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Figure S1. Calculated number of H-bond as a function of time during the 45 ns of the MD simulations in different solvents.



Figure S2. The number of H-bond of the CPNT during the 45 ns of the MD simulations.



Figure S3. Calculated number of H-bonds of the dimer with guest molecule in methanol, water and chloroform.



Figure S4. RMSD (A) and Rg (B) of CPNT with guest molecules in methanol, water and chloroform during the 45 ns of MD simulations.



Figure S5. Optimized structure of CP monomer at M06-2X/6-31G(d) level in the gas phase.



Figure S6. Optimized structure of CP dimer at M06-2X/6-31G(d) level in the gas phase.



Figure S7. Optimized structure of CP dimer with a guest molecule (dimer-complex) at M06-2X/6-31G(d) level in the gas phase.



Figure S8. Calculated electronic density of sates (DOS) for the CP monomer and dimer in different solvents and gas phase (red, green and blue lines shows DOS spectrum, occupied orbitals and unoccupied orbitals, respectively.)



HOMO-Complex

LUMO- Complex

Figure S9. HOMO and LUMO orbitals of the CP monomer, dimer and dimer with guest molecule (Complex).

Table S1. Theoretical (gas phase) and experimental structural parameters of the CP monomer and

| CP dimer (bond length and angles are in Å and degree, respectively) | | | | | | |
|---|---------|---------|---------|--|--|--|
| Parameters | Monomer | Dimer | X-ray | | | |
| C77O78 | 1.22 | 1.23 | 1.23 | | | |
| C49O50 | 1.21 | 1.22 | 1.23 | | | |
| C2O1 | 1.22 | 1.23 | 1.29 | | | |
| C29O30 | 1.21 | 1.21 | 1.21 | | | |
| N3H4 | 1.01 | 1.02 | 0.99 | | | |
| N3C2O1 | 123.14 | 123.07 | 131.23 | | | |
| N81C29O3 | 124.11 | 123.39 | 122.59 | | | |
| N75C77O78 | 123.13 | 123.12 | 125.03 | | | |
| N47C49O50 | 124.16 | 123.07 | 122.73 | | | |
| O1C281H82 | 164.37 | -177.98 | 173.08 | | | |
| O30C29N81H82 | -168.46 | 166.65 | 174.26 | | | |
| O78C77N75H76 | 164.32 | 169.82 | -179.27 | | | |
| O50C49N48H47 | -168.78 | -174.34 | 178.19 | | | |

Table S2. Calculated IR vibrational frequencies (cm⁻¹) for CPs monomer and dimer in the gas phase and different solvents.

| Bond | structures | Gas | Water | Chloroform | Methanol |
|--------|------------|---------|---------|------------|----------|
| N47H48 | Monomer | 3641.36 | 3642.94 | 3642.82 | 3642.94 |
| | Dimer | 3564.51 | 3543.60 | 3545.35 | 3543.78 |
| N3H4 | Monomer | 3641.36 | 3642.94 | 3642.82 | 3642.94 |
| | Dimer | 3497.08 | 3504.71 | 3504.83 | 3504.79 |
| N75H76 | Monomer | 3632.76 | 3641.20 | 3641.07 | 3641.19 |
| | Dimer | 3650.19 | 3641.56 | 3641.67 | 3641.57 |
| N81H82 | Monomer | 3632.76 | 3641.20 | 3641.07 | 3641.10 |
| | Dimer | 3529.02 | 3531.40 | 3533.37 | 3531.59 |
| C49O50 | Monomer | 1844.62 | 1837.82 | 1839.51 | 1837.98 |
| | Dimer | 1817.66 | 1805.65 | 1807.66 | 1805.84 |
| C29O30 | Monomer | 1844.62 | 1837.82 | 1839.51 | 1837.98 |
| | Dimer | 1849.68 | 1841.10 | 1842.68 | 1841.35 |
| 01C2 | Monomer | 1814.80 | 1797.63 | 1798.47 | 1797.98 |
| | Dimer | 1754.32 | 1740.64 | 1744.43 | 1741.03 |
| C77O78 | Monomer | 1814.80 | 1798.63 | 1798.47 | 1797.98 |
| | Dimer | 1772.78 | 1755.27 | 1759.47 | 1755.64 |