

**Intensified Effects of Multi-Cu Modification on Electronic Properties of the
Modified Base Pairs Containing Hetero-Ring-Expanded Pyrimidine Bases**

Nan Lu[†], Yuxiang Bu^{*,‡} and Huatian Wang[†]

[†]College of Chemistry and Material Science, Shandong Agricultural University, Taian

City 271018, P. R. China

[‡]The Center of Modeling & Simulation Chemistry, Institute of Theoretical Chemistry,

Shandong University, Jinan 250100, P. R. China

Supporting Information

Part 1. Structural parameters (bond lengths, DI, BO) and optimized geometries of the natural, the size-expanded and the M-x modified base pairs.

Part 2. Electronic properties (orbital composition, spin density distribution) of the natural, the size-expanded and the M-x modified base pairs.

Part 3. Indicators of transverse electronic communication (vertical transition energies, oscillator strengths, and state assignments; Differences of WC H-bond; Absorption spectra; ELF- π isosurface; DOS Plots) for natural and modified base pairs.

Part 1. Structural parameters (bond lengths, DI, BO) and optimized geometries of the natural, the size-expanded and the M-x modified base pairs.

Table S1. The selected structural parameters of A-nU, G-nU and their M-x modified ones (bond lengths in Å) [B3LYP/6-311++G**(C,H,O,N)/LANL2DZ(Cu)].

	A-nU			G-nU			
	N6-H···O4	N1···H-N3		O6-H···O4	N1···H-N3	N2-H···O2	
natural	2.950	2.908		2.749	2.908	3.038	
1	2.951	2.918		2.745	2.920	3.010	
2	2.943	2.916		2.742	2.931	3.020	
3	2.937	2.923		2.735	2.927	3.021	
4	2.964	2.908		2.752	2.923	3.007	
	A-nC			G-nC			
	N6···H-N4	N1-H···N3		O6···H-N4	N1-H···N3	N2-H···O2	
natural	2.928	2.902		2.824	2.967	2.943	
1	2.917	2.890		2.807	2.971	2.895	
2	2.926	2.903		2.825	2.977	2.933	
3	2.930	2.895		2.822	2.979	2.914	
4	2.901	2.900		2.797	2.979	2.906	
	A ₂ Cu _n U			G ₃ Cu _n U			
	N6-Cu-O4	N1-Cu-N3	Cu1-Cu2	O6-Cu-O4	N1-Cu-N3	N2-Cu-O2	Cu1-Cu2-Cu3
1	3.801	3.873	2.579	3.824	3.842	3.816	2.477/2.482
2	3.794	3.881	2.570	3.817	3.851	3.818	2.475/2.490
3	3.795	3.878	2.569	3.819	3.846	3.818	2.473/2.486
4	3.802	3.875	2.581	3.823	3.846	3.816	2.480/2.484
	A ₂ Cu _n C			G ₃ Cu _n C			
	N6-Cu-N4	N1-Cu-N3	Cu1-Cu2	O6-Cu-N4	N1-Cu-N3	N2-Cu-O2	Cu1-Cu2-Cu3
1	3.799	3.878	2.586	3.801	3.861	3.806	2.484/2.485
2	3.795	3.880	2.590	3.802	3.857	3.815	2.491/2.491
3	3.796	3.880	2.580	3.800	3.860	3.811	2.482/2.488
4	3.799	3.879	2.587	3.799	3.865	3.806	2.486/2.486

Table S2. DI and BO (>0.003) values for the natural and modified base pairs [B3LYP/6-311++G**(C,H,O,N)/LANL2DZ(Cu)].

	DI(*10 ⁻³)		BO(*10 ⁻²)			
	C6···C4	C2···C2	C6···C4	C2···C2	N6-N1-N3-N4	
A-C	0.44	0.43	/	/	/	
A-1C	0.71	1.06	0.35	0.68	0.365	
A-2C	0.68	0.95	0.61	0.36	0.585	
A-3C	0.58	0.84	/	0.74	0.56	
A-4C	0.72	1.01	0.35	0.44	0.49	
A _{2Cu} 1C	2.57	2.72	11.84	0.73	2.26	
A _{2Cu} 2C	2.29	2.09	9.1	1.92	1.78	
A _{2Cu} 3C	1.09	2.28	18.69	1.08	2.1	
A _{2Cu} 4C	0.89	0.08	7.07	1.75	2.11	
	C6···C4	C2···C2	C6···C4	C2···C2	N6-N1-N3-O4	
A-U	0.5	0.75	/	/	/	
A-1U	0.49	0.74	2.88	0.41	/	
A-2U	0.49	0.79	/	/	/	
A-3U	0.49	0.75	2.89	/	/	
A-4U	0.49	0.8	/	0.44	/	
A _{2Cu} 1U	0.53	0.13	17.72	0.49	2.525	
A _{2Cu} 2U	1.56	2.34	9.5	1.16	2.615	
A _{2Cu} 3U	0.79	0.41	19.57	0.39	2.585	
A _{2Cu} 4U	2.5	1.36	8.33	0.82	2.44	
	C6···C4	C2···C2	C6···C4	C2···C2	O6-N1-N3-N4	N1-N2-O2-N3
G-C	0.73	1.52	/	/	/	/
G-1C	0.62	0.49	/	0.54	/	/
G-2C	0.59	0.47	/	0.67	/	/
G-3C	0.59	0.48	1.17	/	/	/
G-4C	0.63	0.47	/	0.67	/	/
G _{3Cu} 1C	1.09	1.82	9.07	4.06	3.24	1.75
G _{3Cu} 2C	1.48	1.11	10.65	3.58	2.68	1.84
G _{3Cu} 3C	0.47	0.46	15.28	2.47	3.46	1.925
G _{3Cu} 4C	1.57	0.25	9.58	4.24	3.04	1.905
	C6···C4	C2···C2	C6···C4	C2···C2	O6-N1-N3-O4	N1-N2-O2-N3
G-U	0.64	0.45	/	/	/	/
G-1U	0.64	0.45	/	/	0.52	0.36
G-2U	0.64	0.46	0.31	/	0.52	0.35
G-3U	0.64	0.45	4.93	0.53	/	0.34
G-4U	0.63	0.45	/	/	0.43	0.33
G _{3Cu} 1U	0.4	0.14	29.85	3.04	2.945	1.525
G _{3Cu} 2U	0.72	0.25	12.73	4.44	2.885	1.48
G _{3Cu} 3U	0.51	2.37	13.7	3.23	2.95	1.675

