

Electronic Supplementary Information (ESI)

Gate Modulation of Proton Transport in a Nanopore

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Detailed Derivation of eqn (3) and (5) in the Main Text

The electrical potential, ϕ , within the liquid phase ($0 < r < R_n - \delta_s$) can be described by

$$\frac{1}{r} \frac{d}{dr} \left(r \frac{d\phi}{dr} \right) = -\frac{\rho_e}{\varepsilon_f}, \quad (\text{R1})$$

where ρ_e is the mobile space charge density and can be expressed as

$$\rho_e = \sum_{i=1}^4 Fz_i C_{ic} \exp \left[\frac{z_i F (\phi_c - \phi)}{RT} \right]. \quad (\text{R2})$$

In the above, ϕ_c is the electric potential at the center of the nanopore, and C_{ic} is the concentration of the i^{th} ionic species at the center of the nanopore, which can be described by,

$$C_{ic} = C_{i0} \exp \left[-\frac{z_i F (\phi_c - \phi_0)}{RT} \right]. \quad (\text{R3})$$

Here ϕ_0 is the reference potential in the reservoirs far away from the charged nanopore and typically vanishes (i.e., $\phi_0 = 0$). Substituting eqn (R3) and (R2) into eqn (R1) gets

$$\frac{1}{r} \frac{d}{dr} \left(r \frac{d\phi}{dr} \right) = -\frac{1}{\varepsilon_f} \sum_{i=1}^4 Fz_i C_{i0} \exp \left(-\frac{z_i F \phi}{RT} \right), \quad (\text{R4})$$

which is the eqn (3) in the main text. Suppose that $C_{t0} = C_{10} + C_{40} = C_{20} + C_{30}$ is the bulk concentration of net cations (or anions). Due to the electroneutrality of bulk solution (i.e., $C_{t0} = C_{10} + C_{40} = C_{20} + C_{30}$) in the reservoirs far away from the charged nanopore, expanding the right term of eqn (R4) yields

$$\begin{aligned}
& -\frac{1}{\varepsilon_f} \sum_{i=1}^4 Fz_i C_{i0} \exp\left(-\frac{z_i F \phi}{RT}\right) \\
&= -\frac{F}{\varepsilon_f} \left[z_1 C_{10} \exp\left(-\frac{z_1 F \phi}{RT}\right) + z_2 C_{20} \exp\left(-\frac{z_2 F \phi}{RT}\right) + z_3 C_{30} \exp\left(-\frac{z_3 F \phi}{RT}\right) + z_4 C_{40} \exp\left(-\frac{z_4 F \phi}{RT}\right) \right] \\
&= -\frac{zF}{\varepsilon_f} \left[C_{10} \exp\left(-\frac{zF \phi}{RT}\right) - C_{20} \exp\left(\frac{zF \phi}{RT}\right) - C_{30} \exp\left(\frac{zF \phi}{RT}\right) + C_{40} \exp\left(-\frac{zF \phi}{RT}\right) \right] \\
&= \frac{zF}{\varepsilon_f} \left[(C_{20} + C_{30}) \exp\left(\frac{zF \phi}{RT}\right) - (C_{10} + C_{40}) \exp\left(-\frac{zF \phi}{RT}\right) \right] \\
&= \frac{2zFC_{i0}}{\varepsilon_f} \left[\frac{\exp\left(\frac{zF \phi}{RT}\right) - \exp\left(-\frac{zF \phi}{RT}\right)}{2} \right] \\
&= \left(\frac{RT}{zF}\right) \left(\frac{2z^2 F^2 C_{i0}}{\varepsilon_f RT}\right) \sinh\left(\frac{zF \phi}{RT}\right) \\
&= \frac{RT \kappa^2}{zF} \sinh\left(\frac{zF \phi}{RT}\right)
\end{aligned} \tag{R5}$$

where $z = z_1 = -z_2 = -z_3 = z_4$ and $\kappa^{-1} = (\varepsilon_f RT / 2z^2 F^2 C_{i0})^{1/2}$ is the Debye length. By substituting eqn (R5) into eqn (R4), we obtain eqn (5) in the main text,

$$\frac{1}{r} \frac{d}{dr} \left(r \frac{d\phi}{dr} \right) = \frac{RT \kappa^2}{zF} \sinh\left(\frac{zF \phi}{RT}\right). \tag{R6}$$

Gate Modulation of the Zeta Potential of the nanopore

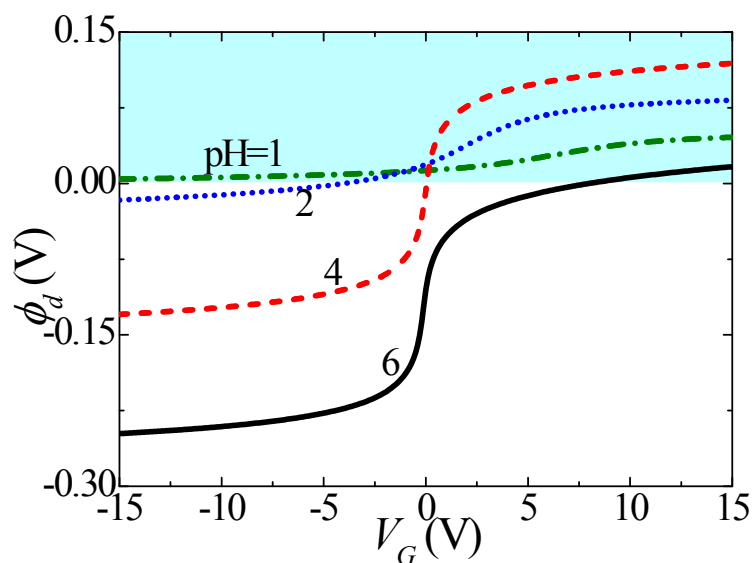


Figure S1. Gate modulation of the zeta potential, ϕ_d , of the nanopore as a function of V_G for various pHs without background salt ($C_{salt} = 0$ mM). The blue region highlights where the modulated ϕ_d is positive.

