

Supporting Information

Geometric and electronic structures of the synthetic Mn₄CaO₄ model compound
mimicking the photosynthetic oxygen-evolving complex

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Contents

S1. Backgrounds

- S1.1 Historical developments of the broken-symmetry methods
- S1.2 More general broken-symmetry solutions and recoveries of the broken symmetries
- S1.3 Broken-symmetry solutions in the Kohn-Sham density functional theory
- S1.4 Spin Hamiltonian models for local spins in diradicals and polyyradicals
- S1.5 Effective procedures for construction of broken-symmetry solutions
- S1.6 Summary of the native OEC structure in the X-ray structures

Figure S1.1 Coordination bond lengths around Mn1 and Mn4

- S1.7 Comparison of Zhang model complex with several OEC models of PSII

Figure S1.2 Early structures of PSII-OEC

S2. Results of **1** calculated at the UB3LYP/DZVP level.

Figure S2.1 Superimposed structures of **1** optimized at the UB3LYP/DZVP level.

Figure S2.2 Localized natural orbitals of **1** in the $S_0(b)$ state

Figure S2.3 Localized natural orbitals of **1** in the $S_0(a)$ state

Figure S2.4 Localized natural orbitals of **1** in the S_1 state

Figure S2.5 Localized natural orbitals of **1** in the $S_2(R)$ state

Figure S2.6 Localized natural orbitals of **1** in the $S_2(C)$ state

Figure S2.7 Localized natural orbitals of **1** in the S_3 state

Figure S2.8 low-lying energy levels of **1** calculated at the UB3LYP/DZVP level.

Table S2.1 Bond valence sum calculation

Table S2.2 Local spin correlation factor of **1** calculated the J values at the UB3LYP/DZVP level.

Atomic coordinates of **1** (in XYZ file format) optimized at the UB3LYP/DZVP level in the S_{-1} , $S_0(a)$, $S_0(b)$, $S_1(DUUU)$, $S_1(UDUD)$, $S_2(R)$, $S_2(C)$, S_3 states

S3. Results of **1** calculated at the UBLYP/DZVP level.

Table S3.1 J values of **1** calculated at the UBLYP/DZVP level.

Table S3.2 Low-lying excitation states of **1** evaluated using the J values calculated at the UBLYP/DZVP level.

Figure S3.1 Coordination bond lengths around Mn1 and Mn4

Figure S3.2 Localized natural orbitals related to the Mn Jahn-Teller distortions.

Figure S3.3 Schematic illustration of the magnetic interactions of 1 calculated at the UBLYP/DZVP level.

S4. Results of **2** calculated at the UB3LYP/DZVP and UBLYP/DZVP level .

Table S4.1 Key bond lengths of the Mn₄CaO₄ cores of **2** calculated at the UBLYP/DXVP level.

Table S4.2 Mn-O bond lengths of the Mn₄CaO₄ cores of **2** calculated at the UBLYP/DXVP level.

Table S4.3 Calculated J values of **2** at the UB3LYP/DZVP (UB3LYP/DZVP) level.

Table S4.4 Low-lying energy levels of **2** calculated at the UB3LYP/DZVP (UBLYP/DZVP) level.

Table S4.5 Local spin correlation factor calculated at the UB3LYP/DZVP (UBLYP/DZVP) level.

Figure S4.1 Superimposed structures of **2**.

Figure S4.2 Localized natural orbitals of **2** in the S0 (b) state

Figure S4.3 Localized natural orbitals of **2** in the S0 (a) state

Figure S4.4 Localized natural orbitals of **2** in the S1 state

Figure S4.5 Localized natural orbitals of **2** in the S2 (R) state

Figure S4.6 Localized natural orbitals of **2** in the S2 (C) state

S5. Theoretical modeling

Figure S5.1 Molecular structures of model **C_a** and **C_b**.

Atomic coordinates of the theoretical models (models A, B, Ca, Cb in XYZ file format) optimized at the UB3LYP/DZVP in the highest spin states (*S*=7).

S6. Ca substitutions

Table S6.1 Mn-Mn distances of the Ca substituted models

Table S6.2 Mn-O distances of the Ca substituted models

Figure S6.1 Superimposed views of the Ca substituted models (Mn₄AO₄, A=Br, Sr, Ca, Mg, Zn and Mn).

S1. Backgrounds

S1.1 Historical developments of the broken-symmetry methods

According to the suggestion of one of the referees, historical developments of the broken-symmetry (BS) methods are briefly reviewed in the supporting material. In early 1970th the extended Huckel molecular orbital (EHMO) method was useful and handy for theoretical studies of organic molecules. The EHMOs are characterized by the point group symmetry (P_n). Spatial symmetries of EHMOs are theoretical foundations of orbital-symmetry conservation rules for concerted reactions by Woodward-Hoffmann and frontier molecular orbital (HOMO, LUMO) theories for chemical reactions by Fukui. However the EHMO method was insufficient for theoretical studies on spins in molecules.

The Hartree-Fock (HF) MO method takes into account for electron spins in molecules. The spin-restricted Hartree-Fock (RHF) solution is usually reliable for MO description of closed-shell electronic states consisted of even number of electrons. Spatial symmetries of RHF MOs are also theoretical foundations for the Woodward-Hoffmann rules and frontier molecular orbital theories. The HF method utilizes one Slater determinant for many particle (electron) systems, indicating one of the mean field (independent particle) theories. Therefore the HF method permits a bifurcation of the closed-shell singlet pair into the open-shell singlet diradical pair in a certain condition: this condition is referred to as the instability condition (ref. s1). Indeed, the closed-shell RHF solution often suffers from the triplet instability when the HOMO-LUMO energy gap becomes smaller than the on-site Coulomb repulsion integral (U) (ref. s2). The symmetry-adapted molecular orbitals of the unstable RHF solution are inevitably reorganized into the unrestricted Hartree-Fock (UHF) molecular orbitals through the HOMO-LUMO mixing (ref. s2) to reduce the effect of the electron repulsion (U) as follows:

$$\begin{cases} \psi_1^\alpha = \cos\theta \psi_+ + \sin\theta \psi_- \\ \psi_1^\beta = \cos\theta \psi_+ - \sin\theta \psi_- \end{cases} . \quad (1.1)$$

where the ψ_+ and ψ_- denote the HOMO and LUMO, respectively.

The HOMO and LUMO have usually different special symmetries such as symmetric (S) and anti-symmetric (A). Therefore the alpha and beta spatial orbitals, ψ_1^α and ψ_1^β , obtained by the HOMO-LUMO mixing inevitably entailed the breakdown of the special symmetry of the molecular orbitals in the UHF solution (ref. s2). Thus the UHF MOs are spatially symmetry-broken in sharp contrast to the RHF orbitals. The HOMO-LUMO crossing often occurs in the course of the symmetry-forbidden

reactions. Therefore the HOMO-LUMO mixing occurs in the orbital crossing region, providing the broken-symmetry (BS) UHF MOs that are responsible for diradical intermediate generated in symmetry-forbidden reactions in organic chemistry. Thus the origin of our concept of the orbital symmetry breaking was closely related to the necessity of the MO-theoretical description of symmetry-forbidden diradical reactions in 1970th (refs. s2 and s8).

The UHF MOs are mainly localized on different spatial region, providing a localized MO picture of diradical electrons, namely local spins, for which chemical notation (\cdot) is often used. Therefore the localized molecular orbitals (LMO) are naturally introduced as the maximum HOMO-LUMO mixing orbitals ($\theta = \pi/4$ in eq. (1.1)) (ref. 17) as

$$\begin{aligned}\psi_+(LMO) &= 1/\sqrt{2}(\psi_+ + \psi_-) \\ \psi_-(LMO) &= 1/\sqrt{2}(\psi_+ - \psi_-)\end{aligned}\quad (1.2)$$

For example, the LMOs are similar to the atomic 1s orbitals for a dissociating hydrogen molecule: $H_a \cdot \dots \cdot H_b$ since the $\psi_+(LMO)$ and $\psi_-(LMO)$ are largely localized on the x- and y hydrogen atoms, respectively. However, they are not identical to the atomic 1s orbitals in the valence bond (VB) model (see eq. (1.14)) except for the dissociation limit since they have small tails at the y- and x-sites, respectively (ref. 17). On the other hand, the LMOs are orthogonal because of this characteristic property, providing a useful LMO basis for the configuration interaction (CI) approach (ref. 18) starting from the BS MO calculations. The LMO CI provided chemical pictures based on the LMOs instead of the VB model. To this end, the BS MOs in eq. (1.1) are rewritten by the LMOs as

$$\left\{ \begin{array}{l} \psi_1^\alpha = \cos\omega \psi_+(LMO) + \sin\omega \psi_-(LMO) \\ \psi_1^\beta = \cos\omega \psi_-(LMO) + \sin\omega \psi_+(LMO) \end{array} \right. \quad (1.3)$$

where the orbital mixing parameter ω is given by $(\theta + \pi/4)$.

The RHF and UHF solutions given by the single Slater determinant can be expanded with the LMOs to obtain the VB picture. For example, the UHF solution is given by

$$|\Psi_{UHF}\rangle = |\psi_1^\alpha \overline{\psi_1^\beta}\rangle = \cos\omega^2 |\psi_+(LMO)^\alpha \overline{\psi_-(LMO)}^\beta\rangle + \cos\omega \sin\omega \{ |\psi_+(LMO)^\alpha \overline{\psi_+(LMO)}^\beta\rangle +$$

$$|\psi_-(LMO)^\alpha \overline{\psi_-(LMO)}^\beta\rangle \}$$

$$= \Phi_S + \cos 2\omega \Phi_T + 1/2 \sin 2\omega \{ \Phi_{ZWa} + \Phi_{ZWb} \} \quad (1.5a)$$

where the pure singlet (S), triplet (T) and zwitterionic (ZW) configurations are defined

by using the LMOs as

$$\begin{aligned}\Phi_S &= 1/\sqrt{2}(\left|\psi_+(LMO)^\alpha \overline{\psi_-(LMO)}^\beta\right\rangle + \left|\psi_-(LMO)^\alpha \overline{\psi_+(LMO)}^\beta\right\rangle) \\ \Phi_T &= 1/\sqrt{2}(\left|\psi_+(LMO)^\alpha \overline{\psi_-(LMO)}^\beta\right\rangle - \left|\psi_-(LMO)^\alpha \overline{\psi_+(LMO)}^\beta\right\rangle)\end{aligned}\quad (1.6)$$

$$\Phi_{ZWa} = \left|\psi_+(LMO)^\alpha \overline{\psi_+(LMO)}^\beta\right\rangle, \Phi_{ZWB} = \left|\psi_-(LMO)^\alpha \overline{\psi_-(LMO)}^\beta\right\rangle \quad (1.7)$$

The UHF solution involves the pure triplet component as shown in eq. (1.5a), indicating the mixing of the singlet and triplet states, namely the spin symmetry breaking. In fact, the half and half mixing occurs at the dissociation limit $\omega=0$.

$$|\Psi_{UHFI}(\omega=0)\rangle = \Phi_S + \Phi_T \quad (1.8)$$

Such symmetry breaking property of the UHF solution was an origin of the critical opinion for the BS method at that time (1970th).

Quantum mechanics (QM) however teach us to construct the other UHF solution by using the BS orbitals in eq. (1.1) as follows:

$$\begin{aligned}|\Psi_{UHFII}\rangle &= \left|\psi_1^\beta \overline{\psi_1^\alpha}\right\rangle = \cos\omega^2 \left|\psi_-(LMO)^\alpha \overline{\psi_+(LMO)}^\beta\right\rangle + \cos\omega \sin\omega \left\{ \left|\psi_-(LMO)^\alpha \overline{\psi_-(LMO)}^\beta\right\rangle + \right. \\ &\quad \left. \left|\psi_+(LMO)^\alpha \overline{\psi_+(LMO)}^\beta\right\rangle \right\} + \sin\omega^2 \left|\psi_+(LMO)^\alpha \overline{\psi_-(LMO)}^\beta\right\rangle \\ &= \Phi_S - \cos 2\omega \Phi_T + 1/2 \sin 2\omega \{\Phi_{ZWB} + \Phi_{ZWa}\}\end{aligned}\quad (1.5b)$$

The UHFI and UHFII solutions are degenerated in energy, and therefore the quantum-mechanical (QM) resonance between them should occur in the finite diradical systems without the magnetic long-range order, providing the pure singlet and triplet states as follows:

$$\begin{aligned}|\Psi_S\rangle &= 1/\sqrt{N_S}(|\Psi_{UHFI}\rangle + |\Psi_{UHFII}\rangle) = 1/\sqrt{N_S}(2\Phi_S + \sin 2\omega\{\Phi_{ZWB} + \Phi_{ZWa}\}) \\ |\Psi_T\rangle &= 1/\sqrt{N_T}(|\Psi_{UHFI}\rangle - |\Psi_{UHFII}\rangle) = \Phi_T\end{aligned}\quad (1.9)$$

where the N_S and N_T denote the normalizing factors (ref. 18). Thus the resonating BS solutions recover the QM requirement (no symmetry breaking), providing the pure singlet and triplet states of diradical states. The pure singlet state in eq. (1.9) can be also obtained by removing out the triplet component from eq. (1.5a or 1.5b), leading to the spin projection (SP) procedure for recovering the spin-symmetry breaking in the BS UHF solution. It is noteworthy that the SP BS solution involves mainly the covalent

term (Φ_s), together with non-negligible zwitterionic terms (Φ_{ZW_a} and Φ_{ZW_b}) in the LMO CI by the use of two LMOs in eq. (1.2) (ref. 18).

The configuration interaction (CI) by the use of the LMOs has been performed to confirm scope and reliability of the spin projected BS UHF solutions in ref. 18 and related papers in 1970th. It was found that the LMO CI wavefunction for the singlet ground state mainly involves the singlet wavefunction, together with small correction terms, as

$$\begin{aligned}\Phi_{LMO-CI}^S &= 1/\sqrt{N_s} (\langle \Psi_s \rangle + (\text{small corrections})) \\ \Phi_{LMO-CI}^T &= 1/\sqrt{N_T} (\langle \Psi_T \rangle + (\text{small corrections}))\end{aligned}\quad (1.10).$$

Thus the BS approach followed by the spin projection was applicable to MO-theoretical descriptions of the singlet and triplet diradical ground states. On the other hand, the LMO CI indicated that the excited singlet states are mainly described by the resonating states of the ZW_a and ZW_b configurations. The above results were our early BS approaches to diradical molecules in 1970th.

The above LMO CI scheme for diradical systems was extended for many electron systems, namely polyyradicals. To this end, the natural orbital (NO) analysis of the BS solutions was performed to determine the natural molecular orbitals (NO) and their occupation numbers. The active natural orbital space (CAS) for polyyradicals are easily selected by using their occupation numbers, leading to the CAS CI for confirmation of the projected BS solutions for the species. The multi-reference (MR) CI by the use of the UHF-NO (=UNO) CAS CI reference was performed for more complex polyyradicals systems (ref. s3).

The multiple metal-metal bonds were topic in inorganic chemistry in late 1970th. The triplet instability condition (ref. s1) for the HF solution was also applicable to the Hartree-Fock-Slater (HFS) solutions for the weak metal-metal bonds such as the δ - δ bond of binuclear transition metal complexes with formal quadruple bonds as shown in ref. 19. The HOMO-LUMO mixing in the HFS solution provided the BS orbitals in eq. (1.1) and localized MOs of the HFS solutions like in the case of eq. (1.3). Therefore the same LMO CI-type expansions of the unrestricted HFS (UHFS) solutions were also feasible as shown in ref. 20. However LMO CI was too difficult for large transition-metal clusters at that time. Therefore the many-body perturbations starting from the UHFS solutions were also proposed in relation to the inclusion of dynamical correlations in ref. 20. Nowadays, the double hybrid DFT involving the second-order perturbations (MP2) was developed. Noddleman utilized the localized MOs for analysis of the BS X α solutions developed by Slater and Johnson in ref. 21.

Our LMO CI approach based on UHF and UHFS crossed the Noddeman's approach near 1980 because the UHFS solution is closely related to the BS X α solution.

In 1970th, generalized valence bond (GVB) method (ref. s4) was exclusively used for theoretical investigations of diradical species. Therefore Noddelman (ref. 21) used a terminology: X α valence bond (VB) model for analysis of the BS X α solution in conformity with the GVB theory. However in 1980th, it was found that the GVB perfect-pairing (PP) method like the VB model could not properly describe the dissociation process of singlet metal-carbene double bond (M=CH₂) into two triplet fragments



The mixing between d π -p π HOMO and d π -p π^* LUMO occurs in an intermediary region, providing BS π -orbitals localized mainly on M and O(CH₂) sites, respectively. Similarly the mixing between d σ -p σ HOMO and d σ -p σ^* LUMO occurs in the dissociation region, providing BS σ -type orbitals localized mainly on M and O(CH₂) sites, respectively. Finally two triplet fragments are generated at the dissociation limit. Therefore GVB PP model was broken, leading to the necessity of GVB CI to describe this type of reactions, for example reactions of transition metal oxo complexes. In order to recover the spin symmetry, a kind of spin optimized (SO) procedure (ref. s11) is required for the VB approach by the use of non-orthogonal orbitals. On the other hand, the full CI by the use of complete active space (CAS) is applicable for refinements of the BS solutions if the CAS were constructed by the orthogonal natural orbitals by the BS solutions as shown in ref. s3. Thus we have developed our own UHF-NO (UNO) CI approach for seamless extension from BS (UHF, UHFB) method to symmetry adapted (SA) post BS methods in general.

S1.2 More general broken-symmetry solutions and recovery of the broken symmetries

In the 1960's, the theory of symmetry breaking has been developed in the elementary particle physics by Y. Nambu and in the solid state physics (magnetism) by Nagamiya, Kanamori and Yoshimori in Osaka. Yoshimori discovered the non-collinear helical (rotating) spin structure for the Mn-oxide in 1959 on the basis of the classical Heisenberg model. One of the authors (K.Y) applied the classical spin vector models for radical clusters such as the equilateral H₃ and tetrahedral H₄ radicals, proposing the triangular (two dimension (2D)) and tetrahedral (three dimension (3D)) spin structures in ref. s5 although the one-dimension (1D) up- and down-spin solutions

in eq. (1.1) were popular in quantum chemistry at that time. The spin structures were characterized by using the magnetic group constructed by the spin rotation (S) and time-reversal (T) symmetries; note that spin is an angular momentum, and therefore the time-reversal (T) operation is necessary for its inversion.

In the field of the theoretical physics, Fukutome developed a general theory of the unrestricted Hartree-Fock (UHF) solutions on the basis of the group-theoretical classification by the use of spin rotation (S) and time-reversal (T) symmetry (ref. s6). The real and complex general spin orbitals (GSO), namely two-component spinors, were introduced, respectively, to describe the 2D and 3D spin structures under the UHF approximation. Later Osaki generalized the Fukutome classification of the UHF solutions by considering the spatial (point group) symmetry (P_n), providing a general UHF theory characterized by the $P_n \times S \times T$ symmetry operations. Furthermore Ozaki developed the Hartree-Fock-Bogoliubov (HFB) theory by considering the gauge symmetry (ϕ) that describes the superconductivity in general (ref.s7). Therefore general broken-symmetry UHF and unrestricted HFB (UHFB) solutions are characterized by using these symmetry operations in infinite systems, for which various types of phase transitions, namely the long-range orders, are feasible.

We were interested in strong electron correlations effects that entail several types of broken symmetry UHF and UHFB solutions even in finite systems where short-range static correlations (not phase transitions) play significant roles in chemistry. The spin correlation functions were introduced to express short-range spin correlations in refs. s8 and s9, elucidating important roles of spin densities in 1D, 2D and 3D UHF solutions (mathematical derivations are abbreviated here). However, recoveries of broken-symmetries are necessary for quantitative purposes like in the case of eq. (1.9) for diradicals. As an example, we have examined the recovery of the broken-symmetries for the equilateral H_3 radical (a tri-radical) that are described by the GSO under the UHF approximation in ref. s10. We have examined the permutation symmetry (S_N) for recovery of spin symmetry, namely the total spin angular momentum $\langle S^2 \rangle$ and the S_z -component of the total wavefunction. The magnetically ordered GSOs for the H_3 radical were also characterized by the time-reversal (T), spin rotation (S) and spatial symmetry (P_n). The total energies of the broken-symmetry (BS) UHF, spin-projected UHF and extended Hartree-Fock (EHF) solutions by the use of the GSOs were obtained analytically by using the Hubbard model, providing the potential curves for the dissociation of the H_3 radical into 3H atoms in ref. s10. The key concept was emerged from these computations, indicating

that the BS solutions provide qualitatively reliable potential curves for the dissociation of covalent bonds.

We have examined the symmetry breaking and recoveries of broken-symmetries in general in ref. s11 by using the symmetry operations, namely S_N , T, S, Pn. These symmetry operations elucidated the interrelationships between several model Hamiltonians used in the field of chemistry as follows:

- a) Huckel and extended Huckel models (spatial symmetry characterized by the point group (Pn))
- b) Classical Heisenberg model (magnetic group, spin rotation (S) x time reversal (T) symmetry)
- c) Quantum Heisenberg model (exchange interaction, permutation group (S_N))
- d) Generalized Hartree Fock and Hartree Fock Slater models (magnetic double group, Pn x S x T)
- e) Extended Hartree Fock model (a unified theory: $S_N \times Pn \times S \times T$).

The number density projection of the UHFB solutions was necessary for the recovery of numbers of electrons in finite systems such as fragments of molecular superconductors. Thus we were concerned with the symmetry breaking and recovery of broken symmetry in general before the crossing with the Noodelman' approach based on the X α model in ref. 21.

