

SUPPORTING INFORMATION

**Possible interstellar formation of Glycine through a concerted mechanism: A
computational study on the reaction of CH₂=NH, CO₂ and H₂**

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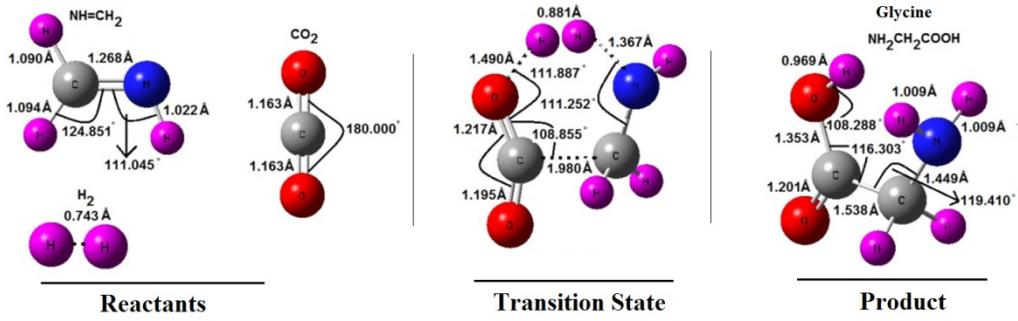


Figure S1: B3LYP/6-31G(3df,2pd) optimized geometries of the reactants, transition state and product.

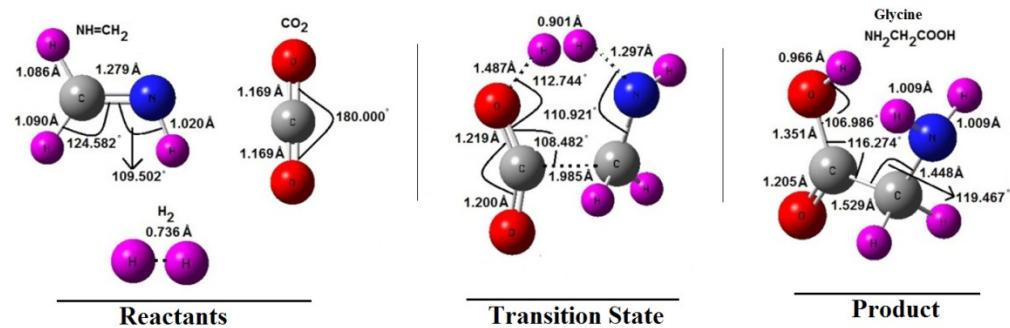


Figure S2: MP2/6-31G(3df,2pd) optimized geometries of the reactants, transition state and product.

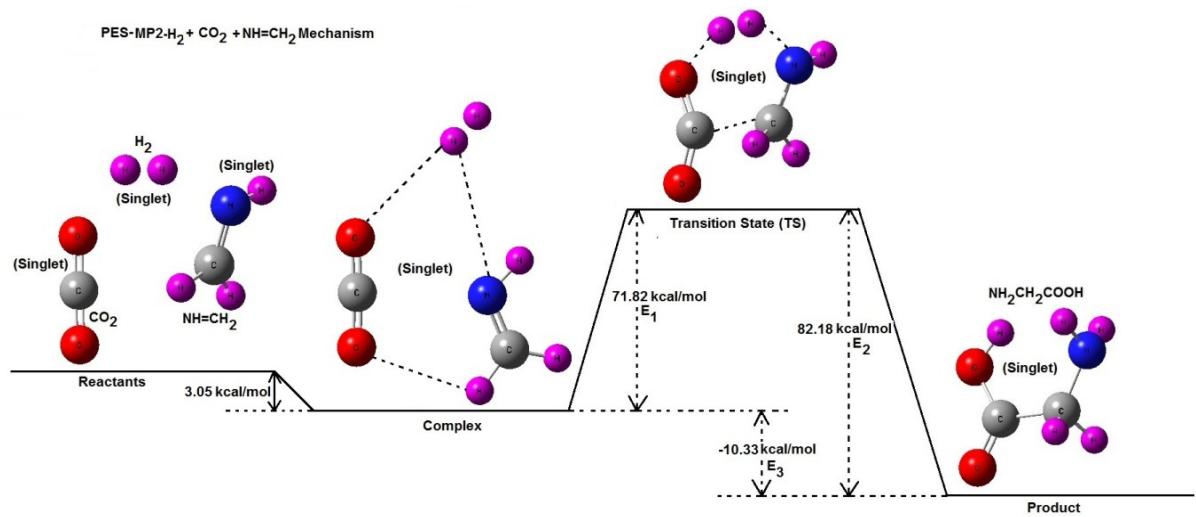
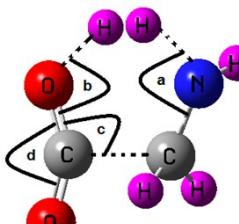


