

Supplementary Information

Enhancement of T_c in atomic phase of iodine-doped hydrogen at high pressures

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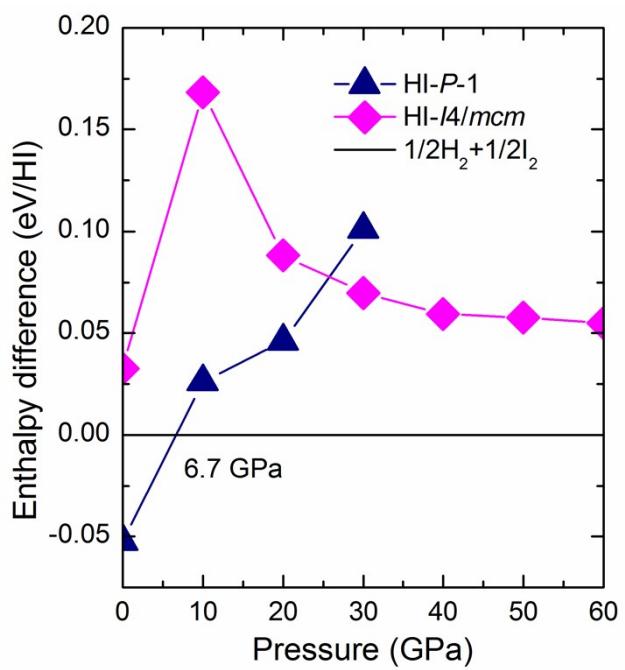


Fig. S1 The calculated enthalpy differences of *P-1* and *I4/mcm* phases in HI with respect to decomposition into constituent elemental solids H_2 and I_2 as a function of pressure.

