

# Theoretical Study of the Oxidation Mechanisms of Naphthalene Initiated by Hydroxyl Radicals: The O<sub>2</sub> Addition Reaction Pathways

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## Electronic Supplementary Information (ESI)

**Table S1:** Effective rate constants (in cm<sup>3</sup> molecule<sup>-1</sup> s<sup>-1</sup>) for the reported reaction channels obtained by means of RRKM theory at different pressures and temperatures, according to the computed UM06-2x/aug-cc-pVTZ energy profiles ( $x=1,3$  and  $y=2,4$ ).

**Table S1a:** [ $T=298$  K; *syn* isomer]

Pressure	Reaction	IM <sub>x</sub> →R1+O <sub>2</sub>	IM1→R1-2OO- <i>s</i>	IM3→R1-4OO- <i>s</i>	$k_{\text{eff}}(\mathbf{1})$	$k_{\text{eff}}(\mathbf{3})$
		$k_{-1}$ (s <sup>-1</sup> )	$k_2(\mathbf{1})$ (s <sup>-1</sup> )	$k_2(\mathbf{3})$ (s <sup>-1</sup> )	R1+O <sub>2</sub> ⇌R1-2OO- <i>s</i> (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )	R1+O <sub>2</sub> ⇌R1-4OO- <i>s</i> (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )
1.00E+04		4.36E+16	1.78E+05	1.07E+06	3.93E-19	2.38E-18
1.00E+02		4.36E+16	1.78E+05	1.07E+06	3.93E-19	2.38E-18
1.00E+00		4.36E+16	1.77E+05	1.07E+06	3.93E-19	2.36E-18
1.00E-02		4.36E+16	1.54E+05	7.55E+05	3.41E-19	1.67E-18
1.00E-04		4.36E+16	2.82E+04	6.65E+04	6.23E-20	1.47E-19
1.00E-06		4.36E+16	7.09E+02	1.12E+03	1.57E-21	2.49E-21
1.00E-08		4.36E+16	8.02E+00	1.17E+01	1.78E-23	2.58E-23
1.00E-10		4.36E+16	8.04E-02	1.17E-01	1.78E-25	2.59E-25
1.00E-12		4.36E+16	8.04E-04	1.17E-03	1.78E-27	2.59E-27

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**Table S1b:** [ $T= 336$  K; *syn* isomer]

Pressure	Reaction	IMx→R1+O <sub>2</sub>	IM1→R1-2OO- <i>s</i>	IM3→R1-4OO- <i>s</i>	$k_{\text{eff}}(\mathbf{1})$	$k_{\text{eff}}(\mathbf{3})$
		$k_{-1}$ (s <sup>-1</sup> )	$k_2(\mathbf{1})$ (s <sup>-1</sup> )	$k_2(\mathbf{3})$ (s <sup>-1</sup> )	R1+O <sub>2</sub> ⇌R1-2OO- <i>s</i> (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )	R1+O <sub>2</sub> ⇌R1-4OO- <i>s</i> (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )
1.00E+04		1.80E+16	7.11E+05	3.57E+06	1.05E-18	5.29E-18
1.00E+02		1.80E+16	7.11E+05	3.57E+06	1.05E-18	5.29E-18
1.00E+00		1.80E+16	7.08E+05	3.53E+06	1.05E-18	5.23E-18
1.00E-02		1.80E+16	5.49E+05	2.03E+06	8.13E-19	3.01E-18
1.00E-04		1.80E+16	6.23E+04	1.12E+05	9.23E-20	1.67E-19
1.00E-06		1.80E+16	1.19E+03	1.59E+03	1.76E-21	2.35E-21
1.00E-08		1.80E+16	1.29E+01	1.62E+01	1.91E-23	2.41E-23
1.00E-10		1.80E+16	1.29E-01	1.62E-01	1.91E-25	2.41E-25
1.00E-12		1.80E+16	1.29E-03	1.62E-03	1.91E-27	2.41E-27

