# **Electronic Supplementary Information (ESI)**

## Surface-Functionalized Monolayered Nanodots of a Transition Metal Oxide and Their

#### Properties

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#### Structures of the precursor layered titanates



**Fig. S1** XRD patterns (a,b) and FESEM images (c,d) of the precursor nanocrystals (a,b) and bulk crystals (c,d). In the panels (a,b), the spectra (i) and (ii) correspond to the Na-TiO<sub>2</sub> nanocrystals and H-TiO<sub>2</sub> after the protonation, respectively. In the pane (c), the spectra (i) and (ii) correspond to the bulk Cs-TiO<sub>2</sub> crystals and H-TiO<sub>2</sub> after the protonation, respectively.

The interlayer distances, namely those of the (020) planes, were shifted after the ion-exchange from the alkaline ions to proton (Fig. S1a,c).

UV-Vis spectra of the µm-TiO<sub>2</sub> and nm-TiO<sub>2</sub>



Fig. S2 UV-Vis spectra and their Tauc's plots (the inset) of the nm-TiO<sub>2</sub> (i) and µm-TiO<sub>2</sub> (ii).

The  $E_g$  of the µm-TiO<sub>2</sub> and nm-TiO<sub>2</sub> 2–5 nm in size was estimated to be  $E_g$ =3.24 eV and  $E_g$ =3.25 eV, respectively. As mentioned in the main text, the effect of the lateral size reduction is calculated to be  $\Delta E_{g,calc}$ =0.019–0.12 eV by the first term of the (eq. 1) on the assumption of  $L_{x,y}$ = 2.0–5.0 nm and  $\mu_{x,y}$ =1.63 $m_e$  as the reduced effective mass of bulk anatase titanium dioxide. In the present work, the experimental  $E_g$  of the nm-TiO<sub>2</sub> would have the slight deviations because of the reproducibility of size distribution and crystallinity resulting from the solution syntheses.