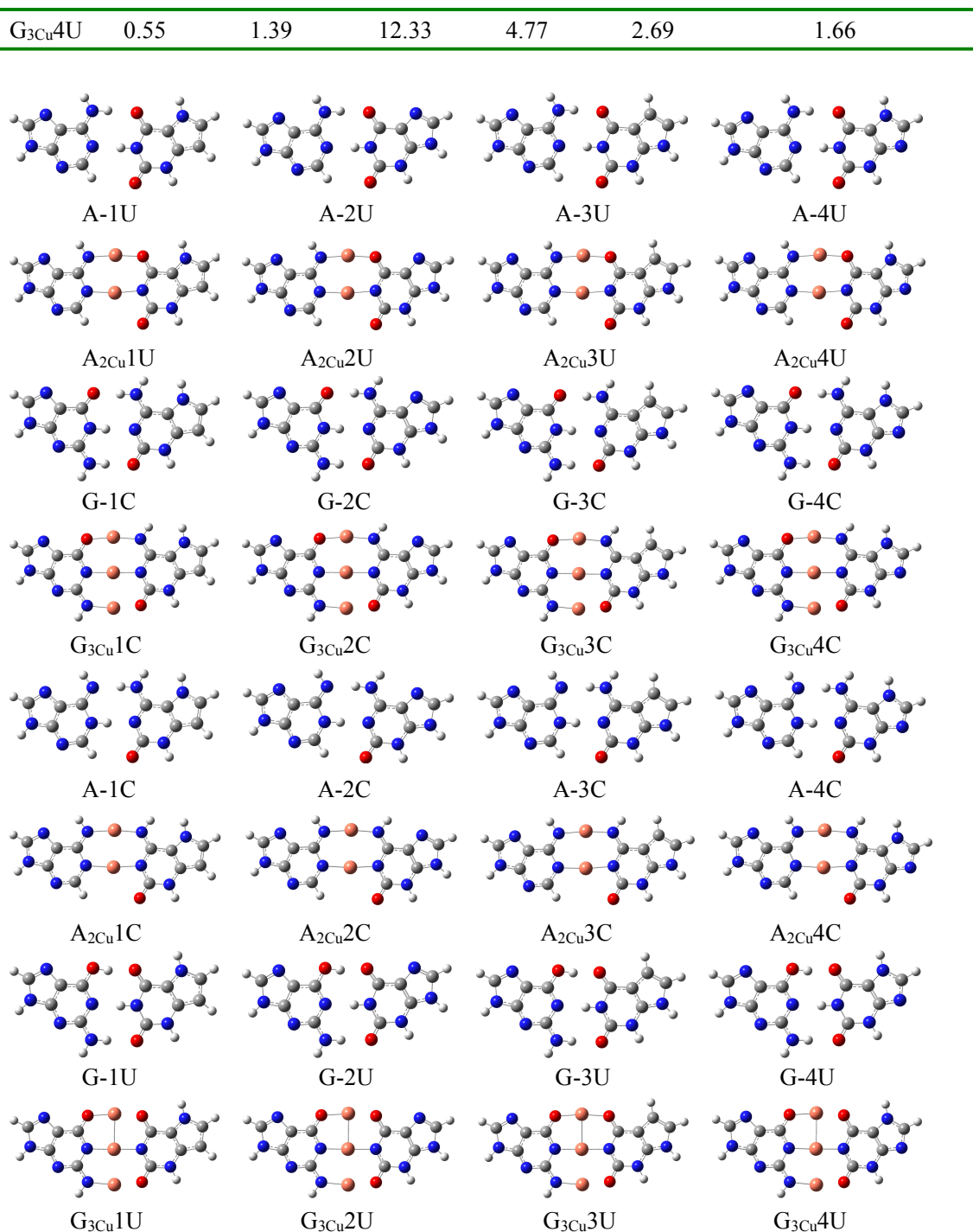


Figure S1. The optimized geometries of the natural, the size-expanded and the M-x modified base pairs [B3LYP/6-31+G*(C,H,O,N)/LANL2DZ(Cu)].

Part 2. Electronic properties (orbital composition, spin density distribution) of the natural, the size-expanded and the M-x modified base pairs.

Table S3. The HOMO, LUMO energy, HOMO-LUMO gaps, the ionization potentials and the electron affinities [eV] of the natural, the size-expanded and the M-x modified base pairs [B3LYP/6-311++G**(C,H,O,N)/LANL2DZ(Cu)].

	HOMO	LUMO	Gaps	VIP	AIP	VEA
A-U	-6.2	-1.63	4.57	7.97	7.74	0.91
A-1U	-6.13	-1.33	4.80	7.62	7.52	1.11
A-2U	-6.13	-1.08	5.05	7.72	7.60	-4.04
A-3U	-6.09	-0.91	5.18	7.50	7.40	-1.42
A-4U	-6.22	-1.60	4.62	7.81	7.70	-0.83
A _{2Cu} 1U	-5.71	-1.43	4.28	7.18		-0.94
A _{2Cu} 2U	-5.71	-1.24	4.47	7.23		-4.21
A _{2Cu} 3U	-5.67	-1.06	4.61	7.09		-2.27
A _{2Cu} 4U	-5.80	-1.67	4.13	7.33		-4.99
G-C	-5.5	-1.74	3.76	7.29	6.92	-1.20
G-1C	-5.43	-1.58	3.85	7.11	6.81	-0.78
G-2C	-5.45	-1.36	4.09	7.18	6.85	-5.12
G-3C	-5.42	-1.17	4.25	7.06	6.80	-2.71
G-4C	-5.52	-1.83	3.69	7.24	6.91	-2.74
G _{3Cu} 1C	-4.61	-1.96	2.64	6.17		-10.51
G _{3Cu} 2C	-4.64	-1.72	2.92	6.21		-11.44
G _{3Cu} 3C	-4.60	-1.55	3.04	6.16		-8.73
G _{3Cu} 4C	-4.69	-2.21	2.47	6.26		-15.32
A-C	-5.69	-1.47	4.22	7.46	7.15	1.19
A-1C	-5.64	-1.36	4.27	7.23	7.04	1.27
A-2C	-5.65	-1.12	4.53	7.30	7.07	-3.33
A-3C	-5.60	-0.93	4.67	7.14	7.01	-1.17
A-4C	-5.73	-1.60	4.14	7.38	7.15	-2.11
A _{2Cu} 1C	-5.54	-1.33	4.21	6.95		0.41
A _{2Cu} 2C	-5.55	-1.08	4.47	6.98		-2.20
A _{2Cu} 3C	-5.50	-0.95	4.55	6.88		-1.62
A _{2Cu} 4C	-5.64	-1.57	4.07	7.09		-3.55
G-U	-5.73	-2.02	3.71	7.54	7.23	-5.63
G-1U	-5.62	-1.68	3.95	7.34	7.09	-2.20
G-2U	-5.66	-1.44	4.22	7.42	7.13	-7.29
G-3U	-5.61	-1.21	4.40	7.28	7.07	-4.18

