Supplementary Information for:
Cyclopentadienyl-based Mg complexes in the Intramolecular Hydroamination of aminoalkenes: Mechanistic evidences for a cationic versus neutral magnesium derivatives.

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Table S1. HSQC ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ spectrum of 2.

| $\boldsymbol{\delta}^{\mathbf{1}} \mathbf{H}$ | $\boldsymbol{\delta}^{\mathbf{1 3}} \mathbf{C}$ |
| :---: | :---: |
| 5.94 | 105.4 |
| 6.17 | 110.2 |
| 6.75 | 116.9 |



Figure S1. HSQC ${ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ spectrum of $\mathbf{2}$.
Table S2. $\mathrm{HMBC}{ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ spectrum of 2 .

| Assignement | $\boldsymbol{\delta}^{\mathbf{1} \mathbf{H}}$ | $\boldsymbol{\delta}^{\mathbf{1 3}} \mathbf{C}$ |
| :---: | :---: | :---: |
| $\left(\mathrm{CH}_{3}\right)_{2} \mathrm{Si}$ | $0.63,0.58$ | 117.5 |
|  |  | 105.9 |
| $\mathrm{C}_{5} \mathrm{H}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}\left(\mathrm{CHMe}_{2}\right)_{2}$ | $2.42-2.36$ | 110.2 |
|  |  | 119.2 |
| $\mathrm{C}_{5} \mathrm{H}_{3} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{~N}\left(\mathrm{CHMe}_{2}\right)_{2}$ | 1.95 | 119.2 |



Figure S2. $\mathrm{HMBC}{ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}$ spectrum of 2.

## Diffusion Ordered Spectroscopy ( $\left.{ }^{1} \mathrm{H}-\mathrm{DOSY}\right)$ experiments.

According to literature procedures ${ }^{2}$ the internal reference method was employed. Three different patrons were chosen. Tetramethylsilane $\left(\mathrm{SiMe}_{4}, \mathrm{Mw}=88.22\right)$, 1,2,3,4-tetraphenylnaphtalene $(\mathrm{TPhN}, \mathrm{Mw}=432.53)$ and dendrimer G2O3A12 (DEN, $\mathrm{Mw}=1424.23$, figure S 4 ). These three patrons and 2 were placed in an NMR tube and 0.5 mL of $\mathrm{C}_{6} \mathrm{D}_{6}$ was added. After the DOSY experiment was recorded Log $\mathrm{D} v s \log \mathrm{Mw}$ of patrons was plotted. The molecular weight can be then obtained by interpolating the D value of the desired complex.


Figure S3. Log D vs Log Mw of $\mathbf{2}$.

Table S3: D, Log D, Mw and Log Mw values of three patrons and 2.

| Compound | $\mathbf{1 0}^{-\mathbf{1 0}} \mathbf{D}\left(\mathbf{m}^{\mathbf{2}} \mathbf{s}^{\mathbf{- 1}}\right)$ | $\mathbf{L o g} \mathbf{D}$ | $\mathbf{M}_{\mathbf{w}}\left(\mathbf{g ~ m o l}^{-\mathbf{1}}\right)$ | $\mathbf{L o g ~ M}$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathbf{w}$ |  |  |  |  |
| $\mathbf{S i M e}$ |  |  |  |  |
| $\mathbf{T P h N}$ | 20.6 | -8.6868 | 88.22 | 1.9455 |
| $\mathbf{D E N}$ | 6.36 | -9.1965 | 432.53 | 2.6360 |
| $\mathbf{2}$ | 2.57 | -9.5897 | 1424.23 | 3.1536 |
|  | 5.84 | -9.2334 | 478.25 | 2.6797 |



Figure S4. Structure of carbosilane dendrimer G2O3A12. ${ }^{1}$
Dendrimer G2O3A12 was synthesized according literature procedure. ${ }^{1}$


Figure S5. DOSY ${ }^{1} \mathrm{H}$ of $\mathbf{2}$ in $\mathrm{C}_{6} \mathrm{D}_{6} / \mathrm{THF}$.


Figure S6. Plot of $\ln \mathrm{v}$ versus $\ln \mathrm{S}$ for the hydroamination of $\mathbf{A}$ catalyzed by 2 at $25^{\circ} \mathrm{C}$ with $[\mathrm{C}]_{0}=20$ mM .


Figure S7. Stoichiometric reaction between $\mathbf{A}$ and $\mathbf{2}$ in a ratio $1: 1$ was carried out at $70^{\circ} \mathrm{C}$.


Figure S8. DOSY ${ }^{1} \mathrm{H}$ of 4 in $\mathrm{C}_{6} \mathrm{D}_{6} / \mathrm{C}_{6} \mathrm{H}_{5} \mathrm{~N}$.


Figure S9. Plot of $\ln S$ versus $t$ for the hydroamination of $A$ catalyzed by $\mathbf{4}$ with $[S]_{0}=1 \mathrm{M}$.


Figure S10. Stoichiometric reaction between A and $\mathbf{4}$ in a ratio 1:1. No reaction was observed.


Figure S11. Stoichiometric reaction between $\mathbf{A}$ and $\mathbf{4}$ in a ratio 1:2 at rt .

1. Sánchez-Nieves, J., Ortega, P., Muñoz-Fernández, M. A., Gómez R., de la Mata, F. J. Tetrahedron, 2010, 66, 9203.
2. Li, D., Keresztes, I., Hopson, R. Williard, P. G. Acc. Chem. Res. 2009, 42, 270.
