

## Electronic Supplementary Information

### **Implications of boron doping on electrocatalytic activities of graphyne and graphdiyne family: A first principles study**

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Table S1: properties of optimized structures, B doping percentage and references of prior works of  $\alpha$ Gy,  $\beta$ Gy,  $\gamma$ Gy,  $\delta$ Gy, 6,6,12Gy, RGy and Gdy.

Structure	Lattice parameters		Space Group	No. of atoms in 2x2 cell	Doping percentage	References of prior works
	a (Å)	b (Å)				
$\alpha$ -Gy	6.96	6.96	P6/mmm(191)	32	3.1%	1, 2, 3, 4
$\beta$ -Gy	9.5	9.5	P6/mmm(191)	72	1.4%	1, 3, 5
$\gamma$ -Gy	6.83	6.83	P6/mmm(191)	48	2.1%	1, 3, 4, 6
$\delta$ -Gy	9.44	10.67	Pm(6)	80	1.3%	7
Gdy	9.46	9.46	P6/mmm(191)	72	1.4%	4, 8, 9
6,6,12-Gy	10.16	6.89	Pmmm(47)	72	1.4%	1, 10
R-Gy	6.02	6.02	P4/mmm(123)	32	3.1%	11

Table S2: Parameters of single B doping in pristine Gy/ Gdy systems for all other considered systems;  $E_{\text{for}}$  is the formation energy of the B doped configurations,  $Q_B$  is the Bader charge transfer from dopant B atom.

B doped configuration	$E_{\text{for}}$ (eV)	$Q_B$ (e)
$\text{Ch}_{\text{sp}}\text{-}\alpha\text{Gy}$	1.874	1.74
$\text{Ch}_{\text{sp}}\text{-}\beta\text{Gy}$	1.466	1.76
$\text{Ch}_{\text{sp}}\text{-}\gamma\text{Gy}$	1.766	1.79
$\text{H}_{\text{sp}2}\text{-}\delta\text{Gy}$	1.155	1.87
$\text{Ch}_{\text{sp}}\text{-}\delta\text{Gy}$	1.915	1.76
$\text{Ch}_{\text{sp}}^{\text{n}}\text{-Gdy}$	1.446	1.77
$\text{Ch}_{\text{sp}}^{\text{f}}\text{-Gdy}$	1.508	1.81
$\text{H}_{\text{sp}2}\text{-6,6,12Gy}$	0.367	1.92
$\text{H}_{\text{sp}2}^{\text{x}}\text{-6,6,12Gy}$	0.615	1.89
$\text{Ch}_{\text{sp}}^{\text{n}}\text{-6,6,12Gy}$	1.490	1.77
$\text{Ch}_{\text{sp}}\text{-6,6,12Gy}$	1.486	1.74
$\text{Ch}_{\text{sp}}\text{-RGy}$	1.717	1.78

Table S3: Parameters for O<sub>2</sub> adsorption on B doped Gy/ Gdy hosts for all other considered structures; E<sup>do</sup><sub>ads</sub> is the O<sub>2</sub> adsorption energy, d<sub>O-O</sub> is the bond length of O<sub>2</sub> in adsorbed state, Δd<sub>O-O</sub> is the amount of stretching occurred after adsorption, d<sub>O(B)-Gy/ Gdy</sub> is the distance of the O atom adsorbed on B from the Gy/ Gdy plane, d<sub>O(C)-Gy/ Gdy</sub> is the distance of O atom adsorbed on C from the Gy/ Gdy plane, Q<sub>O</sub> is the bader charge transferred to the O<sub>2</sub> molecule during adsorption.

Configuration	O <sub>2</sub> adsorption site	Adsorption energy (eV)	Stretched bond length (Å <sup>0</sup> )	Bond stretching (%)	d <sub>O(B)-Gy/ Gdy</sub> (Å <sup>0</sup> )	d <sub>O(C)-Gy/ Gdy</sub> (Å <sup>0</sup> )	Total charge transfer to O <sub>2</sub> molecule (e)
H <sub>sp2</sub> -βGy	B-C <sub>sp2</sub> bridge	-0.03	1.448	16.59	1.544	1.507	0.53
H <sub>sp2</sub> -γGy	B-C <sub>sp</sub> bridge	0.161	1.308	5.31	1.622	-	0.93
H <sub>sp2</sub> -Gdy	B-C <sub>sp2</sub> bridge	-0.037	1.455	17.15	1.540	1.502	0.96
Ch <sub>sp2</sub> -6,6,12Gy	B-C <sub>sp2</sub> bridge	0.429	1.315	5.88	1.607	2.601	0.57
R <sub>sp2</sub> -RGy	B-C <sub>sp2</sub> bridge	-0.098	1.469	18.28	1.522	1.482	1.00