S1.3 Broken-symmetry solutions in the Kohn-Sham density functional theory

The density functional theory (DFT) is the exact theory in principle, and therefore the broken-symmetry (BS) solution should not appear in finite systems without phase transitions. In fact, Becke still stands on this basic position, proposing the symmetry-adapted B13 functional for DFT approaches to strongly correlated electron systems (SCES). On the other hand, the Kohn-Sham (KS) equation permits broken-symmetry (BS) solutions characteristic of the mean field approximation in finite systems with strong electron correlations. Therefore the group-theoretical characterizations mentioned above (section 1.2) are equally applicable to BS KS-DFT solutions and BS hybrid DFT (HDFT) solutions obtained by the mixing of the BS KS DFT and BS HF solutions like in the case of the UB3LYP method (ref. s12). In fact, the natural orbital analysis of the BS HDFT solutions provided the natural molecular orbitals for many electron systems as an extension of LMO in eq. (1.2). Here we do not repeat details. The details are given in our early (ref. s3, s10, s11) and recent papers (refs. 23-25).

S1.4 Spin Hamiltonian models for local spins in diradicals and polyyradicals

As shown in sections 1-3, several BS solutions derived from triplet (ref. s1) and

spin-flip (ref. s6) instabilities provide local spins for strongly correlated electron systems (SCES), for which the temperature-dependent magnetism was often observed by the magnetic susceptibility measurements. The observed magnetism of local spins has been investigated on the basis of the Heisenberg spin Hamiltonian model (ref. s4) that involves important parameters (J) responsible for effective exchange interactions between local spins. The newly appeared magnetic energy levels may be regarded as an indication of quasi-phase transitions like in the case of the newly appeared rotational spectra by deformations of nuclear matters. The Bohr-Mottelson theory (ref. s13) for the nuclear deformation was a guiding principle for us at that time (1970th) since the nuclear matters are finite systems like magnetic clusters in chemistry. Thus the temperature-dependent magnetism is a key evidence for existence of local spins in finite systems.

In analogy with the rotational energy levels, the eight energy levels of the $\pi\pi$, $\sigma\pi$, $\pi\sigma$ and $\sigma\sigma$ diradical configurations of HNOO radical were first obtained as an example of newly appeared spatial and spin-symmetry broken UHF solutions (ref. s14). The successive UHF-NO(UNO) CI calculations were also performed to elucidate singlet and triplet energy gaps for four different diradical structures (ref. s15). For the evaluation of the magnetic interactions (effective exchange J values), three different formulations have been proposed at that time;

$$J_{xy}^{(G)} = \frac{LS_{EX} - HS_{EX}}{S_{max}^2}, \quad (1.12a)$$

$$J_{xy}^{(N)} = \frac{LS_{EX} - HS_{EX}}{S_{max}(S_{max}+1)}, \quad (1.12b)$$

$$J_{xy}^{(Y)} = \frac{LS_{EX} - HS_{EX}}{HS_{\langle S^2 \rangle_X} - LS_{\langle S^2 \rangle_X}}, \quad (1.12c)$$

by Ginsberg [s16], Noddleman [s17,s18] and ours [s15,s19-s21], respectively. Our UHF results for HNOO (ref. s14) and UNO-CI results for CH₂OO correspond, respectively, to the Ginsberg scheme (1.12a) and our later scheme (1.12c). These equations are applicable to local spins at sites x and y. The eq. (1.12c) is the approximate spin projection (AP) scheme, and ${}^Y E_X$ and ${}^Y \langle S^2 \rangle_X$ denote the total energy and total squared-magnitude of total spin at the spin state Y by several BS methods X (X=UHF, UMP, UCC, UDFT). The AP procedure is derived from the approximate spin contamination correction of the low spin (BS) state using the Heisenberg spin Hamiltonian ($H = -2J_{x,y}s_x \cdot s_y$) to reproduce the UNO CI results approximately; the triplet component in eq. (1.5a(b)) is replaced with the high-spin (HS) UHF solution for approximate spin projection. On the other hand, the $J^{(G)}$ and $J^{(N)}$ formulations

correspond to the limits of broken diradical bonds and strong covalent bonds, respectively. The ${}^L\text{S}\langle S^2 \rangle_X$ values at these limits are given by

$$\begin{aligned} {}^L\text{S}\langle S^2 \rangle_X &= S_{\max} \quad \text{for broken diradical bonds,} \\ {}^L\text{S}\langle S^2 \rangle_X &= 0 \quad \text{for strong covalent bonds,} \\ {}^H\text{S}\langle S^2 \rangle_X &= S_{\max}(S_{\max}+1). \end{aligned} \quad (1.13)$$

Therefore our scheme in eq. (1.12c) covers both the limits, and moreover it is equally applicable to symmetry-adapted methods such as CAS-CI, CASSCF, UNO(ULO) CI, MR CI (refs. s15 and s19-s21). Many other computational schemes for J values are derived from eq. (1.12c). The total energy of the LS solution after AP is also given by using the J values as

$${}^{AP-LS}E_X = {}^L\text{S}E_X + J_{xy}^{(Y)}({}^L\text{S}\langle S^2 \rangle_X - 0.0) \quad (1.14)$$

where ${}^L\text{S}\langle S^2 \rangle$ becomes 0.0 for the exact singlet state although the BS methods X provide non-zero values.

These proposed schemes are, however, limited to two-site systems. For multispin system, we proposed a generalized spin projection (GP) scheme in ref. 28, which is a natural extension of the AP scheme, and properly corresponding to the Heisenberg spin Hamiltonian ($H = \sum_{<x,y>} -2J_{x,y} s_x \cdot s_y$). In GP scheme, J values are evaluated as

$$J_{xy}^{(GP)} = \frac{{}^F\text{E}_X - {}^{AF}\text{E}_X}{-2({}^F\langle s_x \cdot s_y \rangle_X - {}^{AF}\langle s_x \cdot s_y \rangle_X)}, \quad (1.12.d)$$

The energy correction scheme after GP was also derived in ref. 28. In this paper we have used the GP scheme for spin projection for UB3LYP solutions for the Mn_4CaO_4 cluster. Thus our computational formula are applicable for both BS and post BS (symmetry adapted) methods because of symmetry projections in contrast to those of other groups.

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S1.5 Effective procedures for construction of broken-symmetry solutions

Broken-symmetry (BS) solutions have been constructed by the HOMO-LUMO mixing procedures in eq. (1.1). However, the SCF processes to obtain the low-spin (LS) BS solutions often exhibit slow convergence or divergence in the case of multi-nuclear transition metal complexes. On the other hand, the highest-spin (HS) solutions are obtained more easily by conventional SCF procedures. Therefore constructions of the trial orbitals for the LS states by using the HS orbitals are practical and convenient for the rapid convergence as shown in ref. 27. To this end localized natural orbital (LNO) transformations were utilized to make good broken-symmetry initial guesses and to clearly depict the Mn d orbitals, which are very variable to understand the Jahn-Teller distortions. Since the theoretical background and their specific procedures have been described in ref. 27, only the concept and supplemental statements for 1 are given here.

In the two-spin 1/2 system, the bonding and antibonding orbitals are easily obtained by the HS triplet solution as

$$\left\{ \begin{array}{l} \psi_+ = \frac{1}{\sqrt{2(1+S)}}(\phi_1 + \phi_2) \\ \psi_- = \frac{1}{\sqrt{2(1-S)}}(\phi_1 - \phi_2) \end{array} \right. \quad (1.15)$$

where ϕ_i is the i -th site atomic orbital and $S = \langle \phi_1 | \phi_2 \rangle$ denotes the overlap integral between them. The broken-symmetry (BS) orbitals are constructed of the mixing of these HS orbitals as shown in eq. (1.1). Using these orbitals, the BS wave function for the low-spin (LS) singlet configuration is given by (see also eq. (1.4a))

$$|\Psi_{\text{BS}}\rangle = |\psi_+^\alpha \psi_-^\beta\rangle \quad . \quad (1.16)$$

The total $\langle S^2 \rangle$ value for the LS BS solution are given by

$$\langle S^2 \rangle_{\text{BS}} = 1 - T^2 \quad (1.17)$$

where T denotes the overlap between the BS orbitals that are non-orthogonal.

$$T = \langle \psi_1^\alpha | \psi_1^\beta \rangle = \cos 2\theta. \quad (1.18)$$

In two-spin systems, natural orbitals are same as their molecular orbitals of eq. (1.1). Interestingly, the orbital localization procedure employed here performs the similar orbital mixing procedure as,

$$\begin{cases} \psi_{\text{LNO1}} = \cos \gamma \psi_+ + \sin \gamma \psi_- \\ \psi_{\text{LNO2}} = -\sin \gamma \psi_+ + \cos \gamma \psi_- \end{cases}, \quad (1.19)$$

though θ and γ are different in general. However, in two-spin 1/2 systems, $\gamma = \pi/4$ is expected for LNOs, these LNO can be utilized as good BS trials (initial guesses) by

$$|\Psi_{\text{BS}}^{\text{LNO}}\rangle = |\psi_{\text{LNO1}} \bar{\psi}_{\text{LNO2}}\rangle. \quad (1.20)$$

It is noted that the initial guess $|\Psi_{\text{BS}}^{\text{LNO}}\rangle$ is not always the same as the true BS state $|\Psi_{\text{BS}}\rangle$. The similarity is calculated to be

$$|\langle \Psi_{\text{BS}} | \Psi_{\text{BS}}^{\text{LNO}} \rangle| = \sin(\theta + \pi/4) \quad (1.21)$$

Therefore, if the spin center is completely localized ($\theta = \pi/4$), the LNO guess is equal to the true BS state. Most BS solutions exist near the localized condition ($\theta \approx \pi/4$), the LNO guesses are expected to be good BS trials (initial guesses). Another important point is that their $\langle S^2 \rangle_{\text{BS}}^{\text{LNO}}$ and $T_{\text{BS}}^{\text{LNO}}$ values can be calculated as

$$\langle S^2 \rangle_{\text{BS}}^{\text{LNO}} = \langle \Psi_{\text{BS}}^{\text{LNO}} | S^2 | \Psi_{\text{BS}}^{\text{LNO}} \rangle = 1 \quad (1.22a)$$

$$T_{\text{BS}}^{\text{LNO}} = \langle \psi_{\text{LNO1}} | \psi_{\text{LNO2}} \rangle = 0, \quad (1.22b)$$

because the LNOs are orthogonalized to each other.

The HS solutions can be obtained for many electron systems. The natural orbital (NO) analyses of the HS solutions are also feasible to elucidate the orthogonal natural orbitals and their occupation numbers for these systems. Therefore the above localization procedures can be applicable for several bonding and antibonding natural orbital pairs. In general, eq. (1.22a) becomes

$$\langle S^2 \rangle_{\text{BS}}^{\text{LNO}} = \langle \Psi_{\text{BS}}^{\text{LNO}} | S^2 | \Psi_{\text{BS}}^{\text{LNO}} \rangle = S(S+1) + N_{\text{LNO},\beta}, \quad (1.23)$$

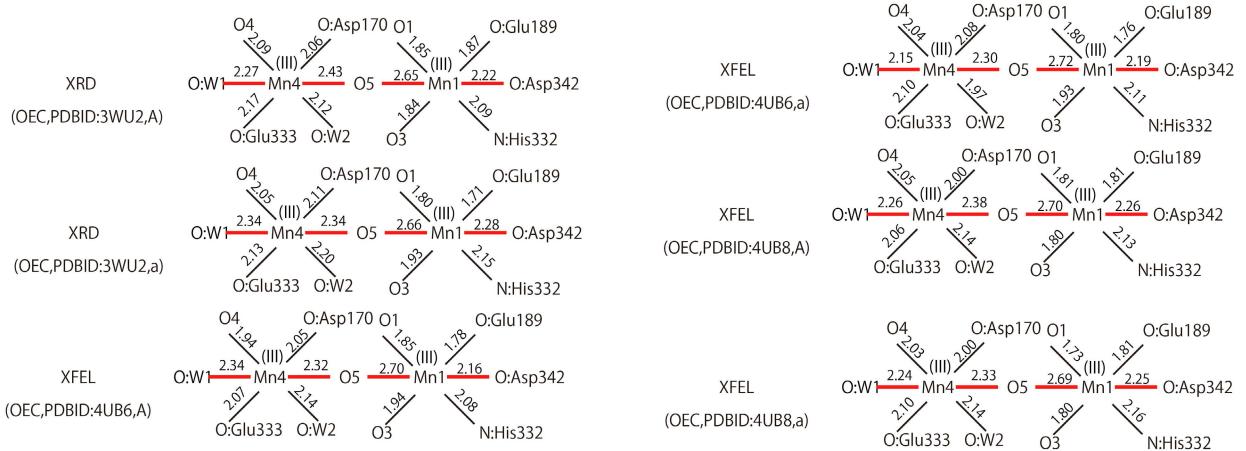
where $N_{\text{LNO},\beta}$ indicates the number of beta spins in the magnetic orbital space of the

BS state. In the present study, the Pipek-Mezey (PM) localization procedure (ref. s22) was used for many electron systems. In the PM method, 2×2 rotational transformations in eq. (1.18) are iteratively performed to maximize the sum of the Mulliken atomic charges. We can construct trial orbitals for LS BS solutions by using the LNOs of the HS solutions as described in ref. 27.

As shown in Figure 4, and SI (Figures S2.2-S2.7,S3.2 and S4.2-S4.6), the LNOs of the HS solutions of the Mn₄CaO₄ cluster are localized on each Mn site according to the orbital symmetry. Therefore, symmetries of the Mn d orbitals, t_{2g} and e_g, are clearly identified from the LNOs. For the Jahn-Teller distortion, orbital distributions of the e_g type are important, LNO transformation is significantly variable as the electronic structure analysis for such complicated multi spin systems (tetra Mn cluster).

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S1.6 Summary of the native OEC structure in the X-ray structures



mono-dentate to Mn₁. Another difference in the Dau and Siegbahn models is the Mn4 coordination, the bridging ligand between Mn4 and Ca is μ_2 -OH instead of Asp 170. These differences may lead to an incorrect Mn valence assignment (Mn1(III), Mn2(III), Mn3(IV), Mn4(IV)) in the S1 state.

Very recent QM computational results in refs. S25, S26 and S27 have demonstrated that W3 is one of substrate molecules for water oxidation in OEC of PSII. Thus the proper ligand coordination of Mn_4CaO_5 may be essential for the full functional models of OEC of PSII. It is noteworthy that the large-scale QM/MM calculations in ref. S29 have elucidated importance of several channels for proton release pathways and water inlet pathway and possible geometrical structures of the redox active $CaMn_4O_5$ cluster catalyst for water oxidation. This in turn indicates that early Dau (S23) and Siegbahn (S24) models are regarded as a modified structural and the cluster model, respectively, in comparison with the native PSII-OEC structures reported in ref 2.

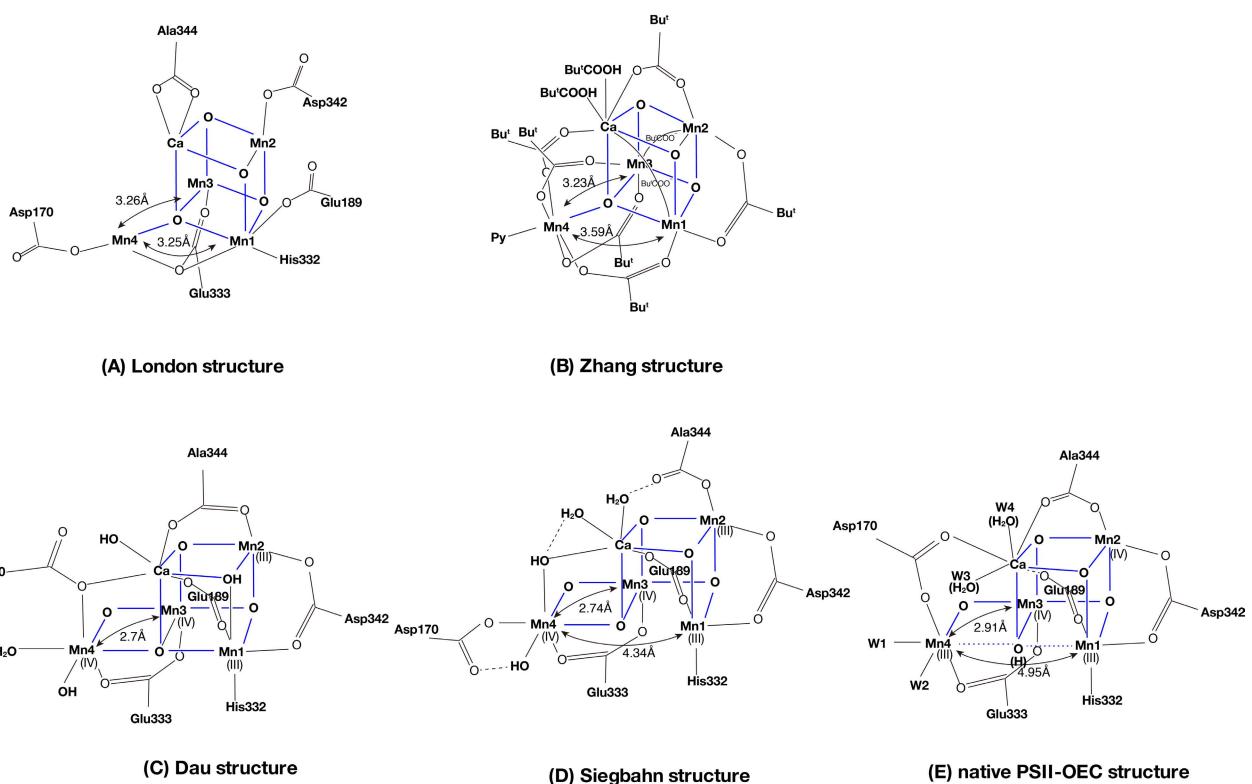


Figure S1.2. Early structures of PSII-OEC: (A) London structure in ref. 16, (B) Zhang structure in ref. 15, (C) Dau structure in ref. 23 and (D) Siegbahn structure in ref. 24. All these models that have shorter Mn4-Mn1 distances than 4.5 Å. (E) Native PSII-OEC (PDBID: 3ARC) in ref. 2

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S2. Results of **1** calculated at the UB3LYP/DZVP level.

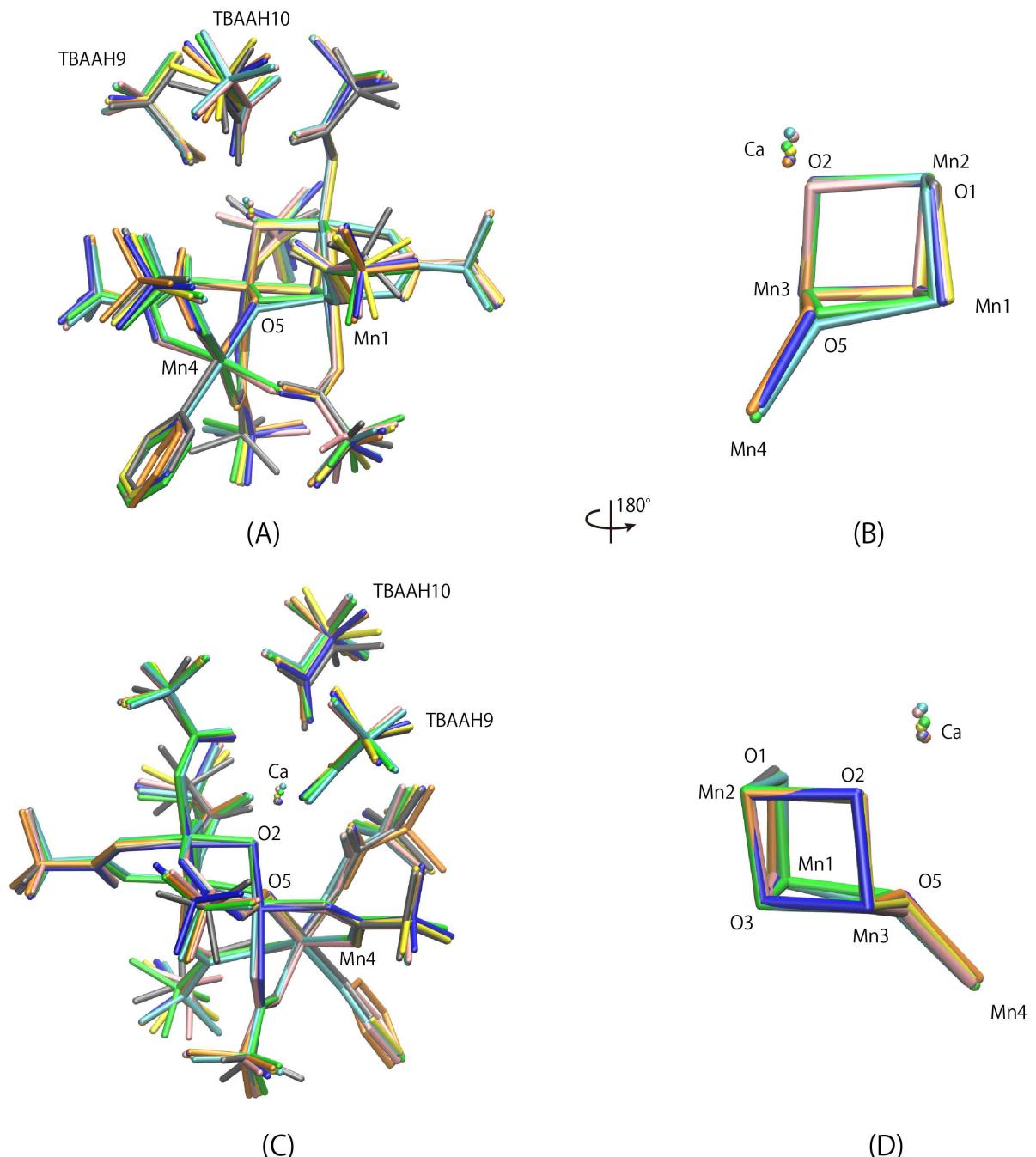


Figure S2.1 Superimposed structures of **1** optimized at the UB3LYP/DZVP level.

