

Electronic Supplementary Information

The degree of π electron delocalization and the formation of 3D-extensible sandwich structure

Xiang Wang,^a Qiang Wang,^b Caixia Yuan,^a Xue-Feng Zhao,^a Jia-Jia Li,^a Debao Li,^b Yan-Bo Wu^{a,*}, and Xiaotai Wang^c

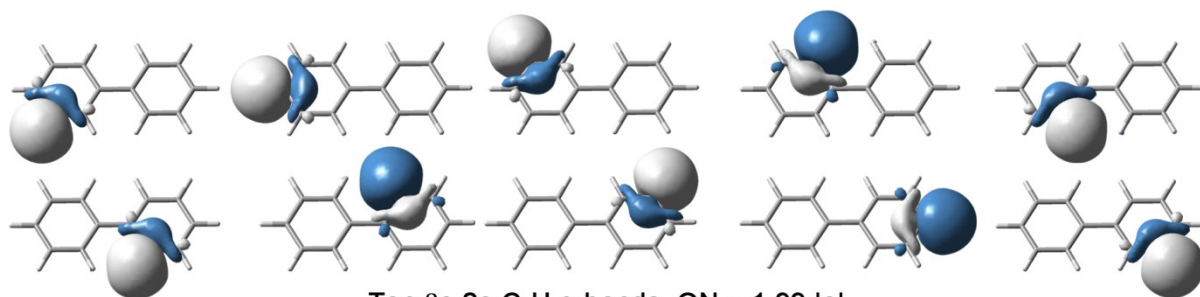
- a. Key laboratory of energy conversion and storage of Shanxi Province, Institute of Molecular Science, Shanxi University, Taiyuan, Shanxi, 030006, People's Republic of China. Email: wzb@sxu.edu.cn
- b. State Key Laboratory of Coal Conversion, Institute of Coal Chemistry, The Chinese Academy of Sciences, Taiyuan, Shanxi, 030001, People's Republic of China.
- c. Department of Chemistry, University of Colorado Denver, Campus Box 194, P.O. Box 173364, Denver, Colorado 80217-3364, United States.

Table of content:

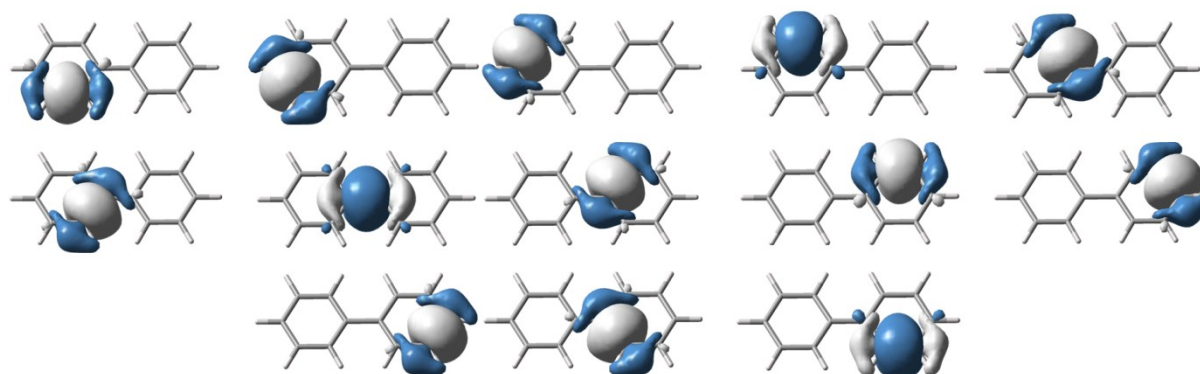
1. Table S1. Comparison between ω B97X-D/6-31G(d) and B3LYP/6-31G(d) results.
2. Figure S1. Full set of AdNDP orbitals of planarized biphenyl.
3. Figure S2. Full set of AdNDP orbitals of sandwich complex with two biphenyl ligands.
4. Figure S3. Full set of AdNDP orbitals of starbenzene dimer.
5. Figure S4. Full set of AdNDP orbitals of sandwich complex with mixed starbenzene dimer-benzene ligands.

