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## Supplementary Information

Effects of oxygen coverage, catalyst size, and core composition on Pt-alloy core-shell nanoparticles for the oxygen reduction reaction Supareak Praserthdam and Perla B. Balbuena\* Department of Chemical Engineering Texas A&M University, College Station, TX 77843 <u>\*balbuena@tamu.edu</u>

## Notation

$$\begin{split} & \mathsf{E}_{\mathsf{sys}} \mathsf{=} \ \mathsf{energy} \ \mathsf{of} \ \mathsf{the} \ \mathsf{selected} \ \mathsf{system} \ (\mathsf{eV}) \\ & \mathsf{E}_{\mathsf{O}/\mathsf{sys}} \mathsf{=} \ \mathsf{energy} \ \mathsf{of} \ \mathsf{the} \ \mathsf{oxygen} - \mathsf{adsorbed} \ \mathsf{selected} \ \mathsf{system} \ (\mathsf{eV}) \\ & \mathsf{E}_{\mathsf{ads},\mathsf{sys}} \mathsf{=} \ \mathsf{adsorption} \ \mathsf{energy} \ \mathsf{of} \ \mathsf{oxygen} \ \mathsf{atom} \ \mathsf{on} \ \mathsf{the} \ \mathsf{selected} \ \mathsf{system} \ (\mathsf{eV}) \\ & \mathsf{E}_{\mathsf{seg}(\mathsf{A}\cdot\mathsf{B})} \mathsf{=} \ \mathsf{segregation} \ \mathsf{energy} \ \mathsf{of} \ \mathsf{the} \ \mathsf{system} \ \mathsf{of} \ \mathsf{A} \ \mathsf{and} \ \mathsf{B} \ \mathsf{segregated} \ \mathsf{pair} \ (\mathsf{eV}/\mathsf{atom/pair}) \\ & \mathsf{E}_{\mathsf{dis},\mathsf{sys}} \mathsf{=} \ \mathsf{dissolution} \ \mathsf{energy} \ \mathsf{of} \ \mathsf{the} \ \mathsf{selected} \ \mathsf{system} \ (\mathsf{eV}) \\ & \mathsf{E}_{\mathsf{sys},0} \mathsf{=} \ \mathsf{energy} \ \mathsf{of} \ \mathsf{the} \ \mathsf{system} \ \mathsf{before} \ \mathsf{dissolution} \ (\mathsf{eV}) \\ & \mathsf{E}_{\mathsf{sys},+} \mathsf{=} \ \mathsf{energy} \ \mathsf{of} \ \mathsf{the} \ \mathsf{system} \ \mathsf{after} \ \mathsf{dissolution} \ (\mathsf{eV}) \\ & \mathsf{\Delta}\mathsf{E}_{\mathsf{dis}} \mathsf{=} \ \mathsf{dissolution} \ \mathsf{energy} \ \mathsf{shift} \ \mathsf{of} \ \mathsf{the} \ \mathsf{system} \ \mathsf{from} \ \mathsf{pure} \ \mathsf{Pt} \ \mathsf{system} \ (\mathsf{eV}) \\ & \mathsf{E}_{\mathsf{dis},\mathsf{alloy}} \mathsf{=} \ \mathsf{dissolution} \ \mathsf{energy} \ \mathsf{of} \ \mathsf{the} \ \mathsf{system} \ \mathsf{from} \ \mathsf{pure} \ \mathsf{Pt} \ \mathsf{system} \ (\mathsf{eV}) \\ & \mathsf{E}_{\mathsf{dis},\mathsf{alloy}} \mathsf{=} \ \mathsf{dissolution} \ \mathsf{energy} \ \mathsf{of} \ \mathsf{the} \ \mathsf{system} \ \mathsf{from} \ \mathsf{pure} \ \mathsf{Pt} \ \mathsf{system} \ (\mathsf{eV}) \\ & \mathsf{E}_{\mathsf{dis},\mathsf{alloy}} \mathsf{=} \ \mathsf{dissolution} \ \mathsf{energy} \ \mathsf{of} \ \mathsf{the} \ \mathsf{system} \ \mathsf{from} \ \mathsf{pure} \ \mathsf{Pt} \ \mathsf{system} \ (\mathsf{eV}) \\ & \mathsf{E}_{\mathsf{dis},\mathsf{alloy}} \mathsf{=} \ \mathsf{dissolution} \ \mathsf{energy} \ \mathsf{of} \ \mathsf{the} \ \mathsf{core} \ \mathsf{system} \ (\mathsf{eV}) \\ & \mathsf{edded} \ \mathsf{system} \ \mathsf{edded} \ \mathsf{system} \ \mathsf{edded} \ \mathsf{edded} \ \mathsf{system} \ \mathsf{edded} \ \mathsf{edded} \ \mathsf{system} \ \mathsf{edded} \ \mathsf{$$

