Electronic Supplementary Information

The Effect of Structure Directing Agents on the Ordering of Fluoride Ions in Pure Silica MFI Zeolites

Shelley L. Brace^a, Philip Wormald^b and Richard J. Darton^{*,a}

a Catalysis and Sustainable Materials Group, Lennard-Jones Laboratories, Keele University, Staffordshire, ST5 5BG, UK

b School of Chemistry, University of St Andrews, St Andrews, Fife, KY16 9ST, UK

1. Experimental Details

1.1 Syntheses

The structure directing agents used in the preparation of the pure silica MFI type zeolites were synthesised via the exhaustive alkylation of tributylamine (Alfa Aesar, 98%) or tripropylamine (Sigma Aldrich \geq 98%) using iodomethane (Acros Organics, 99%) or iodoethane (Acros Organics, 99%) as necessary. The products of the reactions were recrystallized using ethyl acetate and ethanol and the structures and purities verified using solution ¹H and ¹³C NMR spectroscopy. Tetrapropylammonium bromide (Acros Organics, 98%) was used as supplied.

The pure silica MFI zeolite samples were prepared by dissolving the SDAs in deionised water with fumed silica (Sigma Aldrich, 0.007 μ m powder) and ammonium fluoride (Acros Organics, 98+%) in 40 mL Teflon-lined, stainless steel autoclaves. The resulting gels with a molar ratio of 1SiO₂: 0.08SDA: 1NH₄F: 20H₂O were heated in a forced air oven at 180 °C for 5-8 days. The MFI crystals were recovered by repeated centrifugation and washing with deionised water before being dried at 60 °C overnight.

1.2 Powder X-ray Diffraction

Powder X-Ray diffraction patterns were collected on all zeolite samples using a Bruker D8 Advance diffractometer operating with Cu K α radiation. The obtained patterns were compared to those previously published to confirm the materials as the MFI framework type.¹

1.3²⁹Si NMR

²⁹Si{¹H} CP MAS spectra were recorded on a Bruker Avance III HD 400 MHz spectrometer operating at 79.47 MHz for ²⁹Si and 400.13 MHz for ¹H using 2.5 mm rotors. Experiments were conducted using a 7 ms contact time, a linear ramp on the ¹H contact pulse (30% slope), ¹H decoupling (SPINAL-64, nutation frequency of 50 kHz), recycle delay of 3 s and a 6 kHz spinning rate. ²⁹Si chemical shifts were referenced to TMS using Q8M8 as an external secondary standard.

1.4¹³C NMR

 $^{13}C{^{1}H}$ CP MAS spectra were recorded on a Bruker Avance III HD 400 MHz spectrometer operating at 100.59 MHz for ^{13}C and 400.13 MHz for ^{1}H using 2.5 mm rotors. Experiments were conducted using a 3 ms contact time, a linear ramp on the ^{1}H contact pulse 30% slope), ^{1}H decoupling (SPINAL-64, nutation frequency of 50 kHz), recycle delay of 3 s and a 6 kHz spinning

rate. ¹³C chemical shifts were referenced to TMS using adamantane as an external secondary standard.

1.5¹⁹F NMR

¹⁹F MAS spectra were recorded on a Bruker Avance III 400 MHz spectrometer operating at 376.49 MHz for ¹⁹F using 1.9 mm rotors. Experiments were conducted at 280 K using a spin echo, recycle delay of 20 s and a 30 kHz spinning rate. ¹⁹F chemical shifts were referenced to a replacement sample of C_6F_6 (δ_F =-166.4ppm with respect to the signal for CFCl₃).



2. Powder X-ray Diffraction Patterns

Figure S1. Room temperature powder X-ray diffraction patterns of (a) [TPA]-F,MFI, (b) [MTBA]-F,MFI, (c) [MTPA]-F,MFI and (d) [ETBA]-F,MFI

Powder X-ray diffraction patterns were compared to published results and confirmed that all the SDAs produce phase pure MFI materials with no signs of MEL intergrowths, as confirmed by the lack of the additional peaks at 6.2 and 18.7 ° 2-theta. Initial analysis of the patterns show that all the materials can be described in the orthorhombic Pnma spacegroup and show no observable additional peaks and/or splitting to suggest the monoclinic spacegroup P2₁/n or an intergrown material. This confirms the theory that statically disordered fluoride ions only occupy one site within the [4¹5²6²] cage and hence on a local scale shows 24 tetrahedral sites whilst still displaying Pnma symmetry by powder X-ray diffraction.

3. ¹⁹F MAS NMR Spectra



Figure S2. ¹⁹F MAS NMR spectra of (a) [MTBA]-F.MFI, (b) [MTPA]-F,MFI, (c) [ETBA]-F,MFI collected at 30 kHz spinning rate. Spinning sidebands are indicated with an asterisk (*).

The ¹⁹F MAS spectra for all samples showed an isotropic peak with a chemical shift consistent with those previously reported for other fluoride containing zeolites., including [TPA]-F,MFI.² Both [MTBA]-F,MFI and [ETBA]-F,MFI showed a small additional peak at -79.5 ppm, which has been previously observed, and is currently subject to further investigation.³

Sample	δiso / ppm
[MTBA]-F,MFI	-67.68
[MTPA]-F,MFI	-65.09
[ETBA]-F,MFI	-63.97

Table S1. ¹⁹F isotropic chemical shifts of the $[SiO_{4/2}F]^-$ peak in MFI zeolites



Figure S3. ¹³C{¹H} CP MAS spectra of (a) [TPA]-F,MFI, (b) [MTBA]-F,MFI, (c) [MTPA]-F,MFI and (d) [ETBA]-F,MFI at 6 kHz spinning rate.

The ${}^{13}C{}^{1}H$ CP MAS spectra of all the MFI samples show that the SDAs have been successfully incorporated in to the zeolite frameworks and further work to correlate the ${}^{13}C$ spectra with the fluoride disorder is currently underway.

5. Deconvoluted ²⁹Si{¹H} CP MAS NMR Spectra



-112.93

-113.22

-113.89

-115.80

-116.16 -117.77

-117.68

-118.50

-126.42

4

5 6

7

8

48

61

40

63 49

110

110

49

882





Figure S4. Deconvoluted ²⁹Si{¹H} CP MAS spectra of (a) [MTBA]-F,MFI, (b) [TPA]-F,MFI, (c) [MTPA]-F,MFI and (d) [ETBA]-F,MFI at 6 kHz spinning rate.

Deconvolution of the ²⁹Si{¹H} CP MAS spectra for [MTBA]-F,MFI, [ETBA]-F,MFI and [MTPA]-F,MFI at 233 K, all showed 24 Si sites consistent with that previously reported for [TPA]-F,MFI at 223 K where fluoride motion had been frozen out.

6. References

1. Robson, H., Verified Syntheses of Zeolitic Materials. Elsevier: Amsterdam, 2001.

2. H. Koller, A. Wolker, L. A. Villascusa, M. J. Diaz-Cabanas, S. Valencia and M. A. Camblor, *J. Am. Chem. Soc.*, 1999, **121**, 3368-3376.

3. C. A. Fyfe, R. J. Darton, H. Mowatt and Z. S. Lin, Micropor. Mesopor. Mat. 2011, 144, 57-66.