

Supporting Information for

Rational design of ethanol steam reforming catalyst based on analysis of Ni / La₂O₃ metal-support interactions

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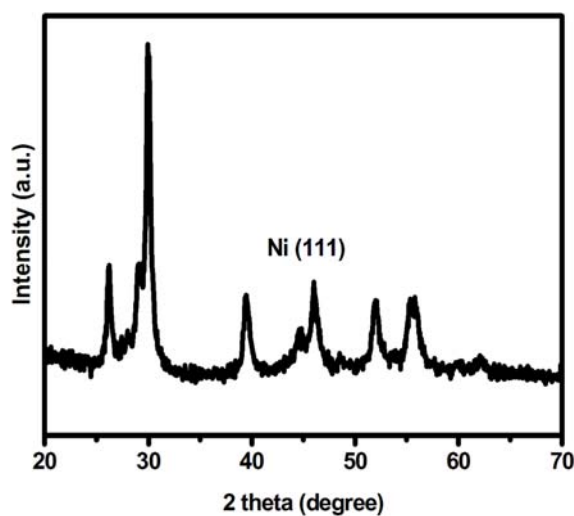


Figure S1 The XRD pattern of Ni/La₂O₃ catalyst.

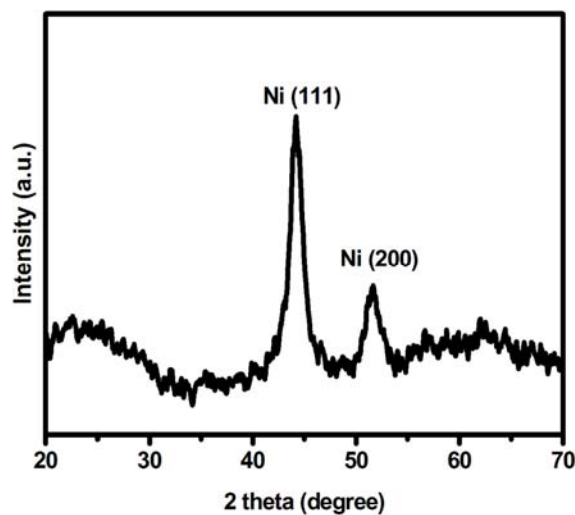


Figure S2 The XRD pattern of Ni/SiO₂ catalyst.