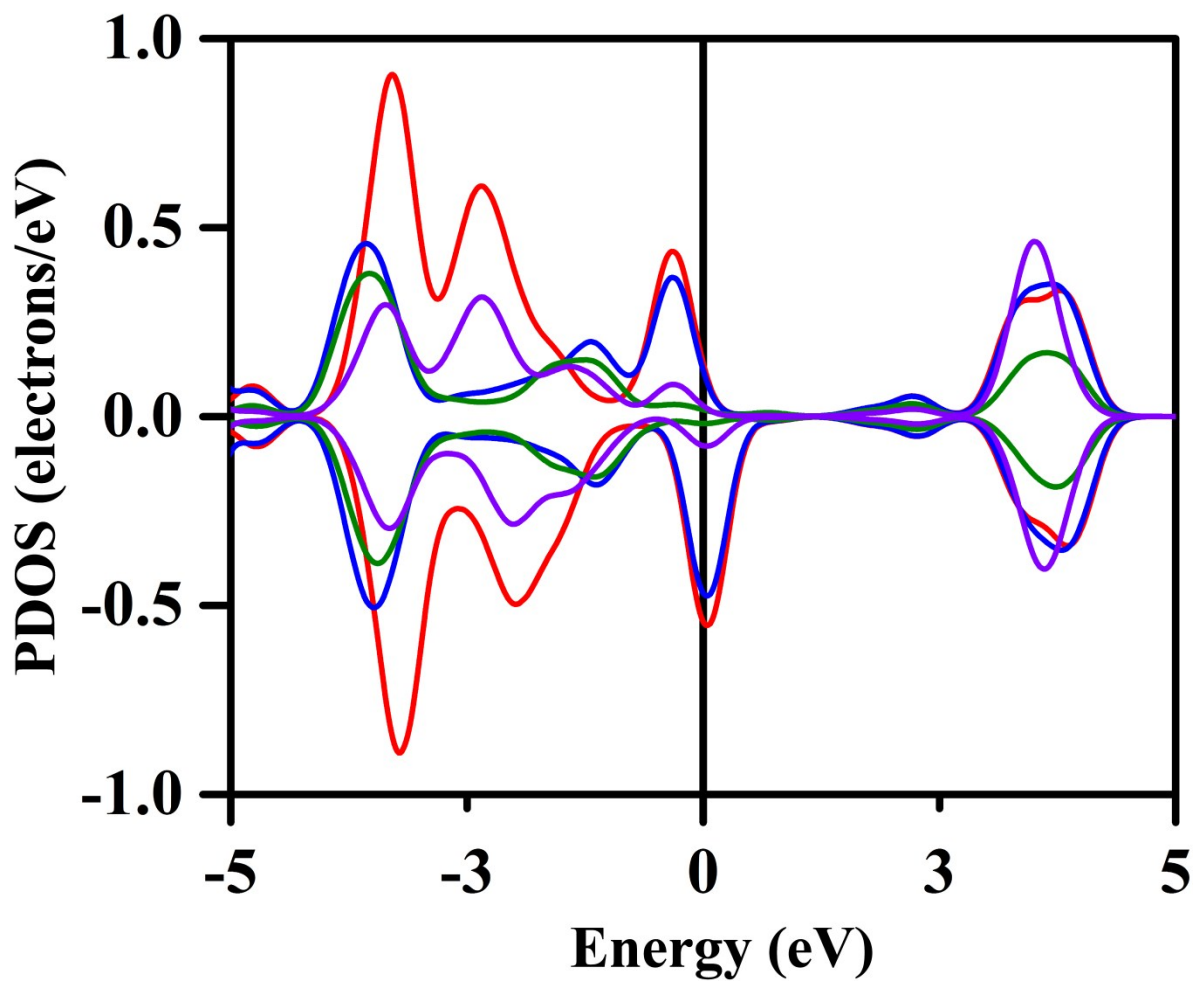


Figure S1. PDOS of dioxygen adsorbed on B-C<sub>sp</sub> bridge of B doped R-S<sub>sp2</sub>Gy configuration. Red and blue lines represent the PDOS of O<sub>B</sub> and O<sub>C</sub> p states respectively. Green and violet lines represent the B and C p states respectively.

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