Supporting information

Selective CO₂ adsorption in four Zinc (II) based metal organic frameworks constructed by using rigid N,N'-donor linker and various dicarboxylate ligands

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Figure S1. FT-IR spectra of compounds 1-4.



Figure S2. Asymetric unit of **1**. Color code: carbon (light gray), nitrogen (blue), oxygen (red) sulfur (yellow) and zinc (green). (Hydrogens and water molecules are omitted for clarity).



Figure S3. A distorted trigonal bipyramidal geometry of Zn(II) atom found in **1**. Color code; same as in Figure S2.



Figure S4. Bridging Modes of tdc ligand found in compound 1 (Harris Notation). Color code; same as in Figure S2.



Figure S5. Illustration of 1D zig-zag chain along the *c*-axis. Color code; same as in Figure S2.



Figure S6. Illustration of 2d sheet like network along the *b*-axis. Color code; same as in Figure



Figure S7. Simplified view of 4-connected single diamondoid (dia) topology



Figure. S8. Asymetric unit of 2. Color code; same as in Figure S2. (Hydrogens and ethanol molecules are omitted for clarity).



Figure S9. Highly distorted octahedral geometry of Zn(II) atom in compound **2**. Color code; same as in Figure S2.



Figure S10. Bridging Mode of 2-NH₂BDC found in compound 2 (Harris Notation). Color code; same as in Figure S2.



Figure S11. Illustration of 1D zig-zag chain along the *c*-axis. Color code; same as in Figure S2.



Figure S12. Illustration of 2d sheet like network found in compound **2** along the *b*-axis. Color code; same as in Figure S2.



Figure S13. Asymmetric unit of 3. Color code: carbon (light gray), nitrogen (blue), oxygen (red) and zinc (green). (Hydrogens are omitted for clarity).



Figure S14. A distorted tetrahedral geometry of Zn(II) ion found in compound **3**. Color code; same as in Figure S12.



Figure S15. Bridging Mode of NDC found in compound **3** (Harris Notation). Color code; same as in Figure S12.



Figure S16. Illustration of 1D zig-zag chain along the *c*-axis. Color code; same as in Figure S12.



Figure S17. Illustration of 2d sheet like network found in compound **3** along the *b*-axis. Color code; same as in Figure S7.



Figure S18. Asymmetric unit of compound 4. Color code: carbon (light gray), nitrogen (blue), oxygen (red) and zinc (green). (Hydrogens, methanol, perchlorate molecules are omitted for clarity).



Figure S19. (a) A octahedral geometry of Zn(II) ion (b) distorted octahedral geometry of Zn (II) ion found in compound **4**. (b) Color code; same as in Figure S17.



Figure S20. Bridging Mode of ADPA found in compound **4** (Harris Notation). Color code; same as in Figure S17.



Figure S21. PXRD patterns of compound 1.



Figure S22. Pxrd patterns of compound 2.



Figure S23. PXRD patterns of compound 3.



Figure S24. PXRD patterns of compound 4.



Figure S25. TGA graph of Compounds 1-4.



Figure S26. N_2 gas adsorption isotherms of compounds 1-4 mesured at 77 K



Figure S27. CH₄ gas adsorption isotherms of compounds 1-4 mesured at 195 K



Figure S28. Gas adsorption isotherms of compounds 3 mesured at 273 K



Figure S29. Gas adsorption isotherms of compounds 4 mesured at 273 K.



Figure S30. Isosteric heats (Q_{st}) of CO₂ adsorption are calculated based on the adsorption data collected at 273 K and 298 K by using Clapeyron method.



Figure S31. Isosteric heats (Q_{st}) of CH_4 adsorption are calculated based on the adsorption data collected at 273 K and 298 K by using Clapeyron method.