**Table S1c:** [ $T= 400$  K; *syn* isomer]

Pressure	Reaction	IMx→R1+O <sub>2</sub>	IM1→R1-2OO- <i>s</i>	IM3→R1-4OO- <i>s</i>	$k_{\text{eff}}(\mathbf{1})$	$k_{\text{eff}}(\mathbf{3})$
		$k_{-1}$ (s <sup>-1</sup> )	$k_2(\mathbf{1})$ (s <sup>-1</sup> )	$k_2(\mathbf{3})$ (s <sup>-1</sup> )	R1+O <sub>2</sub> ⇌R1-2OO- <i>s</i> (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )	R1+O <sub>2</sub> ⇌R1-4OO- <i>s</i> (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )
1.00E+04		6.11E+15	4.00E+06	1.59E+07	3.69E-18	1.47E-17
1.00E+02		6.11E+15	4.00E+06	1.59E+07	3.69E-18	1.47E-17
1.00E+00		6.11E+15	3.96E+06	1.55E+07	3.65E-18	1.43E-17
1.00E-02		6.11E+15	2.31E+06	5.74E+06	2.13E-18	5.29E-18
1.00E-04		6.11E+15	1.25E+05	1.67E+05	1.15E-19	1.54E-19
1.00E-06		6.11E+15	1.73E+03	1.95E+03	1.60E-21	1.79E-21
1.00E-08		6.11E+15	1.79E+01	1.96E+01	1.65E-23	1.81E-23
1.00E-10		6.11E+15	1.79E-01	1.96E-01	1.65E-25	1.81E-25
1.00E-12		6.11E+15	1.79E-03	1.96E-03	1.65E-27	1.81E-27

**Table S1d:** [ $T= 298$  K; *anti* isomer]

Pressure	Reaction	IMy→R1+O <sub>2</sub>	IM2→R1-2OO- <i>a</i>	IM4→R1-4OO- <i>a</i>	$k_{\text{eff}}(\mathbf{2})$	$k_{\text{eff}}(\mathbf{4})$
		$k_{-1}$ (s <sup>-1</sup> )	$k_2(\mathbf{2})$ (s <sup>-1</sup> )	$k_2(\mathbf{4})$ (s <sup>-1</sup> )	R1+O <sub>2</sub> ⇌R1-2OO- <i>a</i> (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )	R1+O <sub>2</sub> ⇌R1-4OO- <i>a</i> (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )
1.00E+04		1.88E+19	1.18E+06	9.39E+06	4.92E-21	3.90E-20
1.00E+02		1.88E+19	1.18E+06	9.38E+06	4.92E-21	3.90E-20
1.00E+00		1.88E+19	1.17E+06	8.92E+06	4.85E-21	3.71E-20
1.00E-02		1.88E+19	6.54E+05	2.82E+06	2.72E-21	1.17E-20
1.00E-04		1.88E+19	4.44E+04	9.44E+04	1.85E-22	3.92E-22
1.00E-06		1.88E+19	7.47E+02	1.19E+03	3.11E-24	4.95E-24
1.00E-08		1.88E+19	7.83E+00	1.20E+01	3.26E-26	4.99E-26
1.00E-10		1.88E+19	7.84E-02	1.20E-01	3.26E-28	4.99E-28
1.00E-12		1.88E+19	7.84E-04	1.20E-03	3.26E-30	4.99E-30

**Table S1e:** [ $T= 336$  K; *anti* isomer]

Pressure	Reaction	IM <sub>y</sub> →R1+O <sub>2</sub>	IM <sub>2</sub> →R1-2OO- <i>a</i>	IM <sub>4</sub> →R1-4OO- <i>a</i>	$k_{\text{eff}}(\mathbf{2})$	$k_{\text{eff}}(\mathbf{4})$
		$k_{-1}$ (s <sup>-1</sup> )	$k_2(\mathbf{2})$ (s <sup>-1</sup> )	$k_2(\mathbf{4})$ (s <sup>-1</sup> )	R1+O <sub>2</sub> ⇌R1-2OO- <i>a</i> (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )	R1+O <sub>2</sub> ⇌R1-4OO- <i>a</i> (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )
1.00E+04		3.92E+18	5.00E+06	3.19E+07	2.45E-20	1.56E-19
1.00E+02		3.92E+18	4.99E+06	3.18E+07	2.45E-20	1.56E-19
1.00E+00		3.92E+18	4.85E+06	2.90E+07	2.37E-20	1.42E-19
1.00E-02		3.92E+18	2.00E+06	6.01E+06	9.79E-21	2.94E-20
1.00E-04		3.92E+18	8.70E+04	1.42E+05	4.26E-22	6.97E-22
1.00E-06		3.92E+18	1.23E+03	1.64E+03	6.04E-24	8.05E-24
1.00E-08		3.92E+18	1.27E+01	1.65E+01	6.21E-26	8.09E-26
1.00E-10		3.92E+18	1.27E-01	1.65E-01	6.21E-28	8.09E-28
1.00E-12		3.92E+18	1.27E-03	1.65E-03	6.21E-30	8.09E-30

