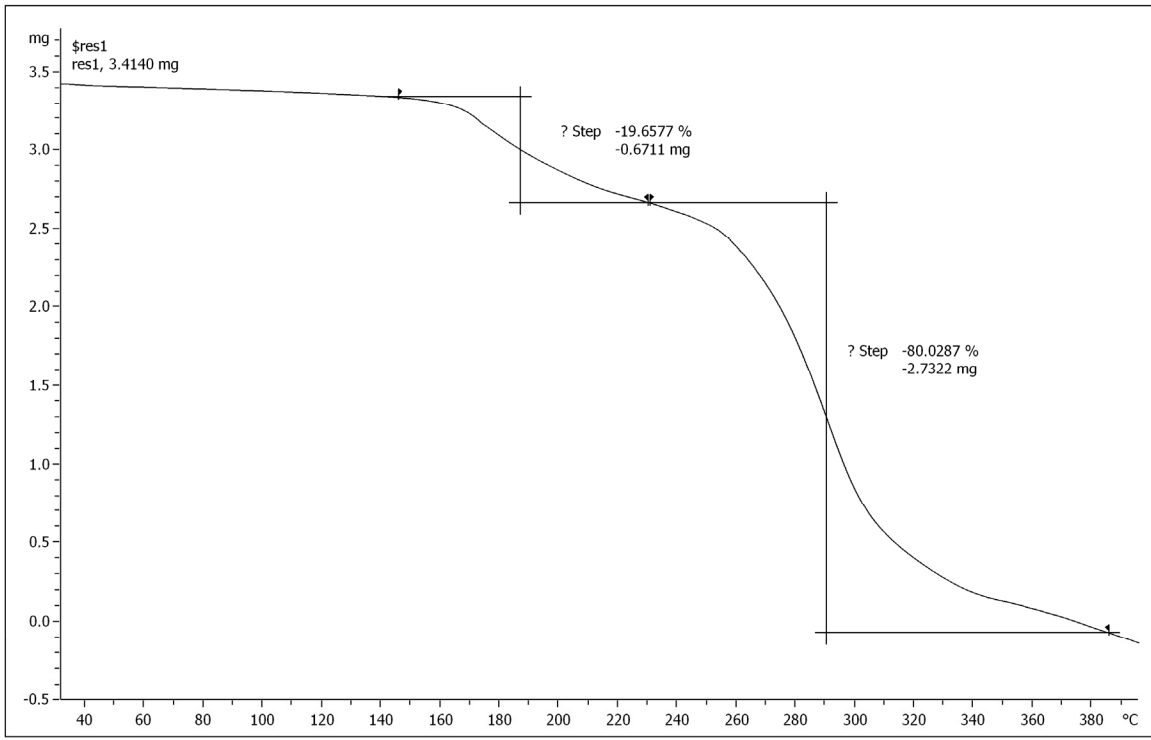


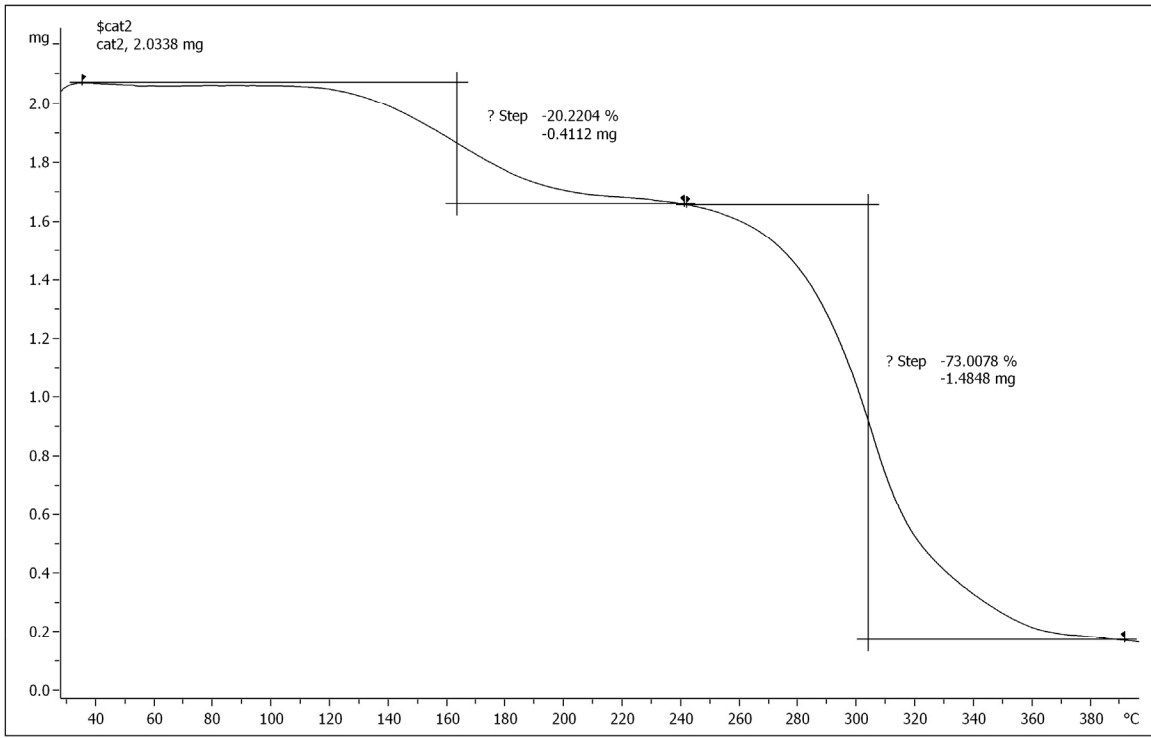
Fig.S2 Comparison of simulated PXRD patterns with experimental