Structures are colored in gray (X-ray structure), blue ($S_0(a)$), orange ($S_0(b)$), yellow (S_1), green ($S_2(C)$), pink ($S_2(R)$), and cyan (S_3). (A) aligned whole molecular structures of **1** and (B) aligned CaMn_4O_4 cores were shown. (C) and (D) are 180 degrees rotated views of (A) and (B). The (B) and (D) drawings show that the Ca atoms becomes far from O5 in the following order; $S_0(b)$ (orange), $S_0(a)$ (blue), X-ray(gray), S_1 (yellow), $S_2(C)$ (green), $S_2(R)$ (pink) and S_3 (cyan). This order is roughly consistent with the RMSD values calculated for the CaMn_4O_4 moiety.

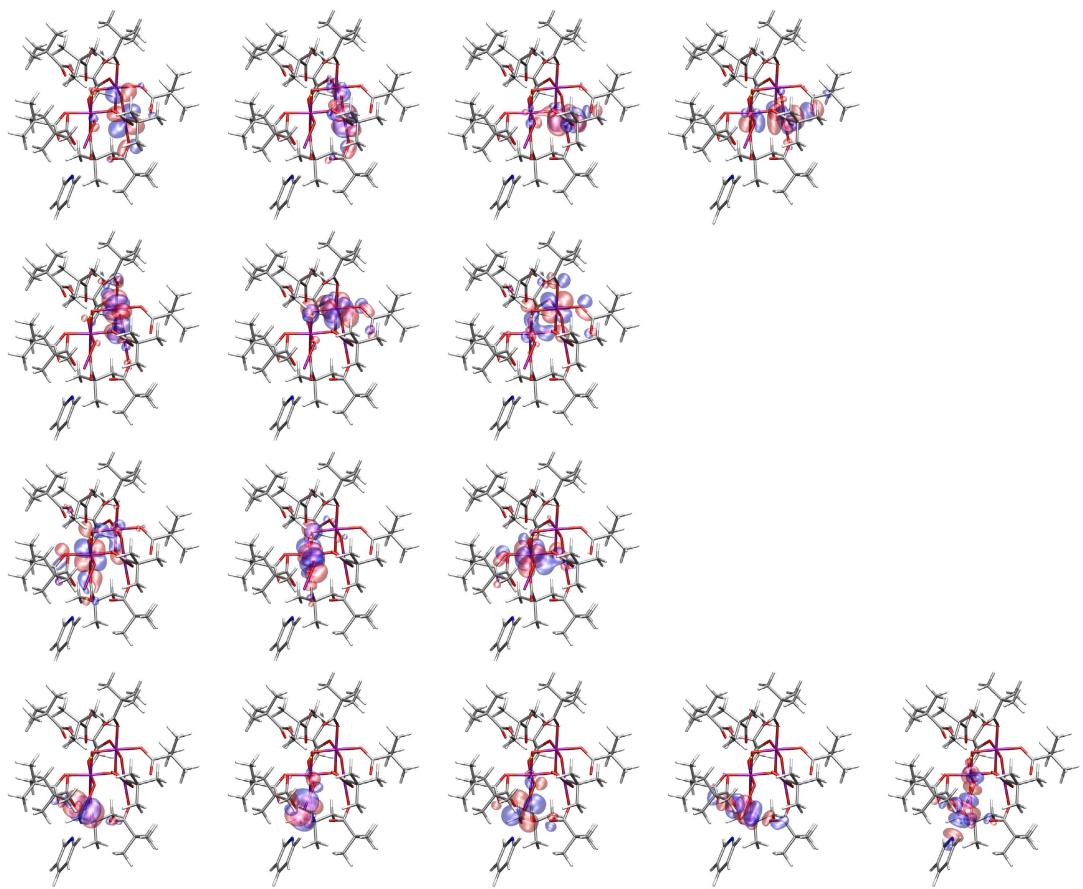


Figure S2.2 Localized natural orbitals of **1** in the highest spin $S_0(b)$ state calculated at the B3LYP/DZVP level. 15 d-type orbitals; 4 for Mn1, 3 for Mn2, 3 for Mn3 and 5 for Mn4 are shown.

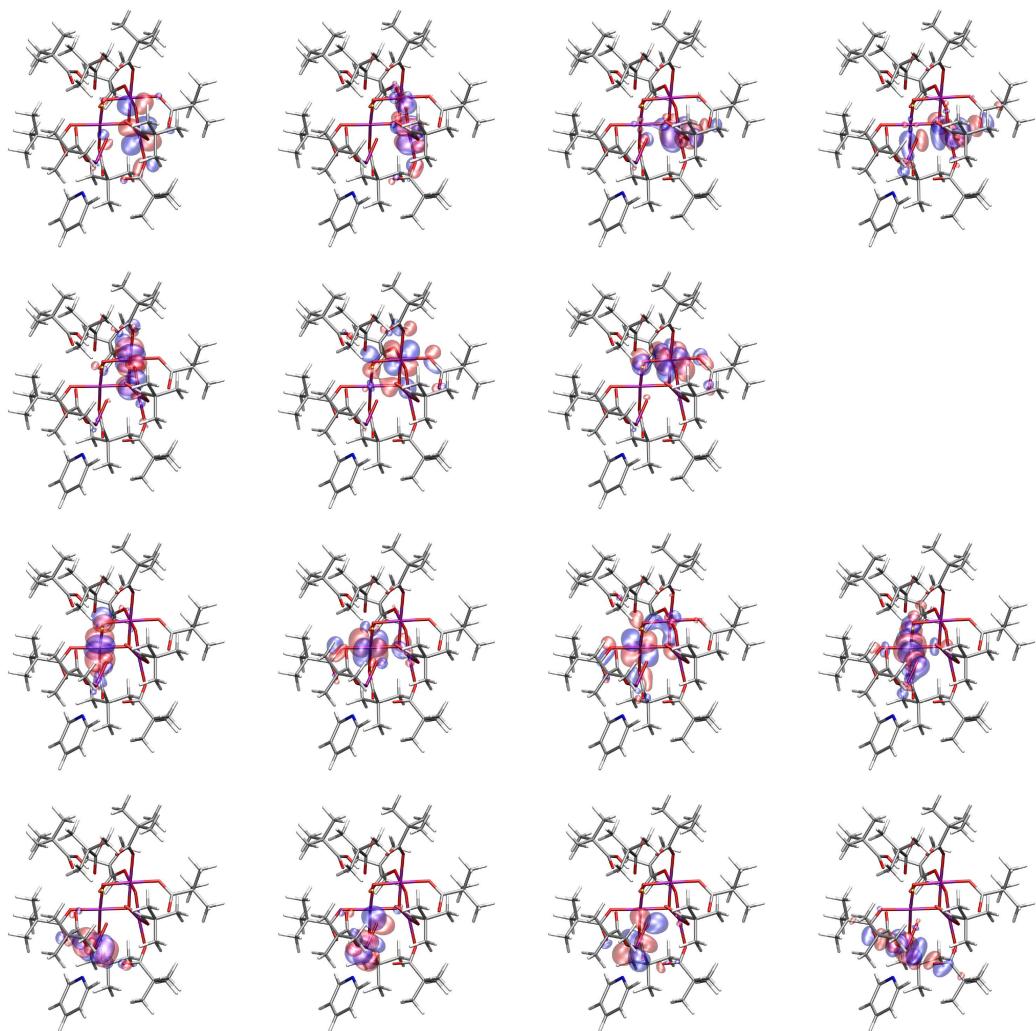


Figure S2.3 Localized natural orbitals of **1** in the highest spin $S_0(a)$ state calculated at the B3LYP/DZVP level. 15 d-type orbitals; 4 for Mn1, 3 for Mn2, 4 for Mn3 and 4 for Mn4 are shown.

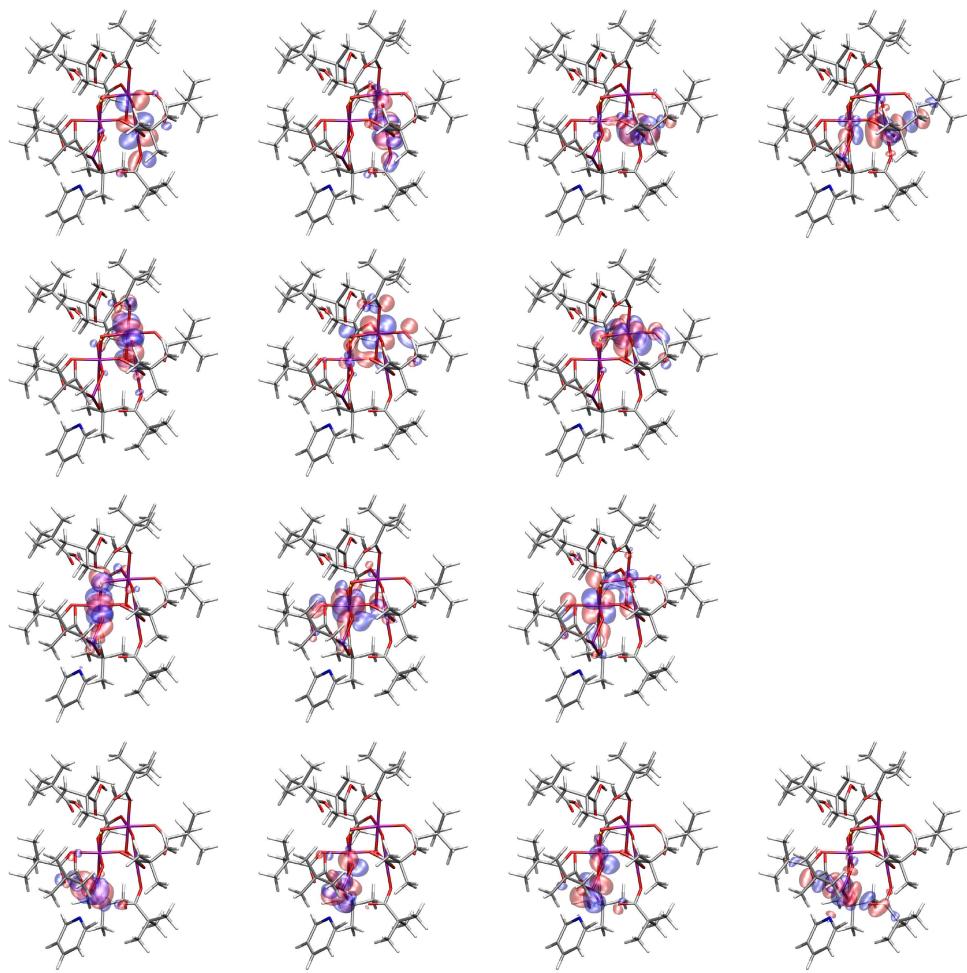


Figure S2.4 Localized natural orbitals of **1** in the highest spin S_1 state calculated at the B3LYP/DZVP level. 14 d-type orbitals; 4 for Mn1, 3 for Mn2, 3 for Mn3 and 4 for Mn4 are shown.

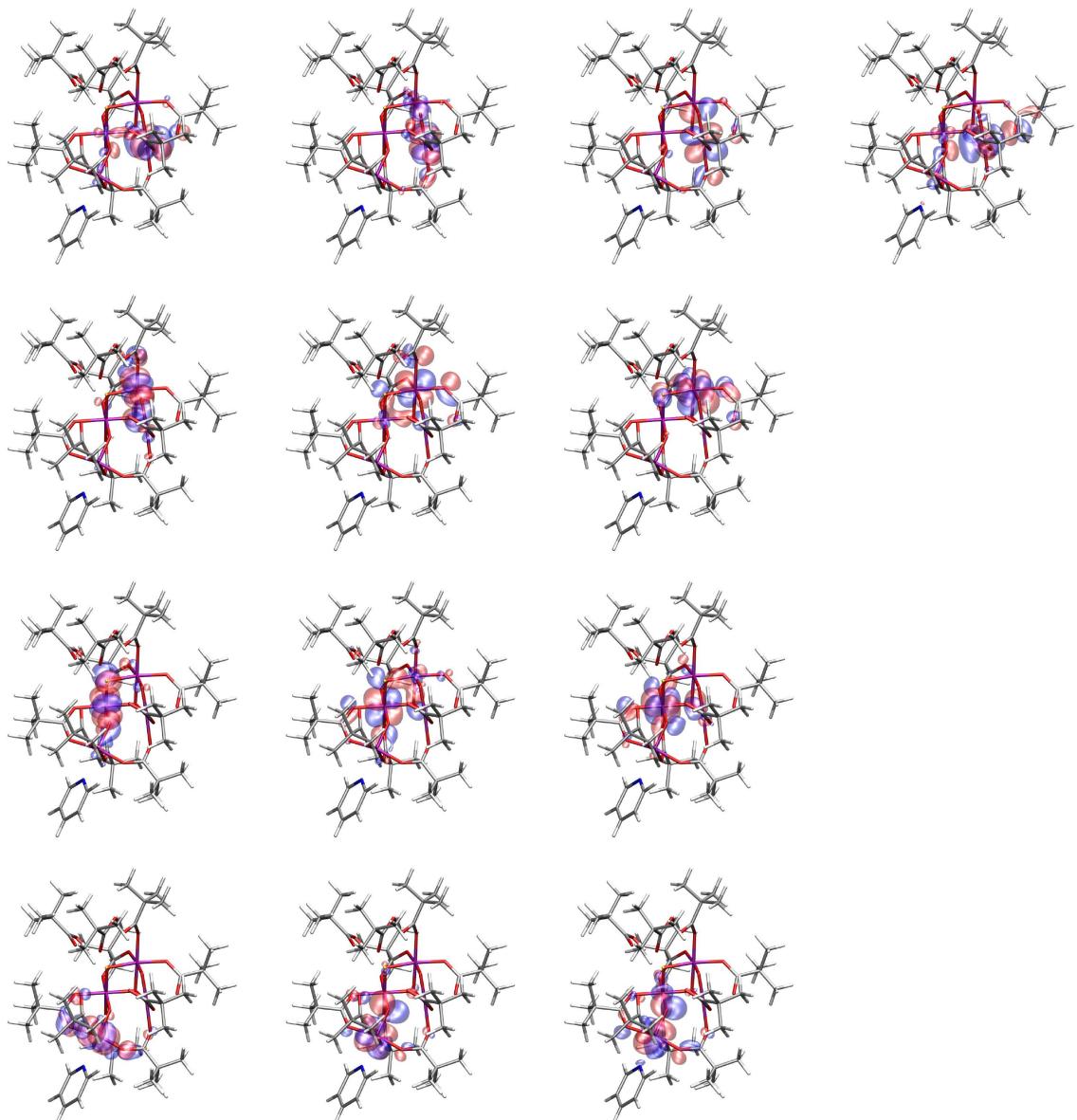


Figure S2.5 Localized natural orbitals of **1** in the highest spin $S_2(R)$ state calculated at the B3LYP/DZVP level. 13 d-type orbitals; 4 for Mn1, 3 for Mn2, 3 for Mn3 and 3 for Mn4 are shown.

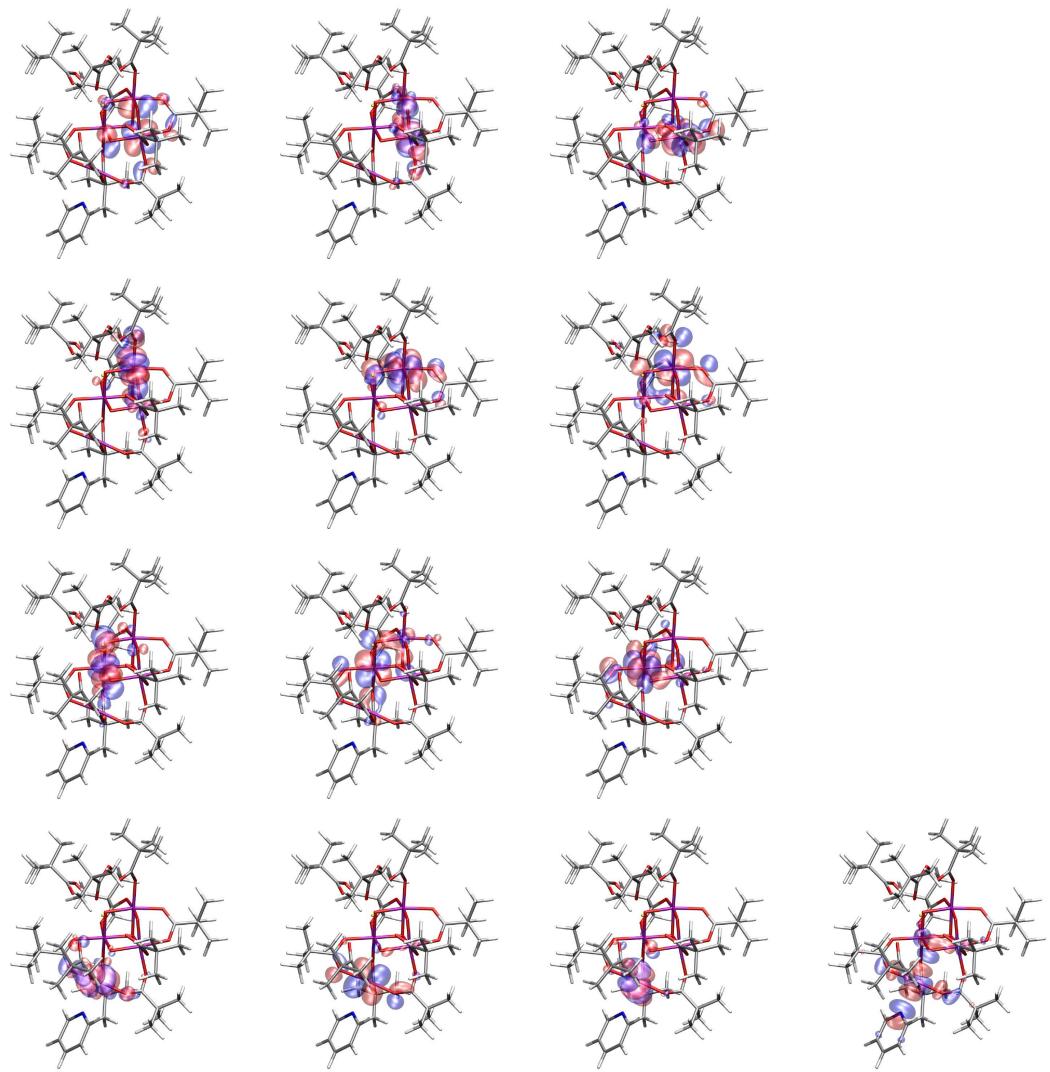


Figure S2.6 Localized natural orbitals of 1 in the highest spin $S_2(C)$ state calculated at the B3LYP/DZVP level. 13 d-type orbitals; 3 for Mn1, 3 for Mn2, 3 for Mn3 and 4 for Mn4 are shown.

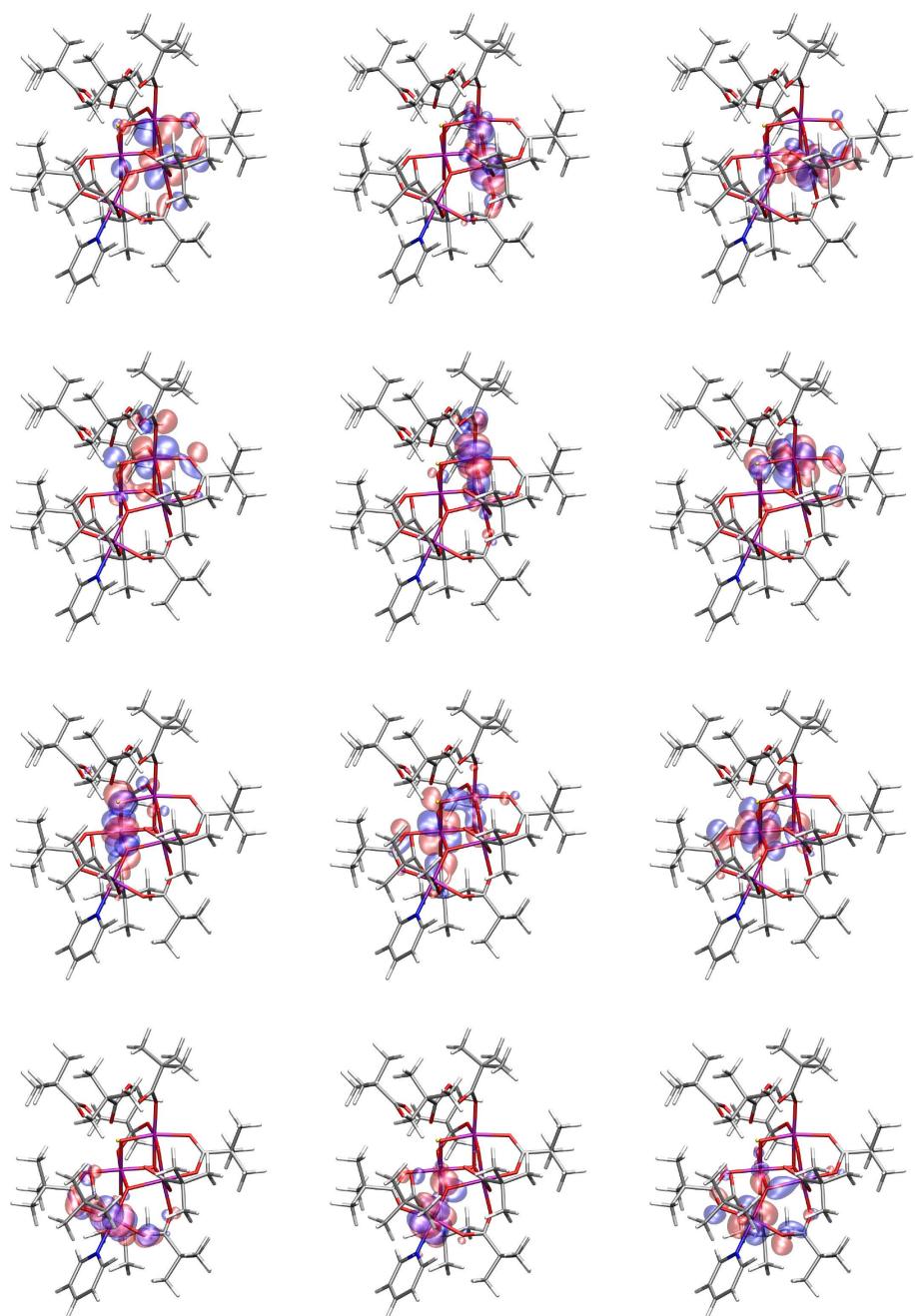


Figure S2.7 Localized natural orbitals of 1 in the highest spin S_3 state calculated at the B3LYP/DZVP level. 12 d-type orbitals; 3 for Mn1, 3 for Mn2, 3 for Mn3 and 3 for Mn4 are shown.

