

**New insight into the mechanism of the Ag, Au and AgAu
nanoparticles formation in aqueous alkaline media:
alkoxides from alcohols, aldehydes and ketones as the
universal reducing agent**

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Supporting Information

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Figure captions

SI 1 - UV-vis spectra of the colloidal Au-Nps produced with 0.50 mol L⁻¹ methanol, 0.010 mol L⁻¹ NaOH, 10 g L⁻¹ PVP and 0.40 mmol L⁻¹ AuCl₃ at 25 °C.

SI 2 – UV-vis spectra of the colloidal Au-Nps produced with 0.50 mol L⁻¹ dihidroxiacetone (black line) or 10 g L⁻¹ PVP (blue line), 0.010 mol L⁻¹ NaOH and 0.40 mmol L⁻¹ AuCl₃ at 25 °C.

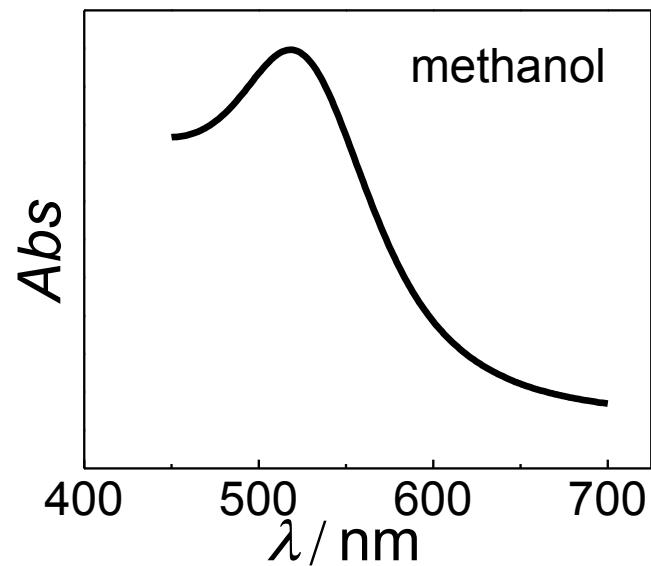
SI 3 – HPLC results of glyoxylic acid, hydroxypyruvic acid, dihydroxyacetone, glyceraldehyde, mesoxalic acid, formic acid, glycolic acid, glyceric acid, tartronic acid and oxalic acid in NaOH solution.

SI 4 – HPLC results of 0.01 mol L⁻¹ glycerol in 0.01 mol L⁻¹ NaOH and chromatograms taken at 1 min., 30 min. and 60 min. after exposing the Ag/C to 0.01 mol L⁻¹ glycerol in 0.01 mol L⁻¹ NaOH.

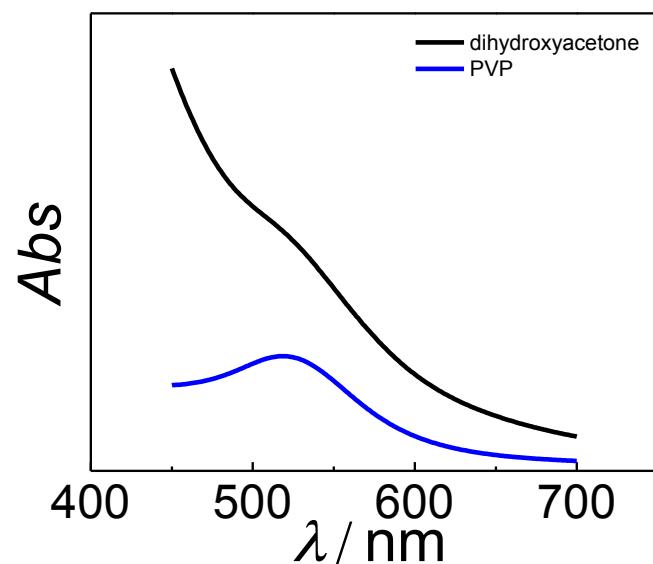
SI 5 – HPLC result of 0.01 mol L⁻¹ glycerol in 0.01 mol L⁻¹ NaOH (bottom), chromatogram taken at 60 min. after exposing the Au/C to 0.01 mol L⁻¹ glycerol in 0.01 mol L⁻¹ NaOH (middle) and chromatogram taken at 60 min. after Au nanoparticles synthesis (top).

