

**Supporting Information for**

**From Lime to Silica and Alumina:**

**Systematic Modeling of Cement Clinkers using a General Force-Field**

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**Table S1.** Structural data for selected crystalline phases of the lime-silica-alumina system. Buckingham potential MD data in italic and relative deviations between MD and experimental results in parenthesis.

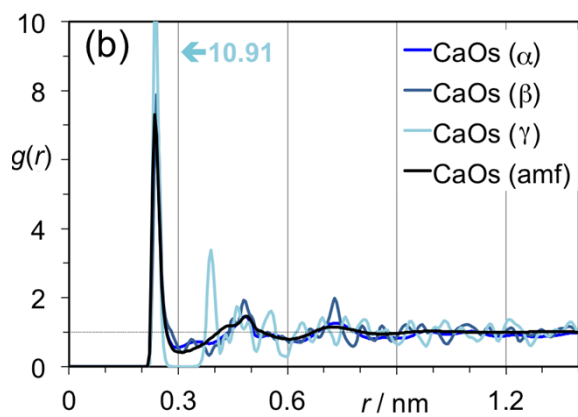
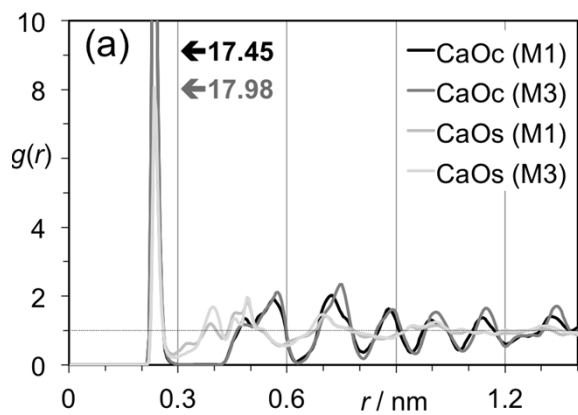
name	alite	alite	belite	belite	belite	rankinite	wollastonite	wollastonite	-	mayenite	krotite	grossite	gehlenite
allotrope	$M_1$	$M_3$	$\alpha$	$\beta$	$\gamma$	-	$\alpha$	$\beta$	-	-	-	-	-
formula	Ca <sub>3</sub> SiO <sub>5</sub>	Ca <sub>3</sub> SiO <sub>5</sub>	Ca <sub>2</sub> SiO <sub>4</sub>	Ca <sub>2</sub> SiO <sub>4</sub>	Ca <sub>2</sub> SiO <sub>4</sub>	Ca <sub>3</sub> Si <sub>2</sub> O <sub>7</sub>	CaSiO <sub>3</sub>	CaSiO <sub>3</sub>	Ca <sub>3</sub> Al <sub>2</sub> O <sub>6</sub>	Ca <sub>12</sub> Al <sub>14</sub> O <sub>33</sub>	CaAl <sub>2</sub> O <sub>4</sub>	CaAl <sub>4</sub> O <sub>7</sub>	Ca <sub>2</sub> Al <sub>2</sub> SiO <sub>7</sub>
CCN	C <sub>3</sub> S	C <sub>3</sub> S	C <sub>2</sub> S	C <sub>2</sub> S	C <sub>2</sub> S	C <sub>3</sub> S <sub>2</sub>	CS	CS	C <sub>3</sub> A	C <sub>12</sub> A <sub>7</sub>	CA	CA <sub>2</sub>	C <sub>2</sub> AS
space group	<i>Pc</i>	<i>Cm</i>	<i>P6<sub>3</sub>mc</i>	<i>P2<sub>1</sub>/n</i>	<i>Pbnm</i>	<i>P2<sub>1</sub>/a</i>	<i>P-1</i>	<i>P2<sub>1</sub>/a</i>	<i>Pa3</i>	<i>I-43d</i>	<i>P2<sub>1</sub>/n</i>	<i>C2/c</i>	<i>P-42<sub>1</sub>m</i>
reference	23	24	25	26	27	28	29	30	31	32	33	34	35
MD cells	2×8×5	2×8×3	10×11×8	10×11×8	10×8×6	5×6×8	7×8×8	4×7×8	4×4×4	5×5×5	6×7×4	4×6×10	7×7×10
MD atoms	12960	15552	12320	13440	11200	11520	13440	13440	16896	14750	14112	11520	11760
<i>a</i> / Å	27.874 <i>28.071</i> (0.70)	33.083 <i>33.373</i> (0.88)	5.579 <i>5.593</i> (0.26)	5.502 <i>5.617</i> (2.09)	5.081 <i>5.247</i> (3.27)	10.600 <i>10.667</i> (0.64)	7.940 <i>7.934</i> (-0.08)	15.426 <i>15.445</i> (0.12)	15.263 <i>15.270</i> (0.05)	11.979 <i>12.015</i> (0.29)	8.700 <i>8.673</i> (-0.31)	12.840 <i>12.957</i> (0.91)	7.690 <i>7.722</i> (0.42)
