

Electronic Supplementary Information

Twice as smart behavior of tert-butylthiacalix[4]arene derivative in glassy and crystalline form

K. V. Gataullina,^a M. A. Ziganshin,^a I. I. Stoikov,^a A. T. Gubaidullin^b and V. V. Gorbatchuk^{*a}

^{a.} *A.M. Butlerov Institute of Chemistry, Kazan Federal University, Kremlevskaya 18, 420008 Kazan, Russia. E-mail: Valery.Gorbatchuk@kpfu.ru*

^{b.} *A.E. Arbuzov Institute of Organic and Physical Chemistry, Akad. Arbuzova, 8, 420088 Kazan, Russia.*

TG/DSC, DSC, TM-DSC and TG/DSC/MS curves for the studied calixarene samples in glassy and crystalline form and TG/DSC/MS curves for the products of their saturation with guest vapors and for metastable β_0 polymorph formed after guest release; visual changes of glassy calixarene in this process and at the heating; molar enthalpies of fusion and $\alpha \rightarrow \alpha'$ polymorphic transition

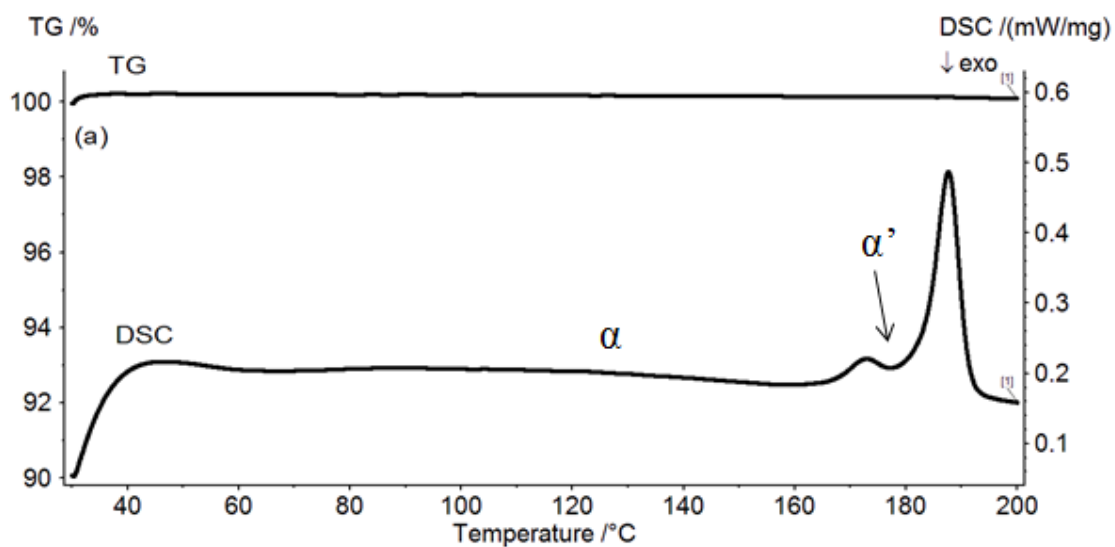


Figure 1S – TG/DSC curves for **1a** sample.

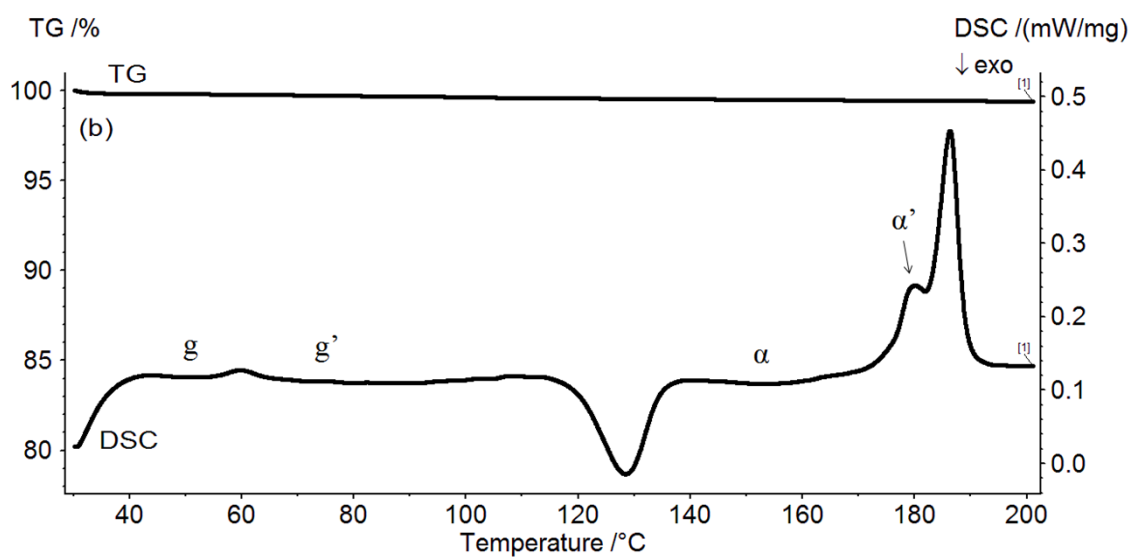


Figure 2S – TG/DSC curves for **1g** sample.