**Table S1f:** [ $T= 400$  K; *anti* isomer]

Pressure	Reaction	IM <sub>y</sub> →R1+O <sub>2</sub>	IM <sub>2</sub> →R1-2OO- <i>a</i>	IM <sub>4</sub> →R1-4OO- <i>a</i>	$k_{\text{eff}}(\mathbf{2})$	$k_{\text{eff}}(\mathbf{4})$
		$k_{-1}$ (s <sup>-1</sup> )	$k_2(\mathbf{2})$ (s <sup>-1</sup> )	$k_2(\mathbf{4})$ (s <sup>-1</sup> )	R1+O <sub>2</sub> ⇌R1-2OO- <i>a</i> (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )	R1+O <sub>2</sub> ⇌R1-4OO- <i>a</i> (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> )
1.00E+04		5.61E+17	3.01E+07	1.46E+08	1.87E-19	9.10E-19
1.00E+02		5.61E+17	3.01E+07	1.46E+08	1.87E-19	9.08E-19
1.00E+00		5.61E+17	2.77E+07	1.17E+08	1.72E-19	7.29E-19
1.00E-02		5.61E+17	6.23E+06	1.19E+07	3.87E-20	7.40E-20
1.00E-04		5.61E+17	1.50E+05	1.86E+05	9.32E-22	1.16E-21
1.00E-06		5.61E+17	1.76E+03	1.97E+03	1.09E-23	1.22E-23
1.00E-08		5.61E+17	1.78E+01	1.97E+01	1.11E-25	1.23E-25
1.00E-10		5.61E+17	1.78E-01	1.97E-01	1.11E-27	1.23E-27
1.00E-12		5.61E+17	1.78E-03	1.97E-03	1.11E-29	1.23E-29

**Table S2:** Kinetic rate constants (in  $\text{s}^{-1}$ ), effective rate constants, and branching ratios for all reaction steps involved in the reported chemical pathways at ambient temperature and different pressures using the RRKM theory, according to the computed UM06-2x/aug-cc-pVTZ energy profiles ( $z=1,2,3,4$ ).

**Table S2a:**

Parameter Reaction pathway	$P = 10^4 \text{ bar}$				
	Rate constant			effective rate constant	Branching ratio ( $R$ )
	$K_p$	$k_{-1}$ IMz $\rightarrow$ R1+O <sub>2</sub>	$k_2$ IMz $\rightarrow$ R2	$k_{\text{eff}}$ (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> ) R1+O <sub>2</sub> $\rightleftharpoons$ R2	
<b><math>T=298 \text{ K}</math></b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	5.38E-05	4.36E+16	1.78E+05	3.93E-19	13.95
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	5.38E-05	4.36E+16	1.07E+06	2.38E-18	84.49
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.01E-07	1.88E+19	1.18E+06	4.92E-21	0.17
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.01E-07	1.88E+19	9.39E+06	3.90E-20	1.38
<b><math>T=336 \text{ K}</math></b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	3.60E-05	4.36E+16	7.11E+05	1.05E-18	16.10
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	3.60E-05	4.36E+16	3.57E+06	5.29E-18	81.13
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.19E-07	1.88E+19	5.00E+06	2.45E-20	0.38
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.19E-07	1.88E+19	3.19E+07	1.56E-19	2.39
<b><math>T=400 \text{ K}</math></b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	2.24E-05	4.36E+16	4.00E+06	3.69E-18	18.94
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	2.24E-05	4.36E+16	1.59E+07	1.47E-17	75.43
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.51E-07	1.88E+19	3.01E+07	1.87E-19	0.96
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.51E-07	1.88E+19	1.46E+08	9.10E-19	4.67

