

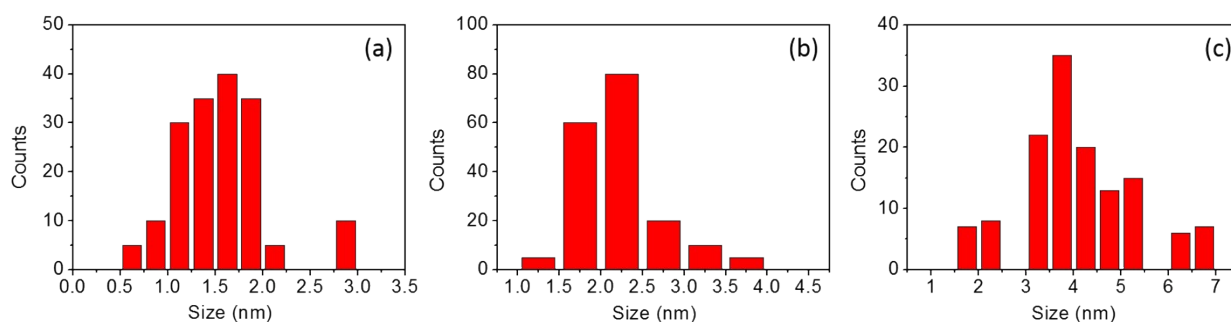
**Supporting Information**

**Production of Acrylic Acid from Biomass-Derived Allyl Alcohol  
by Selective Oxidation Using Au/Ceria Catalysts**

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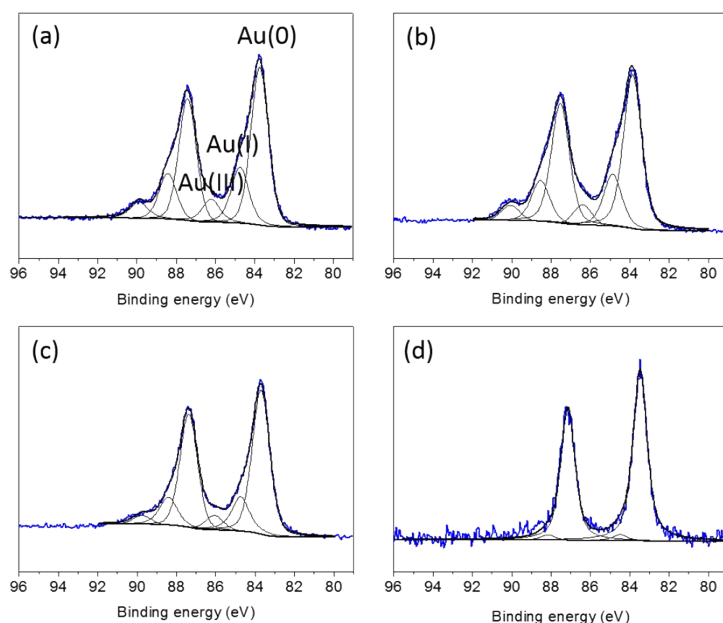
Additional Data; Figure S1 ~ Figure S5, Table S1



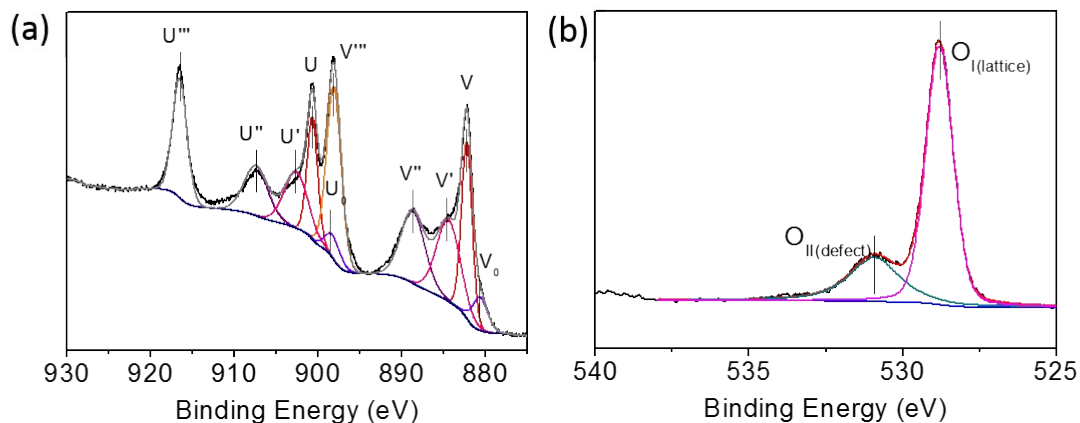
**Figure S1.** Au nanoparticle size distribution for (a) 1.5, (b) 2.2, and (c) 4.1 nm-sized Au/ceria (DP) catalysts.

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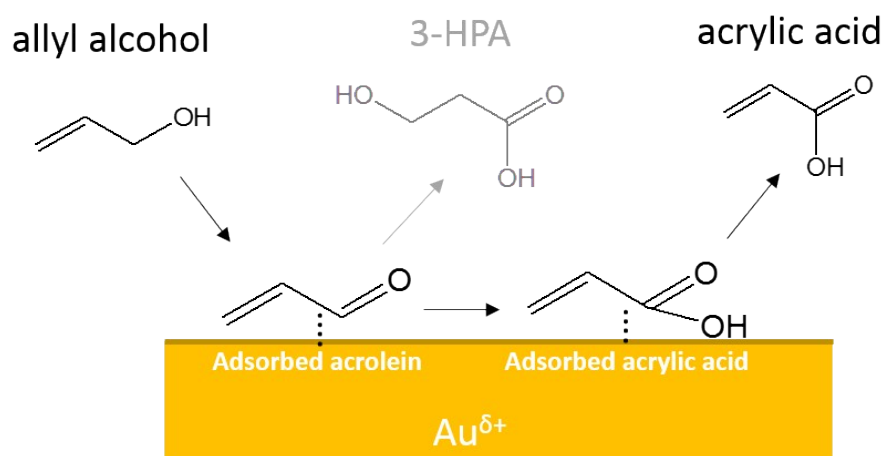
<sup>†</sup> These authors contributed equally.



