

## Supporting Information

### Synthesis of hollow cobalt oxide nanopowders by salt-assisted spray pyrolysis process applying nanoscale Kirkendall diffusion and their electrochemical properties

Hyeon Seok Ju<sup>a</sup>, Jung Sang Cho<sup>a</sup>, Jong Hwa Kim<sup>b</sup>, Yun Ju Choi<sup>c</sup>,

and Yun Chan Kang<sup>a,\*</sup>

<sup>a</sup>Department of Materials Science and Engineering, Korea University, Anam-Dong,  
Seongbuk-Gu, Seoul 136-713, Republic of Korea

<sup>b</sup>Daegu Center, Korea Basic Science Institute, 80 Daehakro Bukgu, Daegu 702-701,  
Republic of Korea

<sup>c</sup>Suncheon Center, Korea Basic Science Institute, Suncheon 540-742, Republic of Korea  
E-mail: yckang@korea.ac.kr

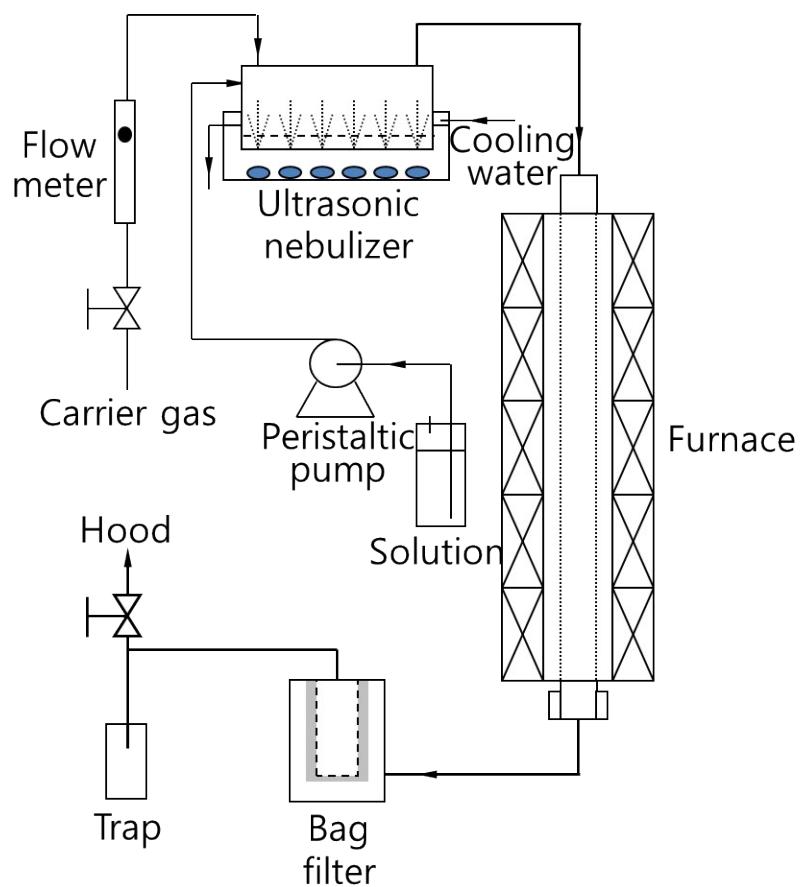


Figure S1. Schematic diagram of the ultrasonic spray pyrolysis system.