PXRD patterns of five cocrystals.



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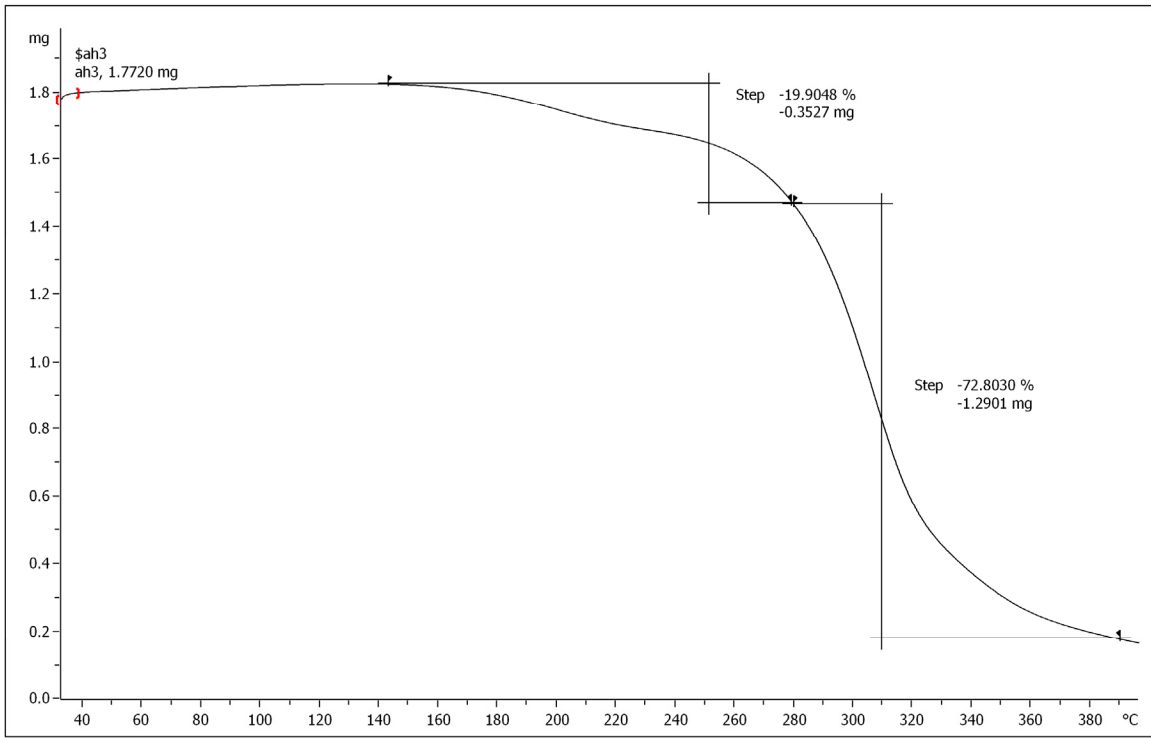
(a)