Figure S3: MP2/6-31G(3df,2pd) PES for the $\text{CH}_2=\text{NH} + \text{CO}_2 + \text{H}_2 \rightarrow \text{Glycine}$, reaction. All the energies are in kcal/mol and the diagram is not to scale.(Colour Code: Pink=Hydrogen, Grey=Carbon, Blue=Nitrogen, Red=Oxygen)

Table S1: Results of effect of various basissets (with B3LYP method) on the potential energy surface of the $\text{CH}_2=\text{NH} + \text{CO}_2 + \text{H}_2 \rightarrow \text{Glycine}$ reaction. Where $\Delta E_1 = E_{\text{TS}} - E_{\text{REACTANTS}}$, $\Delta E_2 = E_{\text{TS}} - E_{\text{PRODUCT}}$ and $\Delta E_3 = E_{\text{REACTANTS}} - E_{\text{PRODUCT}}$. All the energies are ZPE corrected and are in kcal/mol.

Basissets	ΔE_1	ΔE_2	ΔE_3
6-31G	62.0	86.4	24.4
6-31G*	69.8	76.5	6.7
6-31++G(d,p)	69.0	85.3	16.3
6-31++G(2df,2p)	70.8	80.2	9.4
6-31++G(3df,2pd)	70.4	80.1	9.7
6-31++G(3df,3pd)	70.3	80.1	9.8
aug-cc-pvdz	67.0	80.6	13.6
aug-cc-pvtz	71.3	79.8	8.5

Table S2: Effect of various methods on the geometry of the transition state for the reaction, $\text{CH}_2=\text{NH} + \text{CO}_2 + \text{H}_2 \rightarrow \text{Glycine}$. Some important geometric data are shown [Basisset used is 6-31++G(3df,2pd)].



Methods	Bond length	Bond angles
X3LYP	H-H = 0.880 Å	a = 111.334°
	N-H = 1.366 Å	b = 111.940°
	O-H = 1.490 Å	c = 108.819°
	C-C = 1.979 Å	d = 144.866°
B3PW91	H-H = 0.877 Å	a = 111.640°
	N-H = 1.367 Å	b = 112.315°
	O-H = 1.506 Å	c = 108.531°
	C-C = 1.981 Å	d = 145.674°
HF	H-H = 0.866 Å	a = 112.611°
	N-H = 1.394 Å	b = 110.478°
	O-H = 1.419 Å	c = 108.567°
	C-C = 1.957 Å	d = 145.622°
O3LYP	H-H = 0.878 Å	a = 111.337°
	N-H = 1.352 Å	b = 112.300°
	O-H = 1.525 Å	c = 118.587°
	C-C = 1.983 Å	d = 145.134°
MP2=full	H-H = 0.904 Å	a = 111.039°
	N-H = 1.285 Å	b = 113.148°
	O-H = 1.493 Å	c = 108.234°
	C-C = 1.982 Å	d = 146.184°
	H-H = 0.891 Å	a = 110.910°

CBS-QB3	N-H = 1.343 Å	b = 111.813°
	O-H = 1.489 Å	c = 108.681°
	C-C = 1.981 Å	d = 114.584°
G3B3	H-H = 0.881 Å	a = 109.682°
	N-H = 1.374 Å	b = 110.968°
	O-H = 1.500 Å	c = 110.082°
	C-C = 1.942 Å	d = 142.944°
G3MP2B3	H-H = 0.881 Å	a = 109.682°
	N-H = 1.374 Å	b = 110.968°
	O-H = 1.500 Å	c = 110.080°
	C-C = 1.942 Å	d = 142.944°

Table S3: Effect of various Basissets on the geometry of the transition state. Some important geometric data are shown (Method used is B3LYP).