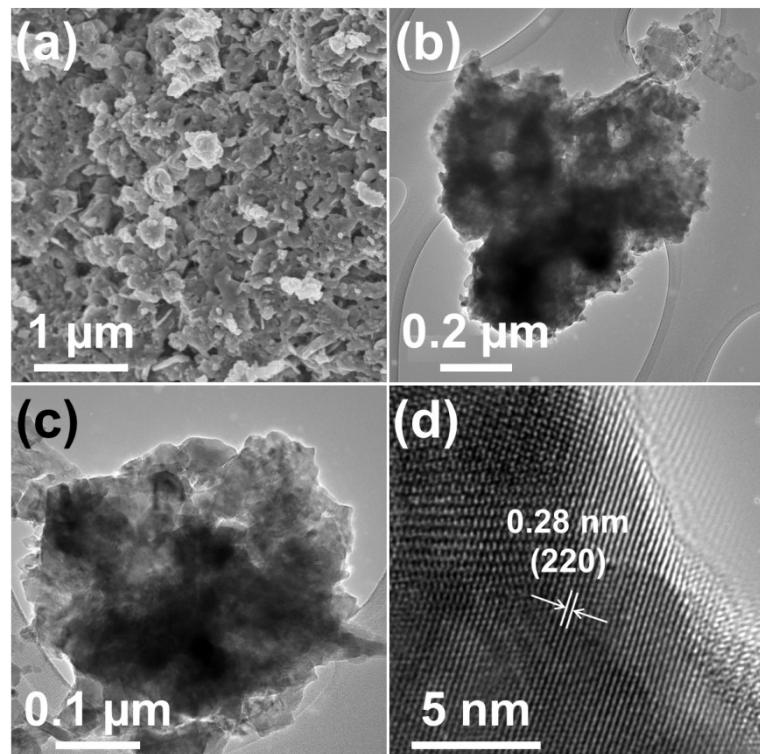
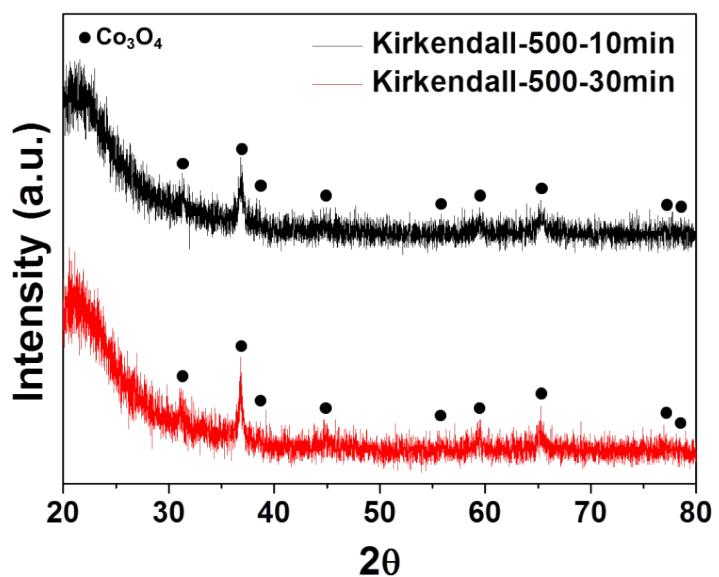
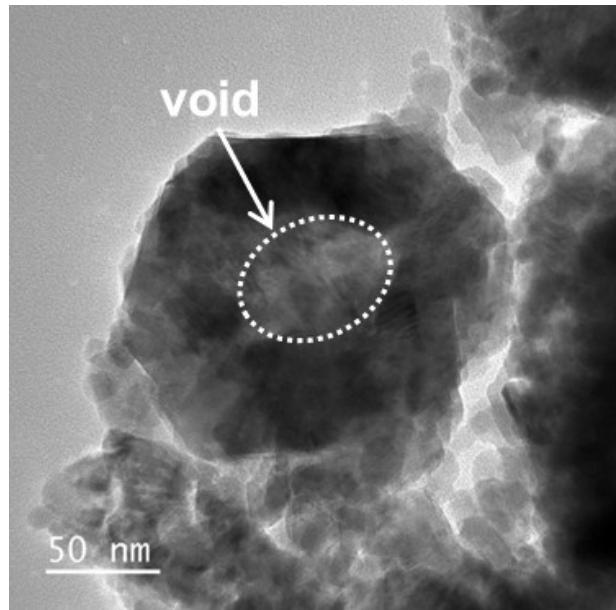


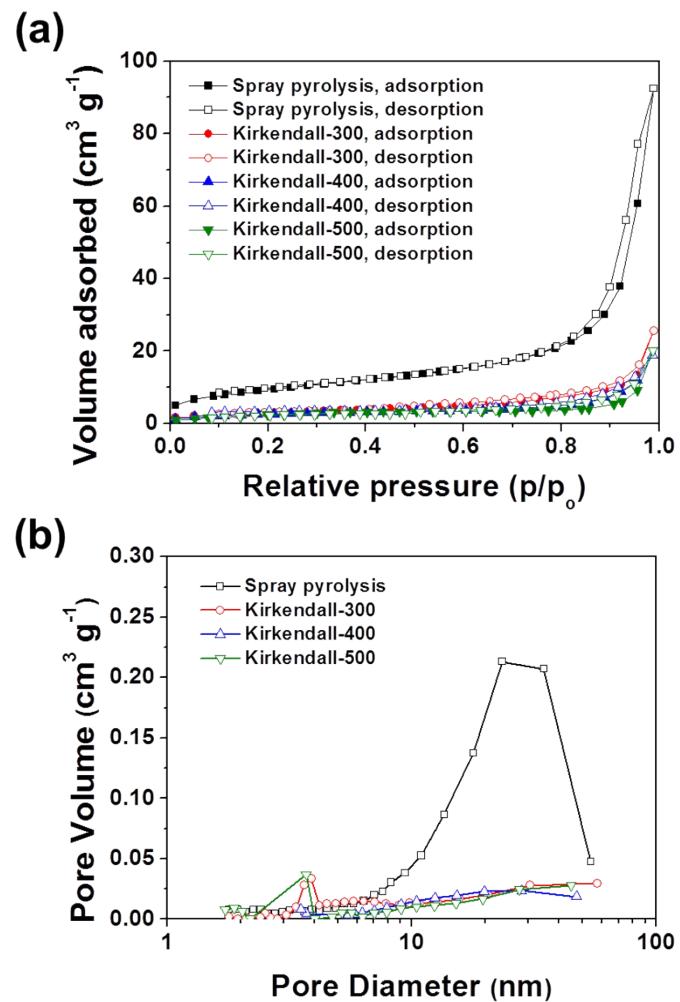
Figure S2. Morphologies of the nanopowders obtained by post-treatment of the metallic Co nanopowder under air atmosphere at a temperature of 300 °C: (a) SEM image, (b) and (c) TEM images, and (d) high resolution TEM image.