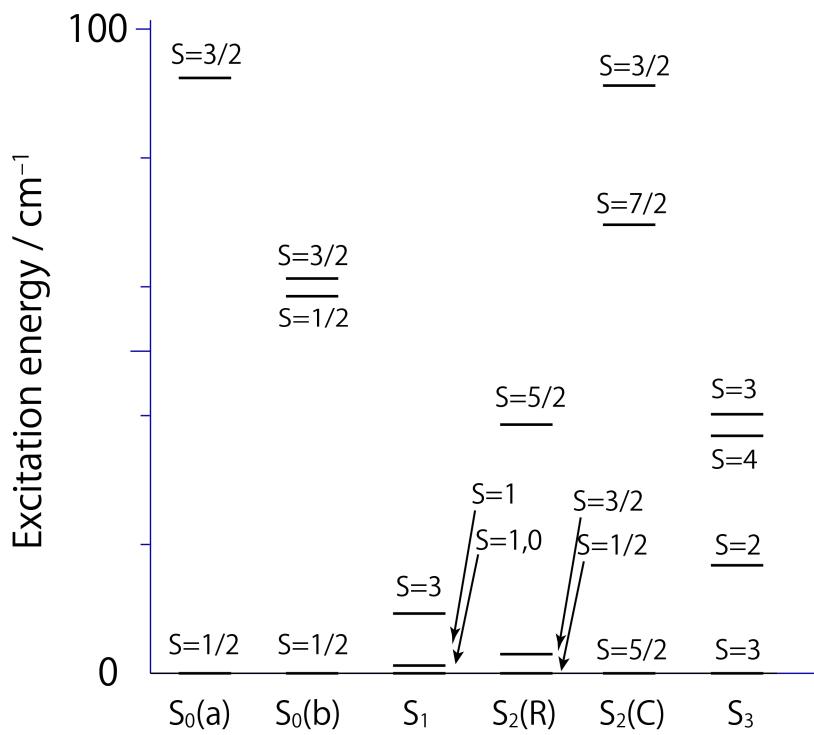


Figure S2.8 Low-lying energy levels of 1 depending on the oxidation states calculated at the B3LYP/DZVP level.

Table S2.1 Bond valence sum (BVS) calculations^a of **1** at the fully optimized structures

	Mn1	Mn2	Mn3	Mn4
S ₀ (a)	2.79	4.07	2.88	2.73
S ₀ (b)	2.80	4.07	3.97	1.94
S ₁	2.81	4.08	4.06	2.83
S ₂ (R)	2.85	4.09	4.08	3.99
S ₂ (L)	3.99	4.14	4.05	2.82
S ₃	3.97	4.14	4.05	3.86

^a Mn oxidation states were determined by using the bond valence sum (BVS) analysis. The BVS values were evaluated using the BVS parameters determined for the closest Mn formal charge states^{S1,S2}.

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Table S2.2 Local spin correlation factor c for low-lying energy levels of **1** calculated at the UB3LYP/DZP level.^a

State	Index	S	c_1	c_2	c_3	c_4
$S_0(a)^b$	0	1/2	1.760	-0.991	1.558	-1.327
	1	3/2	0.616	-0.154	0.467	0.070
	2	5/2	0.447	-0.030	0.317	0.265
$S_0(b)^c$	0	1/2	-0.990	0.691	-0.940	2.239
	1	1/2	1.594	-0.047	0.896	-1.443
	2	3/2	0.260	0.051	-0.204	0.892
S_1	0	1	-0.257	0.380	0.515	0.361
	1	0	—	—	—	—
	2	2	-0.187	0.288	0.478	0.419
$S_2(R)$	0	1/2	0.958	-0.629	1.617	-0.946
	1	3/2	0.179	0.347	0.735	-0.263
	2	5/2	0.177	0.388	0.515	-0.081
$S_2(C)$	0	5/2	0.530	0.546	0.485	-0.562
	1	7/2	0.375	0.389	0.320	-0.084
	2	3/2	0.582	0.819	0.377	-0.778
S_3	0	3	0.397	0.466	0.486	-0.349
	1	2	0.361	0.652	0.463	-0.476
	2	4	0.270	0.351	0.371	0.007

^a $c_i = \langle s_i S \rangle / \langle S S \rangle$

^b Optimized at UDUD spin state at the (B3LYP) method

^b Optimized at DUDU spin state at the (B3LYP) method

S-1, Full, Opted at UB3LYP/DZVP (UUUU)

Mn 14.13647382 4.25171618 15.21347364
 Mn 14.82019196 2.64824624 17.53440195
 Mn 16.81909905 4.58419222 16.82337174
 Mn 15.94722426 7.27879190 15.05078434
 Ca 13.52670440 5.71488257 18.40483223
 O 13.50726935 3.69750613 16.85804570
 O 15.36198167 4.13369322 18.55692339
 O 15.92803929 3.07703859 16.17529554
 O 15.39093557 5.57311747 16.00276694
 O 14.08342469 8.01060909 17.83402722
 O 14.48799220 8.71866090 15.70785739
 O 17.62360943 6.10784357 17.87105379
 O 17.43901296 7.95882612 16.55113163
 O 18.40004564 4.71010144 15.30081645
 O 17.70309038 6.38874803 13.91430619
 O 14.63160735 4.59899413 13.30083081
 O 14.80733145 6.86986019 13.15940593
 O 15.75981756 4.21792140 21.18882288
 O 14.45964995 6.04399930 20.73471730
 O 11.49298843 6.70184960 19.79599775
 O 10.51421162 4.96158881 20.89564831
 O 12.37634769 3.68372632 19.64527046
 O 13.64699399 1.89902003 19.04736527
 O 16.40085680 1.62123355 18.39576618
 O 18.01008098 3.19618185 17.91787639
 O 14.20258052 1.06624657 16.46180968
 O 13.69249986 2.20151424 14.52930327
 O 12.22341041 5.08629072 14.77728825
 O 11.65792602 6.03962678 16.77391895
 N 16.72708076 9.21292737 13.81079432
 C 15.16907316 5.30917124 21.50206448
 C 15.37149061 5.78335766 22.99989789
 C 16.88254971 6.08825333 23.20215402
 C 14.53537285 7.04604911 23.31501509
 C 14.95867277 4.62411058 23.94723506
 C 10.63476919 6.25792221 20.56869562
 C 9.56951620 7.14590016 21.26669619
 C 9.72946992 6.99315702 22.80779693
 C 8.15621623 6.65443722 20.83573270
 C 9.76482957 8.62403293 20.85681880
 C 12.71804879 2.45068430 19.72749445
 C 11.95408824 1.51897736 20.73084790
 C 12.53771598 0.08362205 20.69938679
 C 12.10106279 2.10300039 22.16587388
 C 10.45262015 1.47579613 20.32126492
 C 13.95767433 8.82888517 16.87515035
 C 13.05298782 10.10102392 17.08337329
 C 12.66298843 10.25918810 18.57255107
 C 11.77211126 9.88892923 16.22430805
 C 13.80575091 11.36847797 16.59712627
 C 11.39055498 5.59951280 15.61785794
 C 9.90498580 5.67388096 15.09822243
 C 9.45614911 4.23856358 14.70254337
 C 8.95870726 6.23703715 16.18552299
 C 9.87119094 6.59046971 13.84272340
 C 17.77677105 7.36121582 17.61951806
 C 18.38561450 8.21193375 18.78375279
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 C 20.96292246 5.09216203 13.97714904
 C 19.62356206 5.92429694 11.98648071
 C 17.59173601 2.10723790 18.43819297
 C 18.64908988 1.29739348 19.26297758
 C 19.88291174 1.02183391 18.35985128
 C 19.06551895 2.19391277 20.46677549
 C 18.05872816 -0.03766471 19.77926451

C 13.80553138	1.15377806	15.23300907	H 7.37641656	7.27017737	21.32052262
C 13.44641870	-0.22678954	14.58128414	H 8.02570920	6.73784882	19.74320723
C 14.73764819	-1.09491755	14.56687838	H 15.15517782	4.88611418	25.00561859
C 12.92179476	-0.03523317	13.13760328	H 13.88171550	4.39746727	23.85132348
C 12.36235485	-0.91942021	15.45414392	H 15.52031381	3.71125194	23.69648401
C 14.67895503	5.71576523	12.65454914	H 17.20890906	6.92088885	22.55532136
C 14.50199073	5.58800274	11.10094151	H 17.09285280	6.37174993	24.25229776
C 12.97037519	5.49821455	10.83021029	H 17.48708676	5.20413384	22.94668553
C 15.19216392	4.29700672	10.58627528	H 14.79832425	7.87392219	22.63892749
C 15.09200258	6.82755068	10.38356575	H 13.45780756	6.85005891	23.18628749
C 18.04280217	9.51110992	13.74980985	H 14.70538384	7.37476302	24.35966665
C 18.53485340	10.63367713	13.06178208	H 17.55061578	9.25407150	20.53301093
C 17.62192929	11.47757996	12.40750809	H 16.46457983	9.24970272	19.10567210
C 16.25277269	11.16896682	12.46981912	H 16.64164685	7.77618293	20.07320645
C 15.85016053	10.02791184	13.18502116	H 19.95446023	9.18591406	17.57870376
H 18.69961413	8.81286222	14.27237183	H 18.40767808	10.06956981	17.60940936
H 19.61016509	10.83547095	13.04151683	H 19.48287056	10.09324030	19.04498204
H 17.97096439	12.35994207	11.85994919	H 19.78263079	7.96189556	20.45718421
H 15.50555588	11.79785775	11.97622788	H 18.89346399	6.46681862	20.01787031
H 14.80061748	9.74152132	13.28069814	H 20.23629713	7.03355663	18.99650926
H 14.71203528	11.54738001	17.20336125	H 21.79201498	4.75160382	13.32726120
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H 13.16000272	12.26385498	16.68311269	H 20.94152777	4.46911215	14.88410806
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H 12.14804409	9.36124782	18.94219266	H 19.79556347	6.97578745	12.26961070
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H 10.76381016	8.98765067	21.14229340	H 19.52555091	3.12788312	20.10841559
H 9.66851036	8.75132471	19.76770013	H 18.18916438	2.46727758	21.07842722
H 9.00545671	9.25576909	21.35302488	H 19.79320116	1.66168744	21.10873183
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H 9.59423648	5.94571317	23.11796252	H 17.72786799	-0.67744191	18.94465520
H 7.99928871	5.60236864	21.11947247	H 20.68213985	0.52088015	18.93834398

H 19.61517963	0.36615110	17.51179866	H 11.65244793	3.10443575	22.24620705
H 20.27773237	1.96358382	17.94940258	H 11.59817359	1.44235547	22.89680352
H 15.52501734	-0.61903362	13.95667811	H 11.26075256	4.40608727	20.41289023
H 15.13022225	-1.22072592	15.58770568	H 15.62539613	3.98851692	19.53333710
H 14.52788095	-2.09326885	14.13819426			
H 12.71918020	-1.04416975	16.48780204			
H 11.43610610	-0.31861761	15.48337390			
H 12.10994885	-1.91421650	15.04044894			
H 13.67168103	0.46637085	12.50625718			
H 12.68223358	-1.01762188	12.68795702			
H 12.01167493	0.58579246	13.12373396			
H 16.17561807	6.91282652	10.57099700			
H 14.93383597	6.75097889	9.29089188			
H 14.61968012	7.75368523	10.74425467			
H 16.27995259	4.32973658	10.77029012			
H 14.79242011	3.41090816	11.10020862			
H 15.03340177	4.18476277	9.49663706			
H 12.77539217	5.37106438	9.74821183			
H 12.53168294	4.64455662	11.37141528			
H 12.45670194	6.41466198	11.16730131			
H 10.55522866	6.20924210	13.06945027			
H 8.84866506	6.63891476	13.42061208			
H 10.18388890	7.61864179	14.09597821			
H 9.45733186	3.56939218	15.58122975			
H 8.43114735	4.24994550	14.28381604			
H 10.14350522	3.81390512	13.95541118			
H 9.25506256	7.25472247	16.48478324			
H 7.91827525	6.27319581	15.80783800			
H 8.97816389	5.60841120	17.09043617			
H 10.33602286	1.07303645	19.29977260			
H 9.88976393	0.82049951	21.01198315			
H 9.99295445	2.47534284	20.34929335			
H 13.60403167	0.08330874	20.97378312			
H 11.98906034	-0.56021561	21.41287807			
H 12.45830245	-0.35778305	19.69365873			
H 13.16405895	2.18333514	22.44764913			

S0a, Full, Opted at UB3LYP/DZVP (UDUD)	C 14.37975056 4.76539800 23.50625976
Mn 14.15861710 4.17709055 15.25289822	C 10.88064109 6.38120421 20.57835535
Mn 14.90747339 2.74369117 17.51675668	C 9.81453864 7.26651512 21.23829992
Mn 16.74629384 4.73816828 17.07956678	C 10.16942054 7.43559911 22.73201205
Mn 15.69074102 7.28934445 15.21298637	C 8.44670625 6.55978338 21.10652519
Ca 13.56151641 5.77305655 18.24667071	C 9.76678953 8.63590828 20.54830791
O 13.51303961 3.68173068 16.87252419	C 12.87185352 2.58805669 19.68978294
O 15.51905425 4.25551445 18.39921294	C 11.93700938 1.63675448 20.47913083
O 15.84511852 3.35972693 16.10377715	C 12.34498378 0.17056308 20.26151748
O 15.15595531 5.92419280 16.14293284	C 12.01402747 1.96998425 21.98406031
O 14.06874442 8.13564291 18.00287774	C 10.49571673 1.85193530 19.96551926
O 14.33726274 8.63404808 15.82512486	C 13.90939875 8.87437434 17.01187267
O 17.46004402 6.24706317 18.14165913	C 13.11395703 10.19784805 17.15915695
O 17.22665640 7.94582371 16.67433693	C 12.83478193 10.48508976 18.64137610
O 18.14427487 4.95407814 15.70897180	C 11.78186026 10.01867402 16.39776221
O 17.29775768 6.37272165 14.18403745	C 13.91343115 11.36049238 16.54017018
O 14.80048075 4.56927235 13.42391359	C 11.53969920 5.51390663 15.49253209
O 14.63701586 6.81221594 13.33527486	C 10.11150060 5.60927692 14.89735844
O 15.98463636 4.50015973 20.90069309	C 9.65652187 4.20060993 14.46466082
O 14.49452033 6.15723735 20.57247045	C 9.13291500 6.18142123 15.93393366
O 11.57270237 6.73657139 19.63163270	C 10.16566562 6.53656314 13.66491596
O 10.96763571 5.18044282 21.13655488	C 17.59151167 7.46443661 17.77420968
O 12.66108982 3.83895918 19.72805157	C 18.28018368 8.38454042 18.80864737
O 13.80328374 2.03819819 19.03618165	C 19.71956640 7.86999419 19.02555678
O 16.42423113 1.67471189 18.26980182	C 18.30685899 9.83957511 18.31849385
O 18.09200451 3.17208867 17.97504721	C 17.49039852 8.28922606 20.13063619
O 14.37344634 1.11817095 16.53376636	C 18.12654930 5.50716526 14.56664603
O 13.83929806 2.08319378 14.55801045	C 19.22471222 5.03521808 13.58868168
O 12.39263803 4.91124982 14.74170067	C 18.88121756 3.57692776 13.20542580
O 11.77629850 6.01795154 16.60626771	C 20.58909368 5.07119244 14.30660126
N 16.40015385 9.01891665 14.04449381	C 19.26648524 5.91693172 12.33206215
C 15.23845035 5.50568956 21.30747494	C 17.63986991 2.06674100 18.35621775
C 15.32693597 5.77704895 22.81893688	C 18.64104729 1.06488240 18.98633154
C 16.76451631 5.57315070 23.33304826	C 19.66733164 0.68758561 17.89709423
C 14.85478892 7.20985305 23.11419498	C 19.36061439 1.77793002 20.14943068
	C 17.93288509 -0.19764901 19.50020649

C 13.97702830	1.09602350	15.31353827	
C 13.66385087	-0.32268237	14.77616501	H 7.66594153
C 14.95885233	-1.15866481	14.85441435	H 8.17713085
C 13.16731298	-0.25485636	13.32436984	H 14.38512435
C 12.58438092	-0.95816788	15.67642115	H 13.35211698
C 14.69143242	5.67706384	12.79915633	H 14.69544226
C 14.56960906	5.57386259	11.25834538	H 17.46051982
C 13.06263177	5.42841405	10.94472282	H 16.80046920
C 15.31662630	4.33313703	10.73759035	H 17.11619826
C 15.11805921	6.84349305	10.58716724	H 15.49954918
C 17.68080438	9.40932478	14.07640934	H 13.83591392
C 18.14784620	10.49218928	13.33461634	H 14.87852538
C 17.24963984	11.19183453	12.53023374	H 17.93019369
C 15.91729307	10.78226255	12.50010640	H 16.44145655
C 15.53357106	9.68993381	13.27390314	H 17.51465619
H 18.32747720	8.83462459	14.72696423	H 18.85386831
H 19.19442016	10.77462745	13.39265920	H 17.29416897
H 17.58158557	12.04099090	11.93820102	H 18.79808445
H 15.18155502	11.29515643	11.88867093	H 20.22534571
H 14.51491417	9.32394540	13.28810518	H 19.71167345
H 14.87214889	11.49861888	17.05518350	H 20.30841466
H 14.12310242	11.17665292	15.48362199	H 21.37291468
H 13.34812580	12.29787119	16.62668416	H 20.86420128
H 13.76693451	10.63316764	19.19734921	H 20.56347335
H 12.22604219	11.39356768	18.74251193	H 20.04586261
H 12.30643148	9.64862402	19.10322404	H 18.30978691
H 11.96520407	9.81900547	15.33757973	H 19.48891622
H 11.21047866	9.17740873	16.80534542	H 18.85783552
H 11.16947262	10.92613167	16.48233109	H 17.89855466
H 10.71871626	9.16455584	20.64894829	H 19.63312885
H 9.55749977	8.53425699	19.47957872	H 19.85547059
H 8.98038534	9.25217520	21.00063526	H 18.64968109
H 11.14574244	7.91958710	22.85240360	H 20.11335319
H 9.41793912	8.06549829	23.22393925	H 18.66718135
H 10.20143674	6.46931774	23.24317545	H 17.18852878
H 8.45905014	5.58042070	21.59259625	H 17.40937088
			H 20.42324550

H 19.17747404	0.18543254	17.05345860	H 11.68852796	2.99181339	22.19781122
H 20.17149296	1.58092021	17.51592035	H 11.36586042	1.28649805	22.54762368
H 15.74565415	-0.72364383	14.22593501	H 11.64199479	4.61074449	20.61942318
H 15.32996718	-1.19956509	15.88256968	H 15.85664919	4.34407290	19.88397238
H 14.77212536	-2.18240824	14.50364581			
H 12.92441529	-1.00130734	16.71513059			
H 11.65411841	-0.37724454	15.64535847			
H 12.35730351	-1.97734737	15.33661422			
H 13.91573910	0.20524192	12.67193351			
H 12.95550876	-1.26650050	12.95314343			
H 12.25253556	0.34130614	13.24581290			
H 16.18224396	6.97976371	10.80953356			
H 15.00313714	6.77400544	9.49758481			
H 14.58852473	7.73356412	10.93610871			
H 16.39129103	4.40118853	10.94101362			
H 14.94315019	3.42423437	11.21514297			
H 15.18417526	4.24573434	9.65113255			
H 12.91310881	5.30768191	9.86373992			
H 12.64211933	4.55572890	11.45505989			
H 12.50814171	6.31493737	11.26971476			
H 10.85892070	6.14525171	12.91652149			
H 9.17017918	6.62198511	13.20940533			
H 10.49796600	7.54321605	13.94499175			
H 9.60587575	3.52256117	15.32518157			
H 8.65768125	4.24631290	14.01064589			
H 10.35496879	3.77202584	13.74158471			
H 9.43517029	7.18241387	16.25422351			
H 8.12401944	6.24240619	15.50437511			
H 9.08949924	5.54956253	16.82695934			
H 10.42774528	1.63750205	18.89256528			
H 9.80815237	1.17644003	20.49041709			
H 10.15697933	2.87869600	20.13026869			
H 13.36548158	-0.01402639	20.61107455			
H 11.66166937	-0.48951575	20.81214920			
H 12.31160748	-0.09550129	19.20115559			
H 13.03656447	1.84877307	22.36105444			