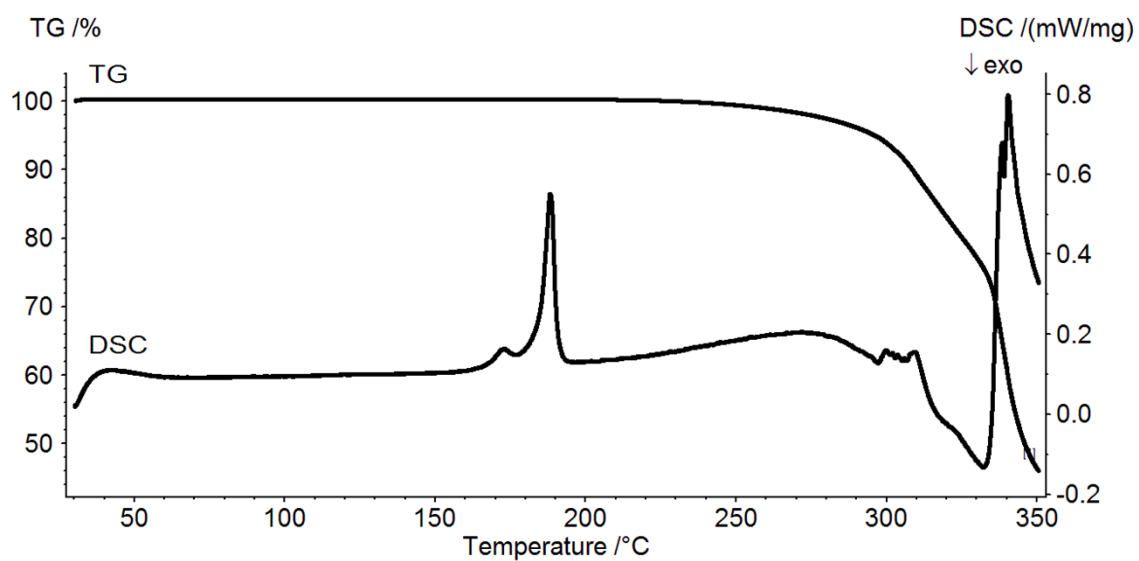


Figure 3S – TG/DSC curves for **1a** sample.

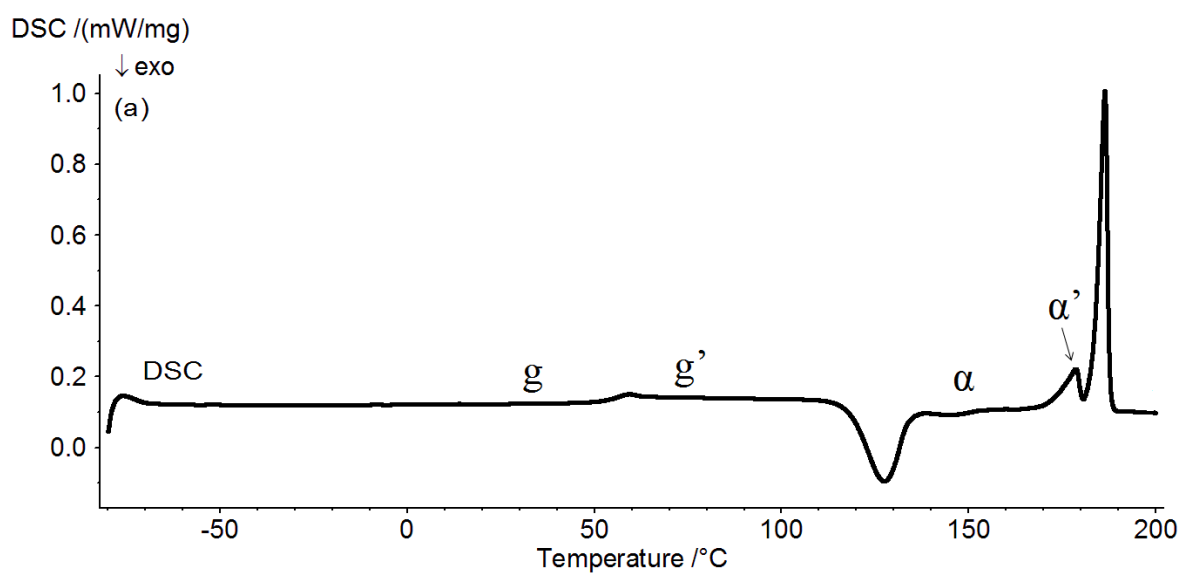


Figure 4S – DSC curves for **1g** sample. Heating rate is 5°C/min.

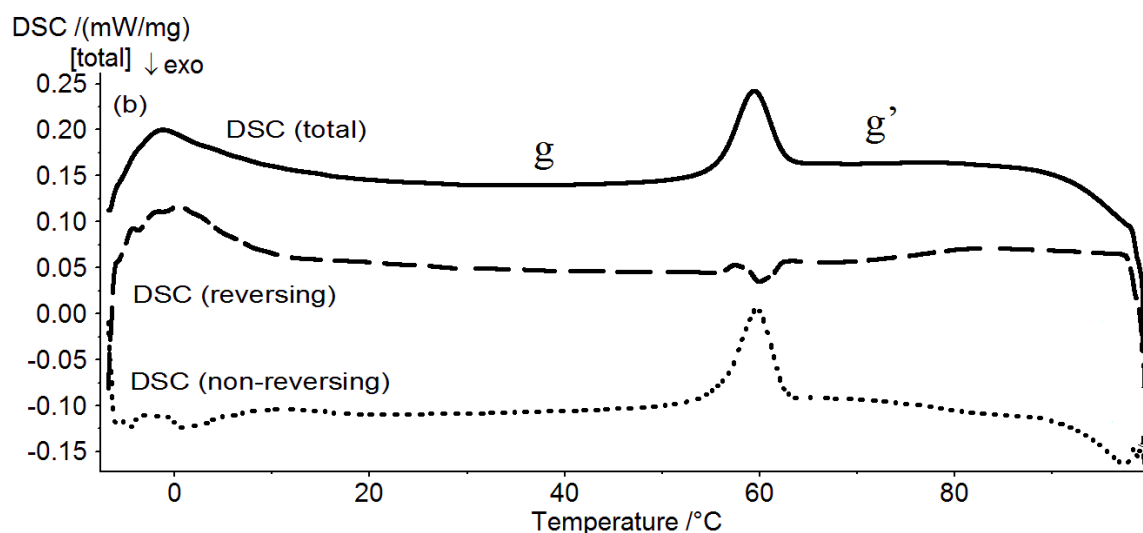


Figure 5S – TM-DSC curves for **1g** sample. Heating rate is 10°C/min.

