Electronic supplementary information

Can HN=NH, FN=NH, or HN=CHOH Bridge the σ-Hole and the Lone Pair at P in Binary Complexes with H<sub>2</sub>XP, for X = F, Cl, NC, OH, CN, CCH, CH<sub>3</sub>, and H?

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Table S1. Structures, total energies, and molecular graphs of complexes of HN=NH, FN=NH, and NHCHOH with  $H_2XP$ , for X = F, Cl, NC, OH, CN, CCH, CH<sub>3</sub>, and H











	0,1.9816202727,1.0818620178,0.
	H.3.29799070760.4307595025.0.
	H.1.0068122815.1.1827594866.0.
	H.1.68870871992.0410326877.0.
	$H_2(OH)P$ ·NHCHOH MP2= -587 42098059 NIMAG=
H	P -1 1496936988 -0 0860815499 0
	$\Omega - 2.7973260638 \ 0.2328602314 \ 0$
Direction of the second s	H -1 1129747082 -1 0316360429 1 0402779193
	H -1 1129747082 -1 0316360429 -1 0402779193
I I I I I I I I I I I I I I I I I I I	N 1 4284800142 -1 1206479577 0
	C = 2.2685789254 = 0.1599811774 0
B	0.1.890586012.1.1200524755.0
	H 3 3517944512 -0 2546359717 0
	H 0 907123677 1 1163422789 0
	$H_{-2} = 0.357073231 + 1.846714683 = 0.0000000000000000000000000000000000$
	H 1 9137555991 _2 01293/0961 0
	$H_1(CN)P \cdot NHCHOH MP2 = 604.33320818 NIMAG=$
, <b>⊭</b> €	0
	P -0 7669014056 -0 159415007 0
· · · · · · · · · · · · · · · · · · ·	N _2 5027714712 0 0770327495 0
	$H_{0.7621018278} = 1.001/20/807 = 1.0/8530/303$
	$H_{-0.7621918278} = 1.0914244897, 1.0485394303$
	N 1 7027385855 1 0345885175 0
с	C = 2.5908764325 = 0.1176749939 0
0	$C_{22}$
	H 3 6665295576 $_{-0}$ 271 $^{1}$
	H 1 3089632723 1 2447855223 0
	$C_{-3} = 6437114214 = 0.3969765237 = 0.0000000000000000000000000000000000$
	H = 2 + 3.67778266 - 1 + 9528718101 + 0
	$H_2(CCH)P(NHCHOH MP2 = 588 26202844 NIM \Delta G = 5588 26202844 NIM \Delta G = 588 2620284 NIM A C = 588 2620284 $
	$n_2(\text{CCH})$ . When on which $2^-$ - 588.20202044 When $A^-$
	P _0 8217269/31 _0 1892753693 0
	$C_{-2}$ 5870189777 0 023081948 0
	H = 0.733088045 + 1.1251204875 + 0.465756747
	$H_{-0.733088045} = 1.1251294875 = 1.0465756747$
L L	N 2 0665076537 -1 0738853572 0
е~ _Я	C = 2.7737734188 = 0.0133922425.0
	0.22368113661121182023880
	H 3 8600610465 0 0378738527 0
	H 1 2643693273 1 0978091208 0
	$C_{-3}$ 77519/0228 0 3058652262 0
	$H_{-4} = 8146594294 \ 0.5310171745 \ 0$
	H 2 6712479521 - 1 8903388723 0
	H,2.6712479521,-1.8903388723,0.



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HN=NH with	$ ho_{ m BCP}$	$ abla^2  ho_{BCP}$	$H_{\rm BCP}$				
H <sub>2</sub> FP	0.035	0.069	-0.005				
H <sub>2</sub> ClP	0.027	0.063	-0.002				
H <sub>2</sub> (NC)P	0.027	0.062	-0.002				
H <sub>2</sub> (OH)P	0.022	0.057	-0.001				
H <sub>2</sub> (CN)P	0.017	0.048	0.000				
H <sub>2</sub> (CCH)P	0.014	0.041	0.001				
H <sub>2</sub> (CH <sub>3</sub> )P	0.009	0.031	0.001				
H <sub>3</sub> P	0.010	0.032	0.001				

Table S2. Electron density properties at intermolecular BCPs (au)  $P \cdots N_1$  pnicogen bonds in H<sub>2</sub>XP:HN=NH complexes