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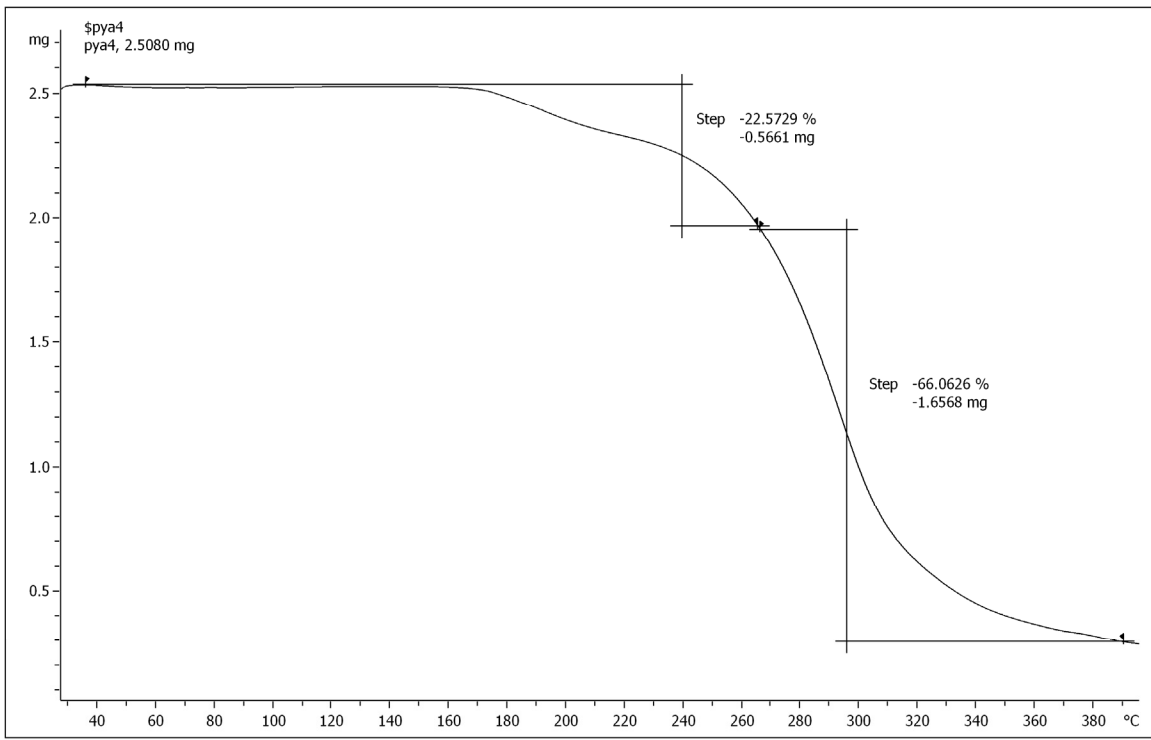
(b)