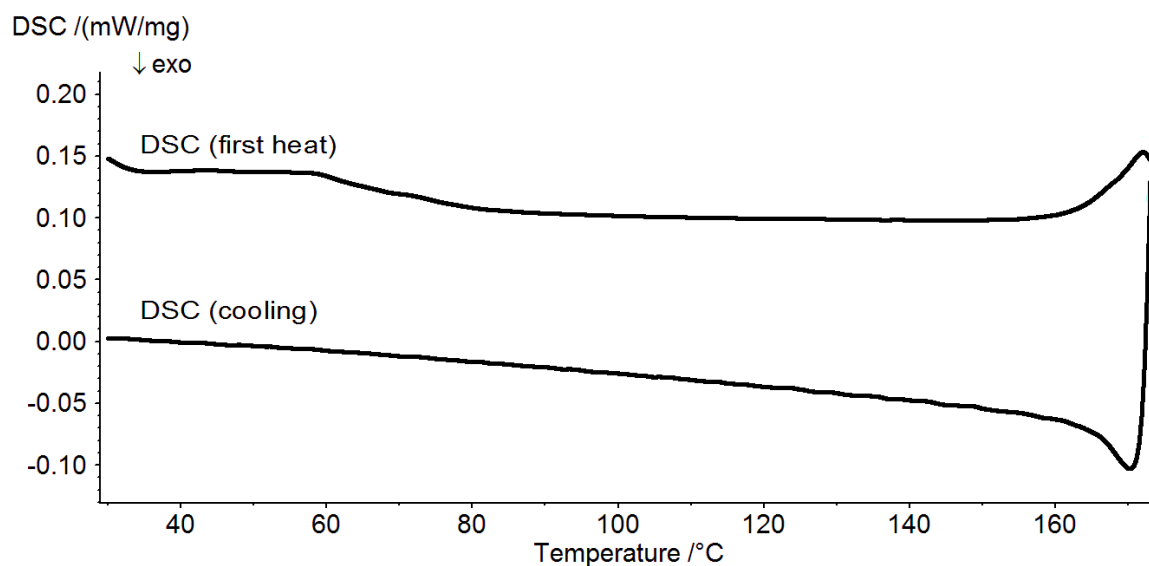


Figure 6S – DSC curves of **1g** heating to 173°C and cooling to 30°C with rate of 4 °C/min.

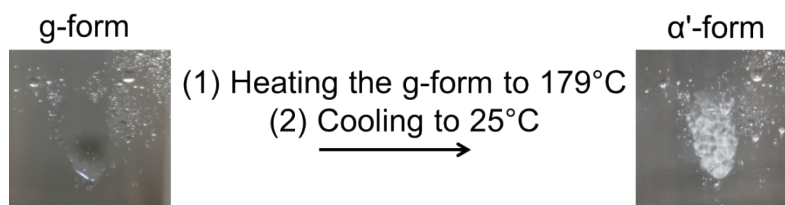


Figure 7S – Visual changes in **1g** sample at the heating to 179°C and cooling to RT of 25°C on the air.

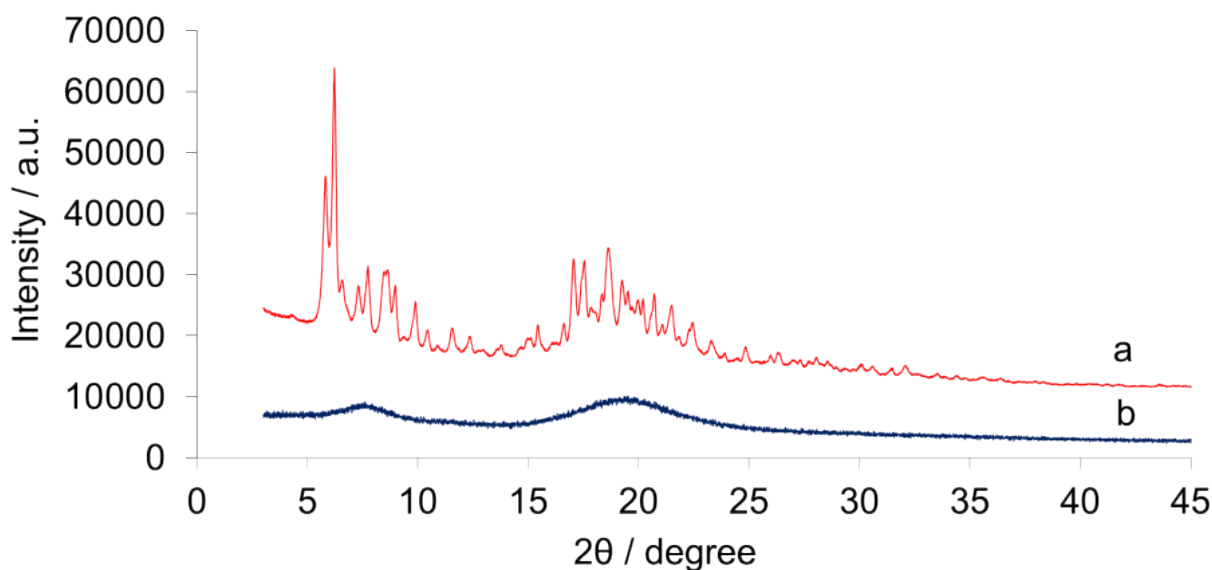


Figure 8S – XRPD data for **1a** prepared by heating of (a) **1g** to 150°C at the heating rate of 10°C/min and immediate cooling to RT on the air; (b) **1g** heated to 80°C and cooled to RT.

