

Electronic Supplementary Information (ESI) for catalysis sciences & Technology.
This journal is © The Royal Society of Chemistry 2015

**Mechanism of a self-assembled Pd (ferrocenylimine)–Si compound-
catalysed Suzuki coupling reaction**

Hui Liu,^{a, b} Tiesheng Li,^{a*} Xiaoxia Xue,^a Wenjian Xu^a and Yangjie Wu^{a*}

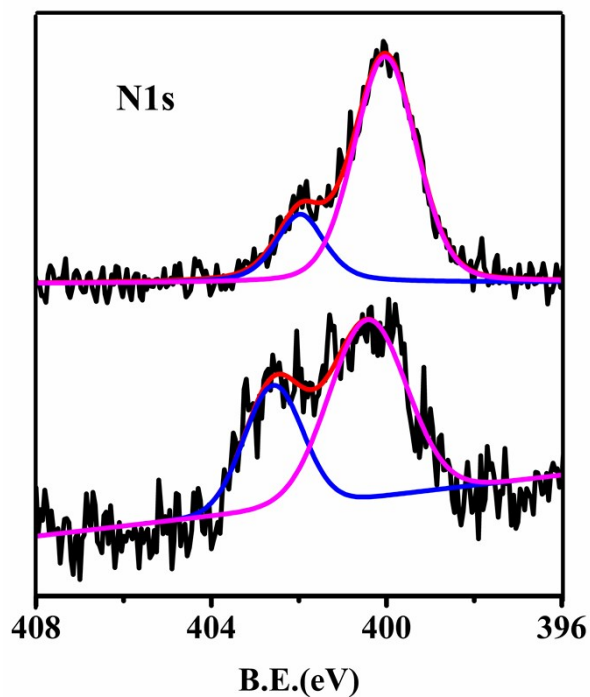


Fig.S1. N1s XPS spectrum of the SAM ferrocenylimine ligand (FcL-Si, curve a) and Pd-ferrocenylimine complex (Pd(FcL)-Si, curve b) on silicon.

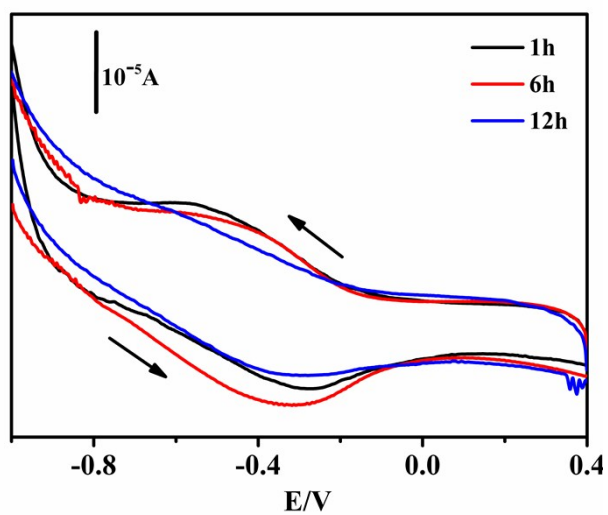


Fig. S2. Cyclic voltammograms of the catalyst-modified ITO at different reaction times in 0.1M N_2 -saturated Tris-HCl.

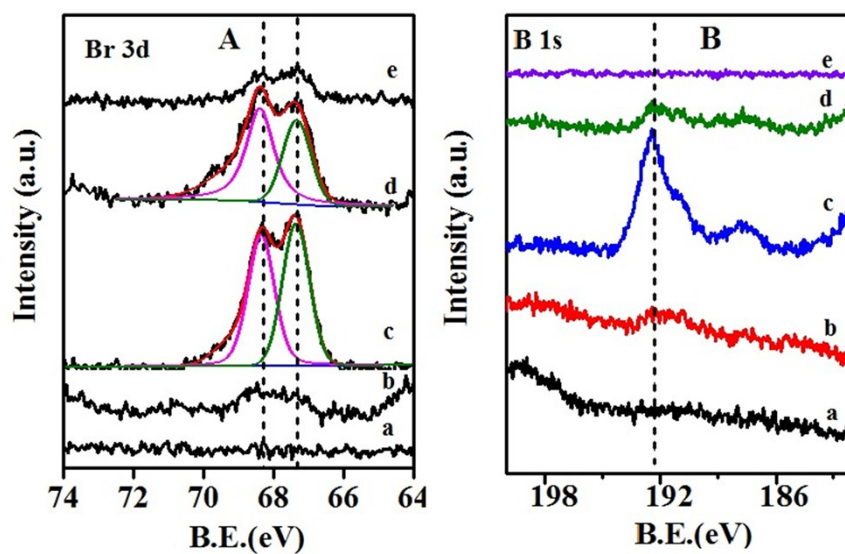


Fig.S3. XPS spectras of Br 3d (A) and B 1s (B) spin orbit doublet of catalytic film at different reaction time. (a) 1 h, (b) 3 h, (c) 6 h, (d) 9 h, (e) 12 h.