Table S1. Comparison between ω B97X-D/6-31G(d) and B3LYP/6-31G(d) results, including the point groups (PG), the lowest vibrational frequencies (ν_{\min} , in cm^{-1}), the HOMO-LUMO gap (Gap, in eV), the total binding energies (BE_{total}) relative to the free starbenzenes, benzenes, and chromium atoms, the binding energies for extending a decker in sandwich chains ($\text{BE}_{1\text{D-ext}}$, in kcal/mol), and the average binding energies for the formation of a hydrogen bridge bond (BE_{HBB}).

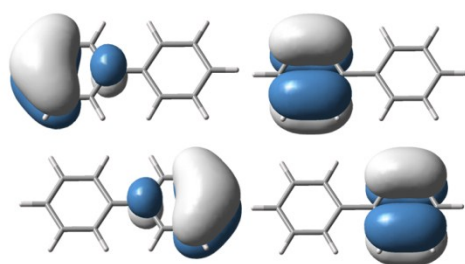
	B3LYP/6-31G(d)					ω B97X-D/6-31G(d)					
	PG	ν_{\min}	Gap	BE_{total}	$\text{BE}_{1\text{D-ext}}$	PG	ν_{\min}	Gap	BE_{total}	$\text{BE}_{1\text{D-ext}}$	
(Bz)2Cr	D_{6h}	39	4.06	-79.3		(Bz)2Cr	D_{6h}	57	8.11	-101.7	
1	D_{6d}	37	3.24	-84.1		1	D_{6d}	41 <i>i</i>	7.28	-118.6	
2'	C_{6v}	16	3.75	-97.9		2	C_{6v}	37	7.94	-127.1	
3''	D_6	8	3.59	-172.0	-74.1	3''	D_6	23	7.44	-228.9	-101.8
4''	C_6	5	3.12	-238.2	-66.3	4''	C_6	29	6.67	-318.5	-89.6
5	D_{6h}	8	2.74	-298.1	-59.9	5	D_{6h}	7	6.23	-399.5	-81.0
	PG	ν_{\min}	Gap	BE_{total}	BE_{HBB}		PG	ν_{\min}	Gap	BE_{total}	BE_{HBB}
2PP2	C_{2v}	11	3.68	-224.8	-29.0	2PP2	C_{2v}	3	7.85	-287.8	-33.7
2PP3	C_{3v}	16	3.64	-380.9	-29.1	2PP3	C_{3v}	18	7.80	-482.1	-33.6
2EE2	C_{2v}	16	3.62	-243.9	-24.0	2EE2	C_{2v}	23	7.75	-309.9	-27.9
2PE4	C_{2v}	10	3.57	-543.8	-25.3	2PE4	C_{2v}	9	7.72	-686.2	-29.7
3PP2	D_{2h}	4	3.56	-405.7	-30.9	3PP2	D_{2h}	15	7.30	-528.2	-35.3
3PP3	D_{3h}	16	3.53	-700.5	-30.8	3PP3	D_{3h}	20	7.25	-898.3	-35.3
3EE2	D_{2h}	21	3.51	-446.5	-25.6	3EE2	D_{2h}	24	7.22	-576.6	-29.7



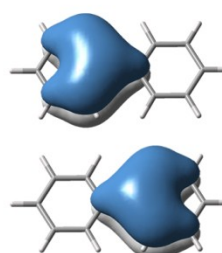
Ten 2c-2e C-H σ bonds, ON = 1.99 |e|



Twelve 2c-2e C-C σ bonds, ON = 1.97 |e|



Four 4c-2e π bonds, ON = 1.99 |e|



Two 6c-2e π bonds, ON = 1.99 |e|

Figure S1. Full set of AdNDP orbitals of planarized biphenyl.