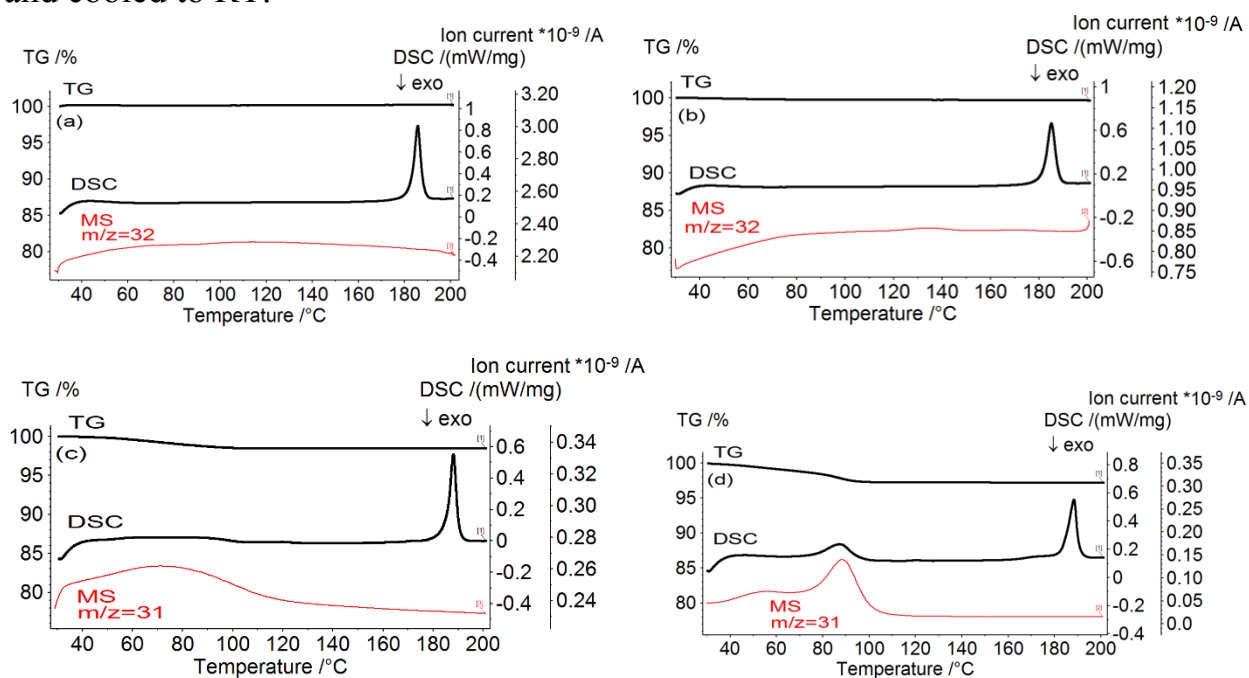


Figure 9S – TG/DSC/MS curves for products of solid calixarene **1** saturation with guest vapors at $P/P_0 = 1$, $T = 25^\circ\text{C}$, in systems: (a) **1a** + MeOH, (b) **1g** + MeOH, (c) **1a** + EtOH, (d) **1g** + EtOH.