Basissets	Bond length	Bond angles
6-31G	H-H = 0.882 Å	a = 110.552°
	N-H = 1.361 Å	b = 111.346°
	O-H = 1.557 Å	c = 110.468°
	C-C = 1.961 Å	d = 142.490°
6-31G(d)	H-H = 0.881 Å	a = 109.682°
	N-H = 1.374 Å	b = 110.968°
	O-H = 1.500 Å	c = 110.082°
	C-C = 1.942 Å	d = 142.944°
6-31+G(d)	H-H = 0.875 Å	a = 110.274°
	N-H = 1.378 Å	b = 111.352°
	O-H = 1.517 Å	c = 110.062°
	C-C = 1.946 Å	d = 143.080°
6-31++G(d)	H-H = 0.877 Å	a = 110.324°
	N-H = 1.375 Å	b = 111.415°
	O-H = 1.516 Å	c = 110.000°
	C-C = 1.948 Å	d = 143.158°
6-31++G(d,p)	H-H = 0.877 Å	a = 111.142°
	N-H = 1.365 Å	b = 111.733°
	O-H = 1.514 Å	c = 109.279°
	C-C = 1.974 Å	d = 144.127°
6-31++G(2df,2p)	H-H = 0.882 Å	a = 111.266°
	N-H = 1.357 Å	b = 111.934°
	O-H = 1.498 Å	c = 108.949°
	C-C = 1.976 Å	d = 144.519°
6-31++G(2df,2pd)	H-H = 0.882 Å	a = 111.317°
	N-H = 1.360 Å	b = 111.954°
	O-H = 1.493 Å	c = 108.824°
	C-C = 1.983 Å	d = 144.726°
6-31++G(3df,2pd)	H-H = 0.881 Å	a = 111.252°
	N-H = 1.367 Å	b = 111.887°
	O-H = 1.490 Å	c = 108.885°
	C-C = 1.980 Å	d = 144.701°
6-31++G(3df,3pd)	H-H = 0.880 Å	a = 111.286°
	N-H = 1.368 Å	b = 111.877°

	O-H = 1.489 Å	c = 108.838°
	C-C = 1.981 Å	d = 144.726°
aug-cc-pvdz	H-H = 0.890 Å	a = 111.651°
	N-H = 1.365 Å	b = 112.087°
	O-H = 1.514 Å	c = 109.056°
	C-C = 1.981 Å	d = 144.570°
aug-cc-pvtz	H-H = 0.885 Å	a = 111.509°
	N-H = 1.357 Å	b = 111.921°
	O-H = 1.486 Å	c = 108.710°
	C-C = 1.987 Å	d = 144.630°

Cartesian Coordinates of stationary points

[From B3LYP/6-31++G(3df,2pd) optimized geometries]

vdW-Complex

C	-2.203213	-0.336688	0.001473
N	-1.280195	0.551229	-0.010608
H	-1.611679	1.525427	-0.022085
H	-3.278983	-0.131151	0.000174
H	-1.907027	-1.383933	0.014057
H	0.595324	3.116494	0.039220
H	0.277356	3.788759	0.052167
C	1.234283	-0.378808	-0.000894
O	0.826328	-1.496510	0.012374
O	1.761166	0.686357	-0.013968