Table S1. Selected	bond I	lengths	and bond	angles (or compound I	

Bond lengths		Bond Angles	
Zn1-O1	2.0089(2)	O1-Zn1-O2	57.44(5)
Zn1-O2	2.505(1)	01-Zn1-O4	108.21(6)
Zn1-O4	1.952(1)	O2-Zn1-O4	91.32(5)
Zn1-N11	2.084(1)	O1-Zn1-N11	95.74(6)
Zn1-N14	2.017(1)	01-Zn1-N14	109.25(6)
		O2-Zn1-N11	153.16(5)
		O2-Zn1-N14	86.43(5)
		O4-Zn1-N11	97.19(6)
		O4-Zn1-N14	133.55(6)
		N14-Zn1-N11	105.32(6)

Bond lengths		Bond Angles	
Zn1-O1	2.013(5)	N1-Zn1-O1	106.3(2)
Zn1-O2	2.427(4)	N1-Zn1-O2	89.9(1)
Zn1-N2	2.085(4)	N1-Zn1-N2	101.8(2)
Zn1-O4	2.058(5)	N1-Zn1-O4	96.1(2)
Zn1-O3	2.324(4)	N1-Zn1-O3	154.8(2)
Zn1-N1	2.077(4)	N1-Zn1-O3	154.8(2)
		O1-Zn1-O2	58.2(2)
		O1-Zn1-N2	96.2(2)
		01-Zn1-O4	145.2(2)
		O1-Zn1-O3	94.7(2)
		O2-Zn1-N2	154.1(2)
		O2-Zn1-O4	96.4(2)
		O2-Zn1-O3	89.4(1)
		N2-Zn1-O4	105.1(2)
		N2-Zn1-O3	89.4(2)
		O4-Zn1-O3	59.0(2)

Table S2. Selected bond lengths and bond angles of compound $\mathbf{2}$

Table S3. Selected bond lengths and bond angles of compound 3

Bond lengths		F	Bond Angles	
Zn1-O2	1.920(4)	O2-Zn1-O5	128.3(2)	
Zn1-N1	2.059(5)	O2-Zn1-N1	126.0(1)	
Zn1-O5	1.991(3)	O2-Zn1-N3	100.4(1)	
Zn1-N3	2.091(3)	O5-Zn1-N1	98.0(2)	
		O5-Zn1-N3	93.5(2)	
		N1-Zn1-N3	103.0(1)	

Bon	d lengths	Bond	Angles
Zn1-O2	2.000(2)	O2-Zn1-O1	100.14(9)
Zn1-O1	2.109(2)	O2-Zn1-N1	90.89(9)
Zn1-N1	2.153(2)	O2-Zn1-O4	163.08(8)
Zn1-O4	2.116(2)	O2-Zn1-O5	103.93(8)
Zn1-O5	2.269(2)	O2-Zn1-N4	90.00(9)
Zn1-N4	2.184(2)	O1-Zn1-N1	88.7(1)
Zn2-O3	2.115	01-Zn1-O4	96.60(9)
Zn2-N5	2.149	01-Zn1-O5	155.92(9)
Zn2-O5	2.149	O1-Zn1-N4	90.1(1)
Zn2-O5	2.149	N1-Zn1-O4	91.99(8)
Zn2-O3	2.115	N1-Zn1-O5	90.25(8)
		N1-Zn1-N4	178.61(9)
		O4-Zn1-O5	59.40(7)
		O4-Zn1-N4	87.47(8)
		O5-Zn1-N4	90.58(8)
		O3-Zn2-N5	89.93
		O3-Zn2-O5	87.72
		O3-Zn2-O5	92.28
		O3-Zn2-O3	180.00
		O3-Zn2-N5	90.07
		N5-Zn2-O5	88.90
		N5-Zn2-O5	91.10
		N5-Zn2-O3	90.07
		N5-Zn2-N5	180.00
		O5-Zn2-O5	180.00
		O5-Zn2-O5	180.00
		O5-Zn2-O3	87.72
		O5-Zn2-N5	91.10
		O5-Zn2-O3	92.28
		O5-Zn2-N5	88.90
		03-Zn2-N5	89.93
		Zn1-O5-Zn2	118.78

 Table S4. Selected bond lengths and bond angles of compound 4