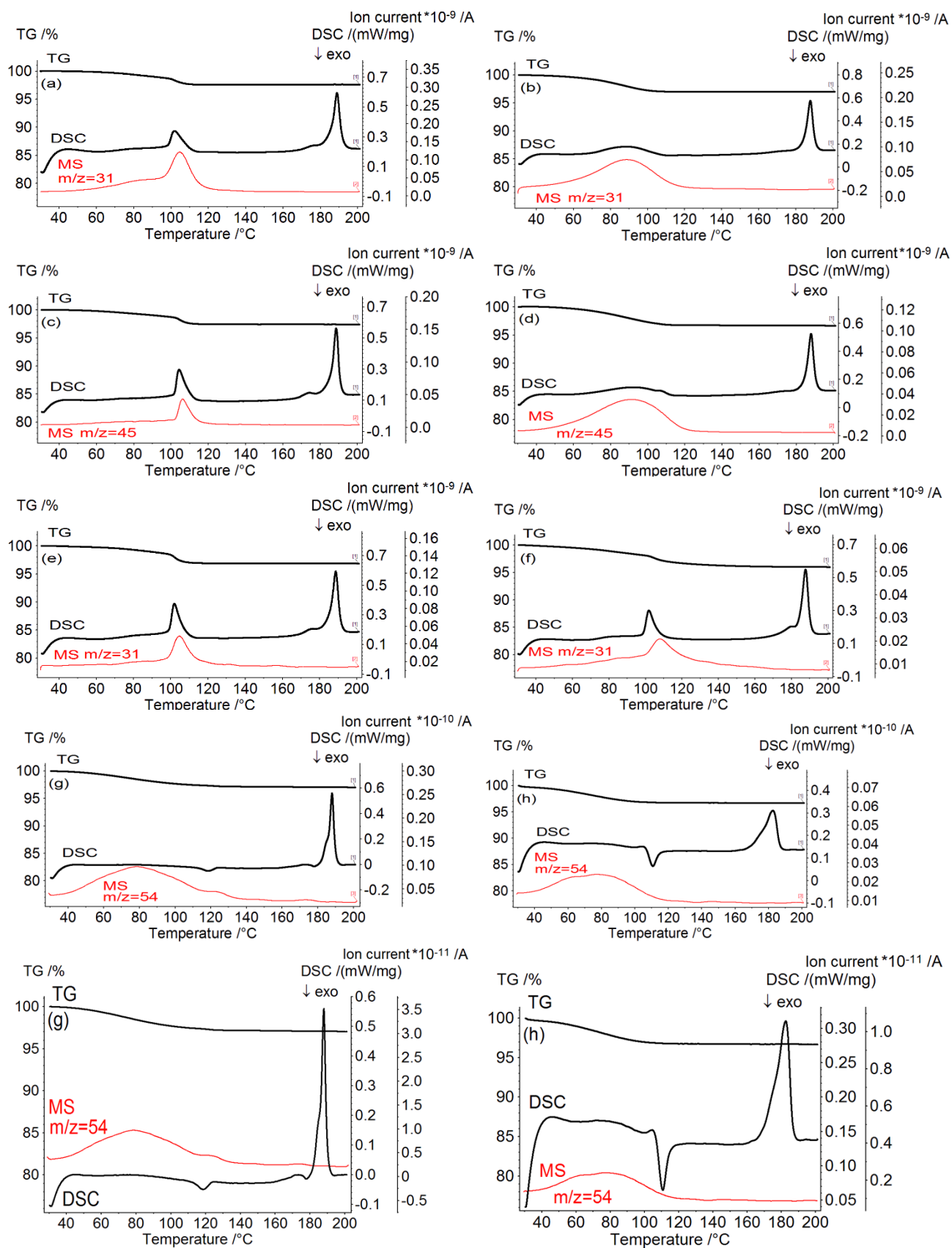


Figure 10S – TG/DSC/MS curves for products of solid calixarene **1** saturation with guest vapors at $P/P_0 = 1$, $T = 25^\circ\text{C}$, in systems: (a) **1a** + n-PrOH, (b) **1g** + n-PrOH, (c) **1a** + i-PrOH, (d) **1g** + i-PrOH, (e) **1a** + n-BuOH, (f) **1g** + n-BuOH, (g) **1a** + EtCN, (h) **1g** + EtCN.