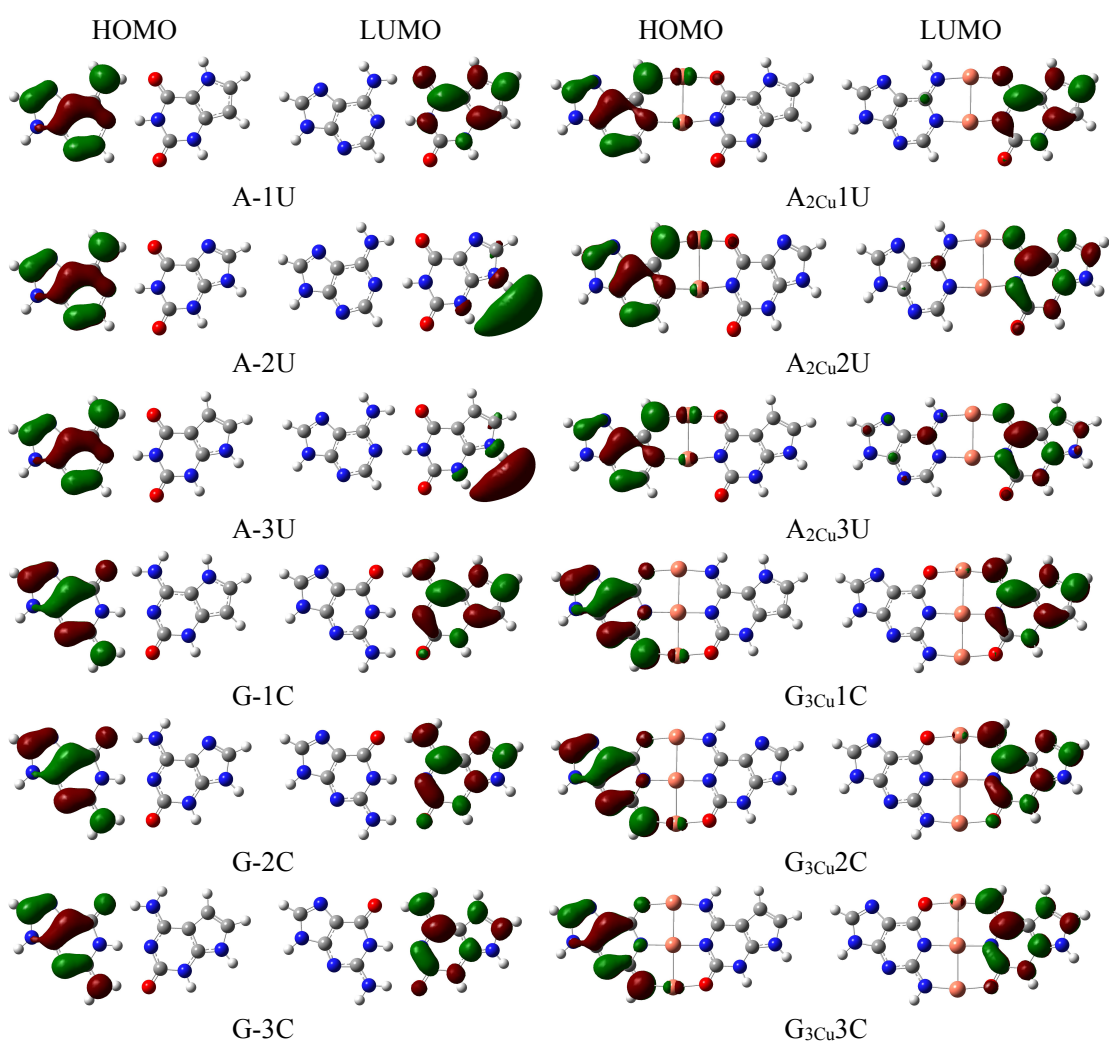
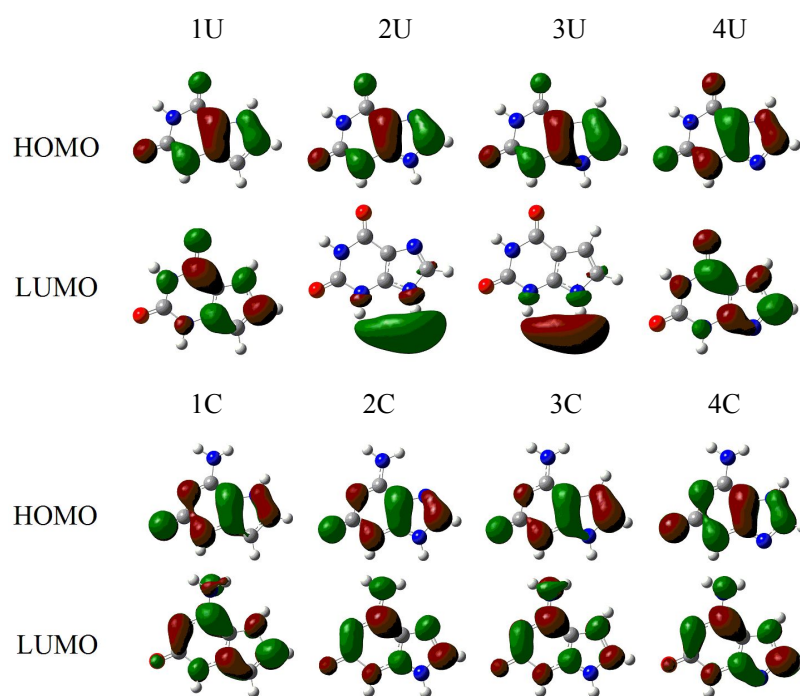
G-4U	-5.70	-1.95	3.75	7.46	7.19	-5.68
G _{3Cu} 1U	-4.72	-2.10	2.62	6.29		-12.45
G _{3Cu} 2U	-4.75	-1.95	2.81	6.33		-13.80
G _{3Cu} 3U	-4.70	-1.74	2.97	6.28		-10.34
G _{3Cu} 4U	-4.80	-2.36	2.44	6.37		-17.46

Table S4. Orbital composition (%) of atoms on the expanded ring for LUMOs of A-nU and A_{2Cu}nU analogs [B3LYP/6-311++G**(C,H,O,N)/LANL2DZ(Cu)]. (Σ is the sum value of listed compositions for each analog.)

	A-1U	A-2U	A-3U	A-4U	A _{2Cu} 1U	A _{2Cu} 2U	A _{2Cu} 3U	A _{2Cu} 4U
C5	7.23	-23.72	-130.46	7.71	3.98	2.28	1.67	4.48
C6	11.84	-45.29	81.86	10.00	11.72	11.06	11.38	10.37
C8	24.55	30.29	16.95	28.05	21.57	7.64	3.58	25.00
C7	1.30	/	/	/	1.03	/	/	/
N7	/	-47.16	-40.41	1.29	/	1.24	1.48	1.07
C9	/	/	-26.33	/	/	/	6.26	/
N9	6.68	-2.25	/	7.43	5.94	6.17	/	6.68
Σ	51.60	-88.13	-98.39	54.48	44.24	28.39	24.37	47.60

Table S5. Orbital composition (%) of atoms on imidazole rings for HOMOs and LUMOs of modified analogs (n=4) [B3LYP/6-311++G**(C,H,O,N)/LANL2DZ(Cu)]. (Σ is the sum value of listed compositions for each analog; Δ is the change value of Σ from expanded base pairs to M-x modified ones)

	A-4U	A ₂ Cu4U	A-4C	A ₂ Cu4C	G-4U	G ₃ Cu4U	G-4C	G ₃ Cu4C
HOMO								
C4	6.09	5.48	9.17	4.72	5.95	4.08	7.96	3.97
C5	16.32	13.31	16.98	11.95	19.45	18.55	21.50	18.54
C8	12.07	8.83	12.24	7.75	14.36	12.3	15.88	12.20
N7	5.44	2.20	2.27	1.85	5.27	3.59	4.83	3.48
N9	1.09	0.33	0.20	0.31	1.26	1.35	0.88	1.36
Σ _H	41.01	30.15	40.86	26.58	46.29	39.87	51.05	39.55
Δ _H	-10.86		-14.28		-6.42		-11.5	
LUMO								
C5	7.71	4.48	4.96	5.12	7.02	3.82	4.59	4.7
C6	10	10.37	9.34	9.41	10.89	11.16	9.74	10.3
C8	28.05	25	29.23	25.87	26.81	24.39	28.48	25.61
N7	1.29	1.07	2.17	1.53	1.05	0.9	1.98	1.33
N9	7.43	6.68	6.89	6.81	7.38	6.44	6.8	6.82
Σ _L	54.48	47.6	52.59	48.74	53.15	46.71	51.59	48.76
Δ _L	-6.88		-3.85		-6.44		-2.83	
Σ(Δ _H +Δ _L)	17.74		18.13		12.86		14.33	



