

## Electronic Supplementary Information

### Nature of the Interaction between Rare Gas Atoms and Transition Metal Doped Silicon Clusters: The Role of Shielding Effects

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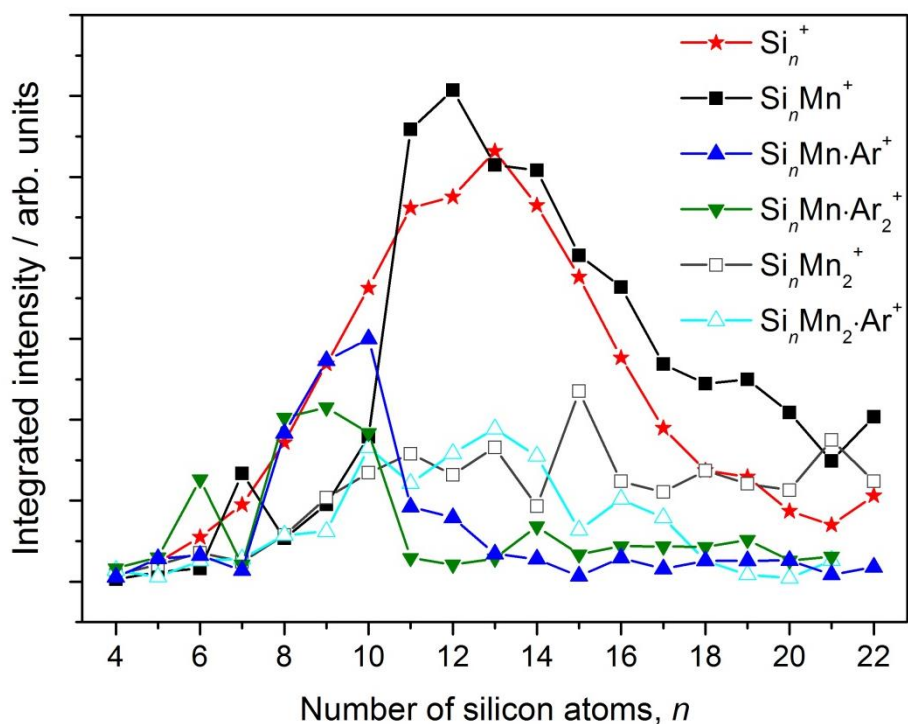
**Part 1. Analysis of the mass spectra**

**Part 2. Dependence of the cluster-RG binding energy on the used functionals**

**Part 3. Interaction of  $\text{Si}_n\text{Mn}^+$  ( $n = 6-10$ ) with rare gas atoms (Ar, Xe)**

**Part 4. Interaction of  $\text{Si}_7\text{TM}^+$  ( $\text{TM} = \text{Cr}, \text{Mn}, \text{Cu}$  and  $\text{Zn}$ ) with rare gas atoms (Ar,  
Xe)**

## Part 1. Analysis of the mass spectra



**Figure S1.** The intensities of  $\text{Si}_n^+$ ,  $\text{Si}_n \text{Mn}^+$ ,  $\text{Si}_n \text{Mn}_2^+$ ,  $\text{Si}_n \text{Mn}^+ \cdot \text{Ar}$ ,  $\text{Si}_n \text{Mn}^+ \cdot \text{Ar}_2$ , and  $\text{Si}_n \text{Mn}_2^+ \cdot \text{Ar}$ , obtained by fitting the natural isotope distribution of those species to the mass spectrum presented in Fig. 1a of the main text.