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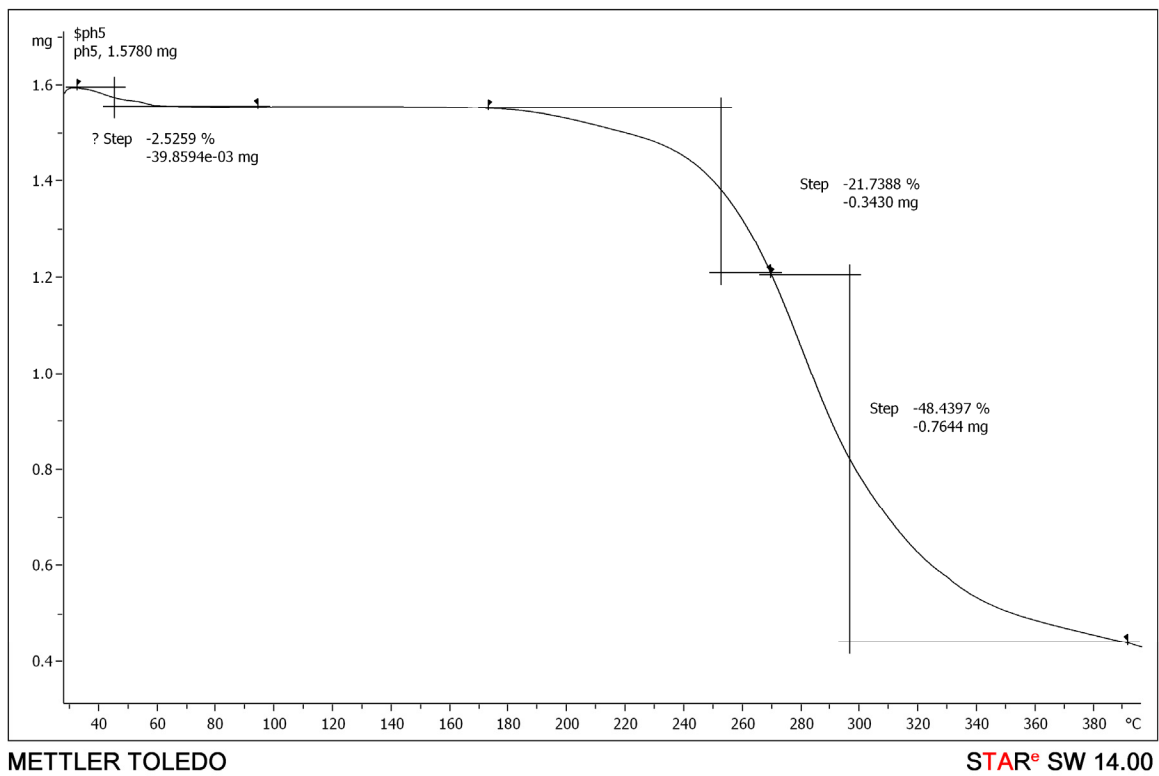
(c)



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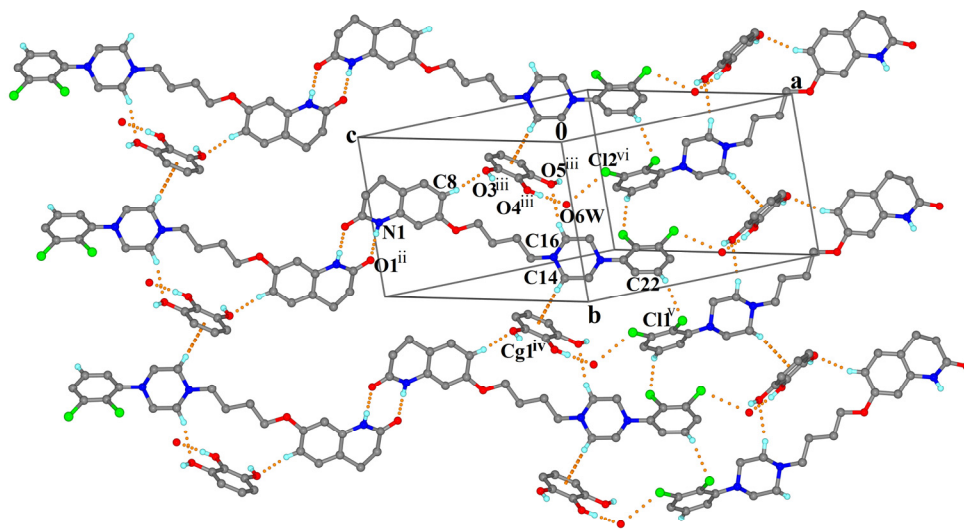
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(d)