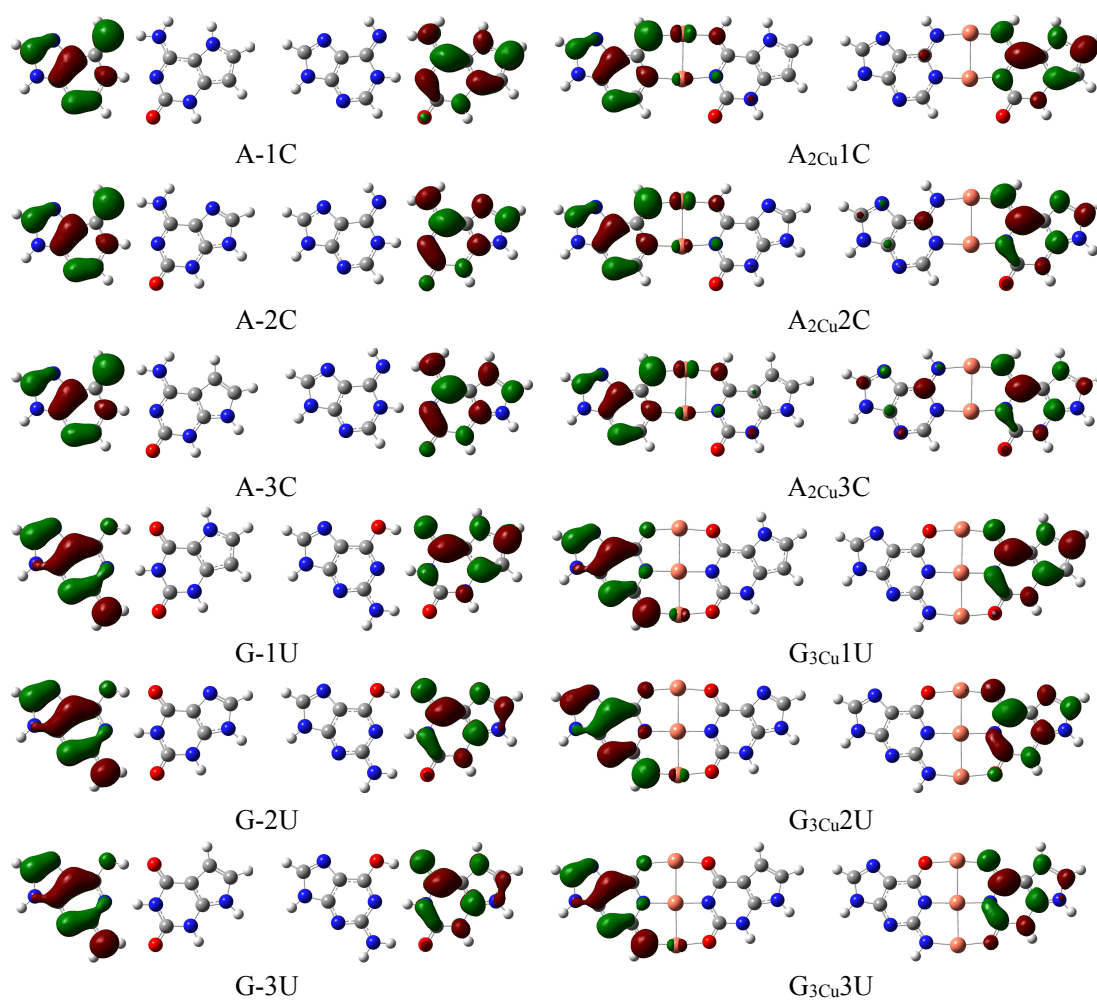


Figure S2. Frontier molecular orbitals of the natural, the size-expanded bases and the M-x modified base pairs (n=1, 2, 3) [B3LYP/6-311++G**(C,H,O,N)/LANL2DZ(Cu)].

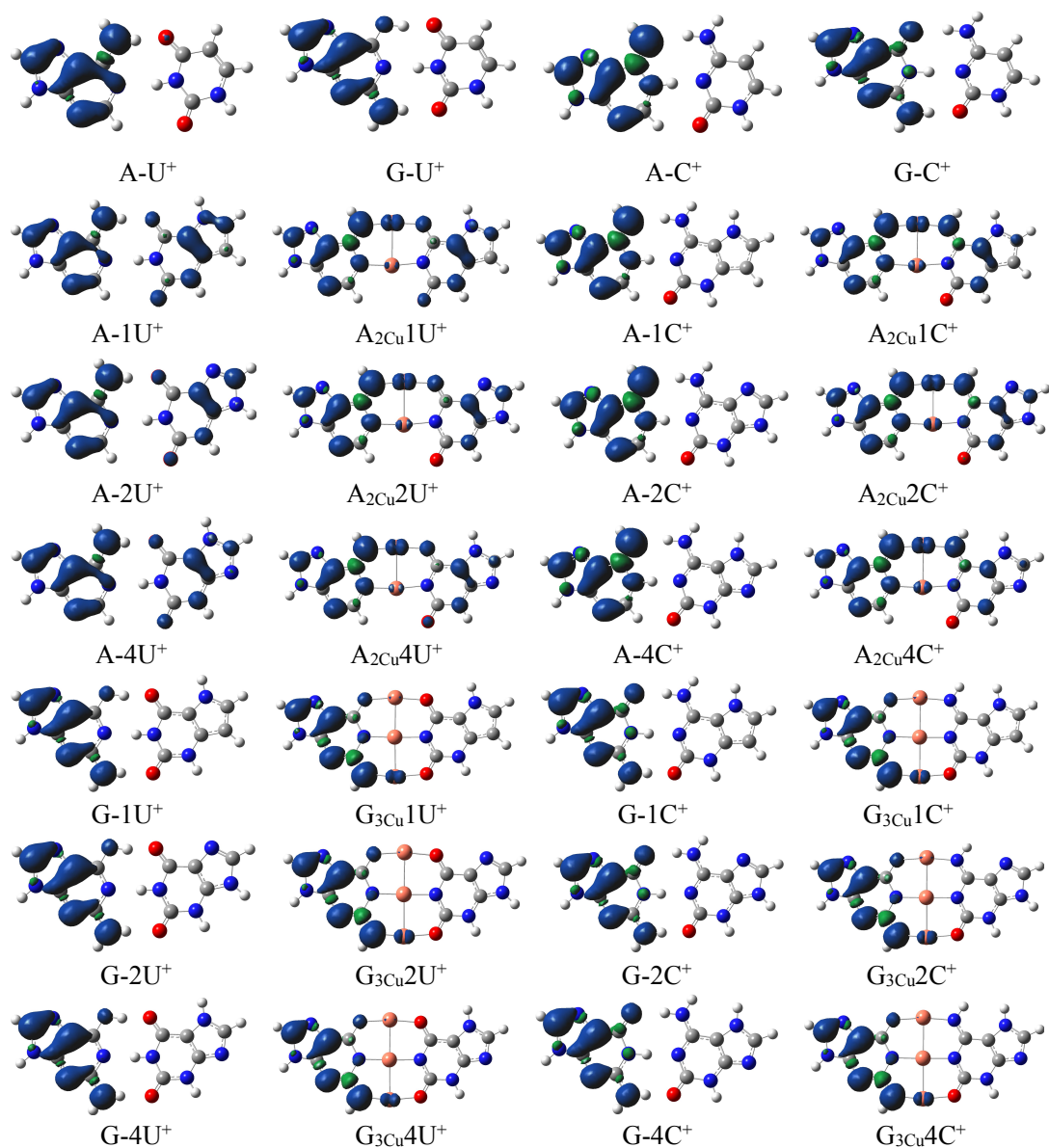


Figure S3. Spin density distributions of the oxidized natural, the size-expanded and the M-x modified base pairs (n=1, 2, 4) [B3LYP/6-311++G**(C,H,O,N)/LANL2DZ(Cu)].

