## **Supporting Information**

Revealing Highly Unbalanced Energy Barriers in Extension and Contraction of the Muscle-like Motion of a [c2]Daisy Chain

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## **Figure and Table Captions**

**Figure S1.** Equilibrated structures in acid and base by SCC-DFTB-D optimizations: (a) Side view of ext<sup>4+</sup>; (b) Central part of ext<sup>4+</sup>; (c) Left half of ext<sup>4+</sup>; (d) Right half of ext<sup>4+</sup>; (e) Top view of ext<sup>4+</sup>; (f) Side view of cont<sup>2+</sup>; (g) One ring of cont<sup>2+</sup>; (h) Left half of cont<sup>2+</sup>; (i) Right half of cont<sup>2+</sup>; (j) Top view of cont<sup>2+</sup>. Selected atoms are labeled (some also magnified) and referred to in the text. Gray, white, red, and blue atom colors represent C, H, O, and N, respectively. To clearly illustrate the geometries in the enlarged sections, hydrogen atoms are omitted except for those forming hydrogen bonds and those on amino N. Lengths are given in nm. (Note that the terpyridine and benzene groups in the bulky stoppers are connected by -C=C- groups, which in this representation incorrectly look like abnormally long single bonds.)

**Figure S2.** Selected frames in side and top views by SCC-DFTB-D constrained geometry optimizations during the contraction in basic environment (shown with a blue background) and during the extension in acidic environment (pink background). Frame labels  $1_b$ ,  $1_a$ , etc., refer to Tables S1 and S2.

**Figure S3.** Configurations of (panels a-h)  $PF_6^-(CH_3CN)_n$  (n=1-8) and (panels i-p)  $PF_6^-$ (CHCl<sub>3</sub>)<sub>n</sub> (n=1-8) complexes, calculated by  $\omega$ B97XD/6-31G\*\*. Grey, white, blue, orange, light blue, and green colors represent C, H, N, P, F, and Cl atoms, respectively. To avoid overloading the figures, distances are only labeled when shorter than 3 Å between F of  $PF_6^-$  and H/Cl of CH<sub>3</sub>CN/CHCl<sub>3</sub> in  $PF_6^-(CH_3CN)_n/PF_6^-(CHCl_3)_n$  (n=1-4).

**Table S1.** Relative total energy and dihedral angle during the contraction in basic environment, as a function of the O1- O2 distance.

**Table S2.** As Table S1, but for stretching in acidic environment.

**Table S3.** Average potential energy  $E_{pot}$ , temperature *T* and their root mean square deviations (RMSD) in the time period of 900-1000 fs during SCC-DFTB-D/MD simulations in solution.

**Movie S1.** An animation is provided of the detailed model of the full contraction and elongation cycle of the [c2] daisy chain.















Figure S2



 $(m) PF_6 (CHCl_3)_5$ 

(n) PF<sub>6</sub>-(CHCl<sub>3</sub>)<sub>6</sub>

(o) PF<sub>6</sub>-(CHCl<sub>3</sub>)<sub>7</sub>

(p) PF<sub>6</sub>-(CHCl<sub>3</sub>)<sub>8</sub>

Figure S3

Table S1. Relative total energy and dihedral angle during the contraction in basic environment, as a function of the O1 - O2 distance. Energy barriers and reaction heats are shown in bold.

frames	01 - 02	$\Delta E_{tot}$ (kcal/mol)				dihadral anala af
	distance	SCC-DFTB-D	SCC-DFTB-D ωB97XD/6-31G**			$C_1 C_2 C_2 N_4 (0)$
III base	(nm)	in vac.	in vac.	in CH <sub>3</sub> CN	in CHCl <sub>3</sub>	(°)
	0.85	0.48	1.25	-1.01	-0.65	116.8
	0.90	0.06	0.67	0.34	0.22	119.6
1 <sub>b</sub>	0.91	0.00	0.17	0.03	0.002	119.6
2 <sub>b</sub>	0.92	0.02	0.00	0.00	0.00	125.1
3 <sub>b</sub>	0.95	0.57	0.047	0.28	0.47	118.6
4 <sub>b</sub>	1.00	3.16	2.95	3.32	3.53	131.3
5 <sub>b</sub>	1.03	5.33	6.95	6.16	6.74	131.0
6 <sub>b</sub>	1.05	3.86	4.89	1.10	1.80	132.4
7 <sub>b</sub>	1.10	2.30	3.63	0.80	1.37	133.6
8 <sub>b</sub>	1.15	-0.48	-5.47	-2.86	-3.25	135.2
9 <sub>b</sub>	1.20	-2.77	-9.34	-3.20	-4.57	134.4
10 <sub>b</sub>	1.25	-3.48	-7.31	-0.43	-2.18	133.8
11 <sub>b</sub>	1.30	-6.84	-8.22	-3.19	-5.61	138.4
12 <sub>b</sub>	1.35	-10.92	-16.49	-12.55	-14.18	142.1
13 <sub>b</sub>	1.40	-13.69	-19.71	-15.87	-17.53	142.4
14 <sub>b</sub>	1.45	-21.80	-28.64	-19.44	-22.17	150.3
15 <sub>b</sub>	1.50	-25.91	-33.81	-23.66	-26.59	150.9
16 <sub>b</sub>	1.55	-33.42	-45.65	-28.76	-32.89	109.0
17 <sub>b</sub>	1.60	-36.83	-50.12	-33.54	-37.97	92.0
18 <sub>b</sub>	1.67	-45.05	-62.17	-40.27	-46.19	78.8
	1.68	-45.01	-61.83	-39.72	-45.71	78.8
	1.70	-44.66	-60.46	-37.93	-44.04	79.0