**Table S2b:**

Parameter Reaction pathway	$P = 10^2 \text{ bar}$				
	Rate constant			effective rate constant	Branching ratio ( $R$ )
	$K_p$	$k_{-1}$ IMz $\rightarrow$ R1+O <sub>2</sub>	$k_2$ IMz $\rightarrow$ R2	$k_{\text{eff}}$ (cm <sup>3</sup> molecule <sup>-1</sup> s <sup>-1</sup> ) R1+O <sub>2</sub> $\rightleftharpoons$ R2	
<b><math>T=298 \text{ K}</math></b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	5.38E-05	4.36E+16	1.78E+05	3.93E-19	13.95
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	5.38E-05	4.36E+16	1.07E+06	2.38E-18	84.49
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.01E-07	1.88E+19	1.18E+06	4.92E-21	0.17
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.01E-07	1.88E+19	9.38E+06	3.90E-20	1.38
<b><math>T=336 \text{ K}</math></b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	3.60E-05	4.36E+16	7.11E+05	1.05E-18	16.10
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	3.60E-05	4.36E+16	3.57E+06	5.29E-18	81.13
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.19E-07	1.88E+19	4.99E+06	2.45E-20	0.38
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.19E-07	1.88E+19	3.18E+07	1.56E-19	2.39
<b><math>T=400 \text{ K}</math></b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	2.24E-05	4.36E+16	4.00E+06	3.69E-18	18.94
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	2.24E-05	4.36E+16	1.59E+07	1.47E-17	75.44
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.51E-07	1.88E+19	3.01E+07	1.87E-19	0.96
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.51E-07	1.88E+19	1.46E+08	9.08E-19	4.66

**Table S2c:**

Reaction pathway	<b><math>P = 1.0</math> bar</b>				
	$K_p$	Rate constant		effective rate constant $k_{\text{eff}}(\text{cm}^3 \text{ molecule}^{-1} \text{ s}^{-1})$	Branching ratio ( $R$ )
		$k_{-1}$ IMz $\rightarrow$ R1+O <sub>2</sub>	$k_2$ IMz $\rightarrow$ R2		
<b><math>T=298</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	5.38E-05	4.36E+16	1.77E+05	3.93E-19	14.06
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	5.38E-05	4.36E+16	1.07E+06	2.36E-18	84.44
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.01E-07	1.88E+19	1.17E+06	4.85E-21	0.17
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.01E-07	1.88E+19	8.92E+06	3.71E-20	1.33
<b><math>T=336</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	3.60E-05	4.36E+16	7.08E+05	1.05E-18	16.29
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	3.60E-05	4.36E+16	3.53E+06	5.23E-18	81.14
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.19E-07	1.88E+19	4.85E+06	2.37E-20	0.37
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.19E-07	1.88E+19	2.90E+07	1.42E-19	2.20
<b><math>T=400</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	2.24E-05	4.36E+16	3.96E+06	3.65E-18	19.36
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	2.24E-05	4.36E+16	1.55E+07	1.43E-17	75.86
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.51E-07	1.88E+19	2.77E+07	1.72E-19	0.91
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.51E-07	1.88E+19	1.17E+08	7.29E-19	3.87

**Table S2d:**

Reaction pathway	<b><math>P = 10^{-2}</math> bar</b>				
	$K_p$	Rate constant		effective rate constant $k_{\text{eff}}(\text{cm}^3 \text{ molecule}^{-1} \text{ s}^{-1})$	Branching ratio ( $R$ )
		$k_{-1}$ IMz $\rightarrow$ R1+O <sub>2</sub>	$k_2$ IMz $\rightarrow$ R2		
<b><math>T=298</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	5.38E-05	4.36E+16	1.54E+05	3.41E-19	16.84
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	5.38E-05	4.36E+16	7.55E+05	1.67E-18	82.45
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.01E-07	1.88E+19	6.54E+05	2.72E-21	0.13
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.01E-07	1.88E+19	2.82E+06	1.17E-20	0.58
<b><math>T=336</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	3.60E-05	4.36E+16	5.49E+05	8.13E-19	21.05
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	3.60E-05	4.36E+16	2.03E+06	3.01E-18	77.94
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.19E-07	1.88E+19	2.00E+06	9.79E-21	0.25
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.19E-07	1.88E+19	6.01E+06	2.94E-20	0.76
<b><math>T=400</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	2.24E-05	4.36E+16	2.31E+06	2.13E-18	28.28
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	2.24E-05	4.36E+16	5.74E+06	5.29E-18	70.23
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.51E-07	1.88E+19	6.23E+06	3.87E-20	0.51
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.51E-07	1.88E+19	1.19E+07	7.40E-20	0.98

Table S2e:

Parameter Reaction pathway	$P = 10^{-4}$ bar				
	Rate constant			effective rate constant $k_{\text{eff}}(\text{cm}^3 \text{ molecule}^{-1} \text{ s}^{-1})$	Branching ratio ( $R$ )
	$K_p$	$k_{-1}$ IMz $\rightarrow$ R1+O <sub>2</sub>	$k_2$ IMz $\rightarrow$ R2		
<b><math>T=298</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	5.38E-05	4.36E+16	2.82E+04	6.23E-20	29.68
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	5.38E-05	4.36E+16	6.65E+04	1.47E-19	70.04
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.01E-07	1.88E+19	4.44E+04	1.85E-22	0.09
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.01E-07	1.88E+19	9.44E+04	3.92E-22	0.19
<b><math>T=336</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	3.60E-05	4.36E+16	6.23E+04	9.23E-20	35.44
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	3.60E-05	4.36E+16	1.12E+05	1.67E-19	64.13
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.19E-07	1.88E+19	8.70E+04	4.26E-22	0.16
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.19E-07	1.88E+19	1.42E+05	6.97E-22	0.27
<b><math>T=400</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	2.24E-05	4.36E+16	1.25E+05	1.15E-19	42.42
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	2.24E-05	4.36E+16	1.67E+05	1.54E-19	56.81
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.51E-07	1.88E+19	1.50E+05	9.32E-22	0.34
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.51E-07	1.88E+19	1.86E+05	1.16E-21	0.43

Table S2f:

Parameter Reaction pathway	$P = 10^{-6}$ bar				
	Rate constant			effective rate constant $k_{\text{eff}}(\text{cm}^3 \text{ molecule}^{-1} \text{ s}^{-1})$	Branching ratio ( $R$ )
	$K_p$	$k_{-1}$ IMz $\rightarrow$ R1+O <sub>2</sub>	$k_2$ IMz $\rightarrow$ R2		
<b><math>T=298</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	5.38E-05	4.36E+16	7.09E+02	1.57E-21	38.59
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	5.38E-05	4.36E+16	1.12E+03	2.49E-21	61.21
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.01E-07	1.88E+19	7.47E+02	3.11E-24	0.08
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.01E-07	1.88E+19	1.19E+03	4.95E-24	0.12
<b><math>T=336</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	3.60E-05	4.36E+16	1.19E+03	1.76E-21	42.68
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	3.60E-05	4.36E+16	1.59E+03	2.35E-21	56.98
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.19E-07	1.88E+19	1.23E+03	6.04E-24	0.15
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.19E-07	1.88E+19	1.64E+03	8.05E-24	0.20
<b><math>T=400</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	2.24E-05	4.36E+16	1.73E+03	1.60E-21	46.88
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	2.24E-05	4.36E+16	1.95E+03	1.79E-21	52.44
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.51E-07	1.88E+19	1.76E+03	1.09E-23	0.32
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.51E-07	1.88E+19	1.97E+03	1.22E-23	0.36

**Table S2g:**

Reaction pathway	$P = 10^{-8}$ bar					Branching ratio ( $R$ )
	Rate constant			effective rate constant $k_{\text{eff}}(\text{cm}^3 \text{ molecule}^{-1} \text{ s}^{-1})$		
	$K_p$	$k_{-1}$ IMz $\rightarrow$ R1+O <sub>2</sub>	$k_2$ IMz $\rightarrow$ R2		R1+O <sub>2</sub> $\rightleftharpoons$ R2	
<b><math>T=298</math> K</b>						
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	5.38E-05	4.36E+16	8.02E+00	1.78E-23	40.75	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	5.38E-05	4.36E+16	1.17E+01	2.58E-23	59.06	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.01E-07	1.88E+19	7.83E+00	3.26E-26	0.07	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.01E-07	1.88E+19	1.20E+01	4.99E-26	0.11	
<b><math>T=336</math> K</b>						
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	3.60E-05	4.36E+16	1.29E+01	1.91E-23	44.07	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	3.60E-05	4.36E+16	1.62E+01	2.41E-23	55.60	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.19E-07	1.88E+19	1.27E+01	6.21E-26	0.14	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.19E-07	1.88E+19	1.65E+01	8.09E-26	0.19	
<b><math>T=400</math> K</b>						
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	2.24E-05	4.36E+16	1.79E+01	1.65E-23	47.37	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	2.24E-05	4.36E+16	1.96E+01	1.81E-23	51.96	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.51E-07	1.88E+19	1.78E+01	1.11E-25	0.32	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.51E-07	1.88E+19	1.97E+01	1.23E-25	0.35	

