Electronic Supplementary Information

Three new luminescent Cd(II)-MOFs by regulation of tetracarboxylates and auxiliary co-ligands, displaying high sensitivity for Fe³⁺ in aqueous solution[†]

Yunlong Wu, Guo-Ping Yang*, Xiang Zhou, Jiang Li, Yan Ning, and Yao-Yu Wang*

Key Laboratory of Synthetic and Natural Functional Molecule Chemistry of Ministry of Education, Shaanxi Key Laboratory of Physico-Inorganic Chemistry, College of Chemistry and Materials Science, Northwest University, Xi'an 710069, P.R. China. Email: <u>ygp@nwu.edu.cn, wyaoyu@nwu.edu.cn.</u>

Complex 1			
Cd(1)-O(4)#1	2.249(3)	O(3)#2-Cd(1)-N(2)#3	84.48(12)
Cd(1)-N(1)	2.289(3)	O(4)#1-Cd(1)-O(1)	93.39(11)
Cd(1)-O(3)#2	2.325(3)	N(1)-Cd(1)-O(1)	97.87(12)
Cd(1)-N(2)#3	2.325(3)	O(3)#2-Cd(1)-O(1)	152.64(12)
Cd(1)-O(1)	2.408(3)	N(2)#3-Cd(1)-O(1)	86.41(12)
Cd(1)-O(2)	2.420(3)	O(4)#1-Cd(1)-O(2)	146.76(11)
O(4)#1-Cd(1)-N(1)	103.67(12)	N(1)-Cd(1)-O(2)	89.51(12)
O(4)#1-Cd(1)-O(3)#2	111.39(13)	O(3)#2-Cd(1)-O(2)	99.35(12)
N(1)-Cd(1)-O(3)#2	87.66(12)	N(2)#3-Cd(1)-O(2)	85.65(12)
O(4)#1-Cd(1)-N(2)#3	85.03(12)	O(1)-Cd(1)-O(2)	54.19(10)
N(1)-Cd(1)-N(2)#3	169.98(13)		

Table S1. Selected bond lengths [Å] and angles [°] for three MOFs.

Symmetry codes: #1: -x+1, -y+1, -z+2; #2: x-1, y, z; #3: x+1, -y+1/2, z+1/2.

Complex 2					
Cd(1)-O(1)	2.291(2)	O(3)#1-Cd(1)-O(4)#1	55.71(8)		
Cd(1)-N(2)	2.300(3)	N(1)-Cd(1)-O(4)#1	111.45(10)		
Cd(1)-O(3)#1	2.315(3)	O(1)-Cd(1)-O(2)	55.89(7)		
Cd(1)-N(1)	2.330(3)	N(2)-Cd(1)-O(2)	94.45(10)		
Cd(1)-O(4)#1	2.386(2)	O(3)#1-Cd(1)-O(2)	118.95(9)		
Cd(1)-O(2)	2.397(2)	N(1)-Cd(1)-O(2)	143.54(8)		
O(1)-Cd(1)-N(2)	101.88(11)	O(4)#1-Cd(1)-O(2)	102.90(8)		
O(1)-Cd(1)-O(3)#1	117.23(9)	O(1)-Cd(1)-C(11)#1	140.91(10)		
N(2)-Cd(1)-O(3)#1	138.16(10)	N(2)-Cd(1)-C(11)#1	117.10(10)		
O(1)-Cd(1)-N(1)	93.35(9)	O(3)#1-Cd(1)-C(11)#1	27.94(9)		
N(2)-Cd(1)-N(1)	71.35(10)	N(1)-Cd(1)-C(11)#1	101.34(9)		
O(3)#1-Cd(1)-N(1)	91.17(10)	O(4)#1-Cd(1)-C(11)#1	27.84(8)		
O(1)-Cd(1)-O(4)#1	153.57(9)	O(2)-Cd(1)-C(11)#1	114.89(8)		
N(2)-Cd(1)-O(4)#1	94.86(10)				

Symmetry code: #1: x, -y+1/2, z+1/2.