frames in acid	01 - 02		- dihedral angle of			
	distance	SCC-DFTB-D ωB97XD/6-31G**				
	(nm)	in vac.	in vac.	in CH <sub>3</sub> CN	in CHCl <sub>3</sub>	(1)
	1.71	2.98	5.04	3.76	1.27	80.2
1 <sub>a</sub>	1.67	0.51	0.70	0.27	0.47	79.2
2 <sub>a</sub>	1.66	0.00	0.00	0.00	0.00	78.7
3 <sub>a</sub>	1.61	1.85	4.58	1.74	1.27	132.2
4 <sub>a</sub>	1.56	3.06	6.73	4.92	4.26	131.5
5 <sub>a</sub>	1.51	5.28	9.28	9.34	7.94	127.2
6 <sub>a</sub>	1.46	8.53	14.14	20.31	18.70	117.9
7 <sub>a</sub>	1.43	10.28	14.59	24.39	23.03	139.9
8 <sub>a</sub>	1.40	8.95	11.49	23.97	21.79	139.4
9 <sub>a</sub>	1.35	6.79	9.00	28.33	24.15	151.2
10 <sub>a</sub>	1.30	2.91	7.00	29.45	23.80	152.0
11 <sub>a</sub>	1.25	-10.03	-0.33	33.12	22.42	137.0
12 <sub>a</sub>	1.20	-17.95	-10.21	26.19	15.43	148.0
13 <sub>a</sub>	1.15	-28.05	-20.59	20.22	7.89	154.3
14 <sub>a</sub>	1.10	-36.54	-31.32	13.25	-0.39	158.1
15 <sub>a</sub>	1.05	-40.91	-39.25	7.17	-6.52	140.2
16 <sub>a</sub>	1.00	-49.25	-47.17	-0.88	-14.11	140.7
17 <sub>a</sub>	0.95	-53.92	-52.26	-5.59	-18.61	140.4
	0.92	-54.82	-52.77	-6.03	-18.50	139.6
18 <sub>a</sub>	0.91	-54.85	-52.81	-5.99	-18.44	139.6
	0.90	-54.70	-52.74	-5.91	-18.23	140.0
	0.85	-52.08	-49.97	-3.57	-15.73	140.9

Table S2. As Table S1, but for stretching in acidic environment.

Table S3. Average potential energy  $E_{pot}$ , temperature *T* and their root mean square deviations (RMSD) in the time period of 900-1000 fs during SCC-DFTB-D/MD simulations in solution.

rotaxanes in solution	<i>E<sub>pot</sub></i> (kcal/mol)	RMSD ( <i>E<sub>pot</sub></i> ) (kcal/mol)	Т (К)	RMSD (T) (K)
ext <sup>4+</sup>	-2001757.81	27.93	306.25	3.96
proton-cont <sup>4+</sup>	-2001745.67	30.94	305.90	3.86
cont <sup>2+</sup>	-2001585.15	28.57	306.78	5.51
deproton-ext <sup>2+</sup>	-2001553.13	25.29	305.34	5.22

Movie S1

An animation is provided of the detailed model of the full contraction and elongation cycle of the [c2] daisy chain through four states, starting with deproton-ext<sup>2+</sup>:



These correspond to the curves in Figure 3 in the main text and data in Table S1-S2 above. The blue/pink background color indicates a basic/acidic environment, respectively.