**Table S2h:**

Reaction pathway	$P = 10^{-10}$ bar					Branching ratio ( $R$ )
	Rate constant			effective rate constant $k_{\text{eff}}(\text{cm}^3 \text{ molecule}^{-1} \text{ s}^{-1})$		
	$K_p$	$k_{-1}$ IMz $\rightarrow$ R1+O <sub>2</sub>	$k_2$ IMz $\rightarrow$ R2		R1+O <sub>2</sub> $\rightleftharpoons$ R2	
<b><math>T=298</math> K</b>						
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	5.38E-05	4.36E+16	8.04E-02	1.78E-25	40.66	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	5.38E-05	4.36E+16	1.17E-01	2.59E-25	59.16	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.01E-07	1.88E+19	7.84E-02	3.26E-28	0.07	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.01E-07	1.88E+19	1.20E-01	4.99E-28	0.11	
<b><math>T=336</math> K</b>						
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	3.60E-05	4.36E+16	1.29E-01	1.91E-25	44.07	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	3.60E-05	4.36E+16	1.62E-01	2.41E-25	55.60	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.19E-07	1.88E+19	1.27E-01	6.21E-28	0.14	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.19E-07	1.88E+19	1.65E-01	8.09E-28	0.19	
<b><math>T=400</math> K</b>						
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	2.24E-05	4.36E+16	1.79E-01	1.65E-25	47.37	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	2.24E-05	4.36E+16	1.96E-01	1.81E-25	51.96	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.51E-07	1.88E+19	1.78E-01	1.11E-27	0.32	
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.51E-07	1.88E+19	1.97E-01	1.23E-27	0.35	

**Table S2i:**

Reaction pathway	$P = 10^{-12}$ bar				
	Rate constant			effective rate constant $k_{\text{eff}}(\text{cm}^3 \text{ molecule}^{-1} \text{ s}^{-1})$	Branching ratio ( $R$ )
	$K_p$	$k_{-1}$ IMz $\rightarrow$ R1+O <sub>2</sub>	$k_2$ IMz $\rightarrow$ R2		
<b><math>T=298</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	5.38E-05	4.36E+16	8.04E-04	1.78E-27	40.66
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	5.38E-05	4.36E+16	1.17E-03	2.59E-27	59.16
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.01E-07	1.88E+19	7.84E-04	3.26E-30	0.07
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.01E-07	1.88E+19	1.20E-03	4.99E-30	0.11
<b><math>T=336</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	3.60E-05	4.36E+16	1.29E-03	1.91E-27	44.07
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	3.60E-05	4.36E+16	1.62E-03	2.41E-27	55.60
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.19E-07	1.88E+19	1.27E-03	6.21E-30	0.14
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.19E-07	1.88E+19	1.65E-03	8.09E-30	0.19
<b><math>T=400</math> K</b>					
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>syn</i>	2.24E-05	4.36E+16	1.79E-03	1.65E-27	47.37
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>syn</i>	2.24E-05	4.36E+16	1.96E-03	1.81E-27	51.96
R1+O <sub>2</sub> $\rightleftharpoons$ R1-2OO- <i>anti</i>	1.51E-07	1.88E+19	1.78E-03	1.11E-29	0.32
R1+O <sub>2</sub> $\rightleftharpoons$ R1-4OO- <i>anti</i>	1.51E-07	1.88E+19	1.97E-03	1.23E-29	0.35

**Table S3:** Dependence upon the pressure and temperature of the regioselectivities [ $\text{RSI} = R(3) - R(1) / R(1) + R(3)$ ] and [ $\text{RSI} = R(4) - R(2) / R(2) + R(4)$ ] of O<sub>2</sub> addition in *syn* and *anti* modes onto the naphthalene-OH adduct [C<sub>10</sub>H<sub>8</sub>OH]<sup>•</sup>, according to the RRKM estimates of effective rate constants [ $k_{\text{eff}}(1)$ ,  $k_{\text{eff}}(2)$ ,  $k_{\text{eff}}(3)$ ,  $k_{\text{eff}}(4)$ ].

