## First Principles study of Organic sensitizers for Dye Sensitized Solar Cells: Effects of

## Anchoring Groups on optoelectronic properties and Dye aggregation

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Methods	L1		L2	
	λ	Exp. <sup>a</sup>	λ	Exp. <sup>a</sup>
TDB3LYP//B3LYP	560.7	404	615.6	427
TDBHandHLYP//B3LYP	441.0		483.4	
TDCAM-B3LYP//B3LYP	440.3		479.7	
TDM062X//B3LYP	445.6		485.5	
TDωB97XD//B3LYP	426.0		463.7	
TDBHandHLYP//BHandHLYP	414.9		452.2	
TDCAM-B3LYP//CAM-	414.9		449.0	
B3LYP	421.2		455.9	
TDM062X//M062X	397.1		430.6	
ΤDωB97XD//ωB97XD				

**Table S1.** The calculated absorption energies ( $\lambda$ , in nm) for the model dye L1 at different DFT functionals using 6-31G\* basis set at acetonitrile medium.

<sup>a</sup>taken from Ref.

**Table S2.** The calculated most representative absorption energies ( $\lambda$ , in nm), oscillator strengths (*f* in a.u.), and the corresponding MO transitions of designed dyes with PO<sub>3</sub>H<sub>2</sub> anchoring group at TD  $\omega$ B97XD// $\omega$ B97XD/6-31G\* level of theory.

Dyes	State	λ	f	Transition assignment
<b>D</b> π-CDM	S0→S1	642.8	0.12	H-1->L+0 (67%)
	S0→S3	343.0	1.55	H-0->L+1 (28%), H-1->L+1 (25%)
Dπ- <sup>CN</sup> CDM	S0→S1	630.5	0.18	H-1->L+0 (64%), H-0->L+0 (19%)
	S0→S3	357.3	1.47	H-1->L+1 (35%), H-0->L+1 (19%)
<b>D</b> π-TP	S0→S1	487.7	0.84	H-0->L+0 (51%), H-1->L+0 (44%)
	S0→S3	339.5	0.35	H-1->L+0 (27%), H-0->L+0 (23%),
				H-0->L+1 (21%)
	S0→S4	300.2	0.54	H-0->L+1 (37%), H-1->L+1 (25%)
Dπ- <sup>CN</sup> TP	S0→S1	511.1	1.00	H-0->L+0 (52%), H-1->L+0 (43%)
	S0→S4	312.4	0.49	H-1->L+1 (45%), H-0->L+1 (31%)
<b>D</b> π-CDT	S0→S1	385.8	1.99	H-0->L+0 (43%), H-1->L+0 (42%)
Dπ- <sup>CN</sup> CDT	S0→S1	412.0	1.98	H-1->L+0 (52%), H-0->L+0 (32%)

\*H and L represent HOMO and LUMO, respectively.

**Table S3.** The calculated most representative absorption energies ( $\lambda$ , in nm), oscillator strengths (*f* in a.u.), and the corresponding MO transitions of designed dyes with SO<sub>3</sub>H anchoring group at TD  $\omega$ B97XD// $\omega$ B97XD/6-31G\* level of theory.

Dyes	State	λ	f	Transition assignment		
<b>D</b> π-CDM	S0→S1	637.3	0.14	H-1->L+0 (64%)		
	S0→S3	346.8	1.63	H-1->L+1 (26%), H-0->L+1 (24%), H-0->L+2 (21%)		
D- CNCDM	S0→S1	630.8	0.21	H-1->L+0 (61%), H-0->L+0 (19%)		
Dπ- <sup>CN</sup> CDM	S0→S3	364.7	1.48	H-1->L+1 (36%), H-0->L+1 (20%)		
Dπ-TP	S0→S1	493.9	0.91	H-0->L+0 (50%), H-1->L+0 (45%)		
	S0→S3	345.7	0.22	H-1->L+0 (30%), H-0->L+0 (28%)		
	S0→S4	302.7	0.63	H-0->L+1 (36%), H-1->L+1 (30%)		
Dπ- <sup>CN</sup> TP	S0→S1	521.9	1.06	H-0->L+0 (52%), H-1->L+0 (43%)		
	S0→S4	315.1	0.50	H-1->L+1 (47%), H-0->L+1 (29%)		
<b>D</b> π-CDT	S0→S1	392.7	2.02	H-1->L+0 (48%), H-0->L+0 (35%)		
Dπ- <sup>CN</sup> CDT	S0→S1	422.1	1.97	H-1->L+0 (53%), H-0->L+0 (29%)		

