

## Supporting Information

### **Pyridine Derivatives/Graphene Nanoribbon Composites as Molecularly Tunable Heterogeneous Electrocatalysts for Oxygen Reduction Reaction**

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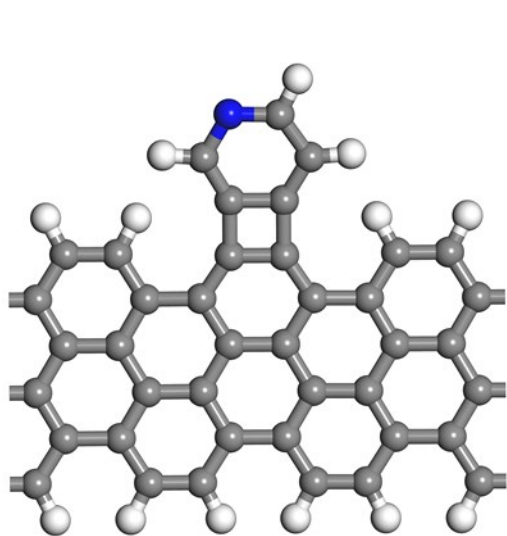
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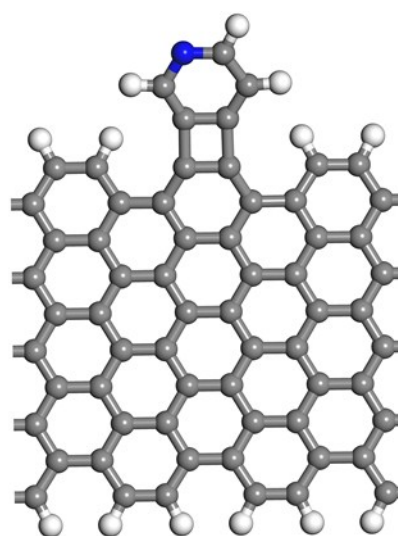
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**Table S1.** Computed adsorption energies of ORR species on pyridine-functionalized graphene nanoribbon with models.

	$E_{\text{ads}}(\text{OOH})$	$E_{\text{ads}}(\text{O})$	$E_{\text{ads}}(\text{OH})$
small model	-0.33	-3.92	-1.64
larger model	-0.33	-3.95	-1.66



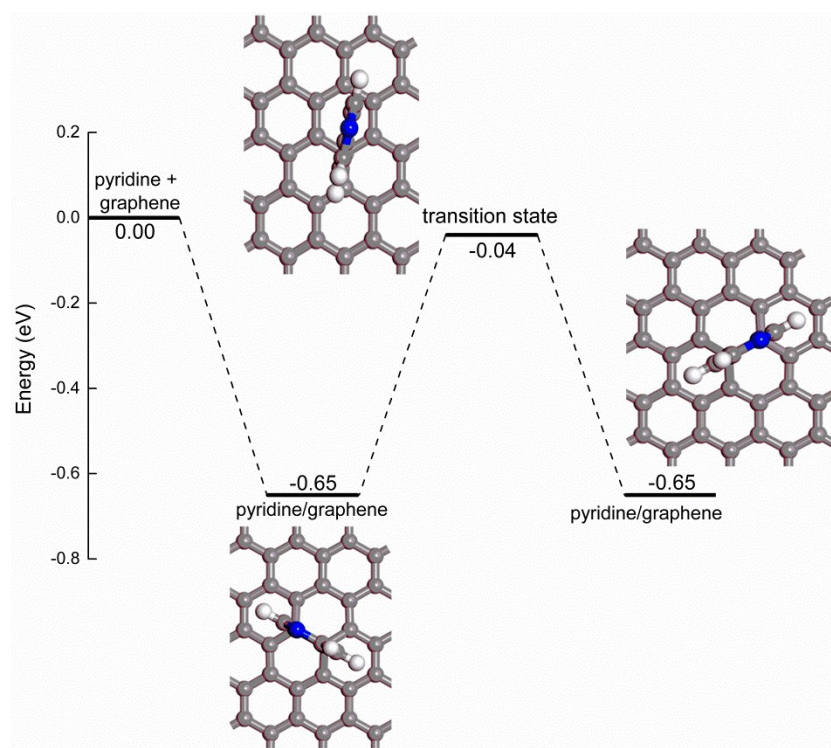
small model



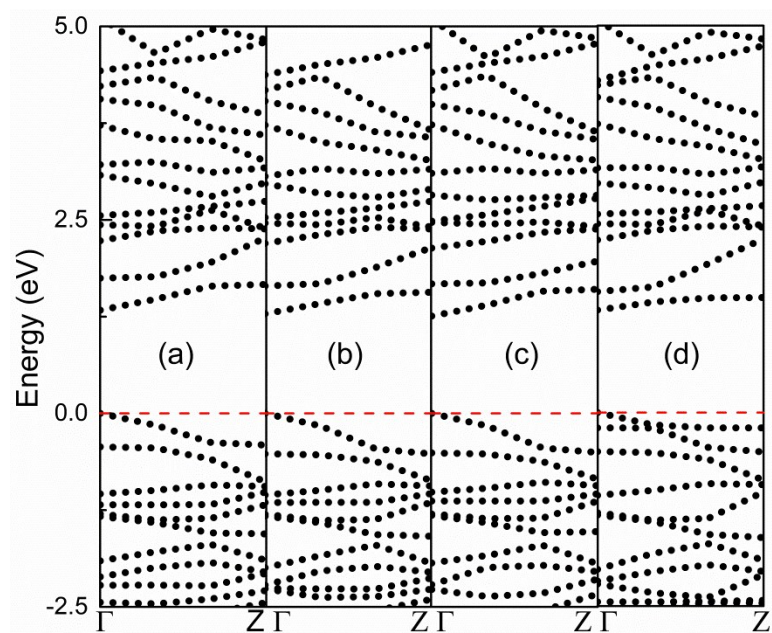
large model

**Table S2.** Computed adsorption energies of ORR species on pyridine-functionalized graphene nanoribbon by employing different  $k$  points.

	$E_{\text{ads}}(\text{OOH})$	$E_{\text{ads}}(\text{O})$	$E_{\text{ads}}(\text{OH})$
$1 \times 1 \times 5$	-0.33	-3.92	-1.64
$1 \times 1 \times 6$	-0.32	-3.90	-1.64
$1 \times 1 \times 7$	-0.32	-3.90	-1.64
$1 \times 1 \times 8$	-0.31	-3.89	-1.64



**Fig. S1.** The diffusion pathway of the adsorbed pyridine on pristine graphene.



**Fig. S2.** The computed band structures of (a) pyridine, (b) pyridazine, (c) pyrimidine, and (d) pyrazine on the edge sites of graphene nanoribbon.