## ELECTRONIC SUPPLEMENTARY INFORMATION

## Graphene oxide–Li<sup>+</sup>@C<sub>60</sub> donor-acceptor composite for photoenergy conversion

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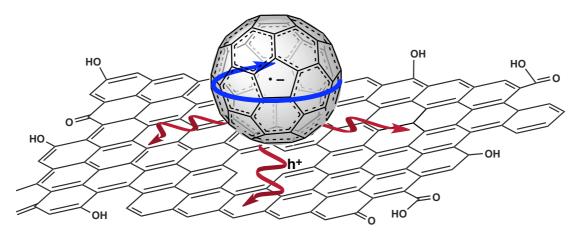


Fig. S1 Representation of the delocalization of radical anion of the fullerene on the  $\pi$ -sphere and the hole migration through the  $\pi$ -conjugations on an assumed portion of a GO surface proposed to occur after the photoinduced electron transfer from GO to fullerene.

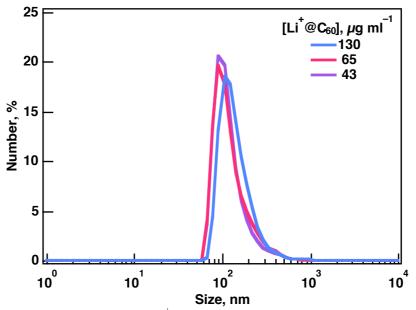


Fig. S2 Size distribution diagram of  $Li^+@C_{60}$  at indicated concentrations in water.

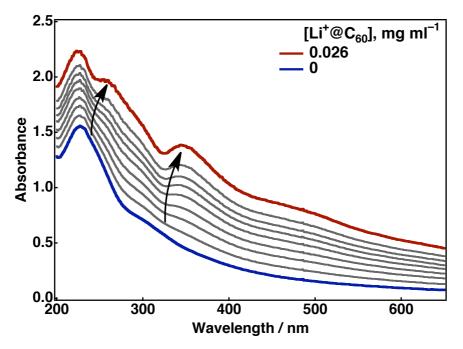


Fig. S3 Absorption spectral changes upon the titration of 0.018 mg ml–1 GO by  $\text{Li}^+@C_{60}$  aqueous dispersion.

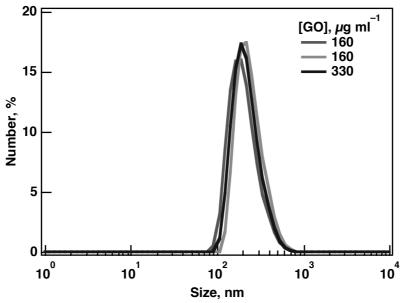
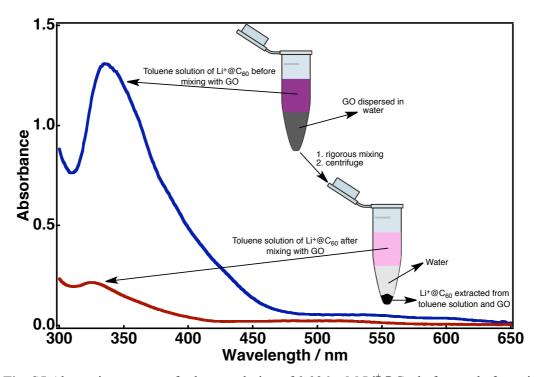
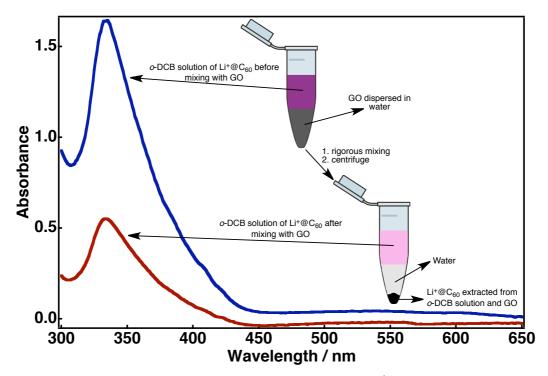


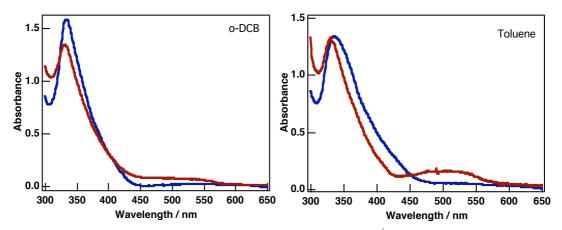
Fig. S4 Size distribution diagram of GO at indicated concentrations in water.