The mass spectra of  $\text{Si}_n \text{Mn}^+ \cdot \text{RG}$  (such as the one shown in Fig. 1) are congested due to the large number of possible compositions and due to the natural isotope distributions of silicon. The intensity of each species can be extracted from the mass spectra by deconvolution. The abundance of each cluster is then obtained by fitting the corresponding isotope distribution to the measured mass spectra. The result of such analysis for  $\text{Si}_n \text{Mn}_m^+ \cdot \text{Ar}$  is shown in Fig. S1 where the intensities of  $\text{Si}_n^+$ ,  $\text{Si}_n \text{Mn}^+$ ,  $\text{Si}_n \text{Mn}_2^+$ ,  $\text{Si}_n \text{Mn}^+ \cdot \text{Ar}$ ,  $\text{Si}_n \text{Mn}^+ \cdot \text{Ar}_2$ , and  $\text{Si}_n \text{Mn}_2^+ \cdot \text{Ar}$  are plotted as a function of  $n$ .

## Part 2. Dependence of the cluster-RG binding energy on the used functionals

**Table S1.** The cluster-RG binding energies (eV) calculated using B3P36 and M06 functionals with the basis set 6-311+G(d) for Ar, Si and TM atoms, aug-cc-pVdZ for Xe atom.

<b>Complex</b>		<b>B3P86</b>	<b>M06</b>
Si <sub>7</sub> Cr <sup>+</sup> .Ar	<b>Com-A</b>	0.189	0.205
Si <sub>7</sub> Cr <sup>+</sup> .Xe	<b>Com-A</b>	0.417	0.513
Si <sub>7</sub> Mn <sup>+</sup> .Ar	<b>Com-A</b>	-0.248	-0.202
	<b>Com-B</b>	0.017	0.110
Si <sub>7</sub> Mn <sup>+</sup> .Xe	<b>Com-A</b>	0.028	0.128
	<b>Com-B</b>	0.143	0.309
Si <sub>7</sub> Cu <sup>+</sup> .Ar	<b>Com-A</b>	0.216	0.264
Si <sub>7</sub> Cu <sup>+</sup> .Xe	<b>Com-A</b>	0.507	0.615
Si <sub>7</sub> Zn <sup>+</sup> .Ar	<b>Com-A</b>	-0.399	-0.596
	<b>Com-B</b>	0.005	0.060
Si <sub>7</sub> Zn <sup>+</sup> .Xe	<b>Com-A</b>	-0.111	-0.291
	<b>Com-B</b>	0.097	0.207