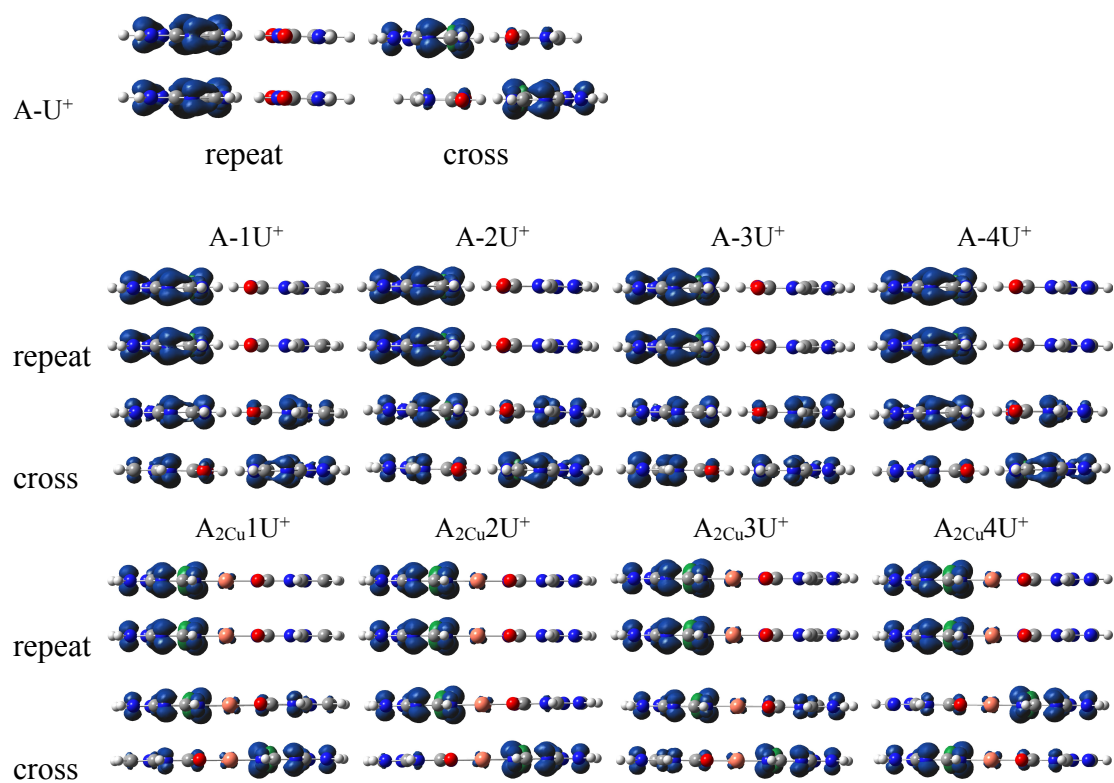


Figure S4. Spin density distributions of the oxidized natural and modified two-layer A-nU pairs

[M06-2X/6-311++G**(C,H,O,N)/LANL2DZ(Cu)].

Part 3. Indicators of transverse electronic communication (vertical transition energies, oscillator strengths, and state assignments; Differences of WC H-bond; Absorption spectra; ELF- π isosurface; DOS Plots) for natural and modified base pairs.

Table S6. The vertical transition energies, oscillator strengths, and state assignments^a to the low-lying singlet states of the natural, the size-expanded and the M-x modified A-nU, G-nU pairs.

calculated at TD-B3LYP/6-311++G** Level.

State		E/eV	f	Assign.	State		E/eV	f	Assign.
A-U	S ₁	4.09	0.003	$\pi_A\pi_U^*$	G-U	S ₂	4.49	0.004	$\pi_G\pi_U^*$
	S ₃	4.91	0.250	$\pi_A\pi_A^*$		S ₅	4.59	0.148	$\pi_G\pi_G^*$
	S ₄	5.05	0.070	$\pi_U\pi_U^*$		S ₉	5.10	0.072	$\pi_U\pi_U^*$
	S ₇	5.16	0.067	$\pi_A\pi_A^*$		S ₁₀	5.21	0.009	$\pi_Gn_G^*$
	S ₁₁	5.45	0.011	$\pi_A\pi_U^*$		S ₁₁	5.25	0.218	$\pi_Gn_G^*$
						S ₁₂	5.29	0.003	$n_G\pi_G^*$
						S ₁₄	5.44	0.018	$\pi_G\pi_U^*$
A-1U	S ₁	4.33	0.010	$\pi_A\pi_U^*$	G-1U	S ₁	3.51	0.004	$\pi_G\pi_U^*$
	S ₂	4.59	0.171	$\pi_U\pi_U^*$		S ₂	4.56	0.216	$\pi_G\pi_G^*$
	S ₃	4.91	0.226	$\pi_A\pi_A^*$		S ₃	4.66	0.071	$\pi_U\pi_U^*$
	S ₉	5.16	0.060	$\pi_A\pi_A^*$		S ₄	4.66	0.004	$\pi_Gn_U^*$
	S ₁₁	5.31	0.135	$\pi_U\pi_U^*$		S ₇	4.88	0.041	$\pi_G\pi_U^*$
						S ₈	4.89	0.004	$\pi_Gn_U^*$
						S ₁₁	5.22	0.003	$\pi_Gn_U^*$
						S ₁₂	5.26	0.259	$\pi_U\pi_U^* \& \pi_G\pi_{GU}^*$
						S ₁₃	5.27	0.070	$\pi_U\pi_U^* \& \pi_G\pi_{GU}^*$
						S ₁₄	5.29	0.002	$n_G\pi_G^*$
						S ₁₅	5.38	0.002	$\pi_Un_U^*$
A _{2Cu} 1U	S ₂	0.028	3.74	$\pi_A\pi_U^*$	G _{3Cu} 1U	S ₁	2.21	0.011	$\pi_G\pi_U^*$
	S ₃	0.302	4.29	$\pi_U\pi_U^*$		S ₃	3.26	0.015	$\pi_G\pi_U^*$
	S ₅	0.099	4.40	$\pi_A\pi_A^*$		S ₅	3.48	0.002	$\pi_G \& n_{Cu}\pi_U^*$
	S ₆	0.002	4.46	$\pi_An_U^*$		S ₈	3.87	0.031	$\pi_G\pi_U^{G*}$
	S ₇	0.078	4.47	$\pi_A\pi_A^*$					
	S ₉	0.040	4.70	$\pi_A\pi_U^*$					
	S ₁₀	0.007	4.74	$n_{Cu}\pi_U^*$					