S0b, Full, Opted at UB3LYP/DZVP (DUDU)	C 14.24161430 4.84538689 23.47199145
Mn 14.02920429 4.02893395 15.24077549	C 10.76879491 6.29690154 20.53241386
Mn 14.76806731 2.65542142 17.53418139	C 9.70248236 7.20253385 21.16048987
Mn 16.46955976 4.72844327 16.95057862	C 10.13816405 7.53762848 22.60468363
Mn 15.63204882 7.37653642 15.14824233	C 8.35931043 6.43965531 21.18783057
Ca 13.45293320 5.73478647 18.21204004	C 9.55988472 8.48994460 20.33728255
O 13.36708975 3.54499903 16.88190541	C 12.82554006 2.51832218 19.73962231
O 15.38172698 4.20288577 18.36187217	C 11.91426633 1.56498998 20.55294593
O 15.69606795 3.25438758 16.07577319	C 12.33540233 0.10027235 20.34962066
O 15.18908881 5.70792441 16.25583683	C 12.00685337 1.92039620 22.05187683
O 13.93817111 8.08258544 18.00816598	C 10.46444087 1.75946736 20.05532087
O 14.23711514 8.80007416 15.89781539	C 13.87505543 8.95605376 17.10970222
O 17.28333306 6.13973255 18.01385824	C 13.28646824 10.35028118 17.46465124
O 17.22291247 7.90838140 16.60668412	C 13.08251576 10.48452616 18.98063611
O 17.86969405 4.90514628 15.60748510	C 11.92748999 10.47290005 16.74134452
O 17.33683854 6.41287040 14.01775946	C 14.23743308 11.45506078 16.96503348
O 14.71259910 4.45657223 13.45695471	C 11.46831276 5.42952186 15.41753695
O 14.47798827 6.69557134 13.39735207	C 10.07005939 5.61952117 14.78149297
O 15.93607496 4.51494632 20.84936079	C 9.53839109 4.25147144 14.30875446
O 14.36962198 6.10216741 20.54569377	C 9.10052895 6.23914930 15.79918756
O 11.42748502 6.60445204 19.54680205	C 10.22700690 6.56218159 13.56865564
O 10.90540605 5.13585401 21.16575433	C 17.53285966 7.35790552 17.68248196
O 12.59996757 3.76100290 19.75460660	C 18.30982497 8.15846491 18.75397562
O 13.76332474 1.96671607 19.08420754	C 19.69925214 7.50715676 18.92516800
O 16.40635487 1.70939808 18.24443781	C 18.46313695 9.62479781 18.32382504
O 17.84945111 3.38979311 17.78456202	C 17.53064134 8.07547201 20.08227255
O 14.29967987 1.00637355 16.58935184	C 18.03692606 5.48872931 14.48049157
O 13.71667251 1.90749473 14.59663920	C 19.25196494 4.95702289 13.68037947
O 12.29774174 4.75367943 14.69794534	C 19.01238073 3.45930158 13.39113351
O 11.71905539 5.93414501 16.52619972	C 20.51454832 5.11564797 14.55420507
N 16.32597250 9.23614756 13.84044052	C 19.42427969 5.72877224 12.36432520
C 15.15611385 5.49104460 21.26856312	C 17.55872693 2.24669366 18.23550213
C 15.25854254 5.78057837 22.77549663	C 18.72987198 1.42312808 18.82154491
C 16.67432642 5.49602285 23.30879380	C 19.71779125 1.13440808 17.67100372
C 14.87423371 7.24628494 23.04034345	C 19.42624218 2.27686917 19.90125711
	C 18.23112880 0.10415700 19.43123764

C 13.89231095	0.94310834	15.36958033	
C 13.61775454	-0.49620837	14.87059191	H 7.58395819
C 14.93399683	-1.29649016	14.97046272	H 8.03280901
C 13.12040570	-0.47803705	13.41771546	H 14.24879822
C 12.55380205	-1.13495277	15.78749158	H 13.22745767
C 14.62622493	5.58097912	12.84563975	H 14.49243413
C 14.65718571	5.51363666	11.29999449	H 17.41927306
C 13.18405271	5.47551259	10.83213936	H 16.71174011
C 15.37874273	4.24517334	10.81420912	H 16.96058842
C 15.34863968	6.76813849	10.73807036	H 15.57291239
C 17.61488787	9.46155390	13.55582829	H 13.87343326
C 18.03683859	10.53890170	12.77702319	H 14.89391422
C 17.07626128	11.41177416	12.26740685	H 18.03416036
C 15.73326549	11.17753031	12.56158106	H 16.50746677
C 15.40404803	10.08001108	13.35630241	H 17.47312050
H 18.31462292	8.74148083	13.96615156	H 18.99670805
H 19.09442288	10.68264883	12.57597963	H 19.02459638
H 17.36900691	12.26004747	11.65336137	H 20.26421508
H 14.95078050	11.83168287	12.18845535	H 19.60086511
H 14.37955844	9.86094941	13.63850784	H 20.28175686
H 15.21346031	11.38939343	17.46202434	H 21.38995152
H 14.40573407	11.37055334	15.88839972	H 20.71527669
H 13.81771801	12.44746823	17.17893471	H 20.39281483
H 14.03658144	10.42071572	19.51506242	H 20.29159829
H 12.61895998	11.45169612	19.21932007	H 18.54120949
H 12.44513372	9.68102488	19.35760999	H 19.58323796
H 12.05346814	10.36129061	15.65977091	H 18.89006071
H 11.23174623	9.69752154	17.08315818	H 18.10622751
H 11.47189008	11.45219769	16.94129237	H 19.86217256
H 10.49532374	9.05525756	20.31706418	H 19.77906112
H 9.28650363	8.26882003	19.30118342	H 18.73822033
H 8.78034127	9.12523323	20.77547234	H 20.28374814
H 11.10104251	8.06164574	22.61381756	H 19.07956668
H 9.39356337	8.19155260	23.07494952	H 17.51874088
H 10.23507630	6.63119072	23.20927523	H 17.72466864
H 8.43926557	5.51622098	21.76786244	H 20.58584796

H 19.24347621	0.53190290	16.88679792	H 11.68073164	2.94405978	22.25320237
H 20.06772868	2.06712013	17.21897526	H 11.36698019	1.24374013	22.63226794
H 15.70954780	-0.85749288	14.33112621	H 11.57705819	4.55956897	20.66025725
H 15.30570776	-1.30340493	15.99891689	H 15.78587090	4.34786221	19.84099123
H 14.77315079	-2.33261590	14.64480243			
H 12.89535029	-1.14732264	16.82633391			
H 11.61045092	-0.57662114	15.74404183			
H 12.35043047	-2.16637526	15.47094274			
H 13.85758812	-0.01699161	12.75353179			
H 12.93414681	-1.50397760	13.07363668			
H 12.19156009	0.09320250	13.32400429			
H 16.38121047	6.83771323	11.09670384			
H 15.36294030	6.73238456	9.64078339			
H 14.82681225	7.67560913	11.05277469			
H 16.42673969	4.23886862	11.13353312			
H 14.90470850	3.34539607	11.21387442			
H 15.35823088	4.20032226	9.71721474			
H 13.13793124	5.41310225	9.73698071			
H 12.66434164	4.60620193	11.25016900			
H 12.65110614	6.37866342	11.14762836			
H 10.91603676	6.13819403	12.83367535			
H 9.25387602	6.72168287	13.08561722			
H 10.61840929	7.53838367	13.87812081			
H 9.41881733	3.56220822	15.15374166			
H 8.55799635	4.37135529	13.82897975			
H 10.22722437	3.79145916	13.59571323			
H 9.46403942	7.20872290	16.15097795			
H 8.11338271	6.38012105	15.33924656			
H 8.98262222	5.59373946	16.67620170			
H 10.38371408	1.52799546	18.98678458			
H 9.79017937	1.08674703	20.60028602			
H 10.12013908	2.78569025	20.20872276			
H 13.36149417	-0.06950456	20.68903087			
H 11.66599354	-0.55871658	20.91789157			
H 12.29032184	-0.18105903	19.29380846			
H 13.03359833	1.80754507	22.41951951			

S1, Full, Opted at UB3LYP/DZVP (DUUU)

Mn 14.03222112 4.11502516 15.17210281
 Mn 14.72281950 2.71040156 17.47815086
 Mn 16.53986255 4.71050086 17.00152337
 Mn 15.80644902 7.28324838 15.20601843
 Ca 13.38532422 5.72710708 18.30157582
 O 13.37005641 3.66353587 16.82488093
 O 15.37358112 4.27496689 18.32065110
 O 15.70908418 3.34714668 16.05283913
 O 15.31639189 5.83224802 16.17901901
 O 14.12174948 7.99480762 18.02386099
 O 14.46805380 8.52082707 15.86712534
 O 17.34935376 6.10322110 18.08761308
 O 17.34197000 7.86606155 16.68403166
 O 17.99118354 4.88226626 15.67396886
 O 17.30817149 6.40928592 14.17597884
 O 14.73345486 4.54049703 13.40830758
 O 14.63112855 6.78379470 13.39238443
 O 15.94612477 4.55436306 20.88449383
 O 14.31862674 6.07525512 20.60145028
 O 11.37590750 6.61954006 19.53783510
 O 10.84081750 5.16845137 21.16155412
 O 12.54039888 3.77977444 19.70371140
 O 13.72281849 2.01914107 18.99242436
 O 16.34165024 1.74334346 18.20639047
 O 17.80875756 3.40924490 17.80286695
 O 14.26191029 1.11648408 16.50494038
 O 13.72685815 2.04949869 14.52069740
 O 12.35187140 4.90323357 14.67058598
 O 11.84593824 6.12341527 16.48868697
 N 16.43875947 9.00579951 14.08068163
 C 15.12889532 5.49707890 21.32500201
 C 15.22713234 5.77859954 22.82897350
 C 16.61696691 5.42515256 23.38772459
 C 14.90951586 7.26290740 23.08617777
 C 14.15207246 4.89542438 23.50886490
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 C 9.25193416 8.33362004 20.16955687
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 C 10.08396697 5.63565438 14.89486109
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C 13.87183662	1.06488829	15.27505318	H 7.82226283	7.22623391	22.23501820
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C 14.56192453	5.60269016	11.26925554	H 16.64197592	5.63374647	24.46375135
C 13.15315571	5.07128788	10.91725243	H 16.85532117	4.36938135	23.23603036
C 15.61977315	4.61675434	10.73355542	H 15.65210762	7.91560207	22.61426695
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C 15.54987583	9.65044771	13.30802373	H 19.14132120	9.65253680	17.48651461
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H 19.16538447	10.88302271	13.48779237	H 19.14284411	10.10793522	19.20141847
H 17.51841913	12.09906635	12.03018530	H 20.36780015	7.93929522	19.79441693
H 15.15274940	11.25557010	11.93821216	H 19.68676531	6.37468882	19.29051818
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H 15.06118737	11.30081591	17.40034959	H 21.41926760	4.79544667	13.91686382
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H 9.81940868	8.71104520	22.84489067	H 17.42442580	0.29169142	20.22778271
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H 15.67966325	-0.70942179	14.19176258	H 11.49828411	4.59002423	20.65707385
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S1, Full, Opted at UB3LYP/DZVP (UDUD)

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C 13.06133062	-0.30604751	13.31507311	H 14.15377964 5.08190667 24.58853747
C 12.52445252	-1.01151217	15.67960681	H 13.15358618 5.11969500 23.12029713
C 14.65860763	5.67488532	12.81267200	H 14.35553968 3.83141250 23.34676798
C 14.56026111	5.59495407	11.27455723	H 17.40203184 6.02257320 22.90580100
C 13.15356183	5.05306031	10.93079060	H 16.64405020 5.63503102 24.46034541
C 15.62272711	4.61323412	10.74059685	H 16.85588327 4.37122804 23.23178262
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H 10.82273112	7.27193202	23.09540609	H 17.43135889 0.29457308 20.21967420
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S2(R), Full, Opted at UB3LYP/DZVP(UDUD)

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C 14.90434710	-1.15191354	14.98782662	H 14.99014290	4.61809482	24.72045485
C 13.14965064	-0.27101082	13.39554608	H 13.86719351	4.24148187	23.40348179
C 12.50340569	-0.93644010	15.74645452	H 15.52894831	3.60973017	23.36290787
C 14.71948632	5.77257192	12.95921108	H 17.22456582	6.93936451	22.76951682
C 14.35701784	5.85291972	11.46766135	H 16.94703736	6.18856316	24.34763664
C 13.65928184	7.18449926	11.13735797	H 17.50478601	5.19991084	22.98410464
C 13.45130993	4.66462579	11.09324290	H 14.79890380	7.86110444	22.80424919
C 15.69178101	5.75914310	10.68676539	H 13.44691795	6.75843975	23.09428910
C 17.96432927	9.32402015	14.01132965	H 14.59489663	7.11413770	24.40021399
C 18.43024240	10.42974249	13.30676730	H 18.15498460	8.87253147	20.67909718
C 17.51590144	11.21204981	12.60174558	H 16.58733098	8.49526805	19.93380258
C 16.16717581	10.85834960	12.63081413	H 17.71058810	7.18409706	20.34498804
C 15.78093816	9.73889172	13.36176868	H 18.73688017	9.91246500	17.12482025
H 18.63334249	8.68519985	14.57355944	H 17.23624426	10.16003700	18.03055762
H 19.48916151	10.66595780	13.31452645	H 18.80117597	10.42393359	18.81986950
H 17.84809659	12.08088220	12.04127115	H 20.33134088	8.46103683	19.36588256
H 15.41915322	11.43693854	12.09900570	H 19.82335490	6.79981788	18.99280847
H 14.74734868	9.42241267	13.42121871	H 20.27026181	7.91480614	17.68269847
H 14.93283841	11.18411672	17.39819729	H 21.37033868	4.37331204	13.74113002
H 14.50817841	11.04251246	15.68040623	H 20.93784303	5.72563868	14.79870473
H 13.49053075	11.99821872	16.77577954	H 20.46112680	4.07326782	15.23853813
H 13.45961981	10.05343472	19.18024294	H 20.27104118	5.52473423	11.78052546
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H 12.04225025	9.11185764	18.69943483	H 19.79596326	6.89538193	12.79486304
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H 9.51635847	8.62494870	19.45943177	H 18.68223779	2.66938482	20.80670597
H 8.82183041	9.24181164	20.97054220	H 20.21265975	1.84305909	20.46159372
H 10.53990833	7.53522156	22.94339311	H 19.02825472	-0.34155088	20.00218811
H 8.79432709	7.80862907	23.06379589	H 17.46754635	0.41428810	20.37905879
H 9.42938215	6.15129820	23.03332388	H 17.68927448	-0.42710964	18.83959227
H 7.88982871	5.59799785	21.06277907	H 20.54725374	0.63620287	18.21205857

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H 20.07390223	2.11091910	17.34111350	H 11.42671710	1.24647391	22.66629936
H 15.71093300	-0.72584722	14.37930073	H 11.14362437	4.45311749	20.51117426
H 15.24302388	-1.19970128	16.02659261	H 15.93266385	4.48839861	19.94390156
H 14.72005244	-2.17237415	14.63386915			
H 12.81819891	-0.98766154	16.79257343			
H 11.57679055	-0.35258455	15.69035296			
H 12.28074455	-1.95196946	15.40058460			
H 13.91587558	0.17019105	12.75086375			
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H 12.23479000	0.31871313	13.28135390			
H 16.20486677	4.81306635	10.88810680			
H 15.48422851	5.80979756	9.61225425			
H 16.36292482	6.58591692	10.94373148			
H 13.94308590	3.70967960	11.29305296			
H 12.51579974	4.68495400	11.66133258			
H 13.20700708	4.71654507	10.02667499			
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H 9.66897893	3.60293187	15.57899575			
H 8.68032584	4.20667670	14.23846985			
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H 9.50163781	7.34768785	16.14339602			
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S2(C), Full, Opted at UB3LYP/DZVP(UUUD)

Mn 14.22274842 4.12373779 15.44948355

Mn 14.76066947 2.62262683 17.67960152

Mn 16.51543885 4.62864875 17.07477610

Mn 15.77977217 7.27233520 15.07800604

Ca 13.43451880 5.70694149 18.52858622

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O 16.98237723 7.84244937 16.55942512

O 17.91836259 4.86925344 15.72291399

O 17.30863940 6.42464835 14.22106572

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C 9.94757150 4.22208300 14.34910145

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C 10.50339946 6.58380057 13.63765362

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C 18.12041625 8.28612072 18.63971448

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C 19.45082330 2.21741230 19.91273560

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C 14.16149353	6.73791189	10.80619821	H 17.48721708	5.02513606	23.11962657
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C 18.17757442	10.70285772	13.20301942	H 18.16177465	8.69310643	20.76740019
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H 13.39836547	12.03690361	16.60304899	H 20.45577400	4.56815430	15.60716341
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S3, Full, Opted at UB3LYP/DZVP(UUUD)

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O 14.46408864 8.26992408 15.65567175
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O 14.92987721 4.46686700 13.57800871
O 14.91723538 6.70924616 13.51645010
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O 16.30538963 1.73091481 18.28187153
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C 18.22906242	10.36667067	13.34249346	H 18.17410324	8.77366677	20.65482127
C 17.30807978	11.10049732	12.59444926	H 16.58002251	8.31089089	20.02808265
C 15.97398204	10.69299301	12.58252624	H 17.81667074	7.06735084	20.31963593
C 15.60500302	9.57066597	13.31630273	H 18.41621382	9.93435615	17.10562240
H 18.46237929	8.65227959	14.64329553	H 16.97758654	10.06236450	18.13345063
H 19.27655148	10.64598565	13.38227357	H 18.58054658	10.40030006	18.80486874
H 17.62439125	11.97345863	12.03158347	H 20.26233859	8.52853179	19.16612874
H 15.22264177	11.23217116	12.01542740	H 19.83317688	6.84579371	18.79932392
H 14.58336259	9.21432445	13.34266718	H 20.10553022	8.01059253	17.48203395
H 14.91573309	11.10584812	17.27286899	H 21.51667348	4.99528802	13.98758040
H 14.43572772	10.94188930	15.56981475	H 20.72943092	6.33223871	14.83626438
H 13.47066722	11.93250596	16.67809338	H 20.63264886	4.68289390	15.49518154
H 13.46584845	10.05485173	19.11970940	H 20.28102307	5.55135148	11.85643917
H 12.03947538	10.87610048	18.46273107	H 18.51634094	5.66277938	11.82023085
H 12.02052561	9.12098524	18.70334384	H 19.46597655	6.90958787	12.64629814
H 12.33069321	9.60391842	14.94376856	H 19.25494544	2.89025150	14.30480981
H 11.34868476	8.84704595	16.21612519	H 18.35664710	3.27546450	12.82269985
H 11.37709588	10.61112985	16.04841905	H 20.12605952	3.20855664	12.79050603
H 10.66924414	9.07379827	20.96306330	H 19.86626775	3.06308337	19.37028102
H 9.60050378	8.75715346	19.58431621	H 18.82781522	2.47212868	20.68589799
H 8.92746016	9.36793714	21.10512563	H 20.27016039	1.55763981	20.21870076
H 10.55374112	7.56108423	23.06123378	H 18.92418908	-0.53989673	19.85419068
H 8.82000248	7.90237528	23.16179086	H 17.44020505	0.29844985	20.33916914
H 9.38523830	6.22205158	23.11337241	H 17.50879558	-0.55087439	18.78614581
H 7.85187835	5.76144622	21.11687469	H 20.35900127	0.34527723	17.95368904

H 18.96385276	0.37770609	16.86434560	H 9.35874545	6.47242376	13.20080247
H 19.92755330	1.85279430	17.12158998	H 10.63691850	7.49963242	13.86661033
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H 14.84663369	-1.03541706	15.49089049	H 8.96231405	4.12261702	14.13882915
H 14.14843505	-1.69757708	13.99883571	H 10.67369700	3.71624332	13.89357889
H 12.50911564	-0.83764833	16.50500324	H 9.51449044	7.22725716	16.18967761
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H 11.80233587	-1.49991154	15.01778782	H 9.24568060	5.60812566	16.85251780
H 13.35000089	0.96462455	12.63328829	H 10.55127606	1.11927362	19.12546484
H 12.31082249	-0.46421598	12.76905543	H 10.01256199	0.73130918	20.76574338
H 11.72850482	1.10912356	13.33679749	H 10.05746706	2.41853295	20.23466444
H 16.52250162	6.79465415	11.03266497	H 13.71266101	0.16748273	20.99960865
H 15.37656830	6.67946180	9.68814322	H 12.11417104	-0.54966179	21.26502571
H 14.98630736	7.67922129	11.09546337	H 12.69056511	-0.24792654	19.61596812
H 16.47859570	4.16458596	11.10021304	H 13.04276058	2.11039362	22.56159709
H 14.92466003	3.33423374	11.29858527	H 11.54536956	3.02164549	22.26203823
H 15.29249760	4.18869110	9.78720482	H 11.46923044	1.33578491	22.78829666
H 13.13503823	5.46187551	10.00126223	H 11.07108119	4.51410939	20.58589246
H 12.74940940	4.68388271	11.54777531	H 16.01231870	4.44943704	20.07819434
H 12.79075782	6.45977852	11.42458057			
H 11.05758888	6.05032186	12.92675822			

S3. Results of **1** calculated at the UBLYP/DZVP level.