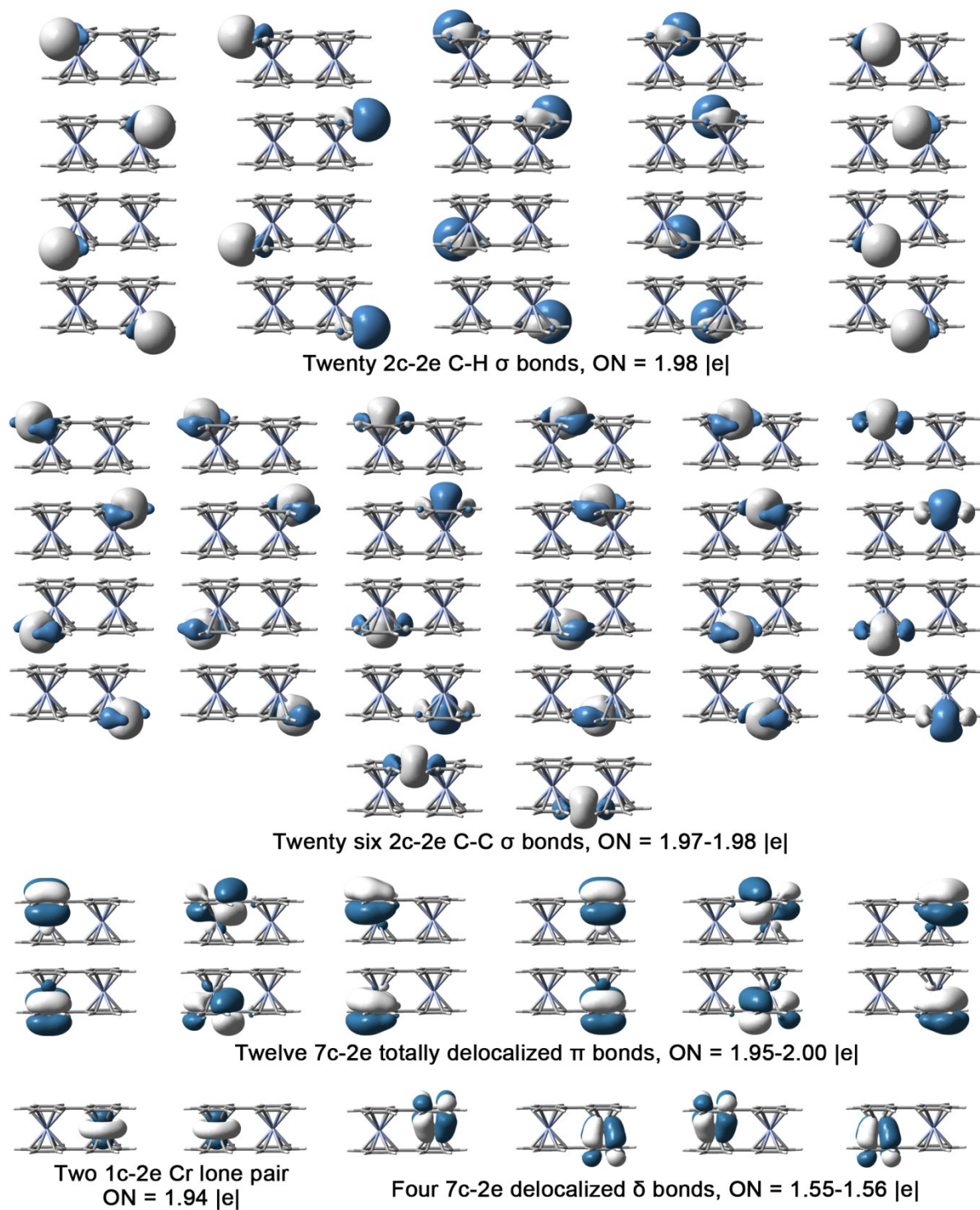
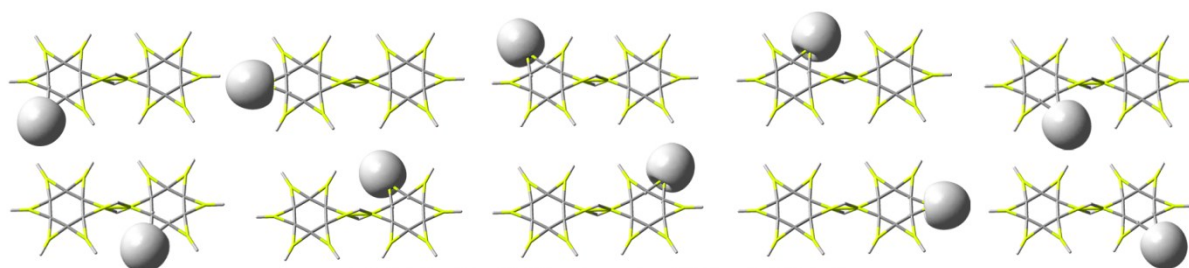
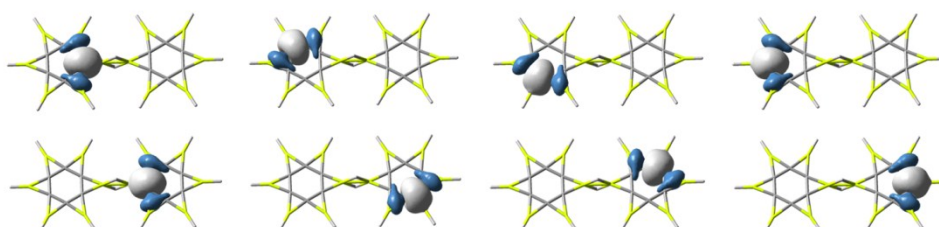