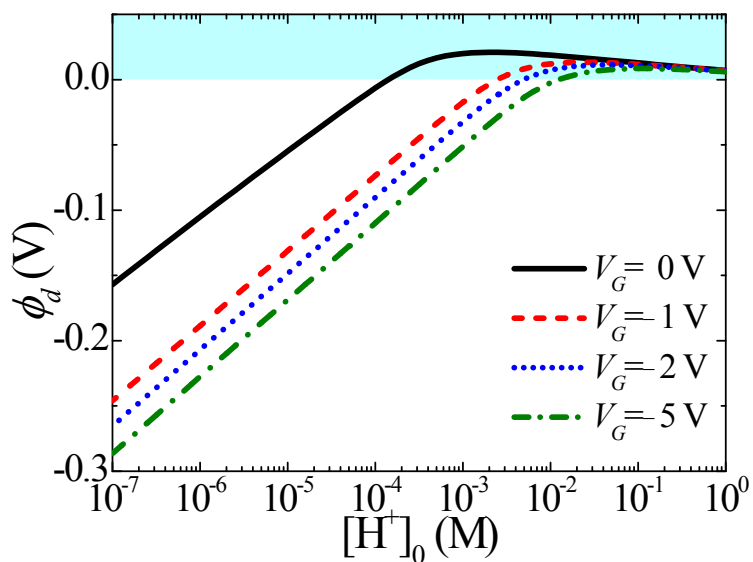


Figure S2. Gate modulation of the zeta potential, ϕ_d , of the nanopore as a function of the molar concentration of protons, $[H^+]_0$, for various levels of V_G without background salt ($C_{salt} = 0$ mM). The blue region highlights where the modulated ϕ_d is positive.

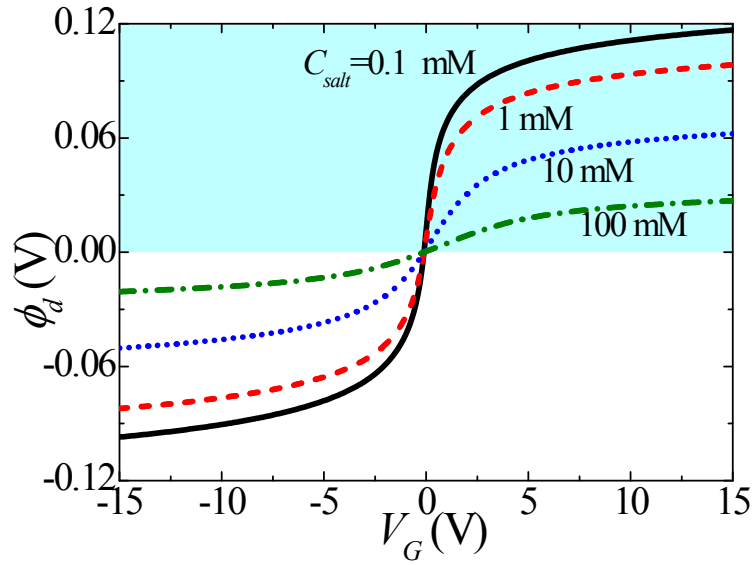


Figure S3. Gate modulation of the zeta potential, ϕ_d , of the nanopore as a function of V_G for various background salt concentrations C_{salt} at pH = 3.5. The blue region highlights where the modulated ϕ_d is positive.

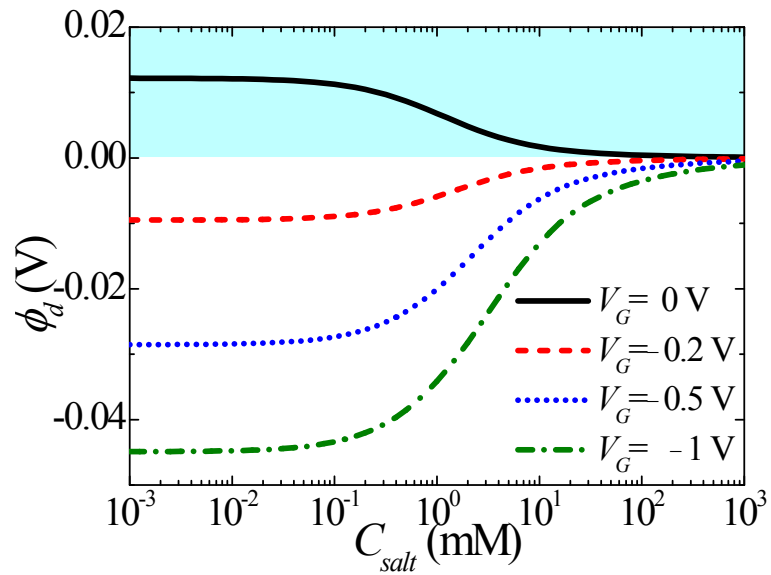


Figure S4. Gate modulation of the zeta potential, ϕ_d , of the nanopore as a function of the background salt concentration C_{salt} for various levels of V_G at pH = 3.5. The blue region highlights where the modulated ϕ_d is positive.