	S ₁₁	0.048	4.79	Ryd on U & $\pi_A\pi_U^*$				
	S ₁₄	0.002	4.95	$\pi_{A\text{N}U}^*$				
A-2U	S ₁	0.014	4.55	$\pi_A\pi_U^*$	G-2U	S ₁	3.75	0.005 $\pi_G\pi_U^*$
	S ₄	0.314	4.90	$\pi_A\pi_A^*$		S ₄	4.60	0.170 $\pi_G\pi_G^*$
	S ₅	0.068	5.02	$\pi_U\pi_U^*$		S ₉	5.04	0.128 $\pi_U\pi_U^*$
	S ₈	0.033	5.15	$\pi_A\pi_A^* & \pi_A\pi_U^*$		S ₁₀	5.11	0.061 $\pi_G\pi_U^*$
	S ₁₀	0.021	5.25	$\pi_A\pi_U^*$		S ₁₂	5.24	0.009 $n_G\pi_G^* & n_G^U\pi_G^*$
	S ₁₁	0.017	5.37	$\pi_U\pi_A^*$		S ₁₃	5.26	0.074 $\pi_G\pi_U^* & \pi_G\pi_G^*$
	S ₁₄	0.211	5.57	$\pi_U\pi_U^*$		S ₁₄	5.30	0.069 $\pi_G\pi_U^* & \pi_G\pi_G^*$
						S ₁₅	5.33	0.006 $n_G\pi_G^*$
A _{2Cu} 2U	S ₂	0.043	3.88	$\pi_A\pi_U^*$	G _{3Cu} 2U	S ₁	2.38	0.016 $\pi_G\pi_U^*$
	S ₅	0.161	4.39	$\pi_A\pi_A^* & \pi_A\pi_A^{U*}$		S ₃	2.98	0.010 $\pi_G\pi_U^*$
	S ₆	0.038	4.44	Ryd on U		S ₆	3.62	0.002 $\pi_G & n_{Cu}\pi_U^*$
	S ₇	0.078	4.51	$\pi_A\pi_A^{U*}$		S ₈	3.68	0.013 $\pi_G\pi_U^*$
	S ₁₀	0.188	4.62	$\pi_U\pi_U^*$		S ₁₂	4.04	0.027 $\pi_G\pi_{UG}^*$
	S ₁₃	0.029	4.88	$\pi_A\pi_U^{A*}$				
	S ₁₄	0.005	4.88	$n_{Cu}\pi_U^*$				
	S ₁₅	0.002	4.92	$n_{Cu}\pi_U^{A*}$				
A-3U	S ₂	0.066	4.71	$\pi_A\pi_U^{A*}$	G-3U	S ₁	3.92	0.008 $\pi_G\pi_U^*$
	S ₃	0.092	4.77	$\pi_U\pi_U^{A*}$		S ₃	4.53	0.037 $\pi_G\pi_U^*$
	S ₅	0.178	4.91	$\pi_A\pi_A^{U*}$		S ₅	4.63	0.153 $\pi_G\pi_G^*$
	S ₆	0.021	4.99	$\pi_U\pi_A^{U*}$		S ₆	4.77	0.066 $\pi_U\pi_U^*$
	S ₁₀	0.050	5.17	$\pi_A\pi_A^{U*}$		S ₁₁	5.23	0.211 $\pi_G\pi_{GU}^*$
	S ₁₂	0.156	5.43	$\pi_U\pi_U^{A*}$		S ₁₃	5.40	0.140 $\pi_U\pi_U^*$
	S ₁₃	0.025	5.48	$\pi_A\pi_U^{A*}$				
	S ₁₅	0.007	5.59	$\pi_U\pi_A^{U*}$				
A _{2Cu} 3U	S ₂	0.056	4.00	$\pi_A\pi_U^*$	G _{3Cu} 3U	S ₁	2.53	0.016 $\pi_G\pi_U^*$
	S ₄	0.225	4.38	$\pi_A\pi_A^*$		S ₃	3.11	0.011 $\pi_G\pi_U^*$
	S ₆	0.094	4.45	$\pi_U\pi_U^*$		S ₈	3.77	0.002 $\pi_G & n_{Cu}\pi_U^*$
	S ₉	0.029	4.51	$\pi_A\pi_A^*$		S ₉	3.83	0.019 $\pi_G\pi_U^*$
	S ₁₀	0.020	4.59	Ryd on U		S ₁₅	4.26	0.117 $\pi_G\pi_{UG}^*$
	S ₁₃	0.025	4.90	$n_{Cu}\pi_U^*$				
	S ₁₄	0.010	4.96	$\pi_U\pi_A^*$				
	S ₁₅	0.040	4.98	$n_{Cu}\pi_U^*$				
A-4U	S ₁	0.005	4.16	$\pi_A\pi_U^*$	G-4U	S ₁	3.32	0.003 $\pi_G\pi_U^*$
	S ₂	0.210	4.63	$\pi_U\pi_U^*$		S ₂	4.55	0.179 $\pi_G\pi_G^*$
	S ₄	0.201	4.93	$\pi_A\pi_A^*$		S ₅	4.70	0.130 $\pi_U\pi_U^*$
	S ₆	0.050	5.16	$\pi_A\pi_A^*$		S ₆	4.75	0.035 $\pi_G\pi_U^*$
	S ₁₁	0.004	5.33	$\pi_U\pi_A^*$		S ₁₁	5.17	0.038 $\pi_G\pi_U^*$
	S ₁₂	0.009	5.51	$\pi_A\pi_U^*$		S ₁₂	5.28	0.097 $\pi_G\pi_G^* & n_G\pi_G^*$
						S ₁₃	5.29	0.055 $n_G\pi_G^* & \pi_G\pi_G^*$
A _{2Cu} 4U	S ₂	0.021	3.61	$\pi_A\pi_U^*$	G _{3Cu} 4U	S ₁	2.04	0.011 $\pi_G\pi_U^*$
	S ₃	0.329	4.31	$\pi_U\pi_U^*$		S ₃	3.16	0.014 $\pi_G\pi_U^*$

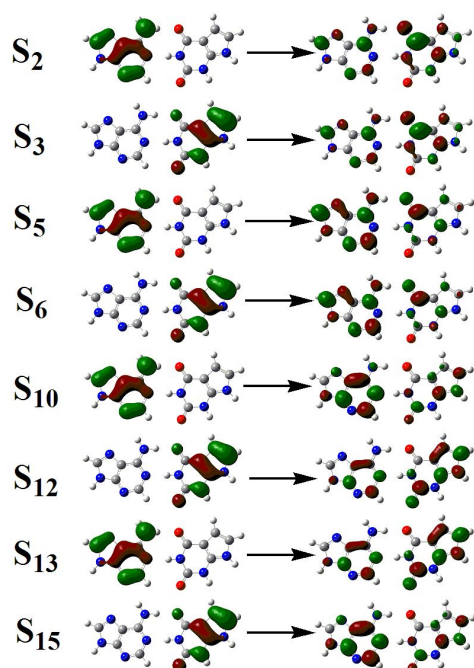
S ₅	0.078	4.41	n_{Cu}π_A[*]	S ₄	3.34	0.002	π_G&n_{Cu}π_U[*]
S ₆	0.100	4.45	π_Aπ_A[*]	S ₇	3.74	0.020	π_Gπ_U[*]
S ₉	0.006	4.64	n_{Cu}π_U[*]	S ₁₄	4.12	0.012	π_Gπ_U^{G*}
S ₁₀	0.026	4.65	π_Aπ_U[*]	S ₁₅	4.13	0.003	π_Gπ_U[*]
S ₁₁	0.011	4.79	Ryd on U				

^aThe charge-transfer transitions are in bold.