\*H and L represent HOMO and LUMO, respectively.

**Table S4.** The calculated dihedral angle values ( $\theta$  in degrees) between different units for the face-to-face dimeric configurations including isolated monomer and individual monomers in dimer. (See figure 8 for dihedral angle definition)

Anchoring group	Acceptor group	Type*	$\theta_1$	θ2	θ3	
СООН		IM	0.51	0.96	-26.49	
	CDM	M1	17.06	6.36	11.67	
		M2	-12.97	-5.94	27.17	
	ТР	IM	-0.02	0.09	-1.14	
		M1	-12.24	-3.43	-10.89	
		M2	-4.19	11.71	2.73	
	CDT	IM	-0.02	0.07	-29.33	
		M1	16.56	2.99	17.59	
		M2	-13.51	-6.73	29.14	
CSSH	CDM	IM	-0.30	0.71	-26.08	
		M1	-1.04	2.34	-8.71	
		M2	-3.82	1.16	10.98	
	TP	IM	-0.12	-0.01	-1.90	
		M1	-5.85	-0.41	-7.30	
		M2	-4.42	-0.19	5.08	
	CDT	IM	1.32	1.21	-23.32	
		M1	-7.82	-1.26	-18.25	
		M2	-5.19	0.58	-4.29	

\*IM, M1 and M2 are Isolated monomer dye, individual monomer dye 1 in dimer and individual monomer dye 2 in dimer, respectively.

System	Acceptor group	R1	R2	R3	R4
	CDM	4.52	5.94	3.83	1.98
	ТР	2.07	3.86	3.53	3.77
	CDT	4.52	5.92	3.83	1.99
St.	CDM	4.62	4.23	3.76	5.50
S <sub>2</sub> H <sub>1</sub>	ТР	4.71	4.29	3.65	5.51
S <sub>4</sub> <sup>S</sup> H <sub>2</sub>	CDT	3.26	4.75	4.83	5.01

**Table S5.** The calculated hydrogen bond distances (in Å) between individual monomers of Head-to-Head dimeric configuration of designed dyes.

R1, R2, R3 and R4 are distances of  $O_1 \cdots H_2$ ,  $O_2 \cdots H_2$ ,  $O_3 \cdots H_1$  and  $O_4 \cdots H_1$  for COOH and  $S_1 \cdots H_2$ ,  $S_2 \cdots H_2$ ,  $S_3 \cdots H_1$  and  $S_4 \cdots H_1$  for CSSH dyes.



Figure S1. Isodensity plots of selected frontier molecular orbitals of the designed dyes with different acceptors and anchoring groups containing CN. The calculations were performed by  $\omega B97XD/6-31G^*$  level of theory and the isovalue is 0.02 a.u.



**Figure S2.** Isodensity plots of selected frontier molecular orbitals of the designed dyes with different acceptors and  $PO_3H_2$  anchoring group. The calculations were performed by  $\omega B97XD/6-31G^*$  level of theory and the isovalue is 0.02 a.u.



**Figure S3.** Isodensity plots of selected frontier molecular orbitals of the designed dyes with different acceptors and SO<sub>3</sub>H anchoring group. The calculations were performed by  $\omega$ B97XD/6-31G\* level of theory and the isovalue is 0.02 a.u.