**Table S3a:** ( $T= 298$  K; *syn* isomer)

$P$ (bar)	$k_{\text{eff}}(1)$	$k_{\text{eff}}(3)$	$k_{\text{eff}}[(1)+(3)]$	$R(1)$	$R(3)$	Log $P$	$[R(3) - R(1)] / [R(3) + R(1)]$
1.00E+04	3.93E-19	2.38E-18	2.77E-18	13.95	84.49	4	0.7166
1.00E+02	3.93E-19	2.38E-18	2.77E-18	13.95	84.49	2	0.7166
1.00E+00	3.93E-19	2.36E-18	2.75E-18	14.06	84.44	0	0.7145
1.00E-02	3.41E-19	1.67E-18	2.01E-18	16.84	82.45	-2	0.6609
1.00E-04	6.23E-20	1.47E-19	2.09E-19	29.68	70.04	-4	0.4047
1.00E-06	1.57E-21	2.49E-21	4.06E-21	38.59	61.21	-6	0.2266
1.00E-08	1.78E-23	2.58E-23	4.36E-23	40.75	59.06	-8	0.1835
1.00E-10	1.78E-25	2.59E-25	4.37E-25	40.66	59.16	-10	0.1854
1.00E-12	1.78E-27	2.59E-27	4.37E-27	40.66	59.16	-12	0.1854



**Table S3b:** ( $T= 336$  K; *syn* isomer)

$P$ (bar)	$k_{\text{eff}}(\mathbf{1})$	$k_{\text{eff}}(\mathbf{3})$	$k_{\text{eff}}[(\mathbf{1})+(\mathbf{3})]$	$R(\mathbf{1})$	$R(\mathbf{3})$	$\text{Log } P$	$[R(\mathbf{3})-R(\mathbf{1})]/[R(\mathbf{3})+R(\mathbf{1})]$
1.00E+04	1.05E-18	5.29E-18	6.34E-18	16.10	81.13	4	0.6688
1.00E+02	1.05E-18	5.29E-18	6.34E-18	16.10	81.13	2	0.6688
1.00E+00	1.05E-18	5.23E-18	6.28E-18	16.29	81.14	0	0.6656
1.00E-02	8.13E-19	3.01E-18	3.82E-18	21.05	77.94	-2	0.5747
1.00E-04	9.23E-20	1.67E-19	2.59E-19	35.44	64.13	-4	0.2881
1.00E-06	1.76E-21	2.35E-21	4.11E-21	42.68	56.98	-6	0.1436
1.00E-08	1.91E-23	2.41E-23	4.32E-23	44.07	55.60	-8	0.1157
1.00E-10	1.91E-25	2.41E-25	4.32E-25	44.07	55.60	-10	0.1157
1.00E-12	1.91E-27	2.41E-27	4.32E-27	44.07	55.60	-12	0.1157

**Table S3c:** ( $T= 400$  K; *syn* isomer)

$P$ (bar)	$k_{\text{eff}}(\mathbf{1})$	$k_{\text{eff}}(\mathbf{3})$	$k_{\text{eff}}[(\mathbf{1})+(\mathbf{3})]$	$R(\mathbf{1})$	$R(\mathbf{3})$	$\text{Log } P$	$[R(\mathbf{3})-R(\mathbf{1})]/[R(\mathbf{3})+R(\mathbf{1})]$
1.00E+04	3.69E-18	1.47E-17	1.84E-17	18.94	75.43	4	0.5987
1.00E+02	3.69E-18	1.47E-17	1.84E-17	18.94	75.44	2	0.5987
1.00E+00	3.65E-18	1.43E-17	1.80E-17	19.36	75.86	0	0.5933
1.00E-02	2.13E-18	5.29E-18	7.42E-18	28.28	70.23	-2	0.4259
1.00E-04	1.15E-19	1.54E-19	2.69E-19	42.42	56.81	-4	0.1450
1.00E-06	1.60E-21	1.79E-21	3.39E-21	46.88	52.44	-6	0.0560
1.00E-08	1.65E-23	1.81E-23	3.46E-23	47.37	51.96	-8	0.0462
1.00E-10	1.65E-25	1.81E-25	3.46E-25	47.37	51.96	-10	0.0462
1.00E-12	1.65E-27	1.81E-27	3.46E-27	47.37	51.96	-12	0.0462