Edis,Pt = dissolution energy of the pure Pt system (eV)

Pt <sub>shell</sub> -(M-Ni) <sub>core</sub>	E <sub>sys</sub> (eV)	E <sub>O/sys</sub> (eV)	E <sub>ads</sub> (sys) (eV)	E <sub>ads</sub> (Pt) (eV)	E <sub>ads,rel</sub> (eV/atom)
Fe	-708.46	-712.65	0.72	-1.23	+1.95
Со	-664.58	-670.02	-0.53	-1.23	+0.70
Ru	-725.96	-731.50	-0.63	-1.23	+0.60
Rh	-660.92	-666.64	-0.80	-1.23	+0.43
Re	-838.57	-844.22	-0.73	-1.23	+0.50
Os	-786.29	-791.87	-0.66	-1.23	+0.57
lr	-705.70	-711.48	-0.86	-1.23	+0.37

## Table I. Adsorption energy of the 116-atom Pt<sub>shell</sub>-(M-Ni)<sub>core</sub> nanoparticle system.

## Table II. Oxygen adsorption energy of Ptshell-(Re-Ni)core system as a function of sized and oxygen coverage and Ptshell-(Fe-Cn-Fe-Ni)core system as a function of carbon content and oxygen coverage per (111) facet. The adsorption sites are defined as in Figure 3.

The label of the adsorption site of E and NE specifies that for the E (Edge) one of the oxygen atoms is adsorbed on the site near the edge of the nanoparticle, while for NE (non-Edge) means that all of the adsorbed oxygen are not located at the site near the edge.