**Fig. S5** Absorption spectra of toluene solution of 0.036 mM  $\text{Li}^+@C_{60}$  before and after mixing with GO aqueous dispersion. The inset describes the mixing process.



**Fig. S6** Absorption spectra of *o*-DCB solution of 0.036 mM  $\text{Li}^+@C_{60}$  before and after mixing with GO aqueous dispersion. The inset describes the mixing process.



**Fig. S7** Absorption spectra of *o*-DCB solution of 0.036 mM  $\text{Li}^+@C_{60}$  before and after mixing with only water.

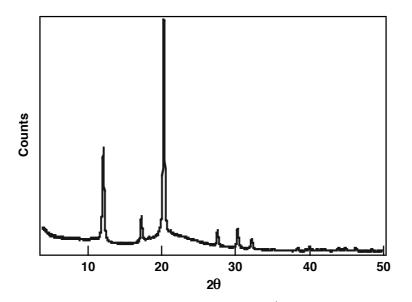


Fig. S8 Powder X-ray diffraction patterns of dried  $Li^+@C_{60}$  nanoparticles obtained from dispersion in water.

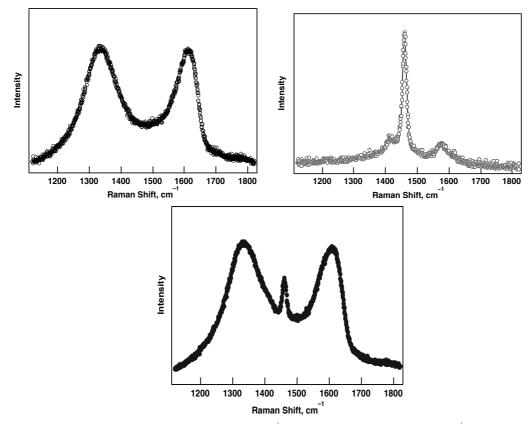
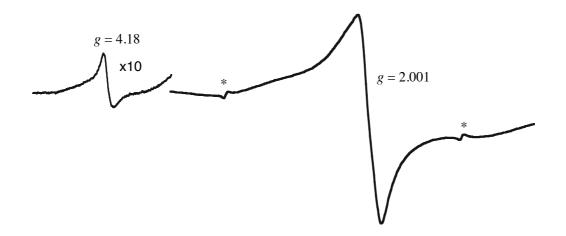


Fig. S9 Raman spectra of GO (top left),  $Li^+@C_{60}$  (top right), and GO– $Li^+@C_{60}$  composite (bottom).



**Fig. S10** EPR spectra of GO–Li<sup>+</sup>@C<sub>60</sub> composite in water showing the triplet signal of  $\text{Li}^+$ @C<sub>60</sub> and the radical ion pair with corresponding *g* values at 4 K observed after photoirradiation.

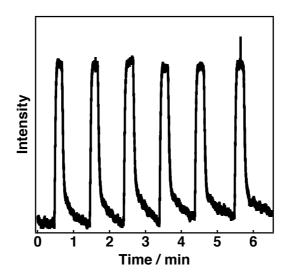


Fig. S11 On-off switching of EPR signal intensity by switching of photoirradiation at 77 K.

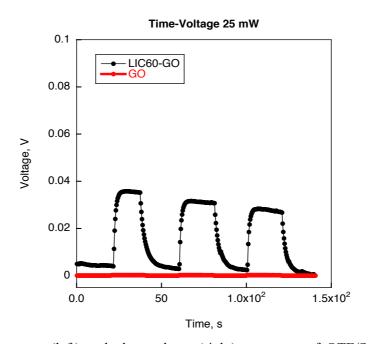


Fig. S12 Photocurrent (left) and photovoltage (right) responses of  $OTE/SnO_2$ -(GO)<sub>n</sub> and  $OTE/SnO_2$ -(GO-Li<sup>+</sup>@C<sub>60</sub>)<sub>n</sub>. Electrolyte: 0.5 M LiI and 0.01 M I<sub>2</sub> in MeCN–PhCN (3 : 1 v/v).