**Table S3d:** ( $T= 298$  K; *anti* isomer)

$P$ (bar)	$k_{\text{eff}}(\mathbf{2})$	$k_{\text{eff}}(\mathbf{4})$	$k_{\text{eff}}[(\mathbf{2})+(\mathbf{4})]$	$R(\mathbf{2})$	$R(\mathbf{4})$	$\text{Log } P$	$[R(\mathbf{4})-R(\mathbf{2})]/[R(\mathbf{4})+R(\mathbf{2})]$
1.00E+04	4.92E-21	3.90E-20	4.39E-20	0.17	1.38	4	0.7760
1.00E+02	4.92E-21	3.90E-20	4.39E-20	0.17	1.38	2	0.7760
1.00E+00	4.85E-21	3.71E-20	4.20E-20	0.17	1.33	0	0.7688
1.00E-02	2.72E-21	1.17E-20	1.44E-20	0.13	0.58	-2	0.6227
1.00E-04	1.85E-22	3.92E-22	5.77E-22	0.09	0.19	-4	0.3588
1.00E-06	3.11E-24	4.95E-24	8.06E-24	0.08	0.12	-6	0.2283
1.00E-08	3.26E-26	4.99E-26	8.25E-26	0.07	0.11	-8	0.2097
1.00E-10	3.26E-28	4.99E-28	8.25E-28	0.07	0.11	-10	0.2097
1.00E-12	3.26E-30	4.99E-30	8.25E-30	0.07	0.11	-12	0.2097

**Table S3e:** ( $T= 336$  K; *anti* isomer)

$P$ (bar)	$k_{\text{eff}}(\mathbf{2})$	$k_{\text{eff}}(\mathbf{4})$	$k_{\text{eff}}[(\mathbf{2})+(\mathbf{4})]$	$R(\mathbf{2})$	$R(\mathbf{4})$	Log $P$	$[R(\mathbf{4})-R(\mathbf{2})]/[R(\mathbf{4})+R(\mathbf{2})]$
1.00E+04	2.45E-20	1.56E-19	1.81E-19	0.38	2.39	4	0.7285
1.00E+02	2.45E-20	1.56E-19	1.81E-19	0.38	2.39	2	0.7285
1.00E+00	2.37E-20	1.42E-19	1.66E-19	0.37	2.20	0	0.7139
1.00E-02	9.79E-21	2.94E-20	3.92E-20	0.25	0.76	-2	0.5004
1.00E-04	4.26E-22	6.97E-22	1.12E-21	0.16	0.27	-4	0.2413
1.00E-06	6.04E-24	8.05E-24	1.41E-23	0.15	0.20	-6	0.1427
1.00E-08	6.21E-26	8.09E-26	1.43E-25	0.14	0.19	-8	0.1315
1.00E-10	6.21E-28	8.09E-28	1.43E-27	0.14	0.19	-10	0.1315
1.00E-12	6.21E-30	8.09E-30	1.43E-29	0.14	0.19	-12	0.1315

**Table S3f:** ( $T= 400$  K; *anti* isomer)

$P$ (bar)	$k_{\text{eff}}(\mathbf{2})$	$k_{\text{eff}}(\mathbf{4})$	$k_{\text{eff}}[(\mathbf{2})+(\mathbf{4})]$	$R(\mathbf{2})$	$R(\mathbf{4})$	Log $P$	$[R(\mathbf{4})-R(\mathbf{2})]/[R(\mathbf{4})+R(\mathbf{2})]$
1.00E+04	1.87E-19	9.10E-19	1.10E-18	0.96	4.67	4	0.6591
1.00E+02	1.87E-19	9.08E-19	1.10E-18	0.96	4.66	2	0.6584
1.00E+00	1.72E-19	7.29E-19	9.01E-19	0.91	3.87	0	0.6182
1.00E-02	3.87E-20	7.40E-20	1.13E-19	0.51	0.98	-2	0.3132
1.00E-04	9.32E-22	1.16E-21	2.09E-21	0.34	0.43	-4	0.1090
1.00E-06	1.09E-23	1.22E-23	2.31E-23	0.32	0.36	-6	0.0563
1.00E-08	1.11E-25	1.23E-25	2.34E-25	0.32	0.35	-8	0.0513
1.00E-10	1.11E-27	1.23E-27	2.34E-27	0.32	0.35	-10	0.0513
1.00E-12	1.11E-29	1.23E-29	2.34E-29	0.32	0.35	-12	0.0513