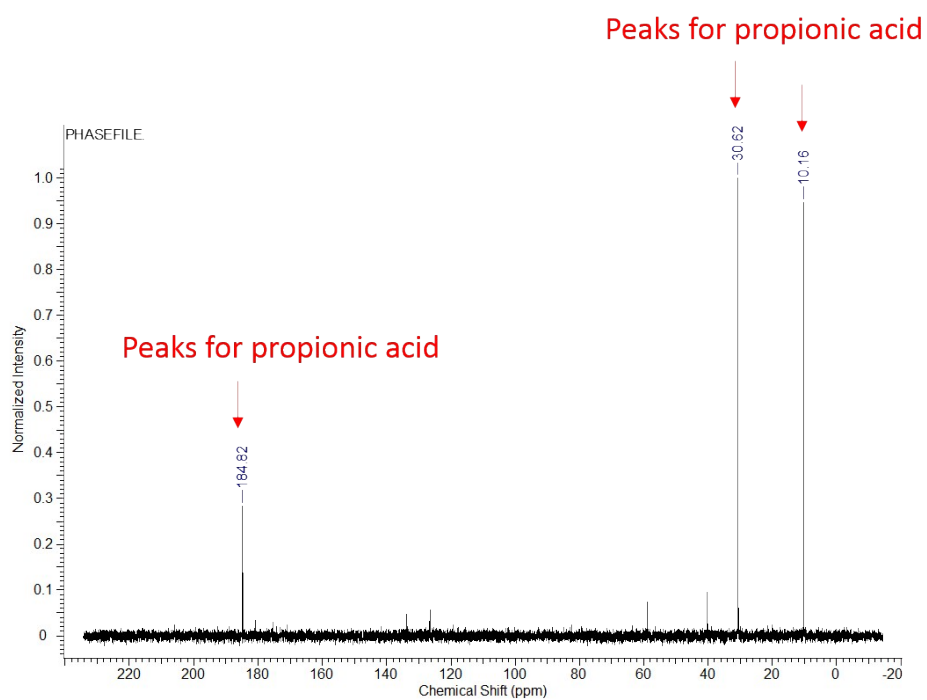
**Figure S2.** XPS data presenting Au 4f peaks for (a) 1.5 nm Au/CeO<sub>2</sub> (DP), (b) 2.2 nm Au/CeO<sub>2</sub> (DP), (c) 4.1 nm Au/CeO<sub>2</sub> (DP), and (d) 2.7 nm Au/CeO<sub>2</sub> (CD).



**Figure S3.** XPS data of Au/Ceria (DP) catalysts presenting (a) Ce 3p peaks and (b) O 1s peaks. The ratio of Ce(III) and Ce(IV) was estimated as 19.1% and 80.9% from the ratio of V<sub>0</sub>, V' and V, V'', V''', and the ratio of O<sub>II(defect)</sub> and O<sub>I(lattice)</sub> was estimated as 27.4% and 72.6%, respectively. Tian's work was used for assigning each peak <sup>S1</sup>.



**Figure S4.** Different pathways for the formation of 3-HPA and acrylic acid.



**Figure S5.**  $^{13}\text{C}$  NMR of the product solution after allyl alcohol oxidation using  $\text{Au}_{82.4}\text{Pd}$  /ceria (DP) catalyst. The reaction condition was the same as Au/ceria (DP) catalyst.

**Table S1.** Allyl alcohol oxidation results when Au-M/ceria (DP) catalysts were used.<sup>a</sup>

Catalyst	Conv. (%)	Yield(%)					
		AcA	3-HPA	Propionic acid	3-APA	GA	Others
Au <sub>82.4</sub> Pd <sub>1</sub> /CeO <sub>2</sub> <sup>b</sup>	100	5.2	8.0	64.6	0	0	22.2
Au <sub>168.8</sub> Pd <sub>1</sub> /CeO <sub>2</sub> <sup>c</sup>	100	9.9	15.3	48.3	0	0	26.5
Au <sub>452.8</sub> Pt <sub>1</sub> /CeO <sub>2</sub> <sup>d</sup>	100	35.2	27.3	11.1	2.9	0.9	19.7
Au <sub>23.6</sub> Cu <sub>1</sub> /CeO <sub>2</sub> <sup>e</sup>	94.2	35.1	27.2	7.8	2.8	0.9	23.4

<sup>a</sup>The reaction was performed at 50°C, O<sub>2</sub> 3 bar, 12 hrs, 3 M NaOH, and a mole ratio of allyl alcohol/Au was 4000. <sup>b</sup>0.352 μmol or <sup>c</sup>0.176 μmol of PdCl<sub>2</sub> dissolved in 0.2 M of HCl solution was added together with Au precursor. <sup>d</sup>0.587 μmol of H<sub>2</sub>PtCl<sub>6</sub> was added together with Au precursor. <sup>e</sup>0.352 μmol of CuCl<sub>2</sub> was added together with Au precursor. The remaining synthetic procedure was the same as the Au/ceria (DP) catalyst. The actual molar ratio of Au versus secondary metal was measured by ICP-MS after the synthesis.

## References

S1. Z. M. Tian, X. Xiang, L. S. Xie and F. Li, *Ind. Eng. Chem. Res.*, 2013, 52, 288-296.