Transition State

C	-0.862934	-0.842765	-0.036846
H	-0.638687	-1.360943	-0.963695
H	-0.628966	-1.362549	0.886707
O	1.721333	-0.754475	0.014891
C	0.881010	0.094701	0.006462
O	0.682131	1.295036	0.013443
H	-0.768861	1.618030	-0.081691
N	-1.926940	-0.024242	-0.067513
H	-2.261122	0.169088	0.878724
H	-1.549954	1.269967	-0.291813

Methanimine ($\text{CH}_2=\text{NH}$)

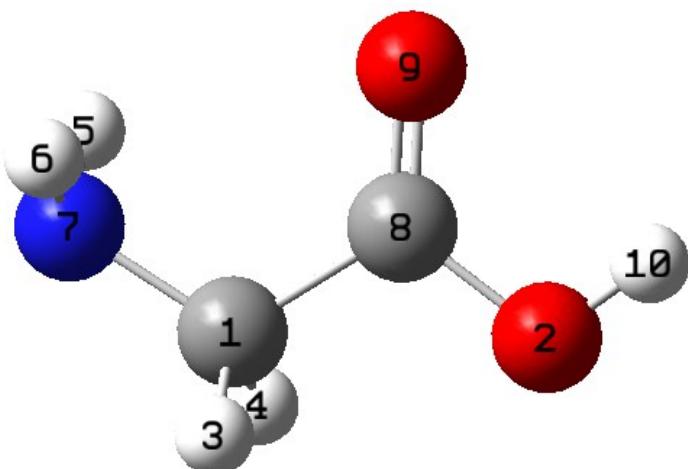
C	0.586885	0.028877	-0.000014
H	1.245720	-0.839259	0.000030
H	1.076099	1.007625	0.000028
N	-0.667824	-0.154020	0.000003
H	-1.168363	0.736515	0.000004

Product (Glycine)

C	0	0.745907	-0.750143	0.000094
O	0	-1.672056	-0.714322	-0.000069
H	0	0.764079	-1.412175	-0.869017
H	0	0.764093	-1.411864	0.869445
H	0	1.949029	0.692680	0.812616
H	0	1.949047	0.692359	-0.812971
N	0	1.931433	0.088161	-0.000058

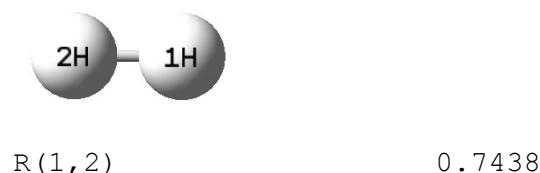
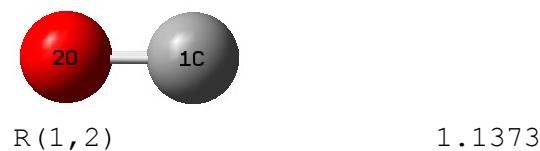
C	0	-0.634976	-0.097365	0.000002
O	0	-0.590688	1.257176	0.000038
H	0	-1.509914	1.564093	-0.000001

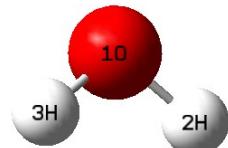
Optimized structure and Cartesian co-ordinates of the most stable structure of the glycine [B3LYP/6-31++G(3df,2pd)]:



C	-0.728457	-0.727536	0.000185
O	1.650900	-0.667486	-0.000017
H	-0.686102	-1.387480	0.870798
H	-0.685992	-1.387893	-0.870104
H	-2.022536	0.621328	-0.809568
H	-2.022468	0.621978	0.808951
N	-1.975150	0.012175	-0.000066
C	0.543433	0.113333	0.000110
O	0.590811	1.318687	-0.000135
H	2.419598	-0.077555	-0.000166

B3LYP/6-31++G(d,p) Optimized structures and the geometric parameters of the various stationary points (in the reaction 1 and 3 (for reaction 2, see the reference 35)