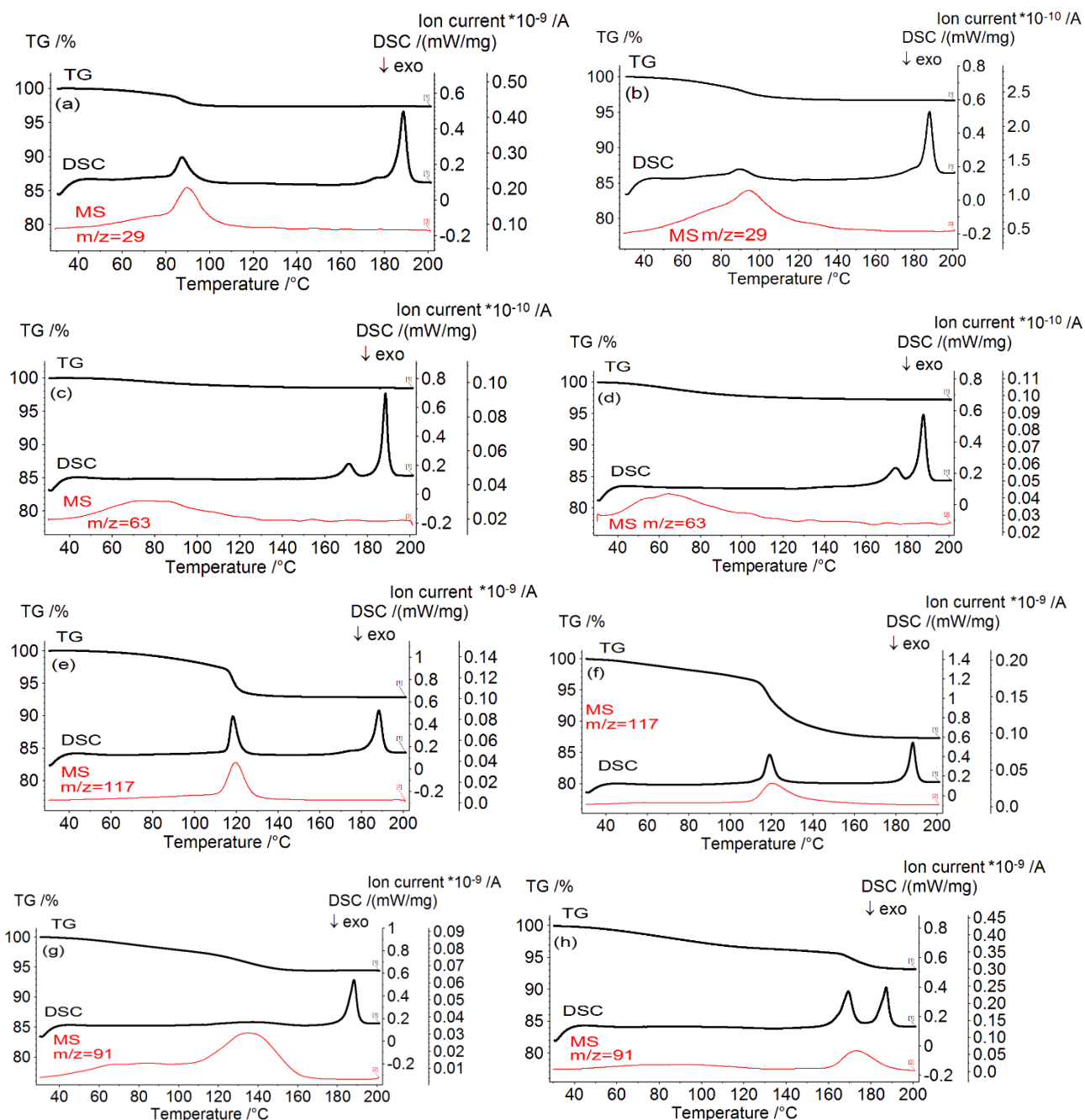


Figure 11S – TG/DSC/MS curves for products of solid calixarene **1** saturation with guest vapors at $P/P_0 = 1$, $T = 25^\circ\text{C}$, in systems: (a) **1a** + PrCN, (b) **1g** + PrCN, (c) **1a** + C₂H₄Cl₂, (d) **1g** + C₂H₄Cl₂, (e) **1a** + CCl₄, (f) **1g** + CCl₄, (g) **1a** + C₆H₅Me, (h). **1g** + C₆H₅Me.

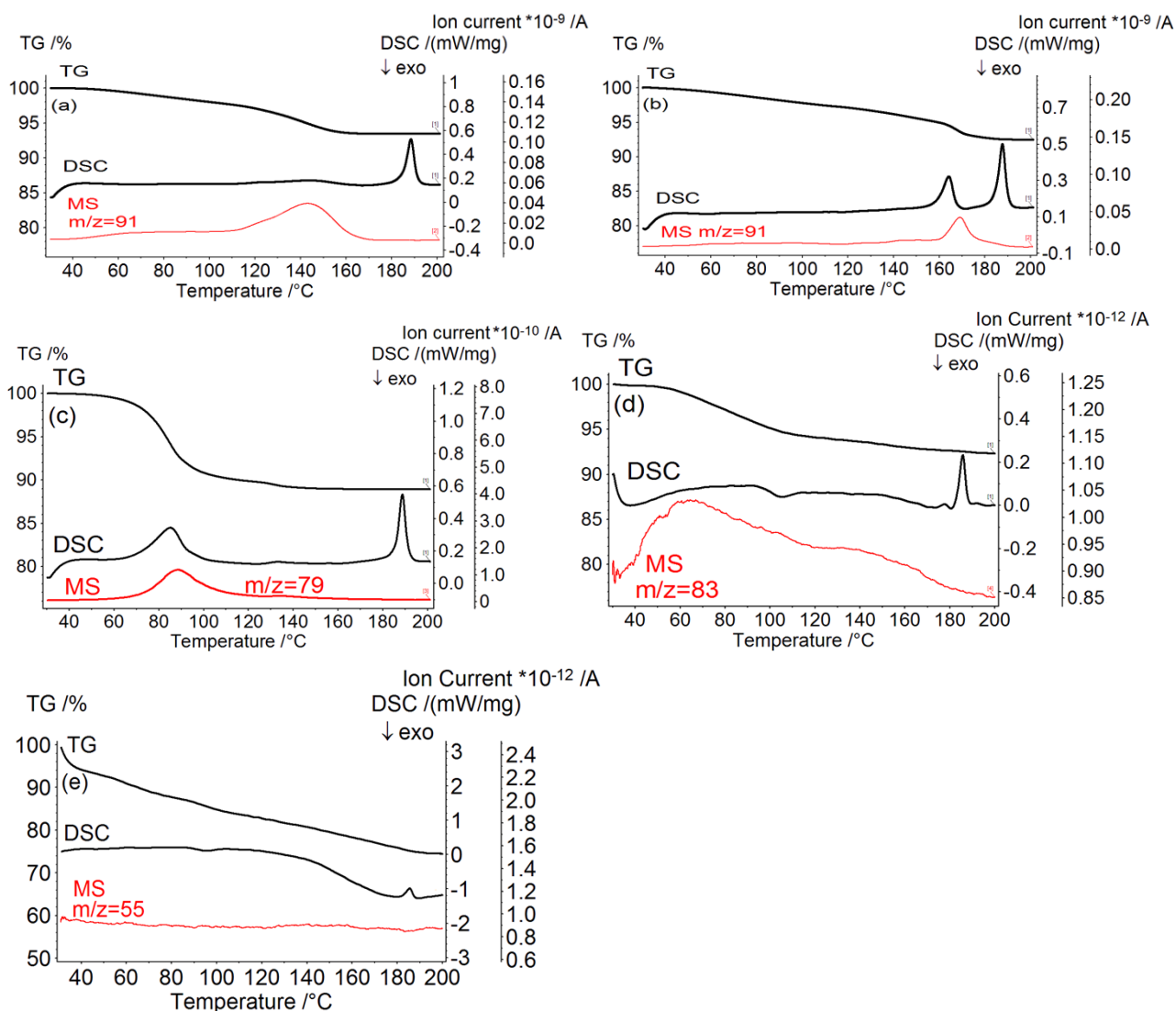


Figure 12S – TG/DSC/MS curves for products of solid calixarene **1** saturation with guest vapors at $P/P_0 = 1$, $T = 25^\circ\text{C}$, in systems: (a) **1a** + $\text{C}_6\text{H}_5\text{Et}$, (b) **1g** + $\text{C}_6\text{H}_5\text{Et}$, (c) **1a** + $\text{C}_5\text{H}_5\text{N}$, (d) **1a** + CHCl_3 , (e) **1a** + HFP.

