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Supplementary information

Why LiFePO₄ is a Safe Battery Electrode: Coulomb Repulsion Induced Electron-State Reshuffling upon Lithiation

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Experimental Methods

Soft x-ray absorption (sXAS) and emission (sXES) spectroscopy. The O *K*-edge and Fe *L*-edge spectroscopy was performed at the undulator beamline 8.0 of the Advanced Light Source (ALS) at Lawrence Berkeley National Laboratory (LBNL). The spherical grating monochromater (SGM) supply a linearly polarized photon beam with resolving power up to 6000. The sXAS spectra were collected in total electron yield (TEY) mode with the data normalized to the photon flux of incident beam measured by an upstream clean gold mesh. The sXES spectra were collected by a Rowland circle geometry spectrometer perpendicular to the incident beam. The experimental resolution is better than 0.2 eV for sXAS. For sXES measurements, the resolution of incident x-ray photon is about 0.9 eV and resolution of the spectrometer is about 0.7 eV.

Hard x-ray Raman scattering (hXRS). The cross-section of hXRS is dominated by the dipole transitions within the low-momentum transfer regime, thus in principle is analogous to sXAS. Stemming from the long penetration depth of high-energy x-ray photon, the hXRS provides the intrinsic bulk information compared to the surface-sensitive sXAS. The O *K*-edge and Fe *L*-edge hXRS measurements were performed at Beamline 6-2 at Stanford Synchrotron Radiation Lightsource (SSRL) with the Si-660 reflection of the XRS spectrometer corresponding to an incident scanning energy of about 10.4 keV. Although the statistics is not as good as the sXAS, the spectral fine structures are still unambiguously shown in the hXRS.

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Multi-peak fitting of sXES spectroscopy. The multi-peak fitting of Fe-L XES spectra was performed with software WinXAS. The L_3 and L_2 were fitted with same group of Gaussian functions. Two broad components were used to fit the background. The main emission peaks were fitted with one Gaussian function for FePO₄ and three Gaussian functions for LiFePO₄. The physics arguments under this analysis were described in the main text.

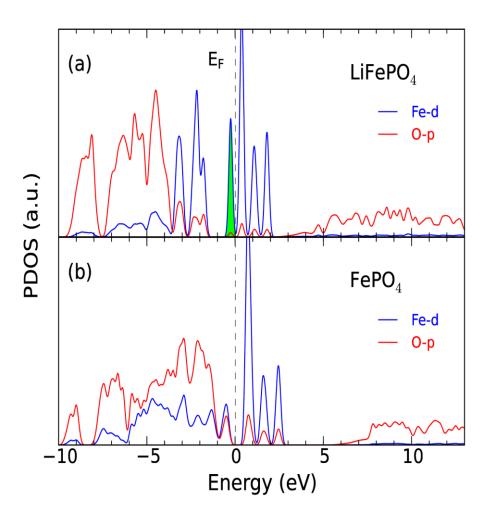


FIG. S1 The GGA (U=0) spin-polarized pDOS of Fe-3*d* (4 atoms) and O-2*p* pDOS (16 atoms) of (a) LiFePO₄ and (b) FePO₄. Blue (red) indicates Fe-d (O-p) states. The distinct occupied Fe-3d state close to the Fermi level (filled with green) is the state of electrons with the minority spin. The Fermi energy is set to be 0 eV. The GGA bandgap of less than 1 eV for both LiFePO₄ and FePO₄ is clearly underestimated. Although the occupied Fe-d and O-p GGA-pDOS of LiFePO₄ shows similar lineshape and distribution compared with MBJGGA+U pDOS shown in Fig. 4, the broad occupied Fe-d states in GGA-pDOS of FePO₄ between -6 eV and Fermi energy is inconsistent with the localized Fe-d states in MBJGGA+U pDOS located at 6 eV below the Fermi energy.

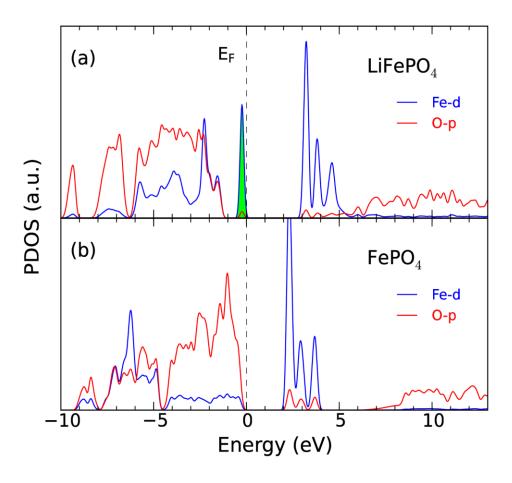


FIG. S2 The GGA+U with U=4 eV spin-polarized pDOS of Fe-3d (4 atoms) and O-2p pDOS (16 atoms) of (a) LiFePO₄ and (b) FePO₄ with the same layout as Fig. S1. By including Hubbard U = 4 eV in the GGA calculations, the bandgap of 3 eV and 2 eV are opened in LiFePO₄ and FePO₄ respectively. In contrast to the GGA calculations, the GGA+U and MBJGGA+U calculations show similar occupied pDOS of FePO₄. However, for GGA+U pDOS of LiFePO₄, the broad Fe-d states at 2 eV below the Fermi energy highly overlap with the O-p states, which feature is not seen in the MBJGGA+U calculations.