<i>b</i> / Å	7.059 <i>7.060</i> (0.02)	7.027 <i>7.192</i> (2.34)	5.579 <i>5.628</i> (0.87)	6.762 <i>6.848</i> (1.27)	11.224 <i>11.296</i> (0.64)	8.920 <i>9.068</i> (1.65)	7.320 <i>7.358</i> (0.52)	7.320 <i>7.364</i> (0.60)	15.263 <i>15.270</i> (0.05)	11.979 <i>12.017</i> (0.31)	8.092 <i>8.171</i> (0.98)	8.862 <i>8.748</i> (-1.29)	7.690 <i>7.722</i> (0.42)
<i>c</i> / Å	12.258 <i>12.233</i> (-0.20)	18.499 <i>18.682</i> (0.49)	7.150 <i>7.029</i> (-1.69)	9.339 <i>9.381</i> (0.45)	6.778 <i>6.805</i> (0.39)	7.890 <i>8.415</i> (3.23)	7.070 <i>7.192</i> (1.73)	7.066 <i>7.191</i> (1.77)	15.263 <i>15.270</i> (0.05)	11.979 <i>12.017</i> (0.32)	15.191 <i>15.109</i> (-0.54)	5.431 <i>5.471</i> (0.74)	5.100 <i>5.069</i> (-0.61)
<i>α</i> / °	90.00 <i>89.99</i>	90.00 <i>90.02</i>	90.00 <i>90.05</i>	90.00 <i>90.00</i>	90.00 <i>90.00</i>	90.00 <i>90.00</i>	90.00 <i>89.95</i>	90.00 <i>90.00</i>	90.00 <i>90.00</i>	90.00 <i>90.01</i>	90.00 <i>90.00</i>	90.00 <i>90.00</i>	90.00 <i>90.00</i>
<i>β</i> / °	116.03 <i>115.81</i>	94.12 <i>93.93</i>	90.00 <i>89.96</i>	94.14 <i>96.94</i>	90.00 <i>90.00</i>	119.60 <i>120.75</i>	95.37 <i>93.32</i>	95.40 <i>93.54</i>	90.00 <i>90.00</i>	90.00 <i>90.02</i>	90.17 <i>90.13</i>	106.83 <i>106.22</i>	90.00 <i>90.00</i>
<i>γ</i> / °	90.00 <i>90.01</i>	90.00 <i>89.98</i>	120.00 <i>119.61</i>	90.00 <i>90.00</i>	90.00 <i>90.00</i>	90.00 <i>90.00</i>	103.43 <i>103.43</i>	90.00 <i>90.00</i>	90.00 <i>90.00</i>	90.00 <i>90.03</i>	90.00 <i>90.00</i>	90.00 <i>90.00</i>	90.00 <i>90.00</i>
<i>V</i> <sub>cell</sub> / Å <sup>3</sup>	2167.24 <i>2182.50</i>	4289.43 <i>4473.14</i>	192.73 <i>192.38</i>	346.55 <i>358.91</i>	386.54 <i>403.31</i>	648.66 <i>677.14</i>	397.82 <i>407.69</i>	794.33 <i>816.44</i>	3555.66 <i>3560.75</i>	1719.12 <i>1735.04</i>	1069.45 <i>1070.71</i>	591.54 <i>595.47</i>	301.59 <i>302.28</i>
<i>d</i> / g.cm <sup>-3</sup>	3.150 <i>3.128</i> (-0.70)	3.182 <i>3.072</i> (-3.47)	3.073 <i>3.079</i> (0.18)	3.301 <i>3.194</i> (-3.24)	2.960 <i>2.838</i> (-4.13)	2.953 <i>2.830</i> (-4.17)	2.909 <i>2.840</i> (-2.38)	2.914 <i>2.836</i> (-2.67)	3.028 <i>3.027</i> (-0.02)	2.690 <i>2.671</i> (-0.85)	2.950 <i>2.942</i> (-0.09)	2.919 <i>2.938</i> (0.65)	3.019 <i>3.037</i> (0.60)

