

Supporting Information

Oxygen Vacancy as Active Site for H₂S Dissociation on Rutile TiO₂ (110) Surface: A First-principles Study

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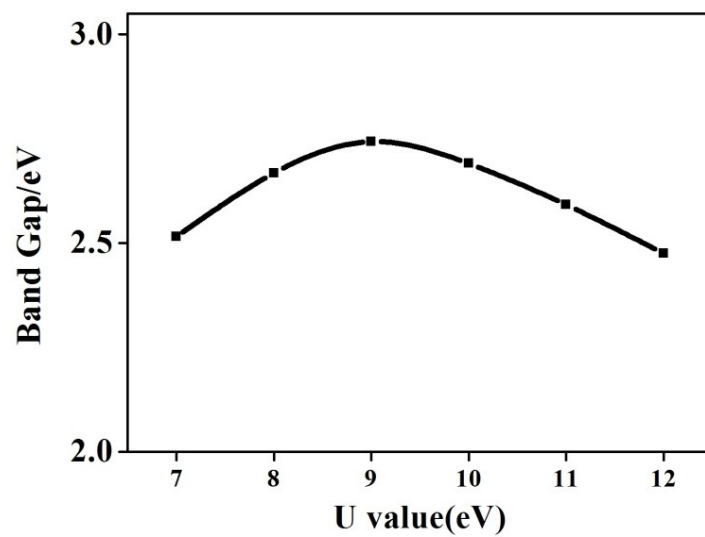


Fig. S1 Variation of band gaps with different U parameters for rutile TiO₂.

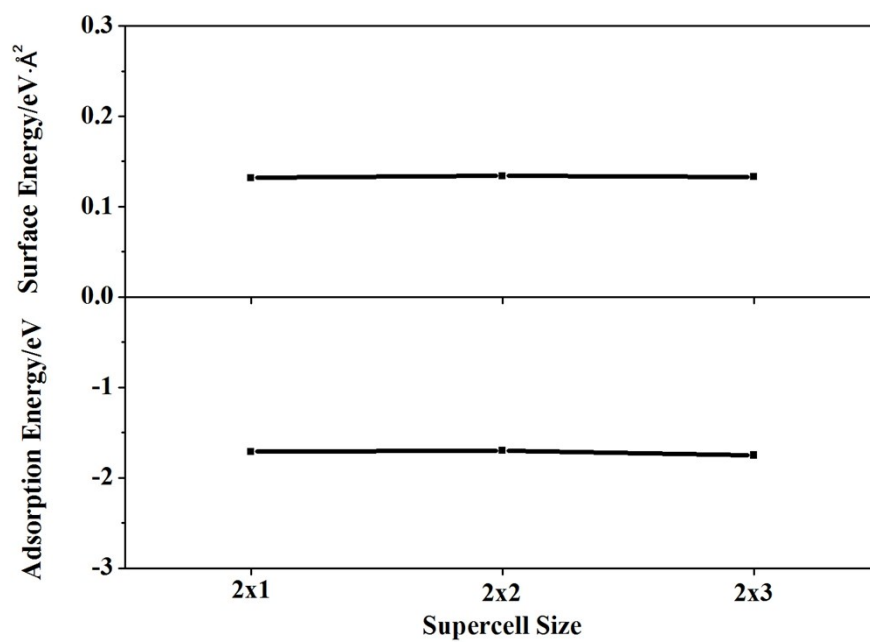


Fig. S2 Convergence tests of Surface energies and H₂S adsorption energies for the perfect surface, respectively.

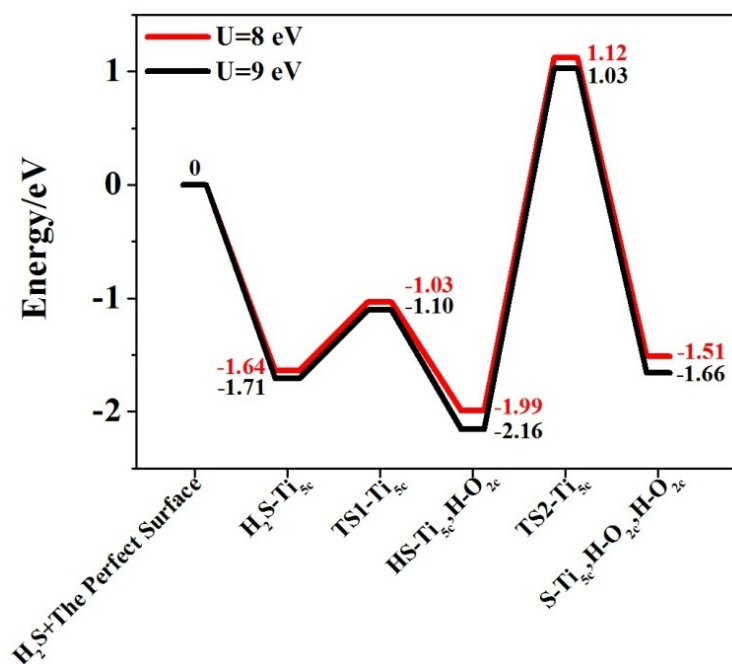


Fig. S3 Energy profiles for the dissociation of H₂S on the perfect surface calculated by different U parameters.