Complex 3			
Cd(1)-O(1)	2.454(3)	N(1)-Cd(1)-O(1)	105.96(13)
Cd(1)-O(2)	2.283(3)	O(2)-Cd(1)-N(2)	121.70(13)
Cd(1)-O(4)#4	2.280(3)	O(4)#5-Cd(1)-N(2)	97.76(12)
Cd(1)-N(1)	2.339(4)	O(4)#4-Cd(1)-N(1)	166.82(12)
Cd(1)-N(2)	2.292(3)	O(2)-Cd(1)-N(1)	98.21(13)
Cd(1)-O(4)#5	2.289(3)	O(4)#5-Cd(1)-N(1)	102.43(12)
O(4)-Cd(1)#2	2.289(3)	N(2)-Cd(1)-N(1)	71.30(13)
O(4)-Cd(1)#1	2.280(3)	O(4)#4-Cd(1)-O(1)	87.16(11)
O(4)#4-Cd(1)-O(2)	90.43(11)	O(2)-Cd(1)-O(1)	54.43(11)
O(4)#4-Cd(1)-O(4)#5	76.63(11)	O(4)#5-Cd(1)-O(1)	86.50(11)
O(4)#4-Cd(1)-N(2)	95.70(12)	N(2)-Cd(1)-O(1)	175.32(13)
O(2)-Cd(1)-O(4)#5	139.67(11)		

Symmetry codes: #1 : x, -y, z+1/2 ; #2 : -x+3/2, y-1/2, -z+3/2 ; #4 : x, -y, z-1/2 ; #5 : -x+3/2, y+1/2, -z+3/2.





(c)

Fig. S1 The FT-IR spectra of complexes 1 (a), 2 (b), and 3 (c).

Table S2. The FT-IR vibrational modes of the functional groups in three MOFs

1	$\upsilon_{O\text{-}Cd} \sim 3488\ cm^{\text{-}1}$	$\upsilon^{as}_{(COO^-)} \sim 1577 \text{ cm}^{-1}, \upsilon^{s}_{(COO^-)} \sim 1390 \text{ cm}^{-1}$	$\omega_{C\text{-H}} \sim 848 \text{ cm}^{-1}$
2	$\upsilon_{O\text{-}Cd} \sim 3415\ cm^{-1}$	$\upsilon^{as}_{(COO^-)} \sim 1580 \text{ cm}^{-1}, \upsilon^{s}_{(COO^-)} \sim 1382 \text{ cm}^{-1}$	$\omega_{C\text{-H}} \sim 760 \text{ cm}^{\text{-1}}$
3	$\upsilon_{O\text{-}Cd} \sim 3423 \text{ cm}^{-1}$	$\begin{split} \upsilon_{c=o} ~\sim~ 1645~cm^{-1};~\upsilon^{as}_{(COO^{-})} ~\sim~ 1405~cm^{-1},\\ \upsilon^{s}_{(COO^{-})} \sim~ 1325~cm^{-1} \end{split}$	$\omega_{C\text{-H}} \sim 759 \ cm^{-1}$



Fig. S2 The 3D framework of 1 viewed along *b* axis.



Fig. S3 The dimeric $[Cd_2(COO)_4]$ SBU unit is simplified as eight-connected node (a), and L_1^{4-} ligand is acted as four-connected nodes (b).



Fig. S4 The 2D grid-like layer of 2.



Fig. S5 The dimeric $[Cd_2(COO)_4]$ motif is considered as four-connected node (a), and L_2^{4-} ligand is acted as four-connected nodes (b).



(b)



Fig. S6 PXRD patterns of complexes 1 (a), 2 (b), and 3 (c) simulated from the X-ray single crystal structures and as-synthesized products.



Fig. S7 TGA curves for three complexes.



Fig. S8 The SEM images of grinded sample of 3 (a) and the original shape of 3 (b).



Fig. S9 Emission spectra of 3 with different metal ions.



Fig. 10 Color change in aqueous solution containing different metal ions.





Fig. S11 The first cyclic emission spectra (a), PXRD patterns (b), and last four cyclic emission spectra of the sensing experiments of **3** (c).