 $N_2$ -H<sub>2</sub>...P hydrogen bonds in H<sub>2</sub>XP:HN=NH complexes

HN=NH with	$ ho_{ m BCP}$	$ abla^2  ho_{BCP}$	$H_{\rm BCP}$
$H_2(CH_3)P$	0.012	0.030	0.001
H <sub>3</sub> P	0.011	0.028	0.001

P····N pnicogen bonds in H<sub>2</sub>XP:HN=NF complexes

HN=NF with	$ ho_{ m BCP}$	$ abla^2  ho_{BCP}$	$H_{\rm BCP}$
H <sub>2</sub> FP	0.019	0.055	0.000
H <sub>2</sub> ClP	0.014	0.044	0.001
H <sub>2</sub> (NC)P	0.016	0.047	0.001
H <sub>2</sub> (OH)P	0.012	0.038	0.001
H <sub>2</sub> (CN)P	0.011	0.036	0.001

N<sub>2</sub>-H<sub>2</sub>…P hydrogen bonds in H<sub>2</sub>XP:HN=NF complexes

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HN=NF with	$\rho_{BCP}$	$ abla^2  ho_{BCP}$	$H_{\rm BCP}$		
H <sub>2</sub> (OH)P	0.011	0.029	0.001		
H <sub>2</sub> (CCH)P	0.012	0.028	0.001		
H <sub>2</sub> (CH <sub>3</sub> )P	0.018	0.034	-0.001		
H <sub>3</sub> P	0.016	0.032	-0.0002		

NH=CHOH with	$ ho_{ m BCP}$	$\nabla^2 \rho_{BCP}$	H <sub>BCP</sub>
H <sub>2</sub> FP	0.034	0.068	-0.005
H <sub>2</sub> ClP	0.029	0.063	-0.002
H <sub>2</sub> (NC)P	0.028	0.063	-0.002
H <sub>2</sub> (OH)P	0.021	0.054	-0.0004
H <sub>2</sub> (CN)P	0.018	0.048	0.000
H <sub>2</sub> (CCH)P	0.014	0.039	0.001
H <sub>2</sub> (CH <sub>3</sub> )P	0.010	0.030	0.001
H <sub>3</sub> P	0.010	0.030	0.001

P...N pnicogen bonds in H<sub>2</sub>XP:NH=CHOH complexes

O-H<sub>2</sub>…P hydrogen bonds in H<sub>2</sub>XP:NH=CHOH complexes

NH=CHOH with	$ ho_{ m BCP}$	$ abla^2  ho_{BCP}$	$H_{\rm BCP}$
H <sub>2</sub> FP	0.023	0.041	-0.002
H <sub>2</sub> ClP	0.020	0.039	-0.001
H <sub>2</sub> (NC)P	0.017	0.037	0.000
H <sub>2</sub> (OH)P	0.023	0.038	-0.002
H <sub>2</sub> (CN)P	0.015	0.033	0.000
H <sub>2</sub> (CCH)P	0.019	0.036	-0.001
H <sub>2</sub> (CH <sub>3</sub> )P	0.023	0.036	-0.002
H <sub>3</sub> P	0.019	0.036	-0.001



Figure S1. Energy densities at bond critical points (au) vs. the P-N distance (Å) for P $\cdots$ N pnicogen bonds in H<sub>2</sub>XP:NH=CHOH complexes



Fig. S2. Energy densities at bond critical points (au) vs the  $P-H_2$  distance (Å) for hydrogen bonds in  $H_2XP:NH=CHOH$  complexes

Table S3. Components of  ${}^{1p}J(P-N_1)$  (Hz) for complexes of H<sub>2</sub>XP with HN=NH, FN=NH, and NH=CHOH