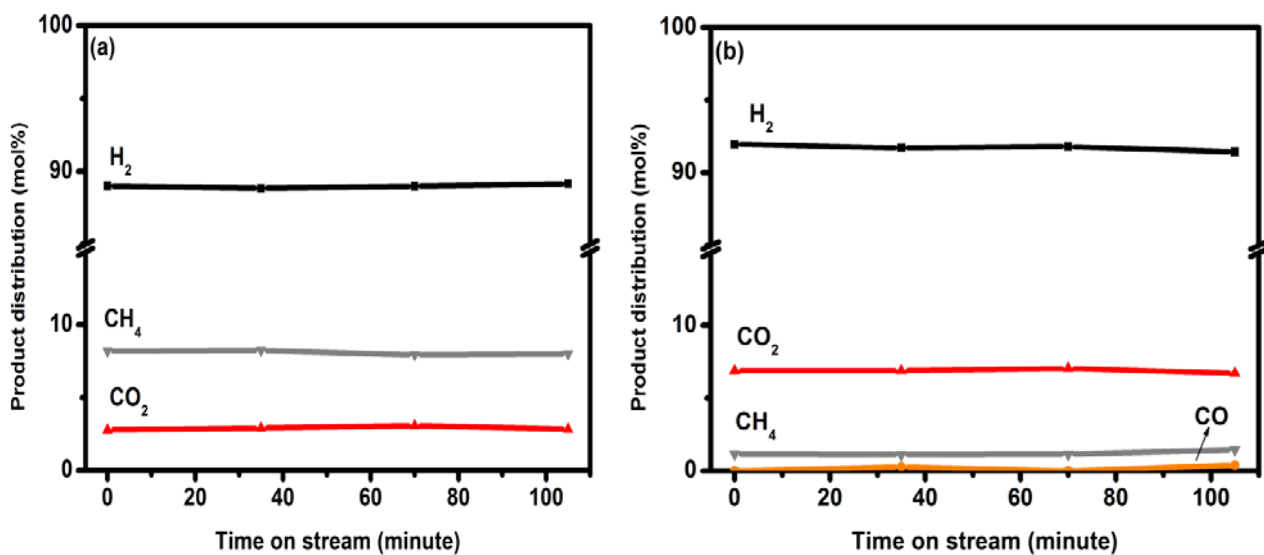


Figure S3 The reaction product distribution of : (a) Ni/SiO₂ and (b) Ni/La₂O₃ catalysts as a function of reaction time at 395°C . Each catalyst was exposed to a well mixture of CO (29.5 torr), H₂ (295 torr) ,and H₂O (334 torr).

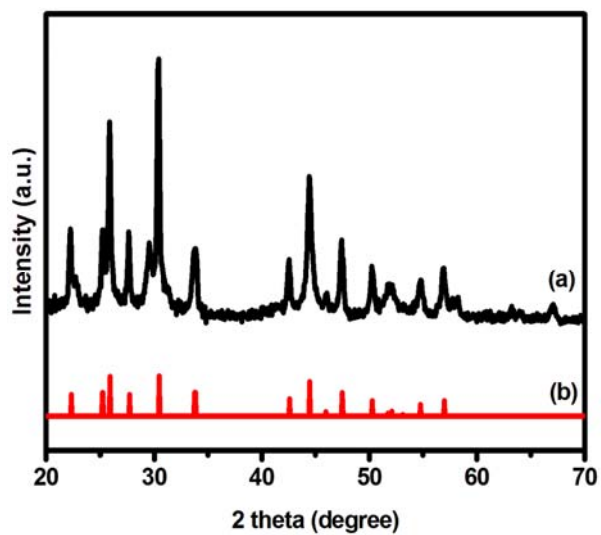


Figure S4 XRD patterns of (a) Ni/La₂O₃ used catalyst and (b) La₂O₂CO₃ JCPDS reference.