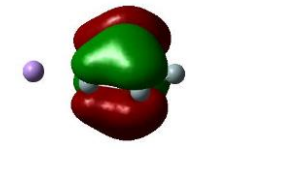
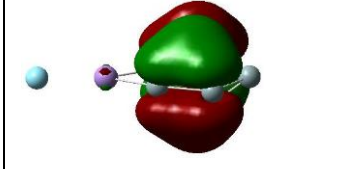
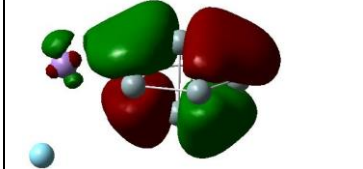
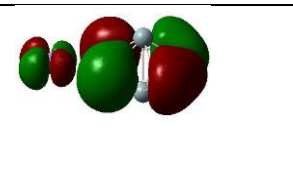
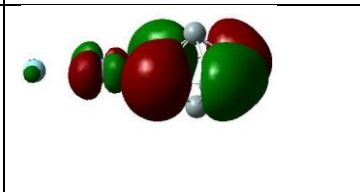
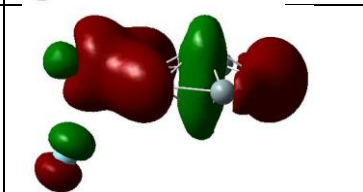
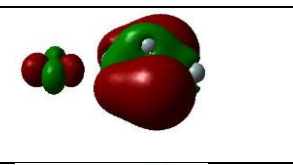
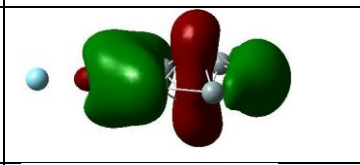
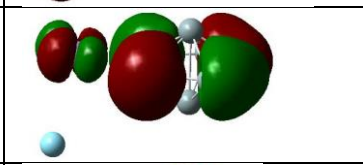
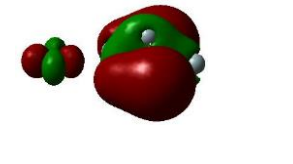
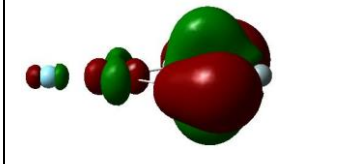
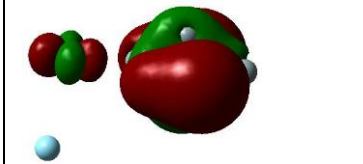
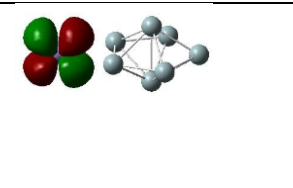
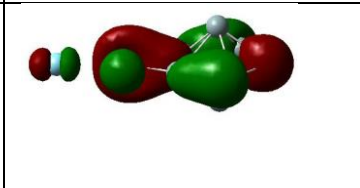
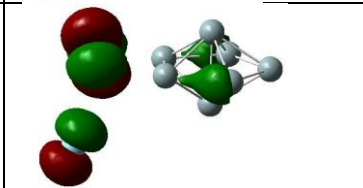
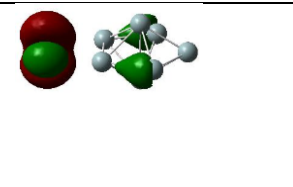
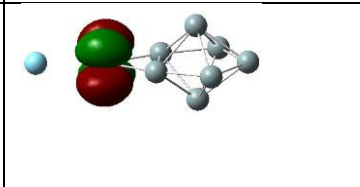
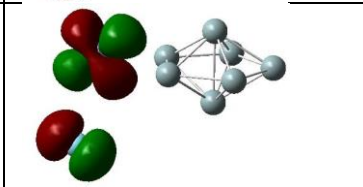
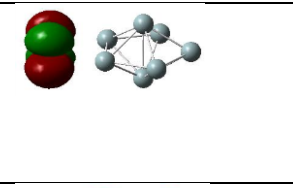
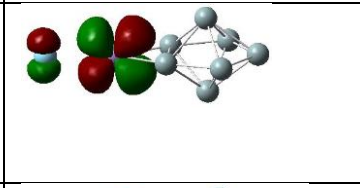
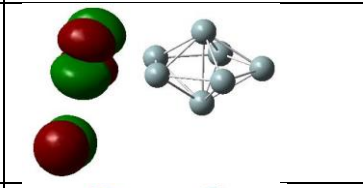
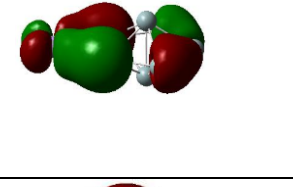
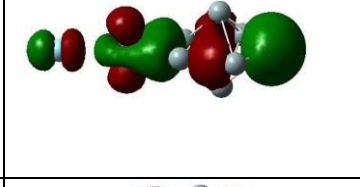
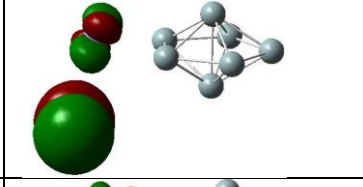
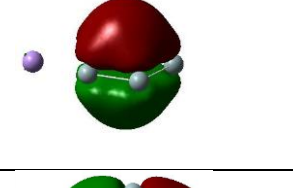
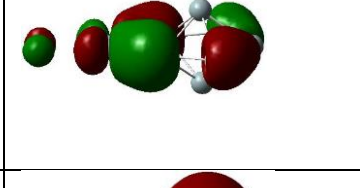
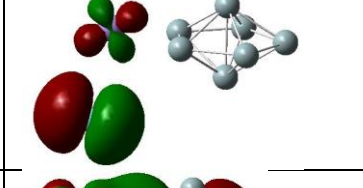
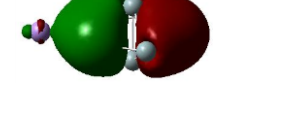
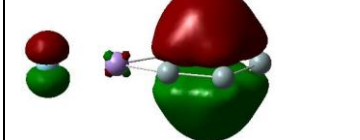
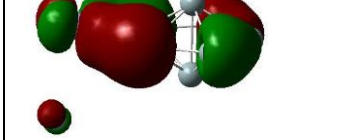
**Part 3. Interaction of  $\text{Si}_n\text{Mn}^+$  ( $n = 6-10$ ) with rare gas atoms (Ar, Xe)**