SI 6 – X-ray diffraction (XRD) pattern of the Au/C, Ag/C and AuAg/C. Condition of synthesis: 0.40 mmol L⁻¹ AuCl₃ (in the case of Au nanoparticles), 0.40 mmol L⁻¹ AgNO₃ (in the case of Ag nanoparticles), 1.0 mol L⁻¹ glycerol and 0.10 mol L⁻¹ NaOH. For the bimetallic nanoparticles the Au³⁺ and Ag⁺ final concentrations were both 0.20 mmol L⁻¹.

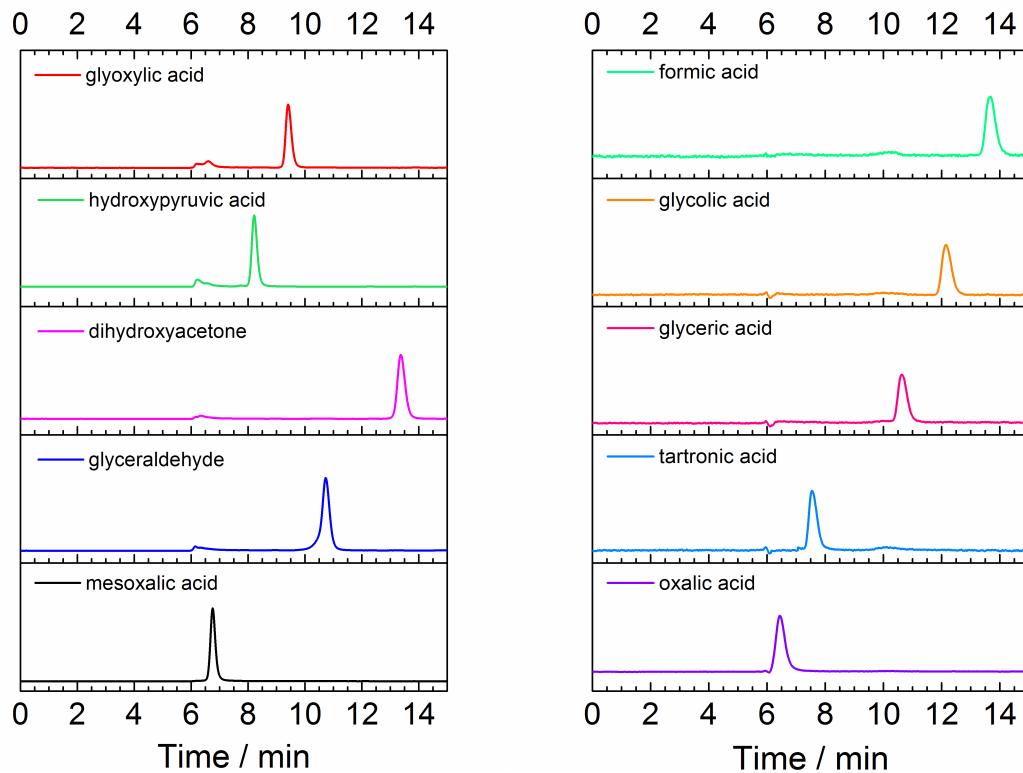
SI 1



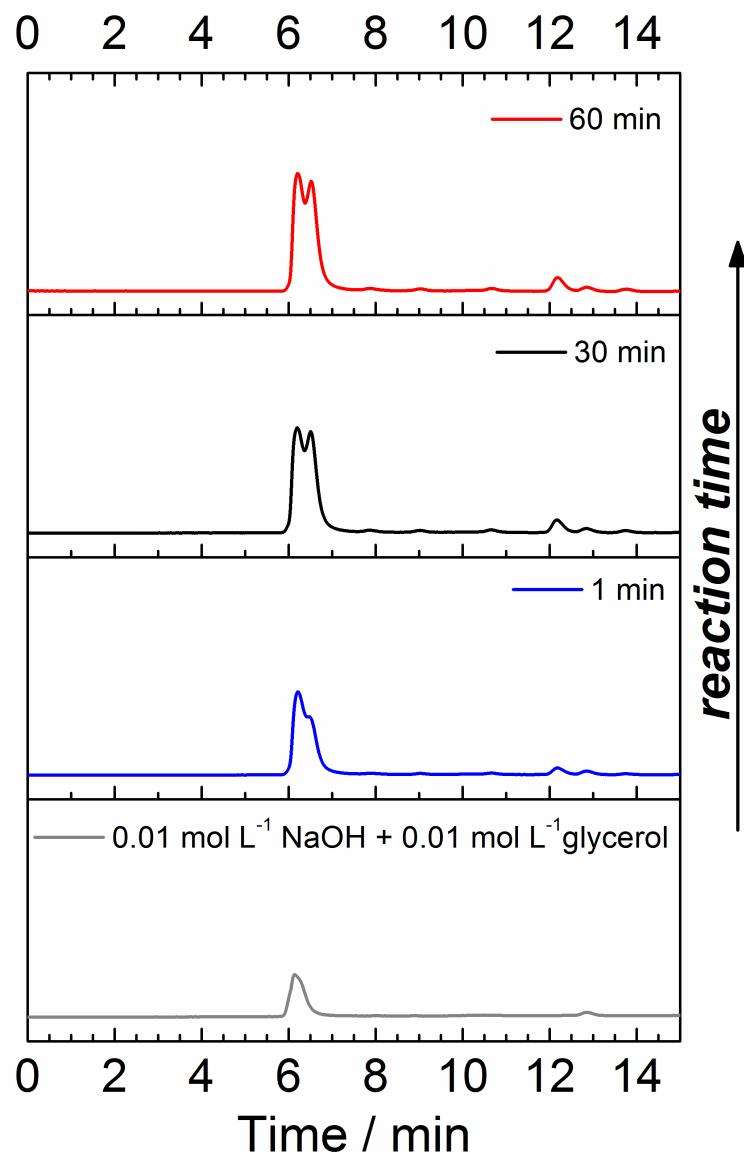
SI 2



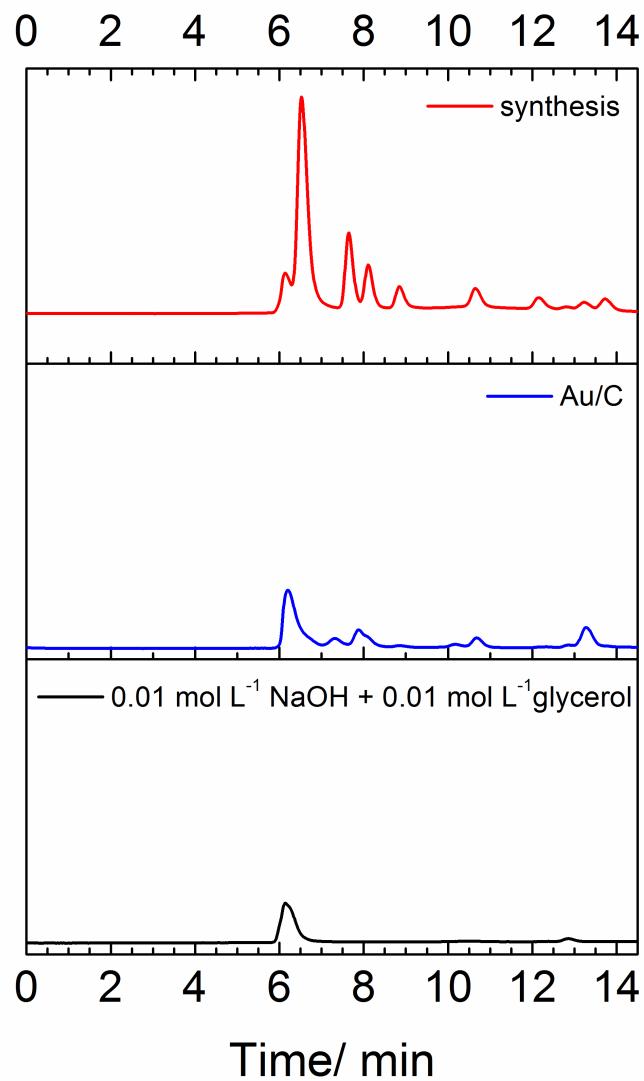
SI 3



SI 4



SI 5



SI 6