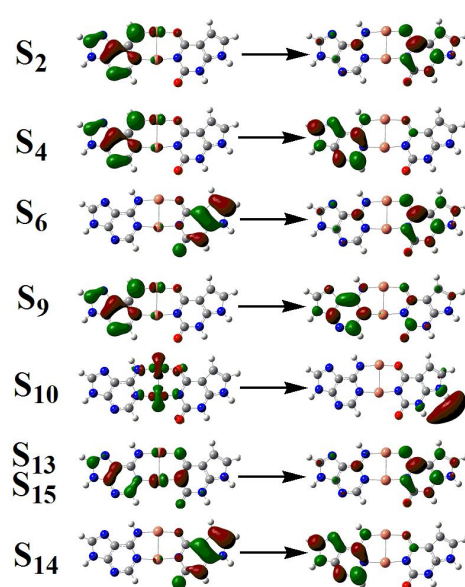
Table S7. Difference value of WC H-bond and N/O-Cu-N/O lengths (Å) between ground states and electronic excited states for G-nU and G_{3Cu}nU series.

	G-nU			G _{3Cu} nU		
	O6-H···O4	N1···H-N3	N2-H···O2	O6-Cu-O4	N1-Cu-N3	N2-Cu-O2
natural	0.002	-0.053	-0.061			
1	0.028	-0.027	-0.047	-0.032	-0.033	-0.024
2	0.027	-0.025	-0.049	-0.039	-0.029	0.002
3	0.028	-0.026	-0.048	-0.058	-0.048	-0.01
4	0.026	-0.028	-0.049	-0.031	-0.033	-0.027

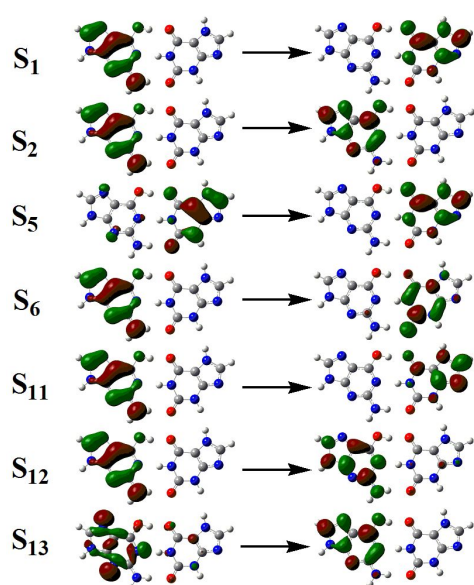
A-3U



A₂Cu₃U



G-4U



G₃Cu₄U

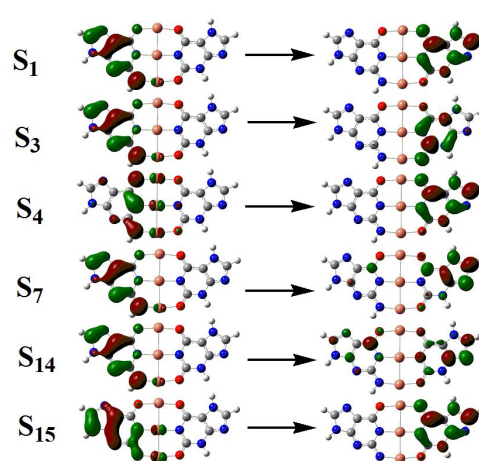


Figure S5. Molecular orbitals involved in several electronic singlet transitions of A-3U, A₂Cu₃U

and G-4U, G₃Cu₄U mentioned in Table S6 calculated at TD-B3LYP/6-311++G** level.

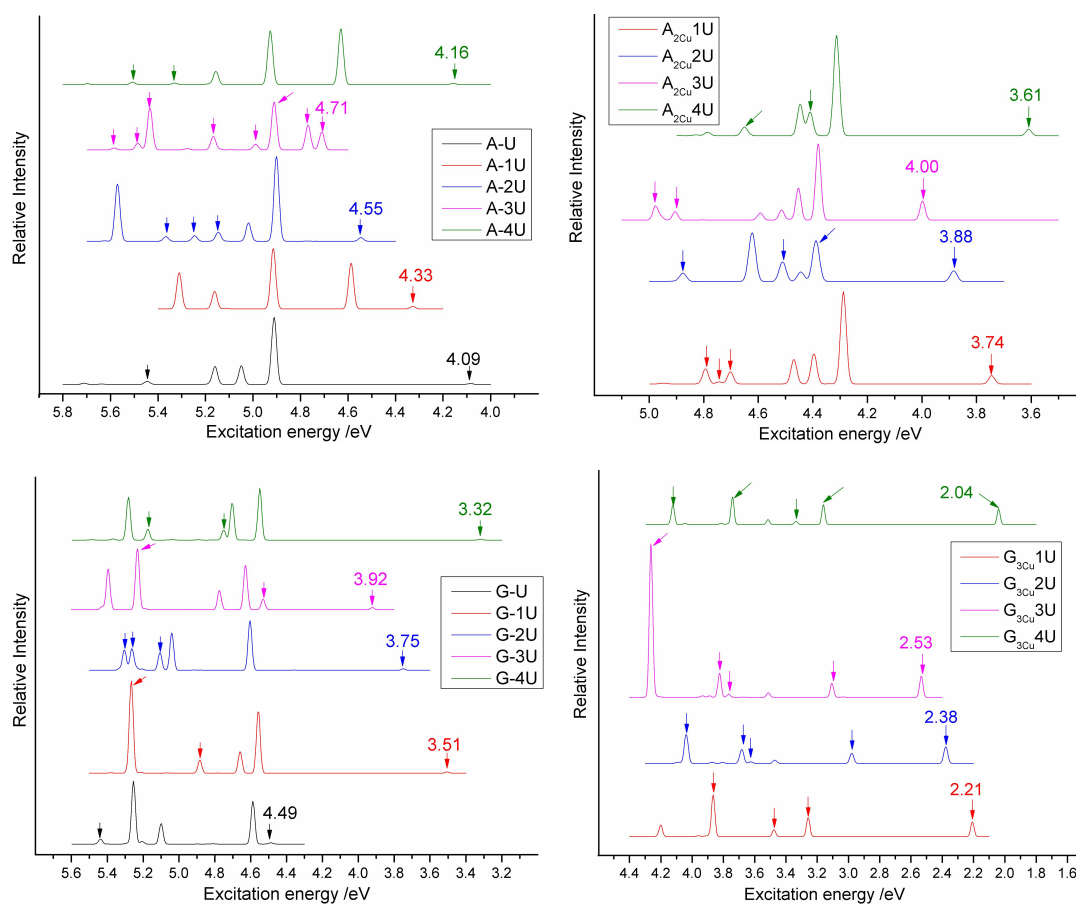


Figure S6. Absorption spectrum of the natural, the size-expanded and the M-x modified A-nU, G-nU pairs in the ultraviolet region obtained at TD-B3LYP/6-311++G** level. (half-bandwidths: 100 cm⁻¹). The charge transfer transitions are identified with arrows.