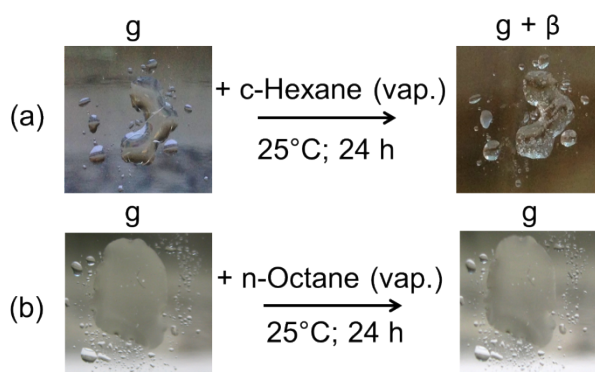


Figure 13S – Visual changes in **1g** sample at the saturation with vapors of (a) cyclohexane, (b) *n*-octane.

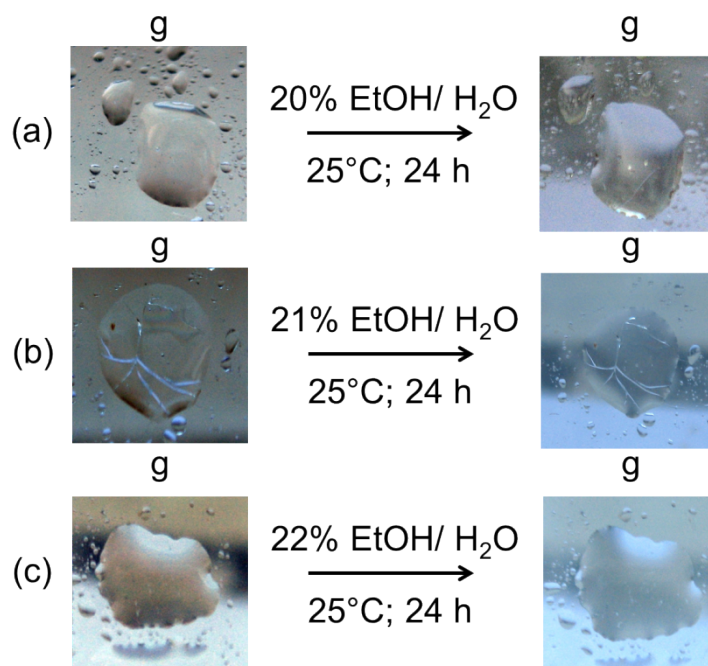


Figure 14S – Visual images of **1g** samples before and after saturation with vapors of binary aqueous mixture having ethanol contents (a) 20 vol. %, (b) 21 vol. %, (c) 22 vol. %; saturation time 24 h, $T=25^{\circ}\text{C}$.

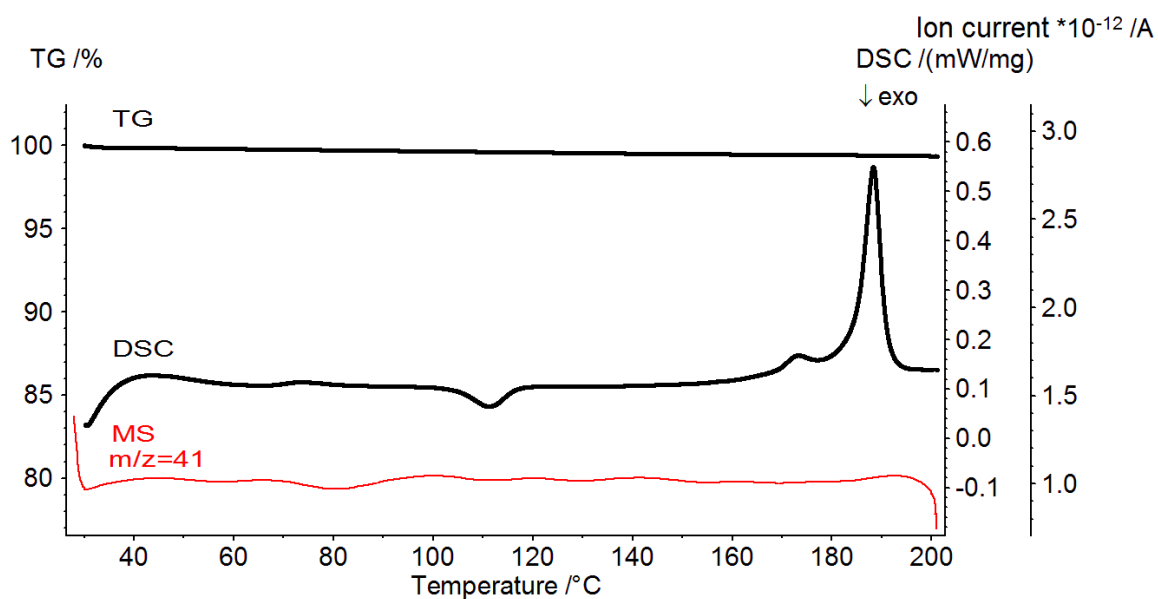


Figure 15S – TG/DSC/MS curves for a sample of β_0 phase prepared from clathrate formed by saturation of **1 α** with acetonitrile vapor.