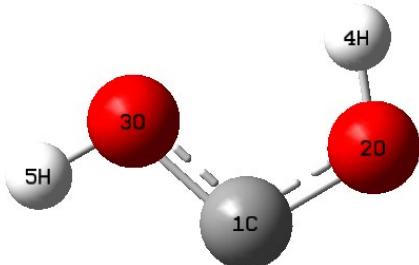




R (1, 2) 0.9653
 R (1, 3) 0.9653
 A (2, 1, 3) 105.7192

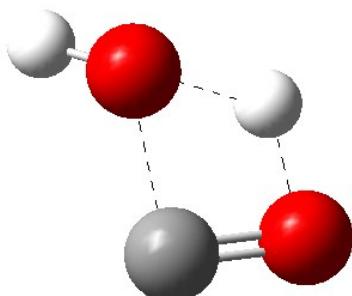


R (1, 2) 1.1694
 R (1, 3) 1.1694
 A (2, 1, 3) 180.0000



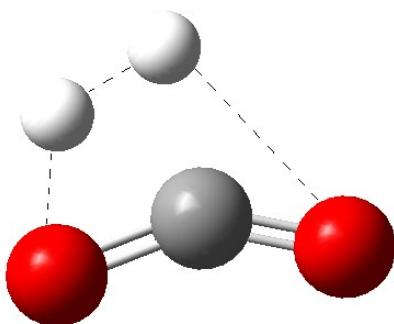
R (1, 2) 1.3132
 R (1, 3) 1.3436
 R (2, 4) 0.9855
 R (3, 5) 0.9679
 A (2, 1, 3) 107.2889
 A (1, 2, 4) 112.1525
 A (1, 3, 5) 107.9239
 D (3, 1, 2, 4) -0.0236
 D (2, 1, 3, 5) 180.0107

Transition state for CO + H₂O → Dihydroxy Carbene in Cartesian coordinates

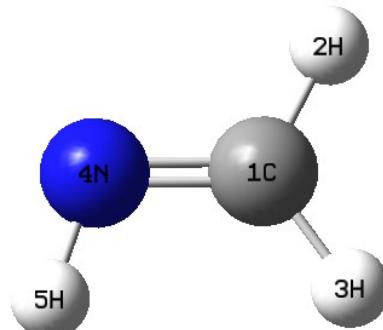


C	0.331554	0.724044	-0.009582
O	1.028719	-0.303161	0.024499
O	-1.053836	-0.138371	-0.092728
H	-0.038883	-0.895925	0.053308
H	-1.749501	0.083921	0.550012

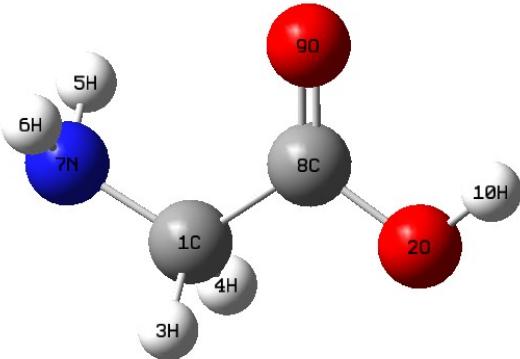
Transition state for $\text{CO}_2 + \text{H}_2 \rightarrow \text{Dihydroxy Carbene}$ in Cartesian co-ordinates



C	0.000000	0.154507	0.000000
O	-0.858268	-0.769092	0.000000
O	1.143441	0.438118	0.000000
H	-1.414272	0.426526	0.000000
H	-0.867110	1.294223	0.000000

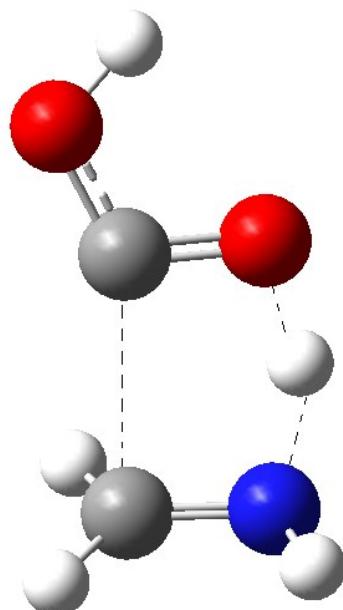


R (1, 2)	1.093
R (1, 3)	1.0985
R (1, 4)	1.272
R (4, 5)	1.0252
A (2, 1, 3)	116.1708
A (2, 1, 4)	118.8487
A (3, 1, 4)	124.9805
A (1, 4, 5)	110.9655
D (2, 1, 3, 4)	180.0
D (2, 1, 4, 5)	180.0
D (3, 1, 4, 5)	0.0