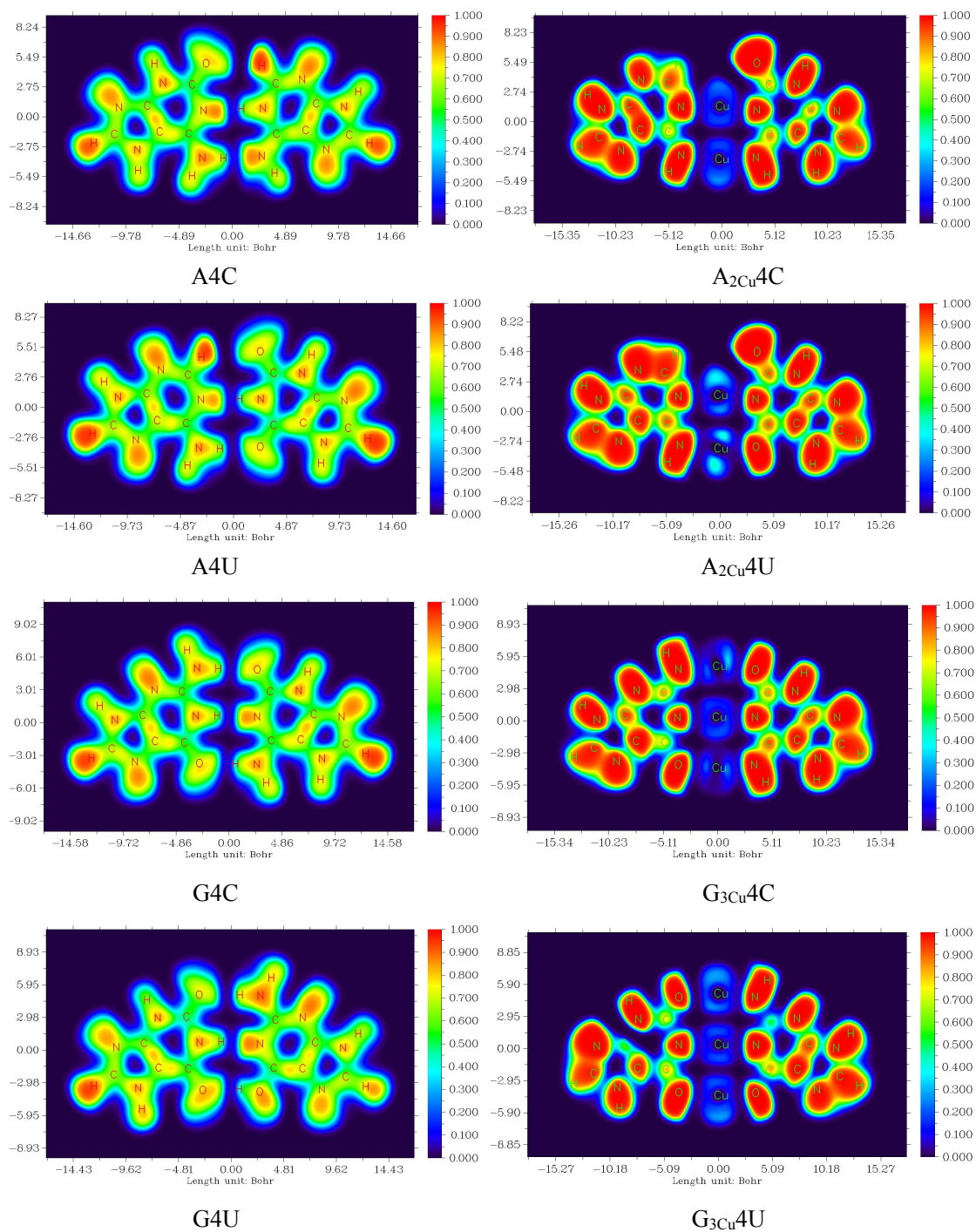
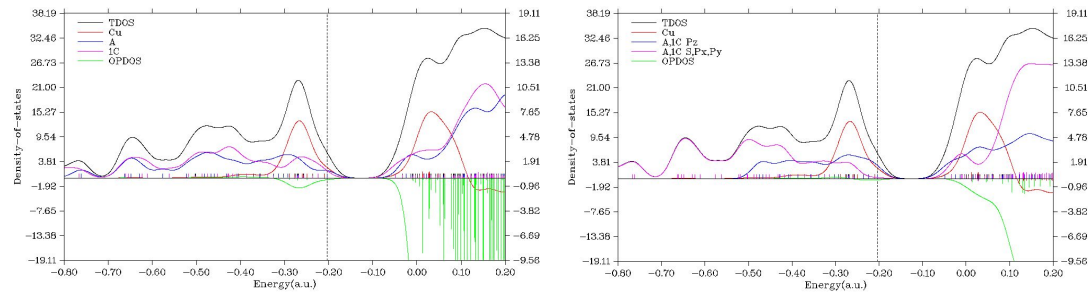
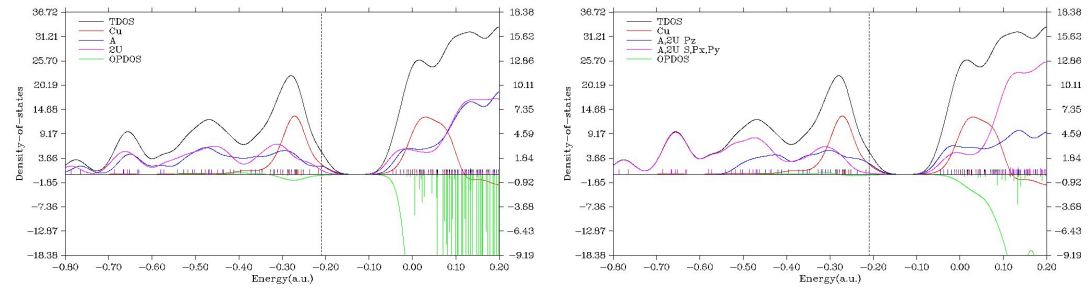


Figure S7. ELF- π isosurfaces above 1.5 bohr of the molecular plane for modified analogs (n=4).

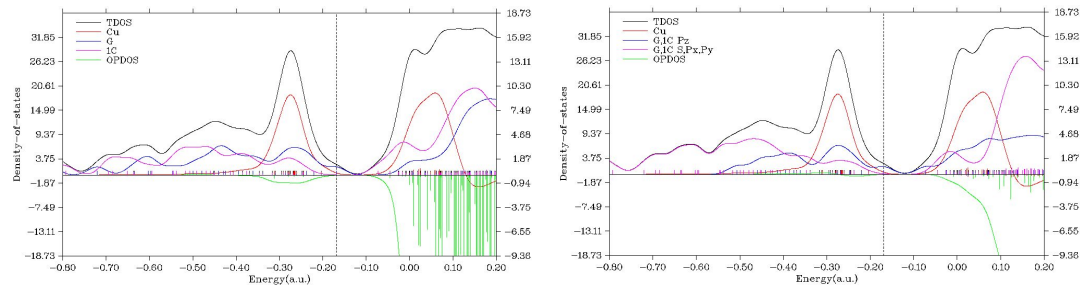
A₂Cu₁C



A₂Cu₂U



G₃Cu₁C



G₃Cu₃U

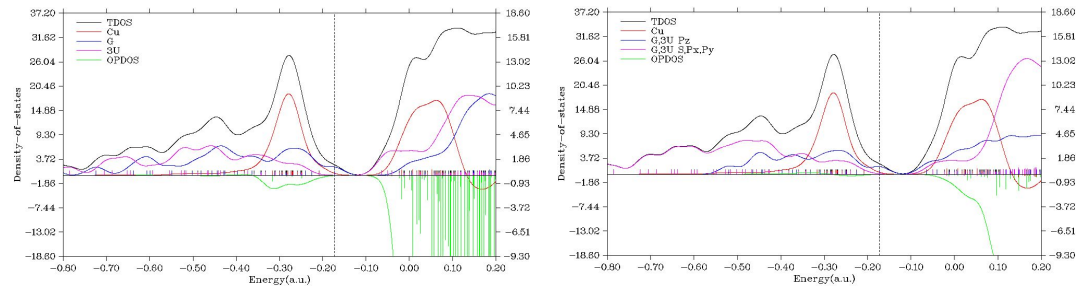


Figure S8. Plots of TDOS, PDOS and OPDOS for A₂Cu₁C, A₂Cu₂U, G₃Cu₁C and G₃Cu₃U.