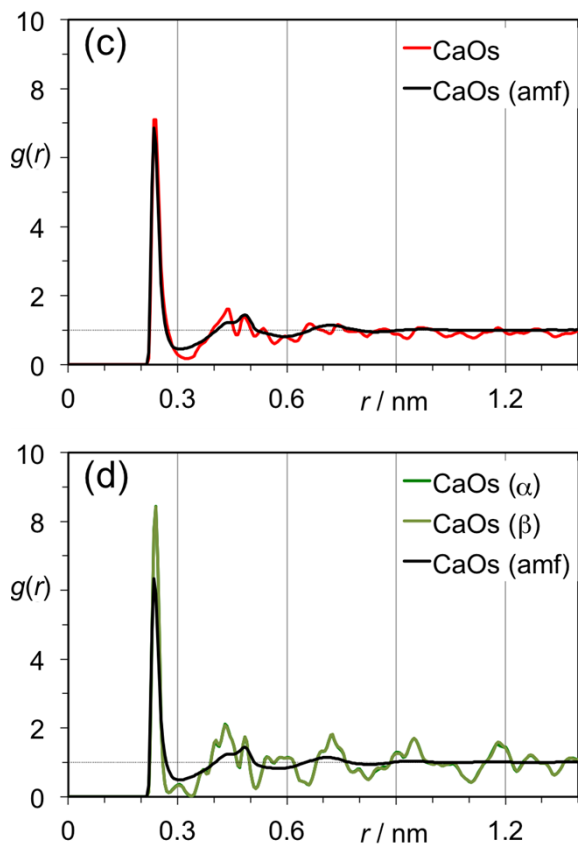
**Table S2.** Statistical distribution of  $Q^n(mAl)$  units in amorphous materials with ratio C/S = 1.5 according to the Loewenstein's rule. The corresponding MD result is reported in italic.