Figure S2. Full set of AdNDP orbitals of sandwich complex with two biphenyl ligands.



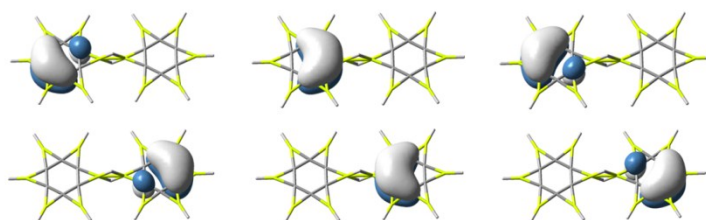
Eight 2c-2e C-H σ bonds, ON = 1.99 |e|



Eight 2c-2e C-C σ bonds, ON = 1.98 |e|

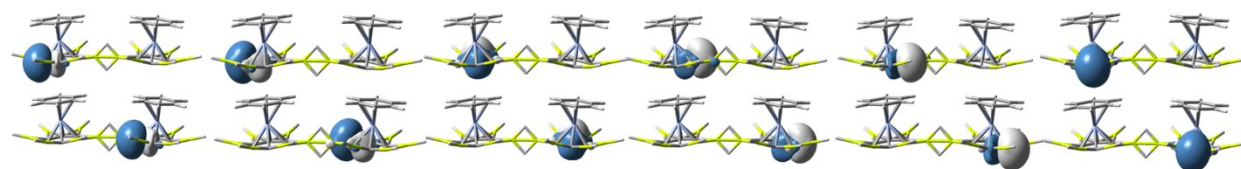


Two 3c-2e σ bonds, ON = 1.98 |e|

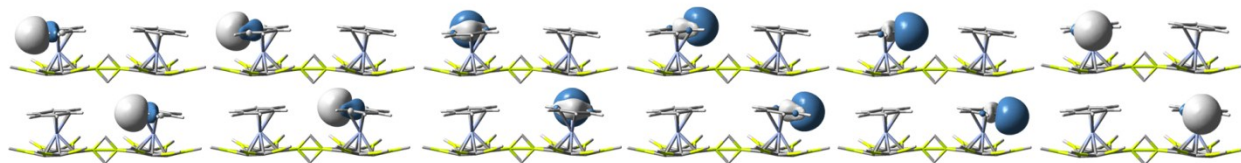


Six 4c-2e π bonds, ON = 1.97 |e|

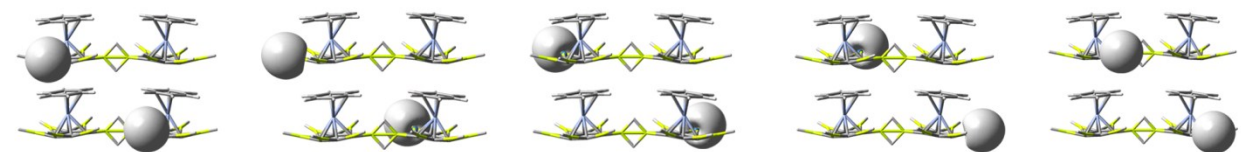
Figure S3. Full set of AdNDP orbitals of starbenzene dimer.