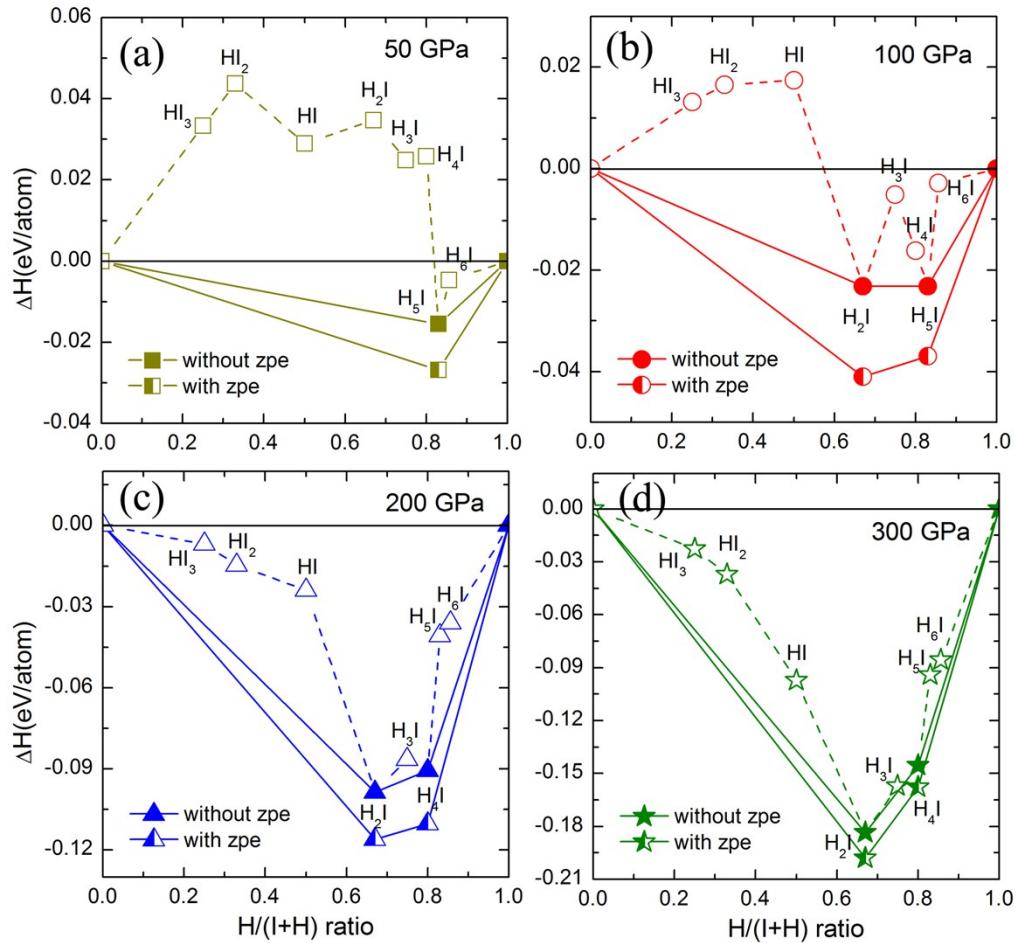


Fig. S2. Formation enthalpies of H-I system with respect to decomposition into constituent elemental solids under pressure. Dashed lines connect data points, and solid lines denote the convex hull. Solid and half-filled symbols indicate formation enthalpies of stable structures without and with ZPE, respectively.

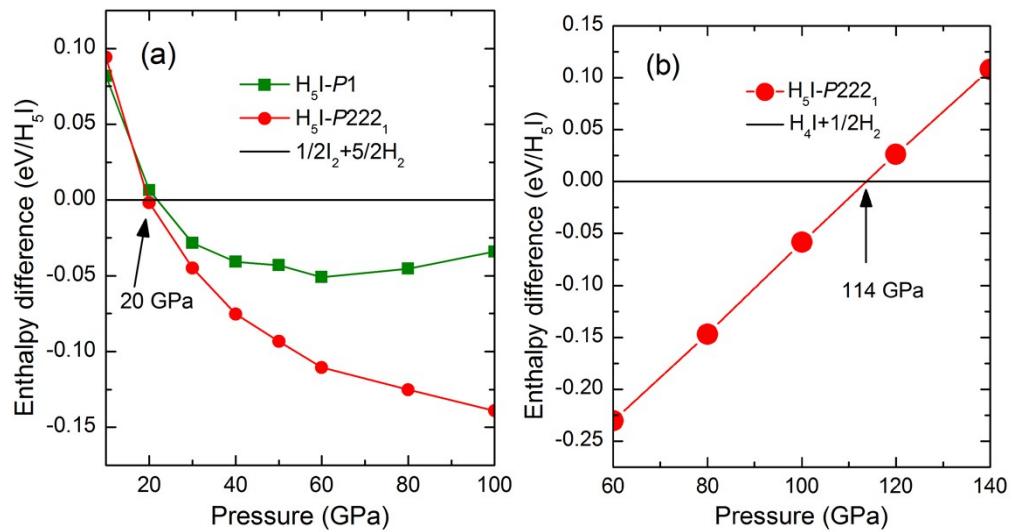


Fig. S3 (a) The calculated enthalpies of formation as a function of pressure for predicted H_5I . (b) The calculated enthalpy differences of $\text{H}_5\text{I}-P222_1$ phase with respect to decomposition into H_2 and H_4I as a function of pressure.

Table S1 Structure parameters of H and I atoms for our predicted stable structures of H-I compounds at the selected pressure.

Space group	Lattice parameters (Å, °)	Atom	Atomic coordinates (fractional)		
			x	y	z
$H_2I-Pmna$ (200 GPa)	$a=5.847$	H(8d)	-0.03489	0.40070	0.87921
	$b=2.727$	I(4c)	0.17825	0.75000	0.62030
	$c=4.389$				
$H_2I-R-3m$ (260 GPa)	$a=3.056$	H(6c)	0.00000	0.00000	0.40676
	$b=5.897$	I(3a)	0.00000	0.000000	0.00000
H_4I-P6/mmm (200 GPa)	$a=2.996$	H(4h)	0.33333	0.66667	0.34832
	$b=2.669$	I(1a)	0.00000	0.00000	0.00000
$H_5I-P222_1$ (100 GPa)	$a=6.553$	H(4e)	0.44116	0.77691	0.08234
	$b=3.178$	H(4e)	0.04076	0.19417	0.93730
	$c=5.524$	H(4e)	0.55080	0.71453	0.08797
		H(4e)	0.03438	0.34861	0.61381
		H(2d)	0.50000	0.72936	0.75000
		H(2b)	0.24729	0.50000	-0.00000
		I(4e)	0.76196	0.21098	0.23417