**Table S2.** Some selected bond lengths ( $d$ , in Å), atomic charge and charge transfer ( $CT$ , in e) in  $\text{Si}_n\text{Mn}^+$  clusters and their RG complexes, as obtained at the B3P86/6-311+G(d) level of theory.

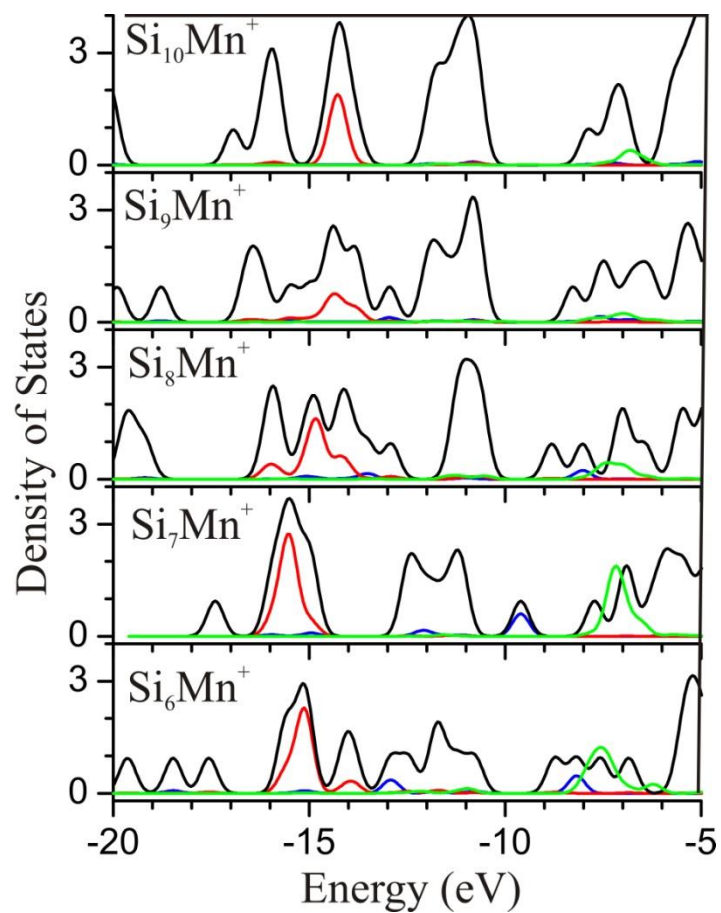
	$\text{Si}_n\text{Mn}^+$		$\text{Si}_n\text{Mn}^+\cdot\text{Ar}$		$\text{Si}_n\text{Mn}^+\cdot\text{Xe}$	
	Shortest $d(\text{Si-Mn})$	Charge on Mn (e)	$d$ (Mn-Ar)	$CT$ (Ar $\rightarrow$ Mn)	$d$ (Mn-Xe)	$CT$ (Xe $\rightarrow$ Mn)
$\text{Si}_6\text{Mn}^+$	2.480	0.96	2.634	0.09	2.848	0.18
$\text{Si}_7\text{Mn}^+(\text{Com-A})$	2.787	0.83	2.719	0.07	2.906	0.14
$\text{Si}_7\text{Mn}^+(\text{Com-B})$	2.787	0.83	3.436	0.01	3.240	0.07
$\text{Si}_8\text{Mn}^+$	2.477	0.93	2.622	0.10	2.837	0.19
$\text{Si}_9\text{Mn}^+$	2.458	0.80	2.640	0.10	2.850	0.20
$\text{Si}_{10}\text{Mn}^+$	2.450	1.01	2.634	0.08	2.844	0.17