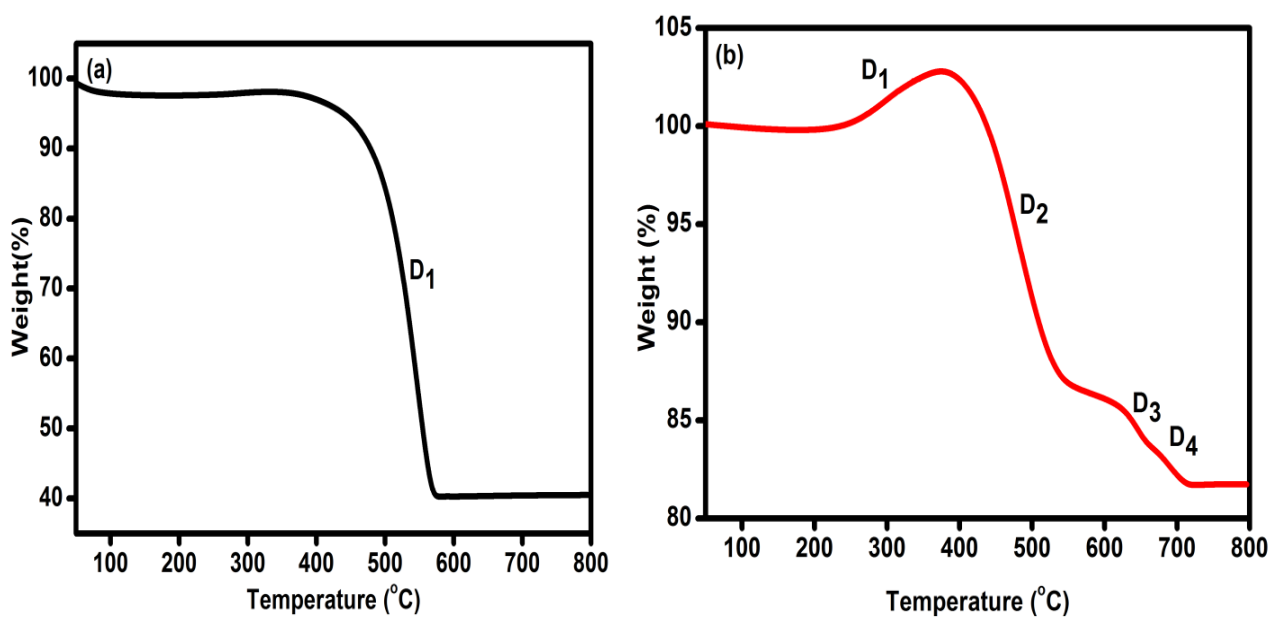


Figure S5 TGA of (a) Ni/SiO₂, and (b) Ni/La₂O₃ used catalysts.

Table S1 the H₂ uptake and grain size of various catalysts

Catalyst	H ₂ uptake ¹ μ mole/g	Grain size ² (nm)
30°C		
Ni/SiO ₂	552.8	6.3
Ni/La ₂ O ₃	54.4	6.6

¹: Calculated based on the H₂ pulse experiment.

²: Calculated based on the XRD results

Table S2 Carbon balance for the Ni/La₂O₃ catalyst with various reaction temperatures

Temp (°C)	300	325	360	395	435	465	500
Input C balance (10 ⁻⁷ mol)	2.49	2.49	2.49	2.49	2.49	2.49	2.49
Output C balance (10 ⁻⁷ mol)	2.49	2.49	2.41	2.45	2.30	2.25	2.36

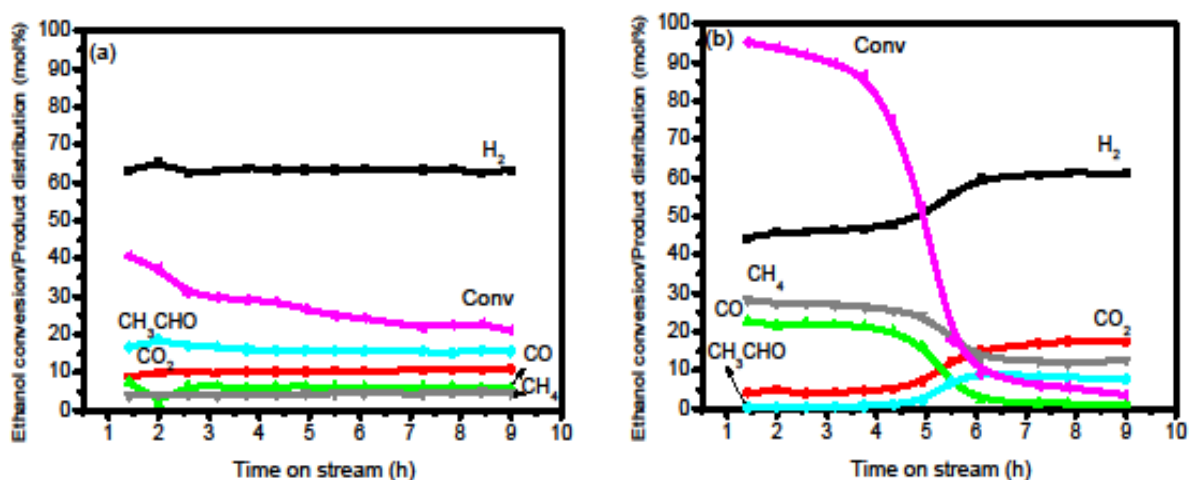


Figure S6 ESR reaction of (a) Ni/La₂O₃ and (b) Ni/SiO₂ catalysts with time on stream (h). Reaction temperature: 395°C. Condition: WHSV = 44.8 h⁻¹ and H₂O/EtOH = 4