	CaO:SiO <sub>2</sub> :Al <sub>2</sub> O <sub>3</sub> 58.3:38.8:2.9					CaO:SiO <sub>2</sub> :Al <sub>2</sub> O <sub>3</sub> 56.5:37.6:5.9					CaO:SiO <sub>2</sub> :Al <sub>2</sub> O <sub>3</sub> 54.6:36.4:9.1				
$mAl$	$Q^0$	$Q^1$	$Q^2$	$Q^3$	$Q^4$	$Q^0$	$Q^1$	$Q^2$	$Q^3$	$Q^4$	$Q^0$	$Q^1$	$Q^2$	$Q^3$	$Q^4$
0	1.00	0.85	0.72	0.61	0.52	1.00	0.69	0.47	0.32	0.22	1.00	0.50	0.25	0.13	0.06
	<i>1.00</i>	<i>0.78</i>	<i>0.63</i>	<i>0.51</i>	<i>0.41</i>	<i>1.00</i>	<i>0.66</i>	<i>0.44</i>	<i>0.27</i>	<i>0.18</i>	<i>1.00</i>	<i>0.51</i>	<i>0.28</i>	<i>0.15</i>	<i>0.05</i>
1		0.15	0.26	0.33	0.37		0.31	0.43	0.44	0.41		0.50	0.50	0.38	0.25
		<i>0.22</i>	<i>0.33</i>	<i>0.38</i>	<i>0.45</i>		<i>0.34</i>	<i>0.44</i>	<i>0.45</i>	<i>0.30</i>		<i>0.49</i>	<i>0.50</i>	<i>0.41</i>	<i>0.31</i>
2			0.02	0.06	0.10			0.10	0.20	0.28			0.25	0.37	0.37
			<i>0.05</i>	<i>0.10</i>	<i>0.10</i>			<i>0.12</i>	<i>0.25</i>	<i>0.38</i>			<i>0.21</i>	<i>0.35</i>	<i>0.35</i>
3				0.00	0.01				0.03	0.08				0.12	0.25
				<i>0.01</i>	<i>0.03</i>				<i>0.04</i>	<i>0.14</i>				<i>0.10</i>	<i>0.24</i>
4					0.00					0.01					0.06
					<i>0.00</i>					<i>0.00</i>					<i>0.06</i>
$Q^n$ %	12.8	35.6	35.3	14.5	1.88	7.12	30.1	37.3	21.4	3.46	4.14	23.7	37.8	27.9	6.50

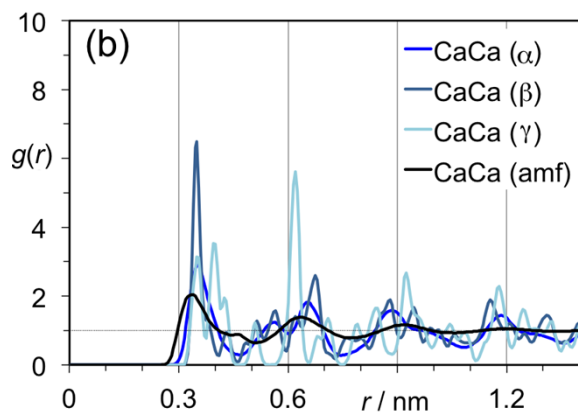
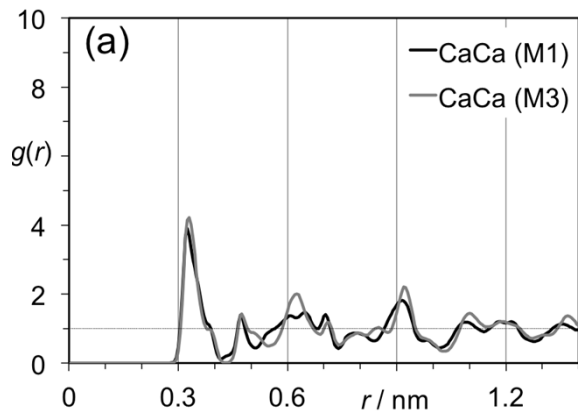
**Table S3.** Statistical distribution of  $Q^n(mAl)$  units in amorphous materials with ratio C/S = 1.0 according to the Loewenstein's rule. The corresponding MD result is reported in italic.