**Figure S3.** XRD patterns of the powders obtained after post-treatment for different oxidation time of the metallic Co nanopowders under air atmosphere at a temperature of 500 °C.



**Figure S4.** TEM image of the powder obtained after post-treatment for 10 min of the metallic Co nanopowders under air atmosphere at a temperature of 500 °C.



**Figure S5.**  $\text{N}_2$  adsorption and desorption isotherms and BJH pore size distributions of the hollow and filled  $\text{Co}_3\text{O}_4$  nanopowders.

**Table S1.** Comparison of the electrochemical performances of the hollow Co<sub>3</sub>O<sub>4</sub> nanopowders prepared by nanoscale Kirkendall diffusion process with the reported results of the nanostructured cobalt oxide materials.

Materials	Voltage range (V)	Current rate	Initial C <sub>dis</sub> /C <sub>cha</sub> [mA h g <sup>-1</sup> ]	Discharge capacity [mA h g <sup>-1</sup> ]	Cycle number	Ref
Co <sub>3</sub> O <sub>4</sub> hollow nanopowders	0.001-3	1000 mA g <sup>-1</sup>	1182/882	775	150	In this work
Co <sub>3</sub> O <sub>4</sub> nanoparticle	0.01-3	178 mA g <sup>-1</sup>	1237/835	~150	100	[40]
Co <sub>3</sub> O <sub>4</sub> nanowire	0.01-3	110 mA g <sup>-1</sup>	1027/755	611	50	[41]
Co <sub>3</sub> O <sub>4</sub> nanoparticle	0.01-3	30 mA g <sup>-1</sup>	1334/~1070	407	100	[42]
Plate-like Co <sub>3</sub> O <sub>4</sub>	0.02-3	178 mA g <sup>-1</sup>	1228/870	618	30	[43]
Hollow Co <sub>3</sub> O <sub>4</sub> @C sphere nanoparticle	0.01-3	74 mA g <sup>-1</sup>	~1050/732	750	20	[44]
Nanosized Co <sub>3</sub> O <sub>4</sub>	0.01-3	220 mA g <sup>-1</sup>	1411/931	913	20	[S1]
Co <sub>3</sub> O <sub>4</sub> Hollow microsphere	0.01-3	60 mA g <sup>-1</sup>	1241/1061	633	25	[S2]
Meso- and macro-porous Co <sub>3</sub> O <sub>4</sub> -carbon composite powders	0.01-3	1400 mA g <sup>-1</sup>	~1150/~920	644	100	[S3]
Mesoporous Co <sub>3</sub> O <sub>4</sub> cubes	0.01-3	89 mA g <sup>-1</sup>	1298/986	980	60	[S4]
Co <sub>3</sub> O <sub>4</sub> @graphene nanocomposite	0.01-3	50 mA g <sup>-1</sup>	1097/~753	~935	30	[S5]
Co <sub>3</sub> O <sub>4</sub> -nanobubble-decorated RGO spheres	0.001-3	2000 mA g <sup>-1</sup>	1471/~1050	1156	200	[S6]
Multishelled Co <sub>3</sub> O <sub>4</sub> hollow microspheres (triple-shelled)	0.05-3	50 mA g <sup>-1</sup>	2064/~1600	1616	30	[S7]

(1 C = 890 mA g<sup>-1</sup>)

## References

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