# total atoms	# O-atoms /O-Coverage	Adsorption site	E <sub>sys</sub> (eV)	E <sub>O/sys</sub> (eV)	E <sub>ads</sub> (eV/atom)	E <sub>ads,rel</sub> (eV)		
		Pt <sub>shell</sub> -(Re	-Ni) <sub>core</sub>					
	1 (0.0128 ML)	Def	-838.57	-844.236	-0.74	-		
		А	-838.57	-844.238	-0.75	-		
		В	orption site $E_{sys}$ (eV) $E_{o/sys}$ (eV) $E_{ads}$ (eV/atom) $t_{shell}$ -(Re-Ni) coreDef-838.57-844.236-0.74A-838.57-844.239-0.75B-838.57-844.239-0.75f & A-838.57-844.239-0.73f & B-838.57-849.85-0.73f & B-838.57-849.86-0.73& B-838.57-849.84-0.72Def*-838.57-882.58-0.59A-1933.92-1939.68-0.84B-1933.92-1939.69-0.85D-1933.92-1939.69-0.85D-1933.92-1939.68-0.61F-1933.92-1939.68-0.66(A+B)-1933.92-1939.68-0.85G-1933.92-1939.50-0.66(A+C)-1933.92-1944.97-0.61(A+C)-1933.92-1944.97-0.61(A+C)-1933.92-1944.99-0.62(A+D)-1933.92-1944.99-0.62(A+D)-1933.92-1944.93-0.59(C+D)-1933.92-1945.14-0.69(E+D)-1933.92-1944.93-0.59P2-299.49-305.007-0.599P2-299.49-305.007-0.599P4-299.49-305.007-0.599P4-299.44-1937.66-3.30Def-1929.44-1937.66-3.30Def	+0.49				
116		Def & A	-838.57	-849.85	-0.73	-		
(NP**)	2 (0.0256 ML)	Def & B	-838.57	-849.86	-0.73	+0.45		
		A & B	-838.57	-849.84	-0.72	-		
	8 (0.103 ML)*	Def*	-838.57	E <sub>O/sys</sub> (eV) E <sub>ads</sub> (eV/atom)   -844.236 -0.74   -844.238 -0.75   -844.239 -0.75   -849.85 -0.73   -849.86 -0.73   -849.86 -0.73   -849.86 -0.72   -882.58 -0.59   -1939.68 -0.84   -1939.69 -0.85   -1939.69 -0.85   -1939.69 -0.61   -1939.68 -0.85   -1939.69 -0.61   -1939.68 -0.85   -1939.50 -0.61   -1939.50 -0.61   -1939.50 -0.62   -1944.97 -0.61   -1945.14 -0.69   -1945.14 -0.69   -1944.96 -0.60   -1944.93 -0.59   -1978.88 -0.70   -305.007 -0.599   -305.007 -0.599   -305.007 -0.599   -305.007 -0.599   -305.007 -0.599 <td>-0.59</td> <td>+0.59</td>	-0.59	+0.59		
		Α	-1933.92	-1939.68	E <sub>ads</sub> (eV/atom)	-		
		В	-1933.92	-1939.44	-0.60	-		
		С	-1933.92	-1939.69	-0.85	-		
	1 (0.007 ML)	D	-1933.92	-1939.82	-0.98	+0.07		
		Е	-1933.92	-1939.45	-0.61	-		
		F	-1933.92	-1939.68	-0.85	-		
		G	-1933.92	-1939.50	-0.66	-		
260 (NP**)	2 (0 014 ML)	E01 (A+B)	-1933.92	-1944.97	-0.61	-		
		E02 (A+C)	-1933.92	-1945.18	-0.71	+0.39		
		E03 (C+F)	-1933.92	-1944.80	-0.53	-		
		E04 (F+G)	-1933.92	-1944.99	-0.62	-		
	2 (0.014 ML)	NE1 (A+D)	-1933.92	-1945.14	-0.69	-		
		NE2 (B+D)	-1933.92	-1944.96	-0.60	-		
		NE3 (C+D)	-1933.92	-1945.14	-0.69	-		
		NE4 (F+D)	-1933.92	-1944.93	-0.59	-		
	8 (0.056 ML)*	A*	-1933.92	-1978.88	-0.70	+0.40		
	1 (0.111 ML)	P1	-299.49	-305.007	-0.599	-		
45		P2	-299.49	-305.008	-0.600	+0.67		
(SLAB)		P3	-299.49	-305.007	-0.599	-		
		P4	-299.49	-305.007	-0.599	-		
Pt <sub>shell</sub> -(Fe-C <sub>n</sub> -Fe-Ni) <sub>core</sub> ***								
284 (Fe <sub>4.6</sub> C <sub>1.0</sub> ) <sup>†</sup>		Def	-1929.44	-1937.66	-3.30	-2.07		
296 (Fe <sub>3.1</sub> C <sub>1.0</sub> ) <sup>+</sup>	1 (0 007 ML)	Def	-2035.20	-2042.31	-2.19	-0.95		
302 (Fe <sub>2.6</sub> C <sub>1.0</sub> ) <sup>†</sup>		Def	-2085.00	-2091.69	-1.77	-0.53		
310 (Fe <sub>2.2</sub> C <sub>1.0</sub> ) <sup>+</sup>		Def	-2154.09	-2157.93	+1.07	+2.30		

\*For the system of 8 oxygen atoms, 1 oxygen atom per (111) facet is adsorbed at the designated position, where each nanoparticle has eight (111) facets.