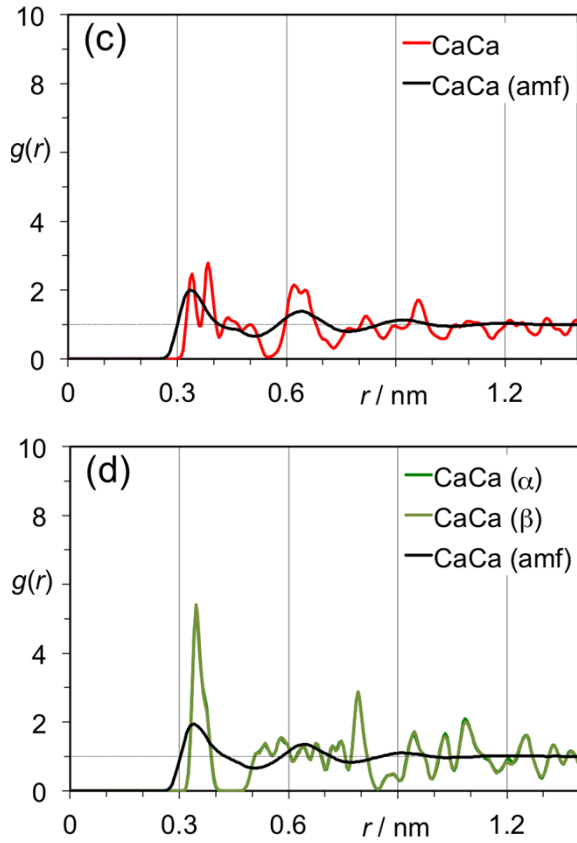
	CaO:SiO <sub>2</sub> :Al <sub>2</sub> O <sub>3</sub> 48.6:48.6:2.9					CaO:SiO <sub>2</sub> :Al <sub>2</sub> O <sub>3</sub> 47.1:47.1:5.9					CaO:SiO <sub>2</sub> :Al <sub>2</sub> O <sub>3</sub> 45.5:45.5:9.1				
$mAl$	$Q^0$	$Q^1$	$Q^2$	$Q^3$	$Q^4$	$Q^0$	$Q^1$	$Q^2$	$Q^3$	$Q^4$	$Q^0$	$Q^1$	$Q^2$	$Q^3$	$Q^4$
0	1.00	0.88	0.78	0.68	0.60	1.00	0.75	0.56	0.42	0.32	1.00	0.60	0.36	0.22	0.13
	<i>1.00</i>	<i>0.85</i>	<i>0.72</i>	<i>0.62</i>	<i>0.48</i>	<i>1.00</i>	<i>0.74</i>	<i>0.53</i>	<i>0.39</i>	<i>0.30</i>	<i>1.00</i>	<i>0.64</i>	<i>0.41</i>	<i>0.26</i>	<i>0.17</i>
1		0.12	0.21	0.28	0.33		0.25	0.38	0.42	0.42		0.40	0.48	0.43	0.35
		<i>0.15</i>	<i>0.26</i>	<i>0.31</i>	<i>0.41</i>		<i>0.26</i>	<i>0.40</i>	<i>0.43</i>	<i>0.39</i>		<i>0.36</i>	<i>0.46</i>	<i>0.45</i>	<i>0.35</i>
2			0.01	0.04	0.07			0.06	0.14	0.21			0.16	0.29	0.34
			<i>0.03</i>	<i>0.06</i>	<i>0.09</i>			<i>0.07</i>	<i>0.16</i>	<i>0.24</i>			<i>0.14</i>	<i>0.25</i>	<i>0.33</i>
3				0.00	0.01				0.02	0.05				0.06	0.15
				<i>0.01</i>	<i>0.02</i>				<i>0.02</i>	<i>0.06</i>				<i>0.05</i>	<i>0.13</i>
4					0.00					0.00					0.03
					<i>0.00</i>					<i>0.00</i>					<i>0.02</i>
$Q^n$ %	2.6	18.0	38.3	32.0	9.07	1.66	13.1	35.1	36.4	13.7	0.85	9.27	31.1	40.3	18.4





**Figure S1.** Radial pair distribution functions between calcium and oxygen atoms of: (a)  $M_1$  and  $M_3$  alite; (b)  $\alpha$ -,  $\beta$ -, and  $\gamma$ -belite; (c) rankinite and (d)  $\alpha$ -,  $\beta$ -wollastonite. The black lines in diagrams b-d represent the corresponding amorphous phases.





**Figure S2.** Radial pair distribution functions between calcium atoms of: (a)  $M_1$  and  $M_3$  alite; (b)  $\alpha$ -,  $\beta$ -, and  $\gamma$ -belite; (c) rankinite and (d)  $\alpha$ -,  $\beta$ -wollastonite. The black lines in diagrams b-d represent the corresponding amorphous phases.