HNNH w.	PSO	DSO	FC	SD	<sup>1p</sup> J(P-N <sub>1</sub> )
$H_2FP$	-0.2	-0.1	-55.5	0.0	-55.8
H <sub>2</sub> ClP	-0.3	-0.1	-55.7	0.1	-55.9
H <sub>2</sub> (NC)P	-0.2	0.0	-49.6	0.0	-49.8
H <sub>2</sub> (OH)P	-0.2	0.0	-37.0	0.0	-37.1
H <sub>2</sub> (CN)P	-0.1	0.0	-30.8	0.0	-30.9
H <sub>2</sub> (CCH)P	0.0	0.0	-20.9	0.0	-21.0
H <sub>2</sub> (CH <sub>3</sub> )P	0.0	0.0	-8.4	0.0	-8.3
$H_3P$	0.0	0.0	-10.2	0.0	-10.2
FNNH w					
H <sub>2</sub> FP	-0.2	-0.1	-57.7	0	-57.9
H <sub>2</sub> ClP	-0.2	-0.1	-41.7	0	-41.9
H <sub>2</sub> (NC)P	-0.1	-0.1	-47.1	0	-47.3
H <sub>2</sub> (OH)P	-0.1	-0.1	-21.5	0	-21.6
H <sub>2</sub> (CN)P	0	-0.1	-27.9	0	-28
H <sub>2</sub> (CCH)P	0	0	-8.7	0	-8.7
H <sub>2</sub> (CH <sub>3</sub> )P	0.1	0	1.1	0	1.2
$H_3P$	0.1	0	1.6	0	1.6
NHCHOH w					
H <sub>2</sub> FP	-0.8	-0.1	-51.3	-0.1	-52.3
H <sub>2</sub> ClP	-0.7	-0.1	-57.8	0.1	-58.5
H <sub>2</sub> (NC)P	-0.5	-0.1	-51.8	0.0	-52.4
H <sub>2</sub> (OH)P	-0.3	-0.1	-39.0	0.1	-39.3
H <sub>2</sub> (CN)P	-0.1	0.0	-36.6	0.1	-36.7
H <sub>2</sub> (CCH)P	0.0	0.0	-27.5	0.1	-27.5
H <sub>2</sub> (CH <sub>3</sub> )P	0.0	0.0	-16.0	0.1	-15.9
H <sub>3</sub> P	0.0	0.0	-16.9	0.1	-16.8

Table S4.	Components	of spin-spin	coupling	constants J	J(N-P) (	Hz) for	complexes	of H <sub>2</sub> XP	with
HN=NH a	und FN=NH, a	and ${}^{2h}J(O-P)$	for compl	lexes with l	NH=CH	IOH			

HNNH w	PSO	DSO	FC	SD	$J(N_2-P)$
$H_2FP$	0.7	0.0	-6.3	0.0	-5.6
H <sub>2</sub> ClP	0.5	0.0	-5.0	0.0	-4.5
H <sub>2</sub> (NC)P	0.4	0.0	-4.4	0.0	-4.0
H <sub>2</sub> (OH)P	0.3	0.0	-5.4	0.0	-5.0
H <sub>2</sub> (CN)P	0.1	0.0	-3.3	0.0	-3.1
H <sub>2</sub> (CCH)P	0.1	0.0	-4.5	0.1	-4.3
H <sub>2</sub> (CH <sub>3</sub> )P	0.1	0.0	-5.6	0.1	-5.3
H <sub>3</sub> P	0.1	0.0	-4.2	0.1	-4.1
FNNH w					
$H_2FP$	0.2	0	0.2	0	0.4
H <sub>2</sub> ClP	0.1	0	-0.2	0	-0.1
H <sub>2</sub> (NC)P	0.1	0	0.4	0	0.4
H <sub>2</sub> (OH)P	0.1	0	-2.7	0.1	-2.5
H <sub>2</sub> (CN)P	0	0	0	0	0.1
H <sub>2</sub> (CCH)P	0.1	0	-4.1	0.1	-3.9
H <sub>2</sub> (CH <sub>3</sub> )P	0	0	-17.8	-0.3	-18ª
H <sub>3</sub> P	0.1	0	-12	-0.2	-12.1ª
NHCHOH w					<sup>2h</sup> J(O-P)
H <sub>2</sub> FP	0.4	0.0	-18.5	0.2	-17.9
H <sub>2</sub> ClP	0.4	0.0	-11.8	0.2	-11.2
H <sub>2</sub> (NC)P	0.3	0.0	-7.3	0.2	-6.9
H <sub>2</sub> (OH)P	0.0	0.0	-19.2	0.2	-19.1
H <sub>2</sub> (CN)P	0.0	0.0	-5.0	0.1	-4.8
H <sub>2</sub> (CCH)P	-0.1	0.0	-14.0	0.1	-14.0
H <sub>2</sub> (CH <sub>3</sub> )P	-0.2	0.0	-23.5	0.0	-23.8
H <sub>3</sub> P	-0.2	0.0	-16.1	0.1	-16.2

a) <sup>2h</sup>J(N-P)



Figure S3. <sup>2h</sup>J(O-P) versus the O-P distance for complexes H<sub>2</sub>XP:NH=CHOH