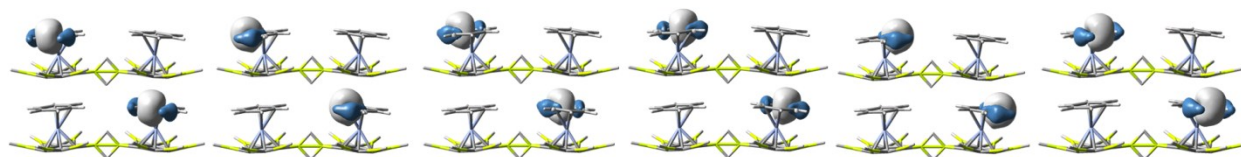
Twelve 1c-2e C lone pairs, ON = 1.82 |e|



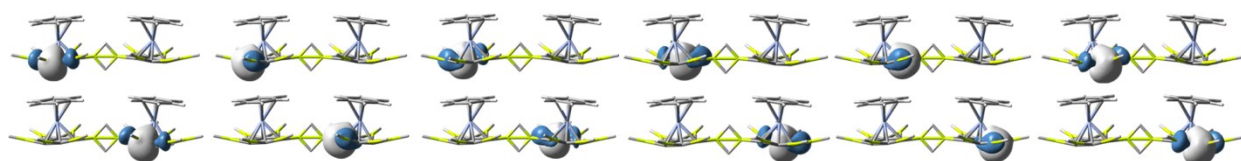
Twelve 2c-2e C-H σ bonds, ON = 1.98 |e|



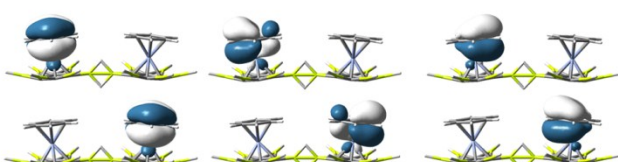
Ten 2c-2e Be-H σ bonds, ON = 1.99 |e|



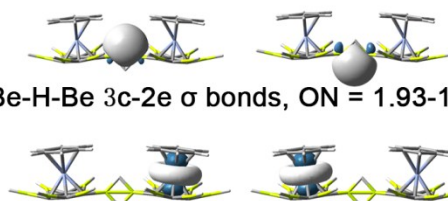
Twelve 2c-2e C-C σ bonds, ON = 1.97 |e|



Twelve 2c-2e C-H σ bonds, ON = 1.93 |e|

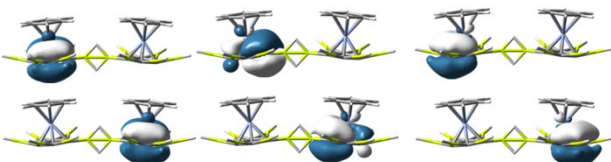


Six 7c-2e totally delocalized π bonds
ON = 2.00 |e|

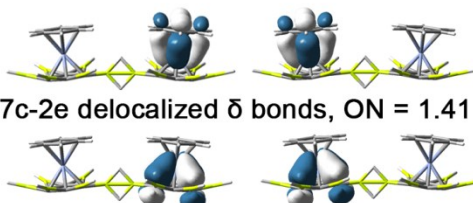


Two Be-H-Be 3c-2e σ bonds, ON = 1.93-1.99 |e|

Two 1c-2e Cr lone pairs, ON = 1.93 |e|



Six 13c-2e totally delocalized π bonds
ON = 1.98-2.00 |e|



Two 7c-2e delocalized δ bonds, ON = 1.41 |e|

Two 13c-2e delocalized δ bonds, ON = 1.71 |e|

Figure S4. Full set of AdNDP orbitals of sandwich complex with mixed starbenzene dimer-benzene ligands.