Table S3.1 Calculated effective exchange integrals (J/cm^{-1}) of **1** at the UBLYP/DZVP level.^a

State	$S_0(\text{b})$	$S_0(\text{a})$	S_1	$S_2(\text{R})$	$S_2(\text{C})$	S_3
$J_{1,2}$	-84.74	-88.99	-66.85	-58.21	34.78	23.49
$J_{1,3}$	-9.32	-6.31	7.50	-2.30	-10.98	-28.81
$J_{1,4}$	-16.97	-43.41	-10.01	-64.19	-35.41	-34.20
$J_{2,3}$	11.25	-98.67	22.39	15.92	30.80	28.21
$J_{2,4}$	-13.45	-1.51	-2.97	-3.37	-7.22	0.28
$J_{3,4}$	-146.68	-55.44	-39.02	-38.98	-53.73	-24.84

^a Full geometrical optimizations were also performed at the UBLYP/DZVP level.

Table S3.2 Low-lying excitation states ($\Delta E / \text{cm}^{-1}$) of **1** by solving the Heisenberg spin Hamiltonian ($H = -2JSS$) with the calculated J values at the UBLYP/DZVP level.

States	$S_0(\text{b})$	$S_0(\text{a})$	S_1	$S_2(\text{R})$	$S_2(\text{C})$	S_3
0	$S = 1/2$	$S = 1/2$	$S = 0$	$S = 1/2$	$S = 5/2$	$S = 2$
1	$S = 3/2$ $\Delta E = 41.0$	$S = 3/2$ $\Delta E = 177.8$	$S = 1,$ $\Delta E = 47.7$	$S = 3/2,$ $\Delta E = 28.9$	$S = 7/2,$ $\Delta E = 207.3$	$S = 3,$ $\Delta E = 0.1$
2	$S = 1/2$ $\Delta E = 162.7$	$S = 5/2$ $\Delta E = 477.4$	$S = 1,$ $\Delta E = 124.0$	$S = 5/2,$ $\Delta E = 143.2$	$S = 3/2,$ $\Delta E = 223.9$	$S = 1,$ $\Delta E = 72.6$

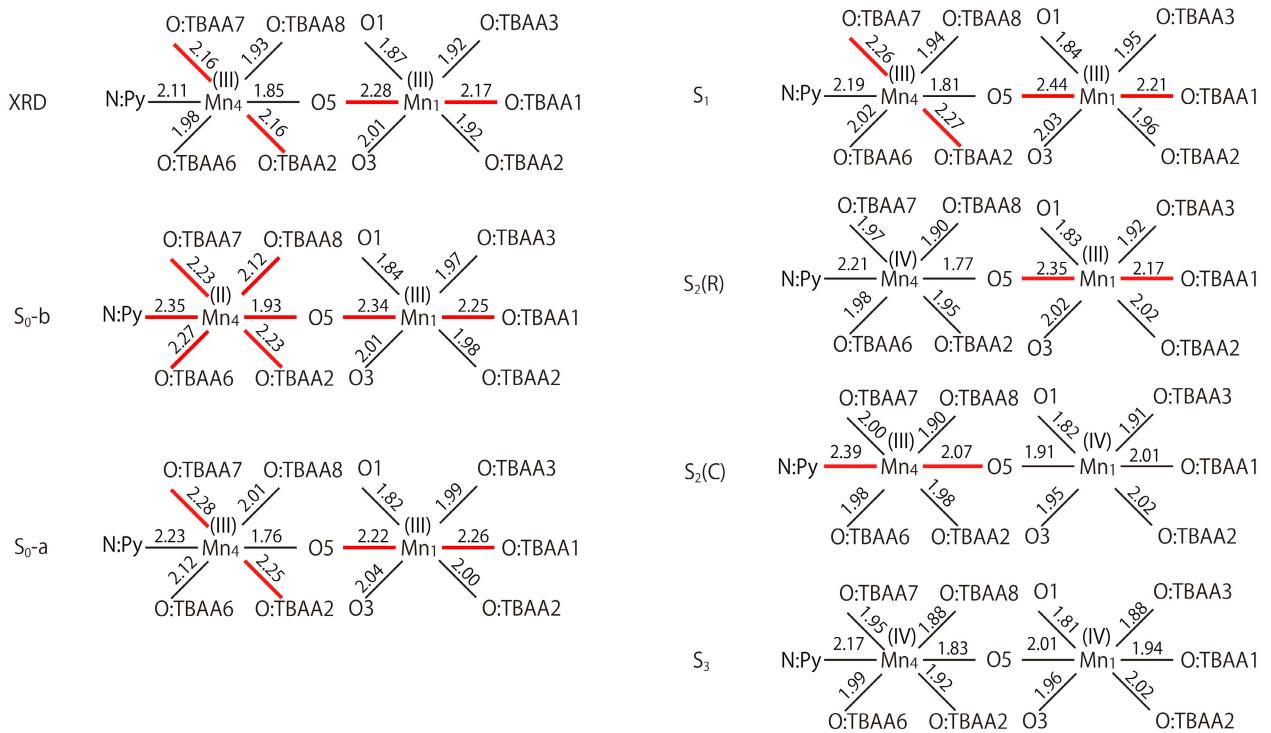
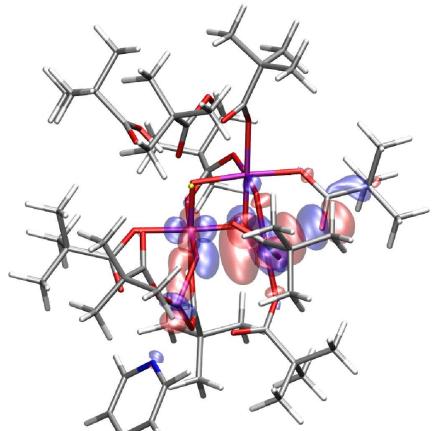
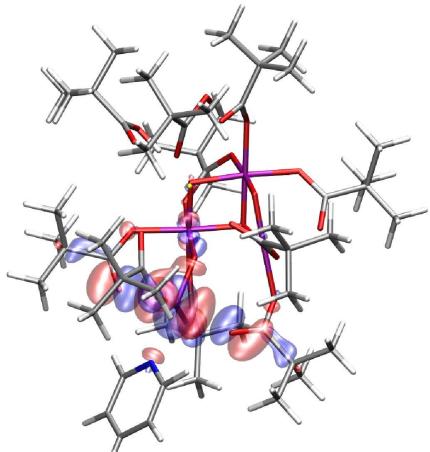


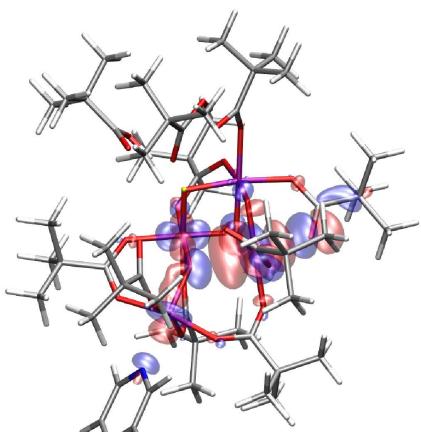
Figure S3.1 The coordination bond lengths of 1 around Mn1 and Mn4 calculated at the UBLYP/DZVP level. The Jahn-Teller axes of Mn(III) atoms are shown as red lines.



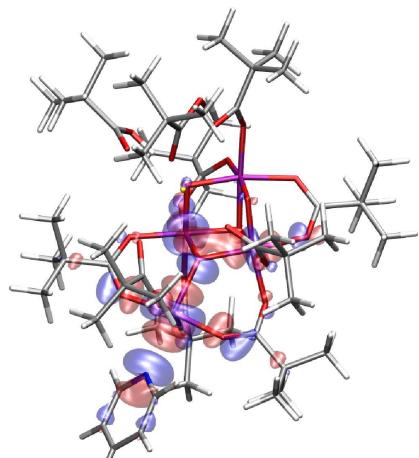
S_1 , LNO(dz^2 , Mn₁)



S_1 , LNO(dz^2 , Mn₄)



$S_2(R)$, LNO(dz^2 , Mn₁)



$S_2(C)$, LNO(dz^2-x^2 , Mn₄)

Figure S3.2 Localized natural orbitals (LNOs) mainly contribute to the Mn Jahn-Teller distortions of **1**. These orbitals were calculated at the UBLYP/DZVP level.

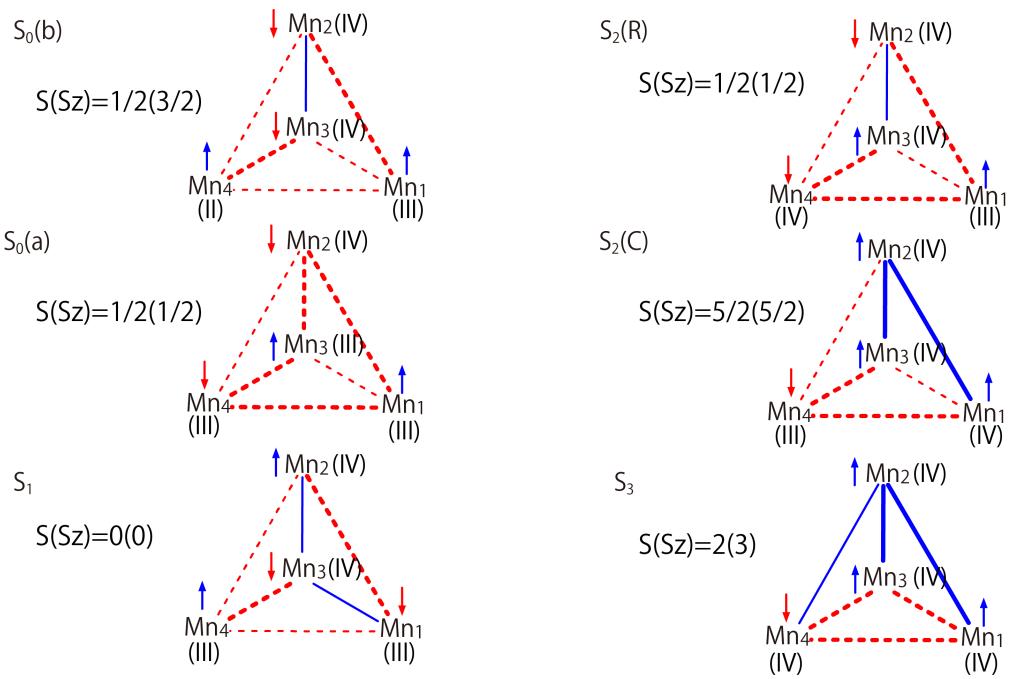


Figure S3.3 Schematic illustrations of the magnetic interactions for the Mn spin sites of **1** in the S_0 , S_1 , S_2 and S_3 states calculated at the UBLYP/DZVP level. Red dashed lines and blue solid lines represent the antiferromagnetic and ferromagnetic interactions, respectively. Bold and thin lines represent strong and weak magnetic interactions, respectively. All the values of the magnetic interactions are summarized in Table S3.1.

S4. Results of **2** calculated at the UB3LYP/DZVP and UBLYP/DZVP levels.

Table S4.1 Key bond distances /Å of **2** optimized at the UBLYP/DZVP level.

System State	Model (2)				Native OEC	
	S ₁	S ₂ -R	S ₂ -L	S ₃	XRD	XRD ^a
Mn1-Mn2	2.79	2.78	2.77	2.78	2.77	2.67
Mn1-Mn3	3.18	3.10	2.88	2.97	3.09	3.24
Mn1-Mn4	3.72	3.64	3.51	3.46	3.60	4.95
Mn2-Mn3	2.76	2.76	2.74	2.75	2.74	2.70
Mn2-Mn4	5.28	5.32	5.41	5.30	5.24	5.17
Mn3-Mn4	3.25	3.29	3.39	3.34	3.23	2.86
O5-Mn1	2.49	2.33	1.91	2.01	2.28	2.70
O5-Mn3	1.85	1.91	1.89	1.97	1.86	2.17
O5-Mn4	1.81	1.77	2.06	1.84	1.85	2.32

^a PDBID: 4UB6, A subunit

Table S4.2 Distances /Å of **2** optimized at the UBLYP/DZVP level.

System	2				Native OEC	
	State	S1	S2-R	S2-L	S3	XRD
Mn1–O1	1.84	1.82	1.82	1.80	1.87	1.85
Mn1–O3	2.02	2.01	1.94	1.95	2.01	1.94
Mn1–O5	2.49	2.33	1.91	2.01	2.28	2.70
Mn2–O1	1.80	1.80	1.82	1.84	1.78	1.83
Mn2–O2	1.89	1.91	1.86	1.87	1.88	1.90
Mn2–O3	1.87	1.90	1.92	1.93	1.86	1.95
Mn3–O2	1.81	1.79	1.82	1.81	1.84	1.78
Mn3–O3	1.89	1.85	1.87	1.86	1.86	2.08
Mn3–O5	1.85	1.91	1.89	1.97	1.85	2.17
Mn4–O5	1.81	1.77	2.06	1.84	1.85	2.32
Ca–O1	2.55	2.58	2.54	2.59	2.50	2.57
Ca–O2	2.49	2.51	2.53	2.58	2.41	2.67
Ca–O5	2.91	3.23	3.00	3.34	2.70	2.59
Ca–Mn1	3.62	3.69	3.60	3.76	3.46	3.51
Ca–Mn2	3.40	3.43	3.45	3.48	3.36	3.35
Ca–Mn3	3.59	3.73	3.63	3.83	3.44	3.47
Ca–Mn4	4.27	4.53	4.48	4.63	4.06	3.76

Table S4.3 Calculated effective exchange integrals (J /cm⁻¹ values) of **2** at the UB3LYP/DZVP (UBLYP/DZVP) level.^a

	S1	S2-R	S2-L	S3
$J_{1,2}$	-26.1 (-66.28)	-18.44 (-55.77)	39.16 (39.68)	36.04 (29.09)
$J_{1,3}$	15.83 (5.70)	13.26 (-4.80)	-6.80 (-6.48)	-16.01 (-32.86)
$J_{1,4}$	-4.62 (-12.00)	-19.43 (-73.98)	-7.80 (-42.80)	-10.89 (-35.69)
$J_{2,3}$	16.59 (20.58)	14.97 (12.13)	31.31 (35.79)	26.72 (26.26)
$J_{2,4}$	-0.09 (-3.90)	2.54 (-4.39)	-1.17 (-8.61)	1.50 (0.27)
$J_{3,4}$	-2.60 (-38.32)	-8.36 (-27.91)	-10.47 (-49.48)	-5.63 (-24.06)

^a Geometrical optimizations were performed at the UBLYP/DZP level.

Table S4.4 Low-lying energy levels of **2** by solving the Heisenberg spin Hamiltonian ($H = -2JS\mathbf{S}$) using the J values calculated at the UB3LYP/DZVP (UBLYP/DZVP) level.

Energy states	S_1	$S_2\text{-R}$	$S_2\text{-L}$	S_3
0	$S = 0$ ($S = 0$)	$S = 3/2$ ($S = 1/2$)	$S = 5/2$ ($S = 5/2$)	$S = 3$ ($S = 2$)
1	$S = 1, \Delta E = 4.03$ ($S = 1, \Delta E = 52.74$)	$S = 1/2, \Delta E = 4.39$ ($S = 3/2, \Delta E$ =8.06)	$S = 7/2, \Delta E = 44.13$ ($S = 7/2, \Delta E$ =223.10)	$S = 2, \Delta E = 22.49$ ($S = 3, \Delta E = 17.32$)
2	$S = 2, \Delta E = 14.11$ ($S = 1, \Delta E$ =123.11)	$S = 5/2, \Delta E = 16.69$ ($S = 5/2, \Delta E$ =70.72)	$S = 9/2, \Delta E$ =101.78 ($S = 3/2, \Delta E$ =273.82)	$S = 4, \Delta E = 37.51$ ($S = 1, \Delta E = 58.21$)

Table S4.5 Local spin correlation factor c for low-lying energy levels at the UB3LYP/DZVP (UBLYP/DZVP) level.^a

State	Index	S	c_1	c_2	c_3	c_4
$S_0(b)^b$	0	(1/2)	(-0.071	-0.194	-0.990	2.255)
	1	(3/2)	(0.843	-0.493	-0.528	1.179)
	2	(5/2)	(0.526	0.058	-0.251	0.666)
$S_0(a)^c$	0	(1/2)	(1.739	-0.998	1.585	-1.326)
	1	(3/2)	(0.609	-0.187	0.490	0.087)
	2	(5/2)	(0.442	-0.069	0.341	0.285)
S_1	0	0	-	-	-	-
		(0)	(-	-	-	-)
	1	1	-0.159	0.361	0.489	0.308
		(1)	(0.649	-0.246	-0.031	0.628)
	2	2	-0.084	0.272	0.445	0.367
		(1)	(-0.432	0.297	-0.201	1.336)
$S_2(R)$	0	3/2	0.170	0.318	0.724	-0.213
		(1/2)	(0.686	-0.593	1.665	-0.759)
	1	1/2	1.068	-0.742	1.609	-0.935
		(3/2)	(-0.171	0.410	0.758	0.002)
	2	5/2	0.121	0.378	0.525	-0.025
		(5/2)	(-0.267	0.437	0.594	0.235)
$S_2(C)$	0	5/2	0.527	0.537	0.503	-0.568
		(5/2)	(0.509	0.558	0.491	-0.558)
	1	7/2	0.370	0.379	0.342	-0.092
		(7/2)	(0.352	0.400	0.329	-0.082)
	2	9/2	0.297	0.304	0.273	0.123
		(3/2)	(0.495	0.848	0.426	-0.771)
S_3	0	3	0.416	0.467	0.478	-0.361
		(2)	(0.290	0.690	0.467	-0.448)
	1	2	0.387	0.654	0.444	-0.486
		(3)	(0.372	0.482	0.480	-0.335)
	2	4	0.292	0.351	0.364	-0.008
		(1)	(0.320	1.140	0.238	-0.700)

^a $c_i = \langle s_i S \rangle / \langle S S \rangle$

^b Optimized at UDUD spin state at the (BLYP) method

^c Optimized at UDDU spin state at the (BLYP) method

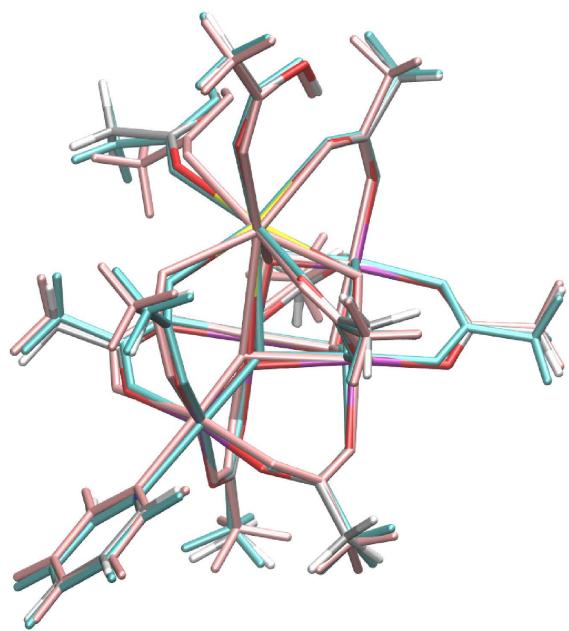


Figure S4.1 Superimposed structures in the S₁(pink), S₂-R(element), S₂-L(cyan) states.

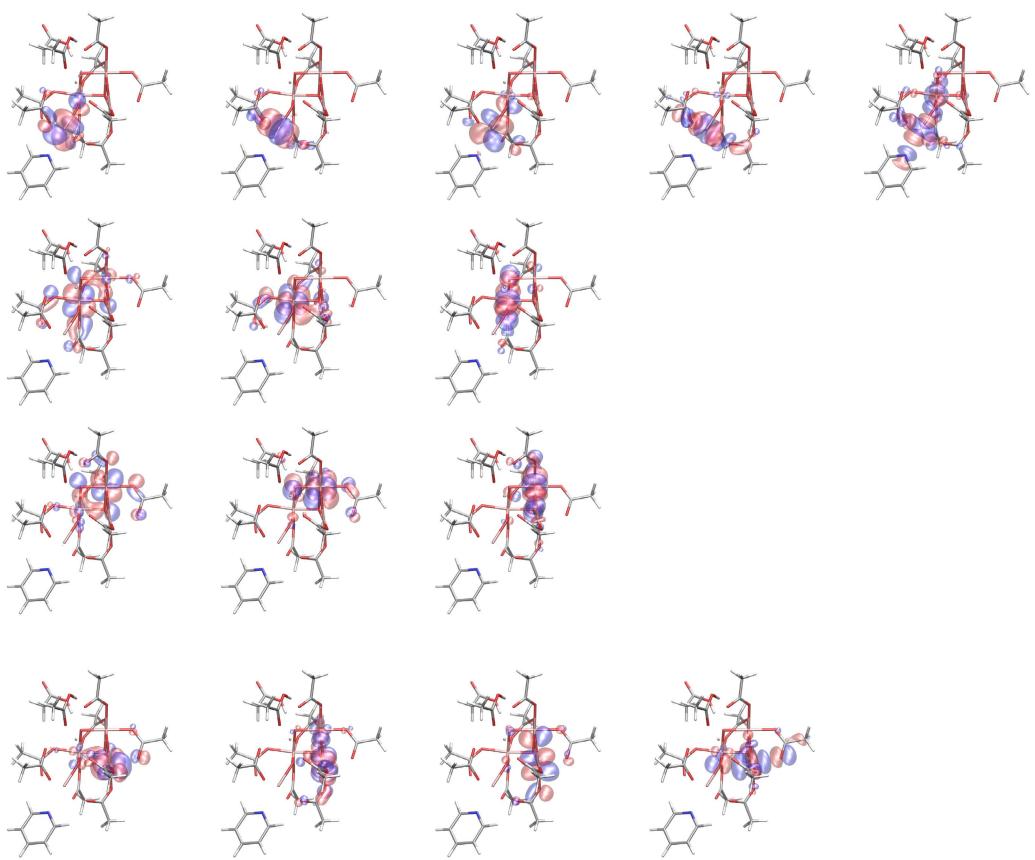


Figure S4.2 Localized natural orbitals of **2** in the highest spin $S_0(b)$ state calculated at the BLYP/DZVP level. 14 d-type orbitals; 4 for Mn1, 3 for Mn2, 3 for Mn3 and 5 for Mn4 are shown.