**Table S3.** Shapes of molecular orbitals (for spin up) of  $\text{Si}_7\text{Mn}^+$  cluster and the  $\text{Si}_7\text{Mn}^+\cdot\text{Ar}$  complexes (**Com-A** and **Com-B**)

	$\text{Si}_7\text{Mn}^+$	<b>Com-A</b>	<b>Com-B</b>
LUMO			
HOMO			
HOMO-1			
HOMO-2			
HOMO-3			
HOMO-4			
HOMO-5			
HOMO-6			

HOMO-7			
HOMO-8			
HOMO-9			
HOMO-10			
HOMO-11			
HOMO-12			
HOMO-13			
HOMO-14			
HOMO-15			
HOMO-16			

HOMO-18			
HOMO-19			
HOMO-20			
HOMO-21			
HOMO-22			
HOMO-23			



**Figure S2.** Total density of states (TDOS) for  $\alpha$  orbitals (black line) together with the partial density of states (pDOS) for  $\alpha$  and  $\beta$  Mn 3d (red and green line, respectively) and  $\alpha$  Mn 4s orbital (blue line) of  $\text{Si}_n\text{Mn}^+$  ( $n = 6-10$ ).



**Part 4. Interaction of Si<sub>7</sub>TM<sup>+</sup> (TM = Cr, Mn, Cu and Zn) with rare gas atoms (Ar, Xe)**

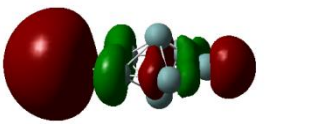
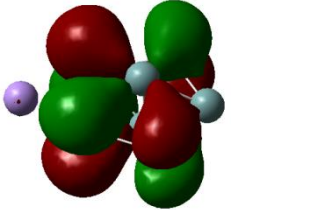
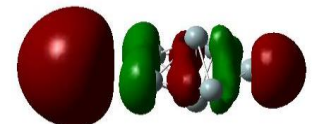
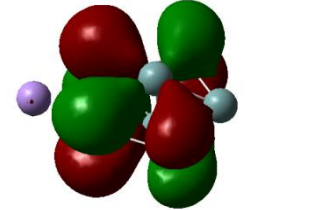
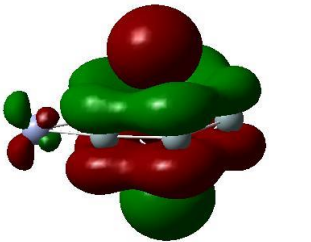
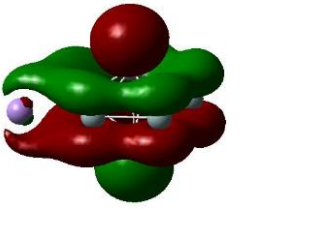
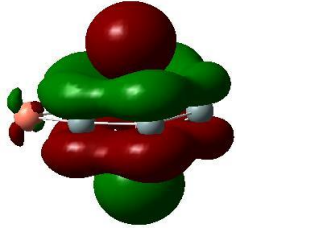
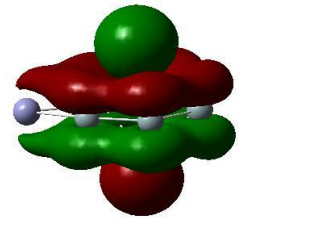
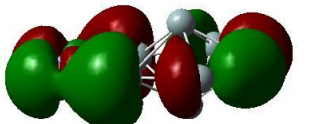
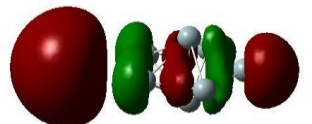
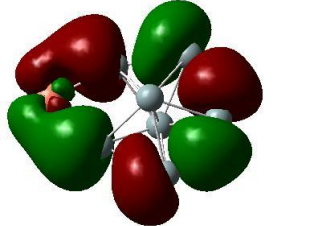
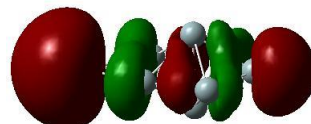
**Table S4.** Electronic state, electron configuration, charge on the dopant atom (in e), HOMO-LUMO gap (in eV) and average Si-dopant bond distance (in Å) of the Si<sub>7</sub>TM<sup>+</sup> (TM = Cr, Mn, Cu and Zn) calculated at the B3P86/6-311+G(d) level. Electron analysis is obtained from the NBO 5.G program.

