

### Supplementary material:

We present the results of the computations of isolated H<sub>2</sub> and of H<sub>2</sub> embedded into an Ar matrix. We used similar methodology as described in the paper for CO@Ar<sub>n</sub>.

**Table S1:** variation of vibrational frequency of H<sub>2</sub> in argon matrix  $\omega_{H_2}$ .

<b>Molecular system</b>	<b><math>\omega_{H_2}</math> (cm<sup>-1</sup>)</b>	<b><math>\Delta\omega_{H_2}</math> (cm<sup>-1</sup>)</b>
<b>H<sub>2</sub></b>	4392.2	
	4401.2 <sup>a)</sup>	
<b>H<sub>2</sub>@Ar<sub>31</sub></b>	4248.0	144.2
<b>H<sub>2</sub>@Ar<sup>b)</sup></b>	4241.6	159.6

a) Ref. [80].

b) Ref. [83].

The spectroscopic properties of dihydrogen embedded in Ar matrix are following the same variations as those noticed for carbon monoxide. See text for more details.