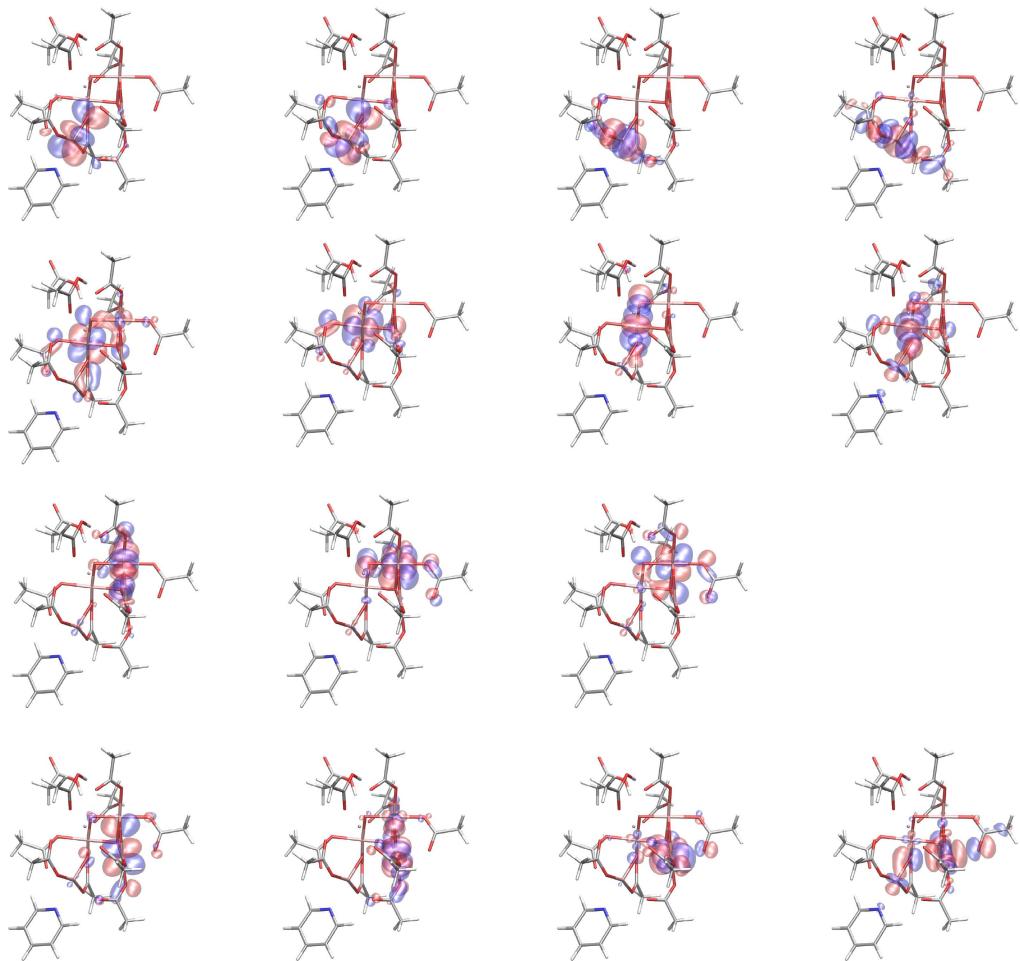


Figure S4.3 Localized natural orbitals of **2** in the highest spin $S_0(a)$ state calculated at the BLYP/DZVP level. 15 d-type orbitals; 4 for Mn1, 3 for Mn2, 4 for Mn3 and 4 for Mn4 are shown.

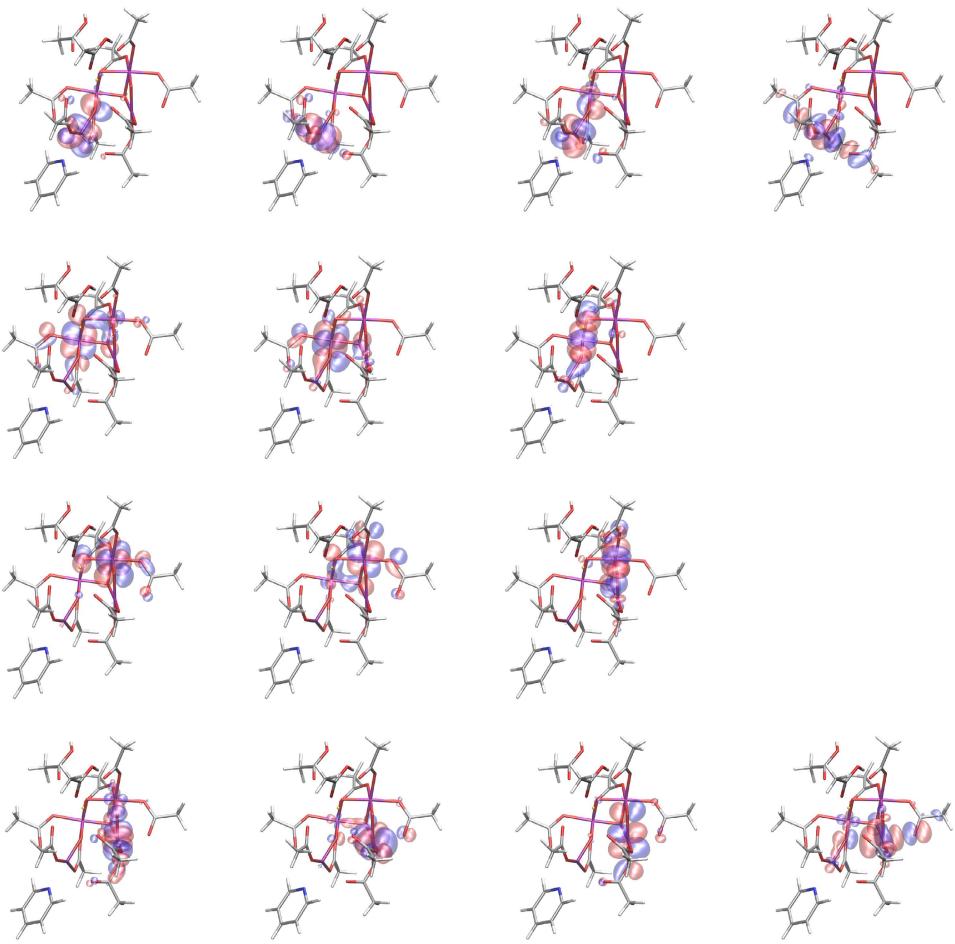


Figure S4.4 Localized natural orbitals of **2** in the highest spin S_1 state calculated at the BLYP/DZVP level. 14 d-type orbitals; 4 for Mn1, 3 for Mn2, 3 for Mn3 and 4 for Mn4 are shown.

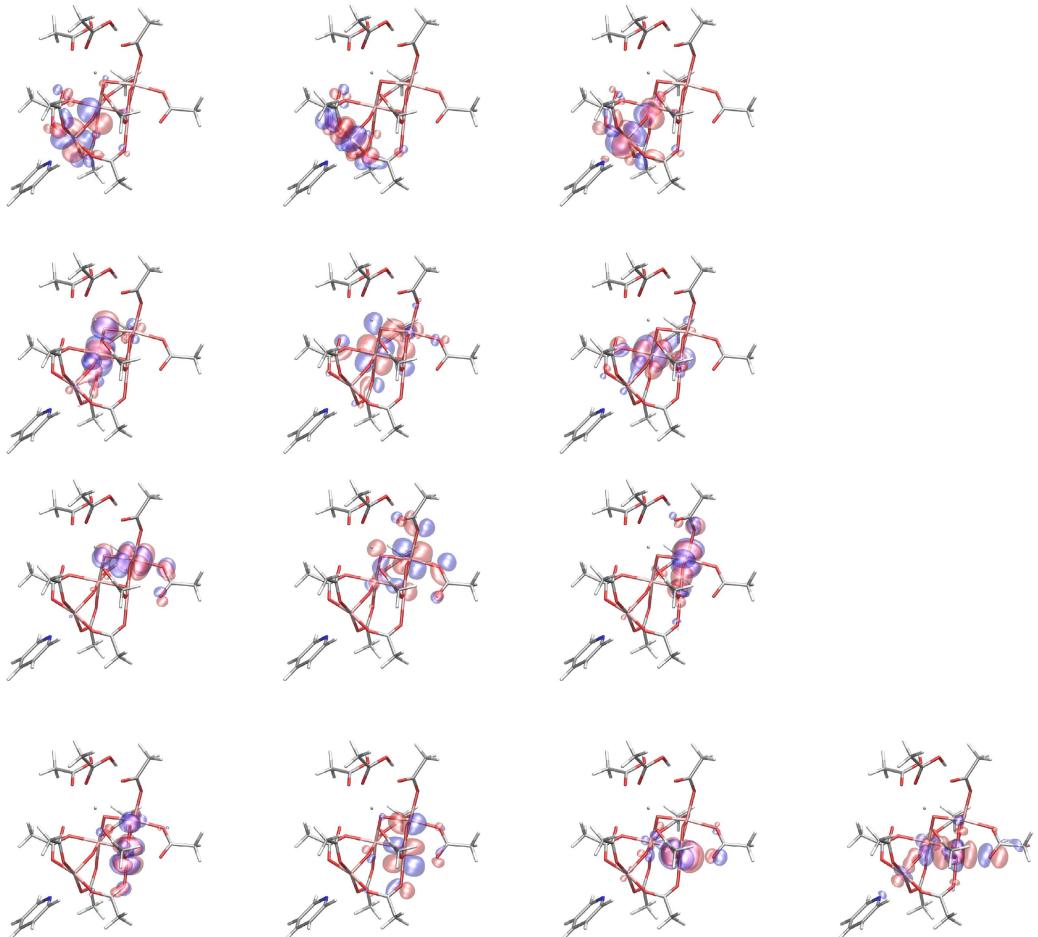


Figure S6. mshoji et al

Figure S4.5 Localized natural orbitals of **2** in the highest spin S_2-R state calculated at the BLYP/DZVP level. 13 d-type orbitals; 4 for Mn1, 3 for Mn2, 3 for Mn3 and 3 for Mn4 are shown.

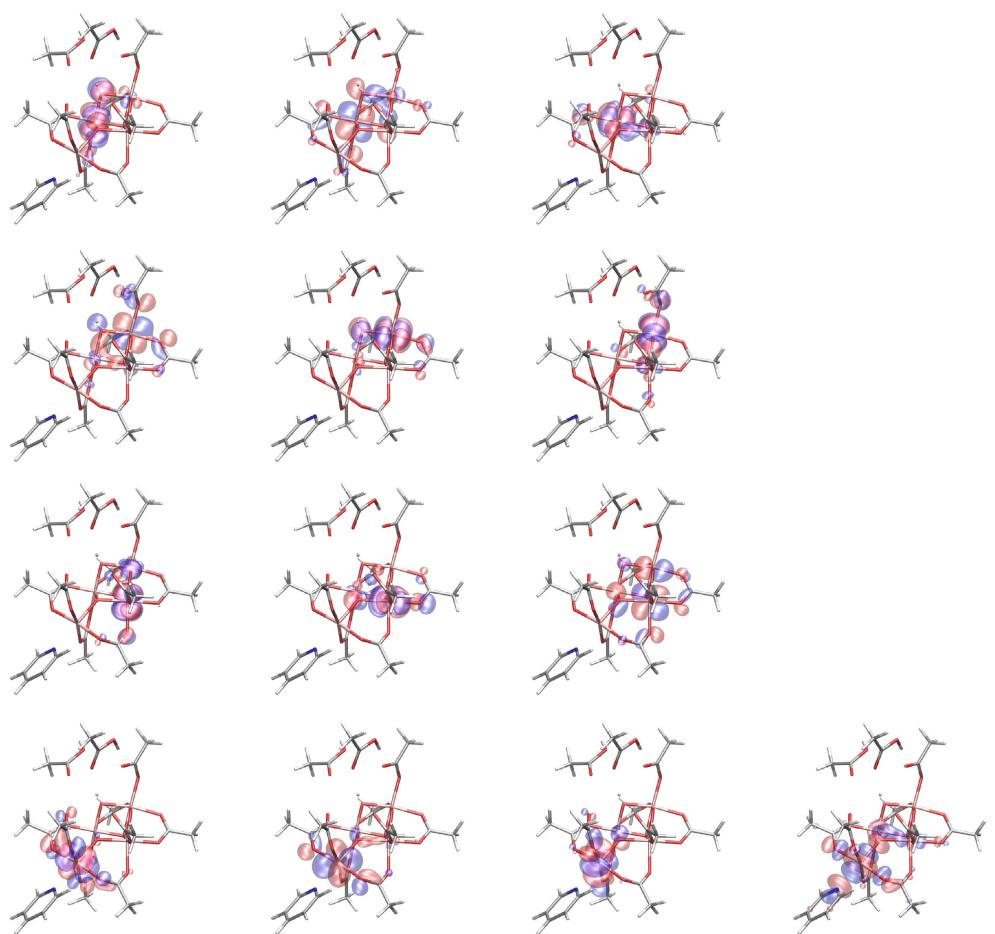


Figure S4.6 Localized natural orbitals of 2 in the highest spin S_2 -C state calculated at the BLYP/DZVP level. 13 d-type orbitals; 3 for Mn1, 3 for Mn2, 3 for Mn3 and 4 for Mn4 are shown.

S5. Theoretical modeling of the synthetic model (**2**)

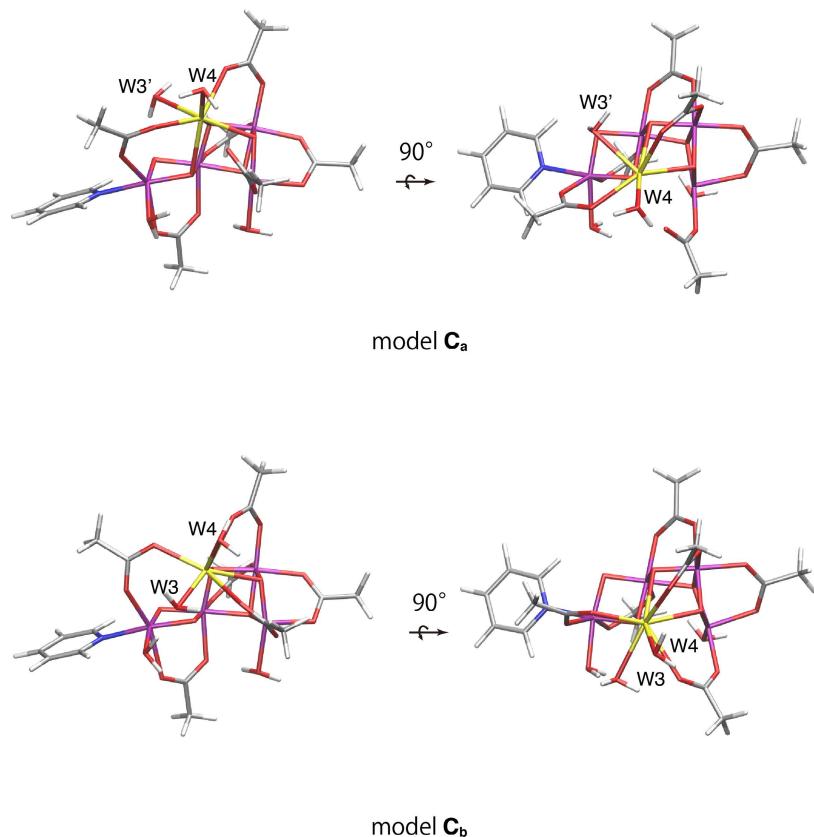


Figure S5.1 Molecular structures of the model **Ca** and model **Cb**.

2, S1, Opted at UB3LYP/DZVP(UUUU)

Mn 14.10881401 4.07108691 15.24148354
Mn 14.75926136 2.71783845 17.59840256
Mn 16.54640248 4.72021313 17.14373855
Mn 15.79044508 7.28776382 15.32446540
Ca 13.37399506 5.70764149 18.33108805
O 13.40793213 3.63975784 16.86929369
O 15.36453949 4.26779356 18.41946016
O 15.80001672 3.33369909 16.19268639
O 15.31608186 5.81798448 16.28834249
O 14.07478131 7.99537148 18.08622844
O 14.35964340 8.46236443 15.90269433
O 17.30071580 6.17171439 18.25349744
O 17.24376553 7.93205771 16.83871995
O 18.02612238 4.91786279 15.82189602
O 17.32311175 6.45133504 14.33001330
O 14.88884639 4.49656358 13.51010270
O 14.67053658 6.73821560 13.48439443
O 14.64491910 4.01387132 21.81921196
O 14.32949550 5.90420085 20.71319436
O 11.43602555 6.65830545 19.57197951
O 10.55573323 5.10170116 20.93830755
O 12.38278963 3.78709470 19.57121392
O 13.74175048 2.05478368 19.11942072
O 16.38359370 1.76206910 18.35654844
O 17.83920756 3.45951047 17.99068909
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O 13.83102710 2.02760241 14.62693387
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O 11.92519914 6.18599186 16.46078490
N 16.38626283 9.02228693 14.19943411
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C 16.58051609 5.45892643 21.42096024
C 10.64648543 6.32943084 20.45396358
C 9.68392873 7.29489623 21.09833621
C 12.77200613 2.60313162 19.74258848
C 12.06914967 1.74077114 20.77284920
C 13.82906620 8.65960406 17.06386007
C 12.79977788 9.77460731 17.11238953
C 11.71031547 5.74537364 15.31706838
C 10.46876809 6.17568967 14.55459547
C 17.48880840 7.40247143 17.94777422
C 18.03261974 8.27706600 19.06245664
C 18.09288441 5.52201988 14.71565204
C 19.18232289 5.07966356 13.76087036
C 17.53300308 2.28720576 18.38324111
C 18.67709099 1.43856656 18.90233224
C 13.95143402 1.04885473 15.39817805
C 13.64652174 -0.34430654 14.88216971
C 14.77288794 5.62755703 12.91260135
C 14.76310581 5.55884570 11.39797028
C 17.64081599 9.49344147 14.26930259
C 18.05144894 10.61066354 13.54792419
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H 18.56710246 9.13354113 18.64586286
H 17.18087684 8.65228000 19.64271417
H 19.41519128 5.86688925 13.04109655
H 20.07484506 4.78266412 14.31645778
H 18.81742130 4.20087991 13.21583470
H 14.76115125 6.55907835 10.96087865
H 15.63667144 4.99842512 11.04988171
H 13.87409922 5.00809503 11.07111478
H 9.80793398 6.76131839 15.19616781

H 10.76945280 6.77853023 13.69012933
H 9.94250852 5.29605299 14.17119542
H 13.38799986 -0.31134516 13.82266427
H 14.51570909 -0.99116007 15.03862488
H 12.81545805 -0.76920195 15.45530319
H 12.46079483 0.72272874 20.76937964
H 10.99398612 1.72583962 20.56518613
H 12.20974873 2.18925939 21.76244436
H 8.66509790 6.89823672 21.04852282
H 9.93816769 7.40965687 22.15779083
H 9.73772341 8.26328765 20.60047950
H 11.81791676 9.32512587 16.92166042
H 12.78114587 10.22393355 18.10783714
H 12.99080989 10.53230501 16.34874334
H 17.12022257 4.73101676 22.03478180
H 16.68874529 6.45186571 21.86754948
H 17.01500809 5.49337186 20.41597644
H 19.39235780 2.05612821 19.45153991
H 18.29973285 0.62792178 19.52870118
H 19.20083629 1.00257512 18.04337529
H 11.23487386 4.51965993 20.46728240
H 15.38185995 3.48395544 22.16354458

A, S1, Opted at UB3LYP/DZVP(UUUU)