\*\*NP = nanoparticle model

\*\*\* For  $Pt_{shell}$ -(Fe-C<sub>n</sub>-Fe-Ni)<sub>core</sub>, all systems are nanoparticles

Highlight = selected system of the most stable configuration at each oxygen coverage  $\dagger$  the composition of the core-anchoring shell as Fe<sub>x</sub>C<sub>y</sub>

# total atoms	# O-atoms /O-Coverage	Adsorption site	E <sub>sys</sub> (eV)	E <sub>NS</sub> (eV)	E <sub>seg</sub> (Pt-M)* (eV/atom/pair)	E <sub>seg</sub> (Re-Ni) (eV/atom/pair)		
Pt <sub>shell</sub> -(Re-Ni) <sub>core</sub>								
		Def	-838.57	-844.24	-1.36	N/D		
	1 (0.0128 ML)	А	-838.57	-844.24	-7.20	N/D		
		В	-838.57	-844.24	-6.52	-0.126		
116 (NP**)		Def & A	-838.57	-849.85	-0.394	N/D		
	2 (0.0256 ML)	Def & B	-838.57	-849.86	-0.22	-0.013		
		A & B	-838.57	-849.84	-0.397	N/D		
	8 (0.103 ML)‡	Def	-838.57	-882.58	-0.27	+0.11		
		А	-1933.92	-1939.68	N/D	N/D		
		В	-1933.92	-1939.44	N/D	N/D		
		С	-1933.92	-1939.69	N/D	N/D		
	1 (0.007 ML)	D	-1933.92	-1939.82	-1.06	-0.566		
		Е	-1933.92	-1939.45	N/D	N/D		
		F	-1933.92	-1939.68	N/D	N/D		
		G	-1933.92	-1939.5	N/D	N/D		
260 (ND**)	2 (0.014 ML)	E01 (A+B)	-1933.92	-1944.97	N/D	N/D		
(INF )		E02 (A+C)	-1933.92	-1945.18	-0.80	-0.147		
		E03 (C+F)	-1933.92	-1944.8	N/D	N/D		
		E04 (F+G)	-1933.92	-1944.99	N/D	N/D		
		NE1 (A+D)	-1933.92	-1945.14	N/D	N/D		
		NE2 (B+D)	-1933.92	-1944.96	N/D	N/D		
		NE3 (C+D)	-1933.92	-1945.14	N/D	N/D		
		NE4 (F+D)	-1933.92	-1944.93	N/D	N/D		
	1 (0.111 ML)	P1	-299.49	-305.007	-1.642	N/D		
		P2	-299.49	-305.008	-1.640	-1.113		
45		P3	-299.49	-305.007	-1.957	N/D		
(SLAB)		P4	-299.49	-305.007	-1.645	N/D		
	1 (0.25 ML)	Def****	-128.8	-138.72	-0.93	-0.481		
Pt <sub>shell</sub> -(Fe-C <sub>n</sub> -Fe-Ni) <sub>core</sub> ***								
272 (Fe <sub>9.2</sub> C <sub>1.0</sub> ) <sup>†</sup>		Def	-1823.07	-1823.07	-1.32	N/D		
$204 (Fe_{4.6}O_{1.0})^{+}$ 296 (Fe_4C4a) <sup>+</sup>	0 (0 ML)	Def	-1929.44	-1929.44	-U.56 _0.61			
$310 (Fe_{22}C_{10})^{\dagger}$		Def	-2035.20	-2035.20	+2.24	N/D		
284 (Fe <sub>4.6</sub> C <sub>1.0</sub> ) <sup>†</sup>		Def	-1929.44	-1937.66	+1.08	N/D		
296 (Fe <sub>3.1</sub> C <sub>1.0</sub> )†		Def	-2035.20	-2042.31	-0.02	N/D		
302 (Fe <sub>2.6</sub> C <sub>1.0</sub> )†	T (0.007 ML)	Def	-2085.00	-2091.69	+0.53	N/D		
310 (Fe <sub>2.2</sub> C <sub>1.0</sub> ) <sup>†</sup>		Def	-2154.09	-2157.93	+0.10	N/D		

Table III. Segregation energy for  $Pt_{shell}$ -(Re-Ni)<sub>core</sub> as a function of sizes and oxygen coverage and for  $Pt_{shell}$ -(Fe-Cn-Fe-Ni)<sub>core</sub> as a function of carbon content and oxygen coverage Definitions of the sites are in Figure 3.