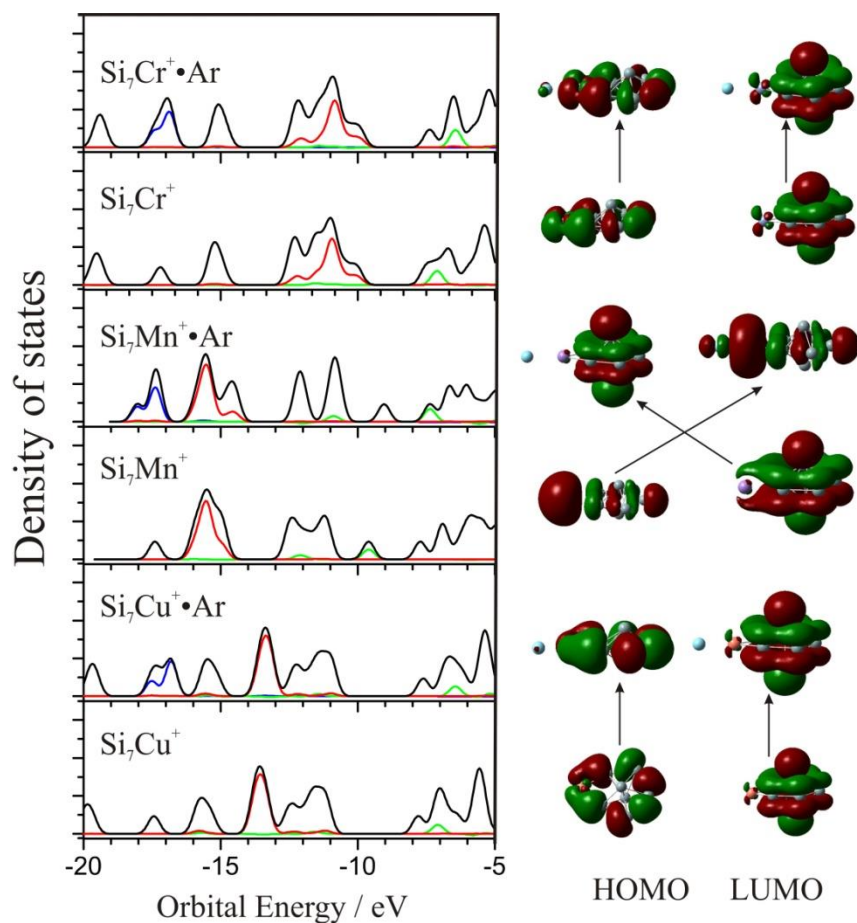
<b>Cluster</b>	<b>Electronic state</b>	<b>Electron configuration of TM</b>	<b>Charge on TM, e</b>	<b>HOMO-LUMO gap, eV</b>	<b>averaged <i>d</i> (Si-TM), Å</b>
<b>Si<sub>7</sub>Cr<sup>+</sup></b>	<sup>6</sup> A <sub>1</sub>	3d <sup>4.96</sup> 4s <sup>0.20</sup> 4p <sup>0.01</sup>	+0.83	2.541	2.5782
<b>Si<sub>7</sub>Mn<sup>+</sup></b>	<sup>7</sup> A <sub>1</sub>	3d <sup>5.10</sup> 4s <sup>1.04</sup> 4p <sup>0.03</sup>	+0.83	1.885	2.7870
<b>Si<sub>7</sub>Cu<sup>+</sup></b>	<sup>1</sup> A <sub>1</sub>	3d <sup>9.91</sup> 4s <sup>0.34</sup> 4p <sup>0.01</sup>	+0.73	3.376	2.3098
<b>Si<sub>7</sub>Zn<sup>+</sup></b>	<sup>2</sup> A <sub>1</sub>	3d <sup>9.99</sup> 4s <sup>1.22</sup> 4p <sup>0.14</sup>	+0.64	2.166	2.6809