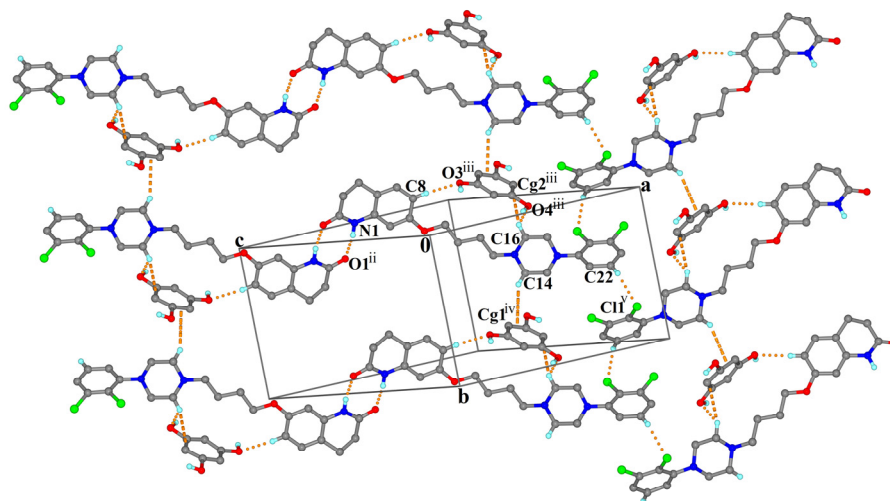


(e)

Fig. S3 (a) Thermo gravimetric curve of cocystal I. (b) Thermo gravimetric curve of cocystal II. (c) Thermo gravimetric curve of cocystal III. (d) Thermo gravimetric curve of cocystal IV. (e) Thermo gravimetric curve of cocystal V.

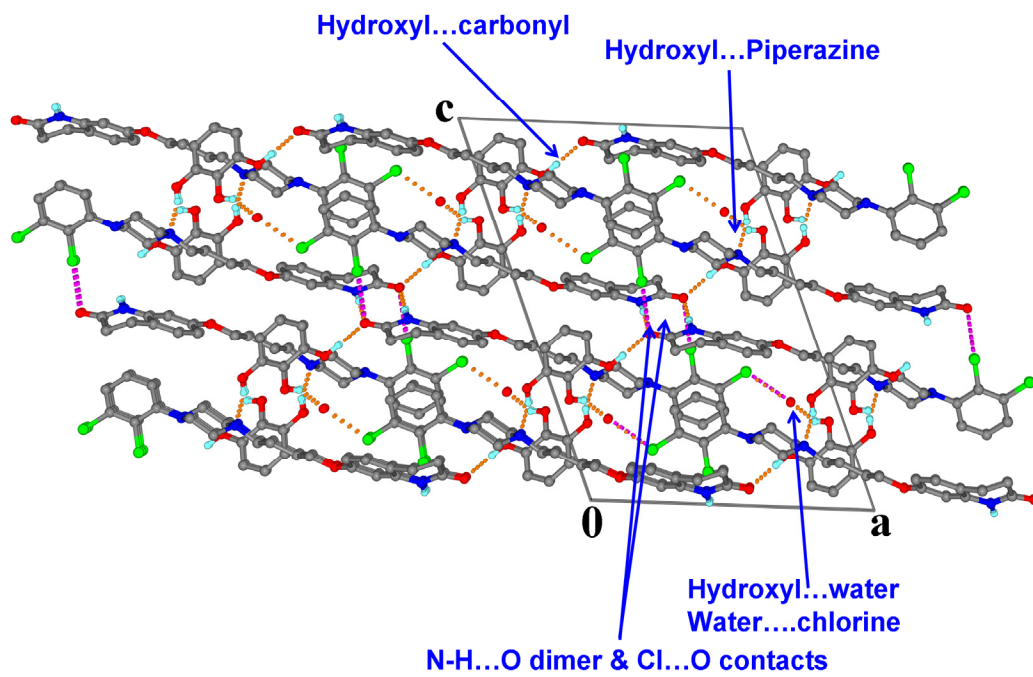


(a)

