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Ligand coordination approach for high reaction stability of an Au-Cu bimetallic carbon-based catalyst in acetylene hydrochlorination process

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Supplementary Information

| Year | Production Capacity | | Congumption | | |
|------|---------------------|-------|------------------|-----------------|-------------|
| | | All | Acetylene Method | Ethylene Method | Consumption |
| 2004 | 676 | 503.2 | 301 | 198 | 701.3 |
| 2005 | 823 | 668.2 | 420 | 239 | 820.05 |
| 2006 | 1122 | 823.8 | 569 | 245 | 918.09 |
| 2007 | 1341 | 971.7 | 710 | 256 | 1026.76 |
| 2008 | 1650 | 822.5 | 555 | 259 | 830.92 |
| 2009 | 1764 | 798 | 527 | 261 | 1026.9 |
| 2010 | 2050 | 1188 | 838 | 342 | 1083.5 |
| 2011 | 2238 | 1295 | 887 | 399 | 1298 |
| 2012 | 2303 | 1595 | 1218 | 374 | 1698 |
| 2013 | 2468 | 1790 | — | | 1915 |

Table S1 The development of PVC industry in China in the last 10 years, unit: 10^4 t.

Table S2 Energy calculation results for each optimized species in Fig. 9 and Fig. S7, energy unit: kJ/mol.

| Au system | | Au/TCCA system | | Au/CA system | | Au/Melamine system | |
|-----------------------------------|-------------|--|-------------|--------------------------------------|-------------|-------------------------------------|-------------|
| Species | Energy | Species | Energy | Species | Energy | Species | Energy |
| Au ₂ Cl ₆ | -1902288.12 | Au ₂ Cl ₅ -TCCA | -2796587.07 | Au ₂ Cl ₅ -CA | -2220033.31 | Au ₂ Cl ₅ -M | -2182410.92 |
| C_2H_2 | -48522.58 | C_2H_2 | -48522.58 | C_2H_2 | -48522.58 | C_2H_2 | -48522.58 |
| Au ₂ Cl ₆ - | 1051272 67 | Au ₂ Cl ₅ -TCCA- | 2844520.20 | Au ₂ Cl ₅ -CA- | -2268373.21 | Au ₂ Cl ₅ -M- | -2230794.24 |
| C_2H_2 | -19312/3.07 | C_2H_2 | -2844339.29 | C_2H_2 | | C_2H_2 | |
| $E_{\rm ads}$ | 62.97 | $E_{\rm ads}$ | -570.35 | $E_{\rm ads}$ | 182.68 | $E_{\rm ads}$ | 139.25 |

*Subscript ads denotes for adsorption which was calculated by *E*(reactant)-*E*(complex).



Fig. S1. Diagram of the experimental equipment (1) Ball valve; (2) Rotor flowmeter; (3) Mass flowmeter; (4) Furnace; (5) Heater; (6) Condenser; (7) Dryer; (8) Gas chromatography; (9) Exhausted gas absorption bottle.



Fig. S2. Typical chromatogram during hydrochlorination reaction.



Fig. S3. XPS spectrum for Au/Cu/TCCA system, Au content 0.25wt%, Au:Cu:TCCA=1:5:20 (mole ratio).



Fig. S4. Typical Ar adsorption/desorption isotherm for Au/TCCA/AC catalysts, Au content 0.25wt%, Au:TCCA=1:20 (mole ratio).



Fig. S5 SEM images of Au/TCCA/AC catalysts, (a)no ligand, (b)Au:TCCA=1:8, (c)Au:TCCA=1:20, (d)Au:TCCA=1:40, Au content 0.5 wt%.



Fig. S6 SEM image of Au/Na $_2S_2O_3$ /AC catalyst, Au content 0.25 wt%.