**Table S5.** Transition metal-rare gas bond distance, average Si-dopant bond distance (in Å), charge transfer from rare gas to metal atom (*CT*, e), dissociation energy of the metal-rare gas bond (*BDE*, in eV), HOMO-LUMO gap (in eV) and electron configuration of metal atom of the Si<sub>7</sub>TM<sup>+</sup>·RG (TM = Cr, Mn, Cu and Zn; RG = Ar, Xe) complexes calculated at the B3P86/6-311+G(d) level. Electron analysis is obtained from the NBO 5.G program.

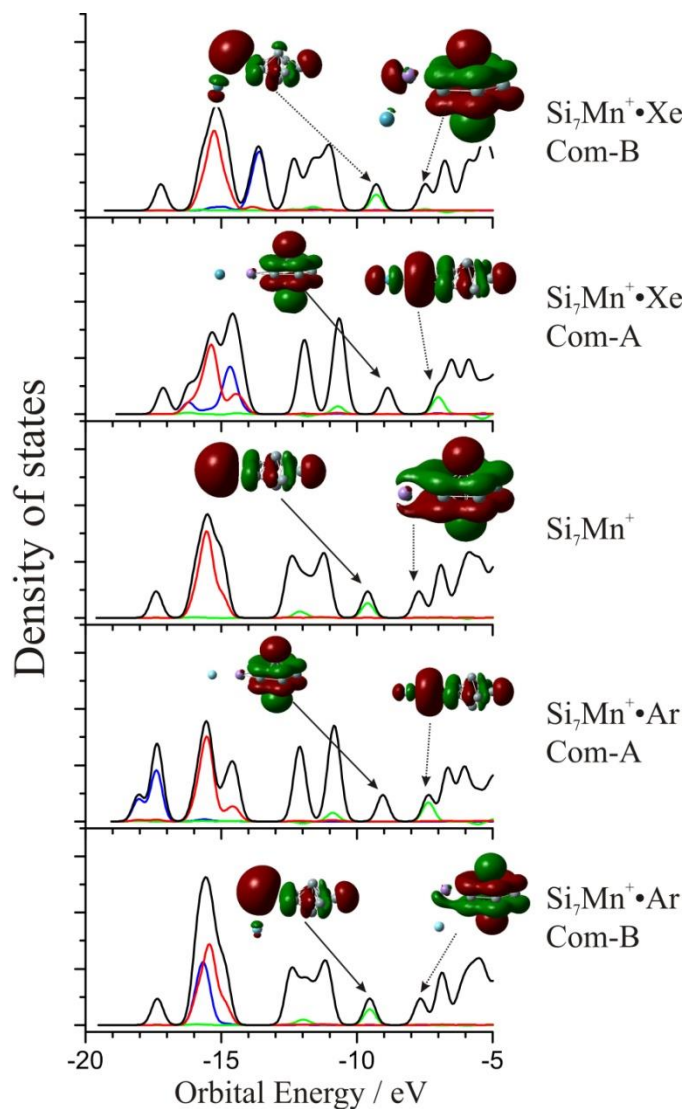
Complex	Structure	<i>d</i> (TM-RG), Å	average <i>d</i> (TM-Si), Å	<i>CT</i> , e	BE, eV	HOMO-LUMO gap, eV	Electron Configuration on TM
Si <sub>7</sub> Cr <sup>+</sup> .Ar	Com-A	2.8169	2.5629	0.066	0.189	2.559	3d <sup>4.97</sup> 4s <sup>0.26</sup> 4p <sup>0.01</sup>
Si <sub>7</sub> Cr <sup>+</sup> .Xe	Com-A	2.8618	2.5906	0.131	0.417	2.522	3d <sup>5.00</sup> 4s <sup>0.30</sup> 4p <sup>0.01</sup> 4d <sup>0.02</sup>
Si <sub>7</sub> Mn <sup>+</sup> .Ar	Com-A	2.7188	2.4779	0.067	-0.248	1.678	3d <sup>5.34</sup> 4s <sup>0.47</sup> 4p <sup>0.03</sup> 4d <sup>0.01</sup>
	Com-B	3.4363	2.7786	0.012	0.017	1.871	3d <sup>5.11</sup> 4s <sup>1.03</sup> 4p <sup>0.03</sup>
Si <sub>7</sub> Mn <sup>+</sup> .Xe	Com-A	2.9057	2.4813	0.144	0.028	1.863	3d <sup>5.36</sup> 4s <sup>0.49</sup> 4p <sup>0.04</sup> 4d <sup>0.02</sup>
	Com-B	3.2402	2.7391	0.065	0.143	1.804	3d <sup>5.13</sup> 4s <sup>0.99</sup> 4p <sup>0.04</sup> 4d <sup>0.01</sup>
Si <sub>7</sub> Cu <sup>+</sup> .Ar	Com-A	2.4573	2.3343	0.074	0.216	3.351	3d <sup>9.91</sup> 4s <sup>0.34</sup> 4p <sup>0.02</sup>