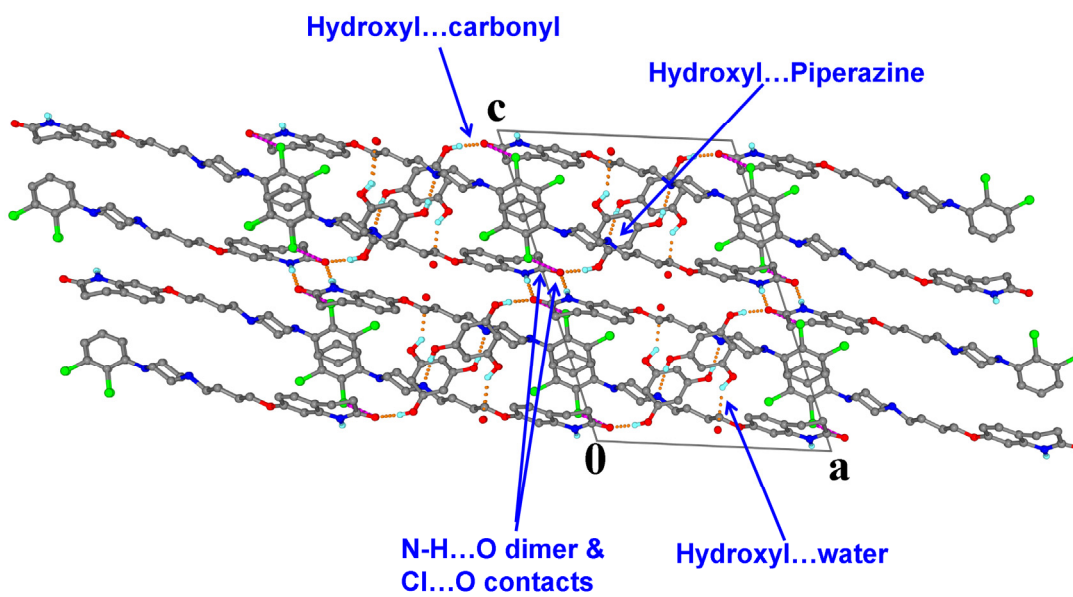


(b)

Fig.S4 (a) Cocystal IV, extension of centrosymmetric amide N1–H1···O1 dimer into a 2D layered structure by a combination of C14–H14B···Cg1, C8–H8···Cg2, C16–H16A···O5, C8–H8···O3, C22–H22···Cl1, O4–H4O···O5W, Cl2···O6W interactions. Symmetry codes are, ⁱ $-1-x, 1/2+y, 1/2-z$; ⁱⁱ $-1-x, 1-y, 1-z$; ⁱⁱⁱ $-x, -1/2+y, 1/2-z$; ^{iv} $-x, 1/2+y, 1/2-z$; ^v $1-x, 1/2+y, 1/2-z$; ^{vi} $-1+x, y, z$; Cg1 is centroid of C24–C29 atoms and Cg2 is centroid of C24/C29 atoms. (b) Cocystal V, extension of centrosymmetric amide N1–H1···O1 dimer into a 2D layered structure by a combination of C14–H14B···Cg1, C16–H16A···Cg2, C17–H17B···O4, C8–H8···O3, C22–H22···Cl1 interactions. Symmetry codes are, ⁱ $-x, 1/2+y, 1/2-z$; ⁱⁱ $-x, -y, 1-z$; ⁱⁱⁱ $1-x, -1/2+y, 1/2-z$; ^{iv} $1-x, 1/2+y, 1/2-z$; ^v $2-x, 1/2+y, 1/2-z$; ^{vi} $-1+x, -1+y, z$; ^{vii} $1-x, -y, 1-z$; Cg1 is the centroid of C28/C29 atoms; Cg2 is of C26/C27 atoms; Cg3 is of C18–C23 atoms.



(a)



(b)

Fig.S5 (a) Cocystal IV. (b) Cocystal V. The three dimensional interlayer bonding in two cocrystals mediated through various interactions is shown in the picture.