Mn 14.06870948 4.11402047 15.26149347
Mn 14.70537549 2.71978665 17.54242954
Mn 16.48273909 4.75120424 17.11079726
Mn 15.75303259 7.34523403 15.31470683
Ca 13.28540075 5.72137716 18.38764153
O 13.33220087 3.67892169 16.82449672
O 15.30559586 4.26769540 18.36309826
O 15.73740853 3.37984532 16.11652332
O 15.23282949 5.85400305 16.22574338
O 14.04696186 7.97242967 18.10050760
O 14.40825991 8.55594803 15.95893510
O 17.23242893 6.17050291 18.21343575
O 17.32332454 7.82200218 16.68304996
O 18.00305278 5.03873217 15.77817716
O 17.03036536 6.20894235 14.12073311
O 14.70020970 4.14502194 21.81963746
O 14.37191774 5.97279566 20.61920135
O 11.32606519 6.58684488 19.57846570
O 10.46780820 5.02507131 20.94741765
O 12.37732858 3.73990176 19.56421543
O 13.72595886 2.03093904 19.01474268
O 16.32141987 1.77763460 18.22761010
O 17.76679433 3.48384697 17.88201675
O 14.25113124 1.14077013 16.49875200
O 13.87029992 2.12884381 14.50230238
O 12.58507174 5.18064543 14.59842366
O 11.80809011 6.12847049 16.48739636
N 16.44229650 9.02064644 14.17892597
C 15.11947142 5.32578777 21.32766460
C 16.52327452 5.76711759 21.66214358
C 10.53129343 6.25349476 20.45821776
C 9.55192959 7.20318288 21.09191805
C 12.76828464 2.55718593 19.69450124
C 12.09417401 1.63647395 20.68629651
C 13.90189604 8.74470800 17.14149501
C 13.05974376 9.99775022 17.27305659
C 11.65267852 5.77390885 15.31614004
C 10.34873220 5.99987460 14.58613549
C 17.54764522 7.35124954 17.82821981
C 18.21893032 8.22719274 18.86201066
C 18.01339870 5.54832328 14.63296133
C 19.24847662 5.38272394 13.77791122
C 17.47307747 2.30145317 18.26365526
C 18.61146890 1.47322557 18.81275103
C 13.94905532 1.11621387 15.25335611
C 13.65799042 -0.25267431 14.68239828
C 17.75419803 9.30681806 14.09973432
C 18.23329005 10.40017532 13.38710074
C 17.32133722 11.22879814 12.73324293
C 15.96137814 10.93448779 12.82327517
C 15.55889320 9.82449597 13.55898182
H 18.41968356 8.64905416 14.64497420
H 19.30002873 10.59525646 13.35373376
H 17.66449108 12.09063240 12.16845590
H 15.21543202 11.55389846 12.33657022
H 14.51688760 9.55308008 13.66812991
H 18.90644334 7.64207235 19.47743441
H 18.74427611 9.05296687 18.37933384
H 17.44267015 8.64249098 19.51594162
H 19.71299066 6.36378732 13.62664256
H 19.96197090 4.70857615 14.25244650
H 18.97129277 5.00620750 12.78800444
H 9.70970184 6.68172663 15.14870446
H 10.52883504 6.38604709 13.57859519
H 9.83831787 5.03570743 14.47524879
H 13.52860689 -0.20142779 13.60041196
H 14.46738790 -0.94207412 14.93975251
H 12.74198506 -0.64078603 15.14141001
H 12.69227407 0.74177441 20.86330937
H 11.12124892 1.33537219 20.27970362

H 11.91518457 2.16880220 21.62461938
H 8.56387939 6.74020618 21.16304510
H 9.88160559 7.42956457 22.11249467
H 9.50193171 8.12773034 20.51647821
H 12.04911340 9.77409688 16.91023283
H 12.99069830 10.29317352 18.32144388
H 13.46416514 10.81242978 16.66751734
H 16.91958046 5.30698665 22.57227194
H 16.53588402 6.85348605 21.76858668
H 17.16392413 5.50668312 20.81116697
H 18.96262344 1.92050760 19.74933018
H 18.28667032 0.44813103 18.99436984
H 19.44850084 1.48927926 18.10864419
H 11.15475703 4.45974164 20.48695176
H 15.39179352 3.73192197 22.36312941
O 15.07017394 4.35886153 13.43193946
H 15.89659262 4.87921721 13.58228141
H 15.31166377 3.44805825 13.18233768
O 14.28744729 7.08144553 13.50130824
H 13.55601275 6.48105743 13.77911684
H 14.66554828 6.65297701 12.71703136

B, S1, Opted at UB3LYP/DZVP(UUUU)

Mn 13.57355751 3.77899561 15.28536082
Mn 14.67324340 2.93548352 17.63881983
Mn 16.14105091 5.03858711 16.63525343
Mn 15.81429616 7.69264257 16.33741770
Ca 12.96613552 5.86088011 18.06289579
O 13.09580513 3.63494376 17.02091314
O 15.27930026 4.56166435 18.17272223
O 15.30126855 3.41969878 15.97698251
O 14.74654859 6.14562053 16.20551412
O 13.02072530 8.33423275 17.69462992
O 15.12939858 8.97376571 18.06331922
O 16.98335522 6.55531340 17.12140307
O 16.83380272 5.12763535 14.66235632
O 16.62034323 7.34147988 14.26803829
O 15.25998672 5.15960196 21.40973167
O 14.19691804 6.62483425 20.14540500
O 10.93636640 6.53420295 19.34457908
O 10.65358946 5.23338906 21.15851540
O 12.58714882 4.07358499 19.81619427
O 14.02813557 2.40373372 19.38439188
O 16.47684548 2.16457638 18.05092165
O 17.68790391 3.85661542 17.14328706
O 14.16069568 1.12477246 16.93800940
O 13.40862077 1.68003579 14.87431539
O 11.96308776 4.49264000 14.56745611
O 11.36557504 5.90741929 16.21425606
N 17.26886555 9.27202508 16.25840604
C 15.25929133 6.24008760 20.60503159
C 16.57765629 6.91048815 20.32861573
C 10.33846490 6.25695179 20.38206415
C 9.16762421 7.05733125 20.89389131
C 13.15111315 2.98859006 20.10715278
C 12.82007326 2.30882876 21.42120884
C 13.89992583 8.99228217 18.35033839
C 13.43278992 9.82270781 19.53015307
C 11.14093076 5.31642395 15.14821627
C 9.83440265 5.52029296 14.40586045
C 16.96159333 6.18660255 13.93567529
C 17.54596674 5.95787517 12.55093397
C 17.56350103 2.72533795 17.70646607
C 18.83930900 1.94933436 17.96847071
C 13.68487085 0.84905965 15.78857196
C 13.39690608 -0.61821820 15.52885777
C 18.56670308 8.98072147 16.43763688
C 19.55812294 9.95495607 16.35865847
C 19.18798461 11.27040507 16.08127418
C 17.83755712 11.56657268 15.89504803
C 16.90667346 10.53700740 15.99531963
H 18.78020464 7.93846596 16.65497741
H 20.59659269 9.68074536 16.51422113
H 19.93840744 12.05321685 16.01247088
H 17.50463580 12.57633461 15.67773480
H 15.84395203 10.70892744 15.86373472
H 17.82260284 6.90786222 12.09039697
H 18.41349776 5.29443925 12.61073438
H 16.79481872 5.46636496 11.92057328
H 9.23824443 6.29635112 14.88824521
H 10.03326747 5.78724005 13.36300823
H 9.27573785 4.57764688 14.39705298
H 13.38274021 -0.82349402 14.45635056
H 14.13156203 -1.24862919 16.03519824
H 12.40803206 -0.85576884 15.93946687
H 13.17401464 1.27684809 21.43044096
H 11.74164397 2.34095940 21.60281679
H 13.31194302 2.86595858 22.22725159
H 8.29113972 6.40870877 20.99373977
H 9.39612943 7.45020667 21.89027137
H 8.95037558 7.87744433 20.20933272
H 12.48736691 10.31736403 19.29218511
H 13.25427821 9.14458161 20.37201311

H 14.18862193 10.55663496 19.81880562
H 17.34706148 6.66565554 21.06818803
H 16.42265782 7.99052479 20.28920426
H 16.90933382 6.60719021 19.32532686
H 19.67019665 2.63357501 18.15528246
H 18.70473346 1.26084095 18.80525174
H 19.07694840 1.36196633 17.07337330
H 11.42878301 4.74147849 20.73231819
H 16.17302544 4.90999942 21.62835366
O 14.62599263 3.93564273 13.45700667
H 15.52685237 4.17218233 13.80793208
H 14.68190084 3.00032594 13.18942359
O 14.24947210 8.73806560 15.38258402
H 13.56610834 8.58839435 16.10959638
H 13.99284103 8.15759485 14.64535533

75

Ca, S1, Opted at UB3LYP/DZVP(UUUU)
 Mn 13.53028411 3.83193132 15.24287942
 Mn 14.76622535 2.87658255 17.48772911
 Mn 16.13573685 5.00064347 16.38768220
 Mn 15.60067606 7.61966284 15.87481822
 Ca 13.19220169 5.81779660 18.06713020
 O 13.16037205 3.60645546 16.99818142
 O 15.39372241 4.52636517 17.95535487
 O 15.29561448 3.38390816 15.77321883
 O 14.63256166 6.01143977 15.89188633
 O 12.67589095 8.18734932 17.89395846
 O 14.55439919 9.05223596 17.03774733
 O 16.78655817 6.63187225 16.85596137
 O 16.86679613 5.01369041 14.40860321
 O 16.60818822 7.16967408 13.81977671
 O 15.18539075 6.85821135 19.32642443
 O 10.93614640 6.51053352 19.06928070
 O 13.35052557 4.32728035 19.98191392
 O 14.24651561 2.38748386 19.24637707
 O 16.63233695 2.17534707 17.77780926
 O 17.72848514 3.93838919 16.85546027
 O 14.21986536 1.10307577 16.79457298
 O 13.31289123 1.71180901 14.80735248
 O 11.83639994 4.50169061 14.66091480
 O 11.21907292 5.70223034 16.45793312
 N 17.00536150 9.23422631 15.75948472
 C 13.67578568 3.13864767 20.13848008
 C 13.40382171 2.42845884 21.45124590
 C 13.47541311 9.13957347 17.70625811
 C 13.15468853 10.50750965 18.28959985
 C 10.96590625 5.14289960 15.37646487
 C 9.56693370 5.17132163 14.79389212
 C 17.03765230 6.00578670 13.61788560
 C 17.84500534 5.73574259 12.35944571
 C 17.67744451 2.80204719 17.43330758
 C 19.01091102 2.15673850 17.74798880
 C 13.66084664 0.85806213 15.67141491
 C 13.38246222 -0.60778745 15.39643170
 C 18.30183217 9.00465908 16.02326334
 C 19.26266303 10.00640032 15.91864797
 C 18.86438086 11.28262405 15.52270139
 C 17.51581045 11.51508246 15.25222610
 C 16.61435400 10.46384951 15.38777538
 H 18.53697643 7.99165907 16.33476991
 H 20.29975917 9.78256663 16.14656387
 H 19.59171566 12.08453903 15.43063075
 H 17.16032312 12.49378928 14.94667164
 H 15.55233667 10.59001841 15.21245845
 H 17.53486600 6.40239258 11.55154527
 H 18.90031460 5.93794403 12.58063991
 H 17.76371594 4.69097941 12.05116538
 H 8.93017911 5.86418111 15.34658429
 H 9.60395575 5.45121391 13.73683000
 H 9.14014822 4.16317383 14.84712515
 H 13.08993442 -0.75883042 14.35580798
 H 14.26352102 -1.20851117 15.63939240
 H 12.56825687 -0.94003200 16.05074896
 H 14.30408115 1.90900067 21.79304679
 H 12.63004447 1.66891913 21.29182823
 H 13.06300799 3.14113445 22.20401295
 H 12.97964629 11.21972766 17.47480385
 H 12.27115627 10.46765729 18.92923701
 H 14.01254165 10.87189603 18.86336280
 H 19.50431252 2.73127248 18.54017189
 H 18.87094190 1.12699981 18.07994765
 H 19.65544201 2.19043586 16.86483796
 O 14.42326200 3.95421787 13.32086621
 H 15.35963296 4.10490973 13.60447246
 H 14.38003538 3.01902496 13.04739790
 O 14.31049805 8.36364467 14.25481149
 H 13.50153273 7.83288618 14.35977740

H 14.83299038 7.92282328 13.55260491

H 11.17107116 7.44787218 18.91574531

H 10.45243660 6.26816720 18.25684405

H 15.87642009 6.92662742 18.62947852

H 15.43246609 6.03900498 19.78856112

Cb, S1, Opted at UB3LYP/DZVP(UUUU)

Mn	13.68997817	3.08343140	14.98307146	C	18.85126488	2.69957383	18.43259770
Mn	14.68315433	2.99647455	17.52612647	C	14.04378000	0.41464274	16.20395732
Mn	16.02507272	4.95229353	16.09909821	C	14.00696888	-1.10026003	16.29079160
Mn	15.50395486	7.46726724	15.27938792	C	18.14660871	8.96524095	15.34798940
Ca	12.86106942	5.74261611	17.11577527	C	19.06941739	9.99696279	15.19751951
O	13.09193754	3.35613998	16.65884478	C	18.63781600	11.21566116	14.67576160
O	15.10106270	4.74914510	17.66217575	C	17.29548818	11.36165617	14.32487256
O	15.37037780	3.13386848	15.80402943	C	16.43434335	10.28434466	14.50879868
O	14.57909281	5.81655068	15.36540929	H	18.40868886	7.99483921	15.75856231
O	13.15317404	7.87403110	18.10783726	H	20.10288595	9.84071016	15.48940780
O	14.49171774	8.83637876	16.58377856	H	19.33446097	12.03952345	14.54698139
O	16.71221884	6.59307396	16.28609441	H	16.91573513	12.29320252	13.91774517
O	16.80685367	4.68736988	14.17114858	H	15.38079304	10.34183794	14.25927234
O	16.45211661	6.75340740	13.32209547	H	18.69833213	4.95468837	12.26051639
O	11.83858972	7.49762445	15.33292670	H	17.22958412	4.23027858	11.60541881
O	10.43078878	5.82946481	17.82062067	H	17.61111236	5.93843388	11.23865487
O	12.40862089	4.55581662	19.23344127	H	8.99442629	4.60627244	13.79639350
O	13.91421199	2.87443090	19.27868726	H	10.02196665	4.04727962	12.44075583
O	16.49513128	2.55246612	18.19550346	H	9.43521627	2.88172634	13.62285819
O	17.61724462	4.07078549	16.93564068	H	13.62788864	-1.53497375	15.36438826
O	14.39456990	1.01824941	17.26779279	H	15.01693928	-1.47456798	16.49066236
O	13.73428657	0.95980017	15.10184672	H	13.37738661	-1.40299695	17.13334240
O	12.02411636	3.36755210	14.02745193	H	13.38161956	3.31450107	21.86946977
O	11.05031990	4.98484594	15.24711345	H	12.32519107	2.07890333	21.18704722
N	16.85631991	9.11122240	15.00656334	H	11.67951665	3.72097607	21.51638798
C	12.96437789	3.58283208	19.78586915	H	14.46727769	10.87550409	18.07886811
C	12.54680526	3.15053249	21.17926570	H	13.13355201	10.26617008	19.11685939
C	13.88666992	8.81394659	17.70452267	H	14.79282524	9.73874159	19.40632427
C	14.07589412	10.01316847	18.62300423	H	19.02976925	3.32294781	19.31707520
C	11.03373945	4.12040909	14.33510850	H	18.78687155	1.65785173	18.75344002
C	9.78384369	3.91280158	13.50100672	H	19.68245684	2.84299290	17.73873793
C	16.90694175	5.59134994	13.25393713	O	14.83088109	2.98958924	13.20704660
C	17.65353103	5.15911073	12.00134141	H	15.05011644	2.04302521	13.14366539
C	17.55023166	3.13937831	17.79417843	H	15.66328410	3.44334961	13.51754542
				O	14.08225583	8.19913733	13.88109631
				H	13.21364027	8.00530875	14.31933537

H 14.18385897 7.55923091 13.15477853
H 10.61627765 5.31665156 18.63199396
H 10.07158843 5.19751337 17.17340686
H 11.37832288 8.12571589 15.91215565
H 11.20301821 6.78862955 15.08842579

S6. Ca substitutions of the synthetic model (**2**)

Table S6.1 Key bond distances / \AA of the model complex (**MMn₄O₄**) in the S1 state.

	Model (MMn₄O₄)							Native OEC XRD ^a
	A=Ba	A=Mg	A=Mn	A=Sr	A=Zn	A=Ca	A=Ca	
	XRD							
Mn1–Mn2	2.80	2.83	2.81	2.79	2.82	2.79	2.77	2.67
Mn1–Mn3	3.17	3.15	3.14	3.16	3.14	3.18	3.09	3.24
Mn1–Mn4	3.66	3.59	3.64	3.64	3.60	3.72	3.60	4.95
Mn2–Mn3	2.72	2.71	2.73	2.72	2.71	2.76	2.74	2.70
Mn2–Mn4	5.22	5.20	5.22	5.21	5.21	5.28	5.24	5.17
Mn3–Mn4	3.24	3.21	3.23	3.24	3.21	3.25	3.23	2.86
O5–Mn1	2.41	2.21	2.27	2.39	2.22	2.49	2.28	2.70
O5–Mn3	1.86	1.84	1.84	1.86	1.83	1.85	1.86	2.17
O5–Mn4	1.82	1.80	1.81	1.82	1.81	1.81	1.85	2.32

^a PDBID: 4UB6, A subunit

Table S6.2 Distances /Å of the model complex (Mn_4AO_4) in the S1 state.

	Model (Mn_4AO_4)						Native OEC
	A=Ba	A=Mg	A=Mn	A=Sr	A=Zn	A=Ca	A=Ca XRD ^a
	XRD						
Mn1–O1	1.82	1.81	1.83	1.83	1.82	1.84	1.87
Mn1–O3	2.08	2.17	2.10	2.07	2.15	2.02	2.01
Mn1–O5	2.41	2.21	2.27	2.39	2.22	2.49	2.28
Mn2–O1	1.79	1.81	1.80	1.79	1.81	1.80	1.78
Mn2–O2	1.85	1.84	1.86	1.85	1.87	1.89	1.88
Mn2–O3	1.85	1.83	1.85	1.85	1.83	1.87	1.86
Mn3–O2	1.79	1.82	1.82	1.79	1.82	1.81	1.84
Mn3–O3	1.83	1.85	1.85	1.84	1.84	1.89	1.86
Mn3–O5	1.86	1.84	1.84	1.86	1.83	1.85	1.85
Mn4–O5	1.82	1.80	1.81	1.82	1.81	1.81	1.85
A-O1	2.80	2.19	2.24	2.66	2.25	2.55	2.50
A-O2	2.73	2.24	2.29	2.58	2.21	2.49	2.41
A-O5	3.00	3.02	2.88	2.90	2.99	2.91	2.70
A-Mn1	3.79	3.35	3.35	3.69	3.37	3.62	3.46
A-Mn2	3.66	3.10	3.13	3.49	3.12	3.40	3.36
A-Mn3	3.74	3.49	3.46	3.63	3.45	3.59	3.44
A-Mn4	4.29	4.31	4.17	4.22	4.28	4.27	4.06

^a PDBID: 4UB6, A subunit

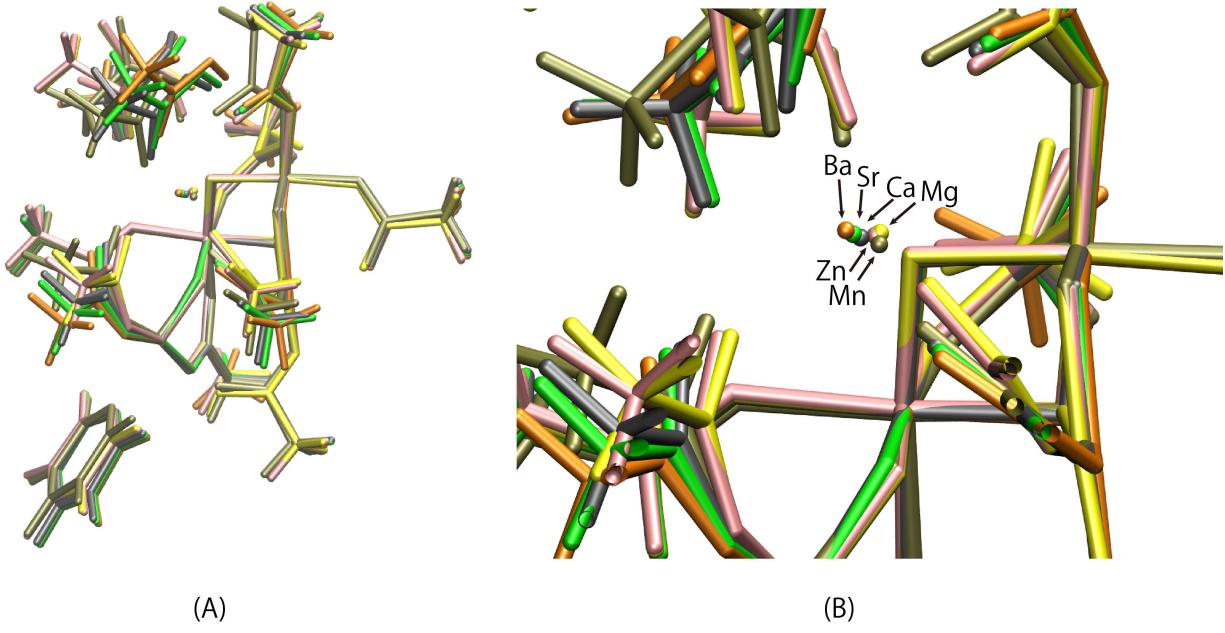


Figure S6.1 Superimposed views of the Ca substituted models (Mn_4AO_4) optimized at the UB3LYP/DZVP level. Structures are colored in orange (A=Ba), green (A=Sr), gray (A=Ca), yellow (A=Mg), pink (A=Zn) and tan (A=Mn). (A) whole molecular structures and (B) enlarged view for the Mn_4AO_4 cores are shown.