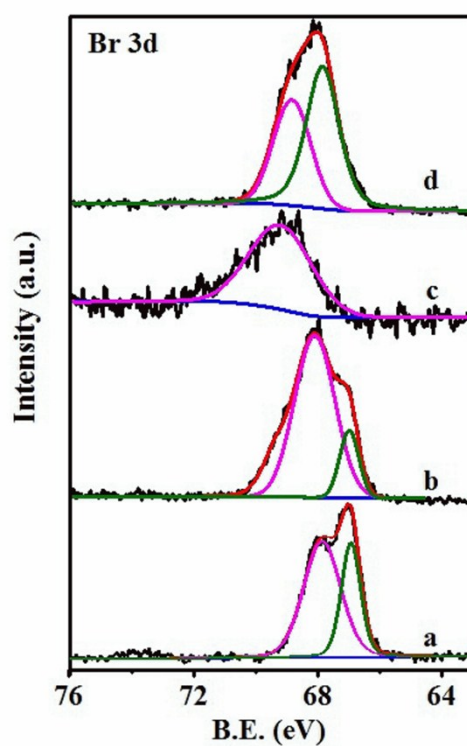
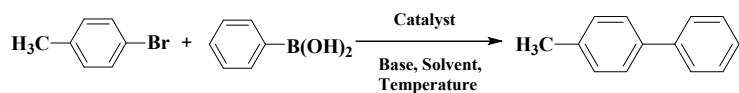


Fig.S4. XPS spectras of Br spin orbit doublet of catalytic film at different aqueous solutions containing (a) K_2CO_3 , (b) K_2CO_3 +TBAB, (c) K_2CO_3 +PhBr, (d) K_2CO_3 +TBAB+PhBr.

Table S1 Effect of solvents, bases and temperature on Suzuki cross-reaction. ^a



Entry	Solvent	Base	Time(h)	Temperature (°C)	Yield ^c (%)
1	DMF	K ₂ CO ₃	12	80	5
2	toluene	K ₂ CO ₃	12	80	43
3	DMSO	K ₂ CO ₃	12	80	0
4	Et ₃ N	K ₂ CO ₃	12	80	0
5	THF	K ₂ CO ₃	12	80	8
6	CH ₃ OH	K ₂ CO ₃	12	80	48
7	CH ₃ OH-H ₂ O(1:1)	K ₂ CO ₃	12	80	50
8	H ₂ O	Na ₂ CO ₃	12	80	81
9	H ₂ O	NaHCO ₃	12	80	69
10	H ₂ O	NaOH	12	80	64
11	H ₂ O	NaOAc	12	80	40
12	H ₂ O	Na ₂ HPO ₄	12	80	32
13	H ₂ O	KH ₂ PO ₄	12	80	72
14	H ₂ O	K ₃ PO ₄	12	80	81
15	H ₂ O	CsCO ₃	12	80	85

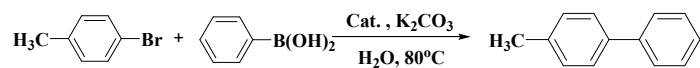
16	H ₂ O	t-BuNa	12	80	91
17	H ₂ O	Et ₃ N	12	80	85
18	H ₂ O	K ₂ CO ₃	12	40	trace
19	H ₂ O	K ₂ CO ₃	12	50	37
20	H ₂ O	K ₂ CO ₃	12	60	54
21	H ₂ O	K ₂ CO ₃	12	70	73
22	H ₂ O	K ₂ CO ₃	12	80	94
23 ^c	H ₂ O	K ₂ CO ₃	12	80	35
24	H ₂ O	K ₂ CO ₃	15	80	96
25	H ₂ O	K ₂ CO ₃	18	80	97

^a Reaction conditions: 0.125 mmol of 4-bromotoluene, 0.15 mmol of PhB(OH)₂, 0.15 mmol of base and catalyst (1.24×10⁻⁸ mol·cm⁻² Pd) in 5.0 mL of solvent at 80°C, 12h. 0. 15 mmol of TBAB as phase transfer catalyst was added in H₂O as a solvent.

^b HPLC yield.

^c In the absence of TBAB

Table S2 Recyclability of Pd (FcL)-Si- catalyzed (1 cm×1cm) Suzuki coupling reaction.



Cycles	1 st	2 nd	3 rd	4 th	5 th
Yields (%)	94	90	91	89	90

Reaction conditions: 4-bromotoluene (0.125 mmol), PhB(OH)₂ (0.15 mmo), K₂CO₃ (0.15 mmo), TBAB (0.15 mmo), catalyst (1.24×10⁻⁸ mol·cm⁻² Pd) and H₂O (5.0 mL); temperature = 80°C, 12h. Yield determined by HPLC.