Supplementary Information

to accompany

Electrochemical and physicochemical properties of small

phosphonium cation ionic liquid electrolytes with high lithium

salt content

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1 Preparation of trimethyl(isobutyl)phosphonium

bis(fluorosulfonyl)imide (P_{111i4}FSI)

Isobutylphosphine (CYTOP®141, Cytec Industries) was reacted with trimethyl phosphate (>3 mol) at a temperature 110-130 °C using the process described in US patent 7,829,744 B2.

The trimethyl(isobutyl)phosphonium dimethylphosphate intermediate was further reacted with 1.05 equivalents of potassium bis(fluorosulfonyl)imide (KFSI) in the presence of dichloromethane. The organic phase was treated with saturated NaHCO₃ in water followed by additional washes with H₂O, hexanes, and a number of washes with deionized H₂O and checked for the presence of any residual halide with the use of AgNO₃ in water. After removal of the volatiles on the Rotovap, the resulting product was additionally purified using short-path evaporator. Its structure was confirmed by ¹H, ¹³C, ¹⁹F and ³¹P NMR analyses (deuterated acetone ((CD₃)₂CO)).

2 Nuclear Magnetic Resonance spectra of P_{111i4}FSI







3 Electrospray mass spectra of P_{111i4}FSI



Figure S7 Mass spectra of P_{111i4}FSI

4 Transport properties - VTF parameters

(a)

(b)

Figure S1 VTF plots of ionic conductivity (a) and dynamic viscosity (b) for P_{111i4}FSI electrolytes and various concentrations of LiFSI

Table S1 VTF equation parameters of ionic conductivity from -40 to 120° C (

$$\sigma = \sigma_0 exp \left\{ \frac{-B}{T - T_0} \right\}_{j}$$

LiFSI	Molar ratio	Ln σ_0 /	B (K)	^T ₀ (K)	R (correlation
concentration	Li+: P+	mS.cm ⁻¹			coefficient)
2.0	0.19:0.31	6.36±0.09	791±24	160±2	0.9999
3.2	0.25:0.25	6.07±0.0.02	768±5	173±0.3	0.9999
3.8	0.27:0.23	6.35±0.0.04	844±8	171±0.5	0.9999

Table S2 VTF equation parameters of dynamic viscosity from 20 to 80° C (

$$\eta = \eta_0 exp\left\{\frac{B}{T - T_0}\right\}_{j}$$

LiFSI	Molar ratio	Ln η_0 / P	B (K)	^T ₀ (K)	R (correlation
concentration	ncentration Li ⁺ : P ⁺				coefficient)
2.0	0.19:0.31	-4.84±0.38	589±100	188±11	0.9998
3.2	0.25:0.25	-5.69±0.39	863±117	173±10	0.9999
3.8	0.27:0.23	-5.16±0.64	742±170	186±15	0.9996

5 Electrochemical measurements of tri(methyl)isobutylphosphonium bis(fluorosulfonyl)imide (P_{111i4}FSI)



Figure S2 Linear sweep voltammograms for neat $P_{111i4}FSI$ at a glassy carbon working electrode with a potential sweep rate of 20mV.s⁻¹ at different temperatures 25-50-75-100°C.