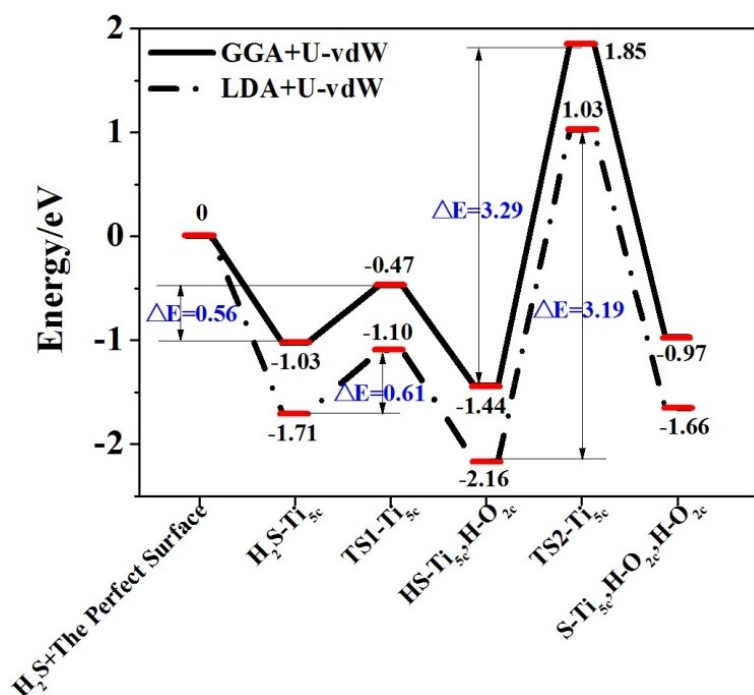


Fig. S4 Energy profiles for the H₂S dissociation on the perfect surface from dispersion corrected GGA+U and LDA+U, respectively.

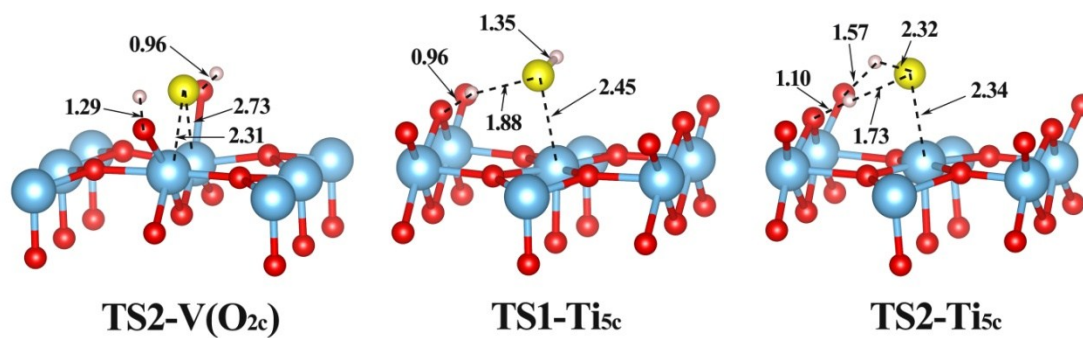


Fig. S5 Configurations of transition states. The bond length is in Å.

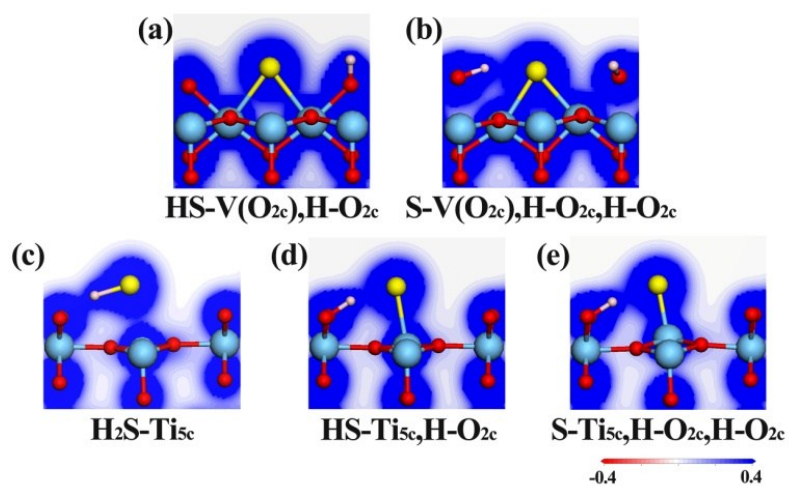


Fig. S6 The total electron densities of HS/H (a) and S/H/H (b) on the defected surface, and H₂S (c), HS/H (d) and S/H/H (e) on the perfect surface, respectively. The electron density is in electrons/Å³.