\*M = Re for  $Pt_{shell}$ -(Re-Ni)<sub>core</sub> and M = Fe for  $Pt_{shell}$ -(Fe-C<sub>n</sub>-Fe-Ni)<sub>core</sub>

\*\*NP = nanoparticle model \*\*\* all systems are nanoparticles

\*\*\*\*This adsorption site is at the same position as P1 in 0.111 ML system

<sup>‡</sup>For the system of 8 oxygen atoms, 1 oxygen atom per (111) facet is adsorbed at the designated position, where each nanoparticle has eight (111) facets.

Highlight = selected system of the most stable configuration at each oxygen coverage

<sup> $\dagger$ </sup> = the composition of the core-anchoring shell as Fe<sub>x</sub>C<sub>y</sub>

# total atoms	Adsorption site	E <sub>sys,0</sub> (eV)	E <sub>sys,+</sub> (eV)	E <sub>dis,alloy</sub> (eV)	E <sub>dis,Pt</sub> (eV)	$\Delta E_{dis}$ (eV)		
Pt <sub>shell</sub> (Re-Ni) <sub>core</sub>								
116 (NP**)	IP	-838.57	-831.58	-6.98	-6.64	-0.34		
	ED1	-838.57	-832.22	-6.35	-5.98	-0.37		
	ED2	-838.57	-831.92	-6.65	-6.24	-0.42		
260 (NP**)	IP1	-1933.92	-1927.27	-6.65	-6.62	-0.03		
	IP2	-1933.92	-1926.74	-7.18	-6.92	-0.26		
	IP3	-1933.92	-1927.07	-6.86	-6.91	+0.05		
	IP4	-1933.92	-1927.26	-6.67	-6.63	-0.04		
	ED1	-1933.92	-1927.01	-6.92	-6.32	-0.60		
	ED2	-1933.92	-1927.73	-6.20	-5.89	-0.30		
	ED3	-1933.92	-1926.97	-6.95	-6.31	-0.64		
	ED4	-1933.92	-1927.22	-6.70	-6.32	-0.38		
45 (SLAB)	P1	-299.49	-292.23	-7.26	-7.29	+0.03		
	P2	-299.49	-292.24	-7.26	-7.29	+0.04		
	P3	-299.49	-292.23	-7.26	-7.29	+0.03		
	P4	-299.49	-292.23	-7.26	-7.29	+0.03		
Pt <sub>shell</sub> -(Fe-C <sub>n</sub> -Fe-Ni) <sub>core</sub> ***								
272 (Fe <sub>9.2</sub> C <sub>1.0</sub> )†	IP2	-1823.07	-1817.13	-5.94	-6.92	+0.97		
284 (Fe <sub>4.6</sub> C <sub>1.0</sub> )†	IP2	-1929.44	-1923.68	-5.76	-6.92	+1.16		
296 (Fe <sub>3.1</sub> C <sub>1.0</sub> )†	IP2	-2035.20	-2028.69	-6.52	-6.92	+0.40		
302 (Fe <sub>2.6</sub> C <sub>1.0</sub> )†	IP2	-2085.00	-2077.33	-7.68	-6.92	-0.76		
310 (Fe <sub>2.2</sub> C <sub>1.0</sub> )†	IP2	-2154.09	-2146.81	-7.28	-6.92	-0.36		

Table IV. Dissolution potential shift for  $Pt_{shell}(Re-Ni)_{core}$  nanoparticles as a function of sizes and  $Pt_{shell}$ -(Fe- $C_n$ -Fe-Ni)<sub>core</sub> nanoparticles as a function of carbon content Definitions of the sites are in Figure 9.

\*\*NP = nanoparticle

\*\*\* all systems are nanoparticles

 $^{\dagger}\,$  the composition of the core-anchoring shell as  $Fe_{x}C_{y}$