R (1, 3)	1.0965
R (1, 4)	1.0965
R (1, 7)	1.4489
R (1, 8)	1.5255
R (2, 8)	1.3566
R (2, 10)	0.9726
R (5, 7)	1.0163
R (6, 7)	1.0163
R (8, 9)	1.2125
A (3, 1, 4)	105.6019
A (3, 1, 7)	109.8916
A (3, 1, 8)	107.5308
A (4, 1, 7)	109.8917
A (4, 1, 8)	107.532
A (7, 1, 8)	115.8688
A (8, 2, 10)	107.4783
A (1, 7, 5)	110.6883
A (1, 7, 6)	110.6881
A (5, 7, 6)	106.4526
A (1, 8, 2)	111.5141
A (1, 8, 9)	125.709
A (2, 8, 9)	122.7769
D (3, 1, 7, 5)	-178.9936
D (3, 1, 7, 6)	63.2173
D (4, 1, 7, 5)	-63.203
D (4, 1, 7, 6)	179.008
D (8, 1, 7, 5)	58.9025
D (8, 1, 7, 6)	-58.8865
D (3, 1, 8, 2)	56.6567
D (3, 1, 8, 9)	-123.3429
D (4, 1, 8, 2)	-56.6458
D (4, 1, 8, 9)	123.3546
D (7, 1, 8, 2)	-179.9951
D (7, 1, 8, 9)	0.0054
D (10, 2, 8, 1)	-179.9986
D (10, 2, 8, 9)	0.001

Transition state for CH₂=NH + Dihydroxy Carbene → Glycine in Cartesian co-ordinates



C	1.559382	0.831178	0.114655
H	1.255435	1.191327	1.091298
H	1.792364	1.589295	-0.633570
N	1.842668	-0.436990	-0.035242
H	2.193751	-0.645546	-0.969736
C	-0.789193	0.223403	-0.068325
O	-0.511287	-1.004661	0.121062
O	-2.091138	0.495747	-0.088378
H	0.669602	-1.007492	0.164719
H	-2.611566	-0.324829	0.054532

Pre-reaction complex for CO₂ + H₂ → Dihydroxy Carbene in Cartesian co-ordinates



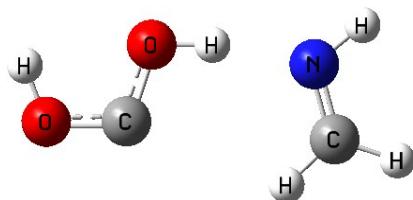
C	0.377868	-0.000850	-0.000039
O	-0.791581	-0.005607	-0.000119
O	1.547023	0.003856	0.000047
H	-4.527355	0.012111	-0.002479
H	-3.783386	0.006997	0.003295

Pre-reaction complex for CO + H₂O → Dihydroxy Carbene in Cartesian coordinates



C	0.984844	0.102639	0.006397
O	2.109682	-0.050449	-0.004060
O	-2.325829	-0.127510	-0.000966
H	-2.795790	0.715065	-0.010819
H	-1.384095	0.092773	0.012644

Pre-reaction Complex for CH₂=NH + Dihydroxy Carbene → Glycine in Cartesian co-ordinates



C	2.293978	-0.713912	0.000111
H	1.500801	-1.462698	-0.000956
H	3.333583	-1.057367	0.001456
N	1.957264	0.515099	-0.000215
H	2.754434	1.154577	0.001053
C	-1.146396	-0.420505	-0.000513
O	-0.731561	0.832590	-0.000012
O	-2.471699	-0.427343	0.000271
H	0.275487	0.832108	-0.000749
H	-2.824561	0.492213	0.001044