Si <sub>7</sub> Cu <sup>+</sup> .Xe	Com-A	2.6389	2.3462	0.150	0.507	3.324	3d <sup>9.92</sup> 4s <sup>0.44</sup> 4p <sup>0.02</sup> 4d <sup>0.01</sup>
Si <sub>7</sub> Zn <sup>+</sup> .Ar	Com-A	2.7305	2.4269	0.054	-0.399	1.518	3d <sup>9.99</sup> 4s <sup>0.88</sup> 4p <sup>0.05</sup>
	Com-B	3.6126	2.6799	0.005	0.005	2.126	3d <sup>9.99</sup> 4s <sup>1.23</sup> 4p <sup>0.05</sup> 5s <sup>0.01</sup>
Si <sub>7</sub> Zn <sup>+</sup> .Xe	Com-A	2.8311	2.4213	0.143	-0.111	1.669	3d <sup>9.99</sup> 4s <sup>0.86</sup> 4p <sup>0.07</sup> 4d <sup>0.01</sup>
	Com-B	3.2534	2.6676	0.050	0.097	2.033	3d <sup>9.99</sup> 4s <sup>1.21</sup> 4p <sup>0.05</sup> 5s <sup>0.01</sup>

**Table S6.** Shapes of some selected frontier orbitals of  $\text{Si}_7\text{TM}^+$  (TM = Cr, Mn, Cu and Zn)

MO	Cr	Mn	Cu	Zn
LUMO+1				
LUMO				
HOMO				



**Figure S3.** Black curve is TDOS of  $\text{Si}_n\text{TM}^+$  (TM = Cr, Mn, Cu) and their Ar-complexes (**Com-A**). Red curve is pDOS of 3d (TM); green curve is pDOS of 4s (TM); blue curve is pDOS of 3p (Ar). The DOS are plotted based on *alpha* orbitals. HOMO and LUMO orbital are depicted on the right.



**Figure S4.** TDOS of  $\text{Si}_7\text{Mn}^+$  (middle panel) and its **Com-A**, **Com-B** complexes with Ar (lower panels) and Xe (upper panels). Red curve is pDOS of  $3d$  (Mn); green curve is pDOS of  $4s$  (Mn); blue curve is pDOS of either  $3p$  orbitals (Ar) or  $5p$  (Xe) in the complexes. The DOS are plotted based on *alpha* orbitals. The solid arrow indicates the position of the HOMO the dashed arrow the position of the LUMO.