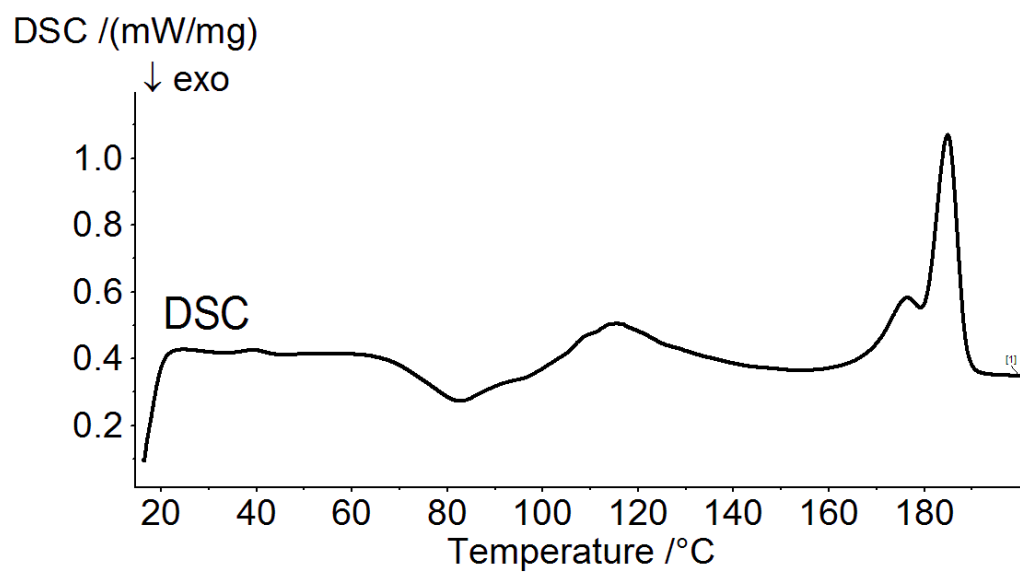


Figure 16S – DSC curve for compact **1g** glass saturated with vapor of liquid aqueous ethanol (25 vol%) for 24 h.

Table 1S. Enthalpies of $\alpha \rightarrow \alpha'$ transition $\Delta H_{\alpha \rightarrow \alpha'}$, fusion ΔH_m , and sum of these two values $\Delta H_\Sigma = \Delta H_{\alpha \rightarrow \alpha'} + \Delta H_m$ for products of **1a** and **1g** saturation with guest vapors from TG/DSC/MS experiment.

| Guest | Host | | | | | |
|---|---|--------------------------|-------------------------------|---|--------------------------|-------------------------------|
| | 1a | | | 1g | | |
| | $\Delta H_{\alpha \rightarrow \alpha'}$, kJ/mol | ΔH_m , kJ/mol | ΔH_Σ , kJ/mol | $\Delta H_{\alpha \rightarrow \alpha'}$, kJ/mol | ΔH_m , kJ/mol | ΔH_Σ , kJ/mol |
| No guest ^a | 6 | 46 | 52 | 14 | 31 | 45 |
| H ₂ O | 7.2 ^b | 43.5 | 50.7 | - | 34.9 | 34.9 |
| CH ₃ OH | - | 57.0 | 57.0 | - | 52.2 | 52.2 |
| C ₂ H ₅ OH | 1.1 | 44.8 | 45.9 | 4.6 | 42.2 | 46.9 |
| 1-C ₃ H ₇ OH | 5.0 | 42.3 | 47.3 | 3.0 | 41.5 | 44.5 |
| 2-C ₃ H ₇ OH | 5.1 ^b | 41.0 | 46.0 | 3.0 | 38.6 | 41.5 |
| 1-C ₄ H ₉ OH | 6.4 ^b | 43.8 | 50.3 | 9.2 ^b | 37.1 | 46.3 |
| CH ₃ CN | 3.1 ^b | 42.8 | 45.9 | 4.3 | 41.6 | 45.9 |
| C ₂ H ₅ CN | 3.2 ^b | 44.7 | 47.9 | - | 35.0 | 35.0 |
| C ₃ H ₇ CN | 4.6 | 41.5 | 46.0 | 3.2 | 42.4 | 45.7 |
| Acetone | 4.6 ^b | 43.8 | 48.3 | 3.8 | 40.1 | 43.9 |
| Pyridine ^c | 3.9 | 43.1 | 47.0 | - | - | - |
| C ₂ H ₄ Cl ₂ | 12.7 ^b | 40.9 | 53.6 | 15.7 ^b | 38.2 | 53.9 |
| CCl ₄ | 5.3 | 41.7 | 47.0 | - | 40.9 | 40.9 |
| CHCl ₃ ^c | 1.1 | 16.4 | 17.5 | - | - | - |
| Benzene | 3.4 ^b | 41.3 | 44.7 | - | 43.9 | 43.9 |
| Toluene | - | 42.7 | 42.7 | - | 29.5 | 29.5 |
| Ethylbenzene | - | 41.2 | 41.2 | 2 | 32.2 | 34.2 |
| Cyclohexane | 7.6 ^b | 44.5 | 52.1 | 4.8 | 41.8 | 46.6 |
| <i>n</i> -Octane | 6.4 ^b | 44.5 | 50.9 | 4.4 | 39.3 | 43.8 |

^a data for initial **1a** and **1g** samples from TM-DSC experiment; ^b a separate peak of $\alpha \rightarrow \alpha'$ transition is observed, in other cases $\Delta H_{\alpha \rightarrow \alpha'}$ values are for shoulders of fusion peak where observed; ^c data for inclusion compounds from dried **1a** solution in liquid guest formed in vapor sorption process. The values of ΔH_Σ are determined with the error of ± 1 kJ/mol.