

Structural, Photophysical and Magnetic Properties of Transition Metal Complexes Based on the Dipicolylamino-chloro-1,2,4,5-tetrazine Ligand

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SUPPORTING INFORMATION

X-ray structures

Coordination geometry polyhedra in the solid state structures of **2a-2d**

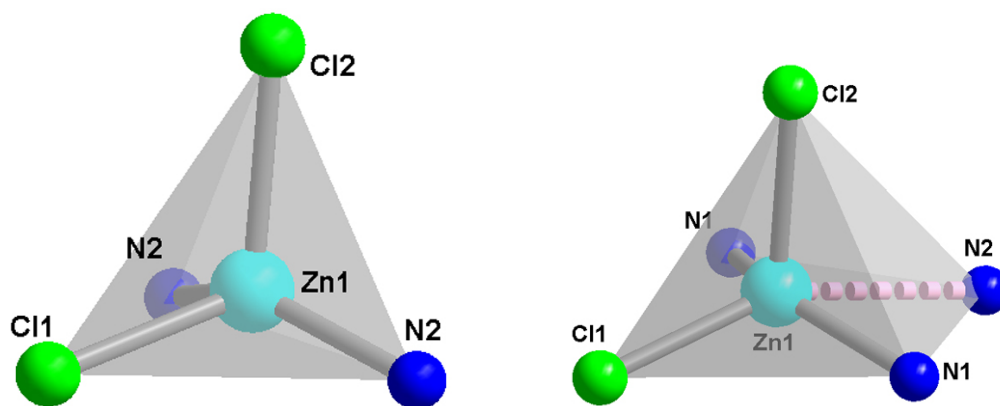


Fig. S1 Complex **2a** with Zn(II) either in tetrahedral (left) or square pyramid (right) environment.

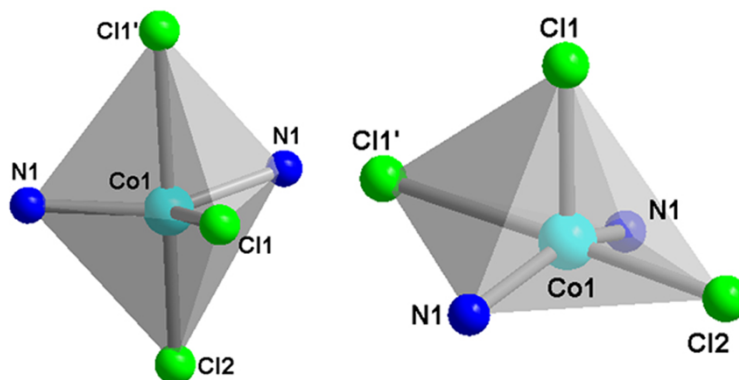


Fig. S2 Complex **2d** with Co(II) penta-coordinated.

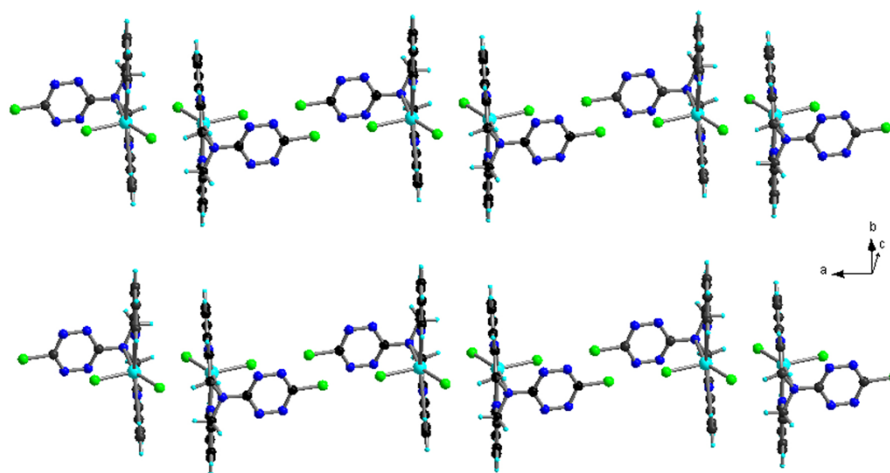
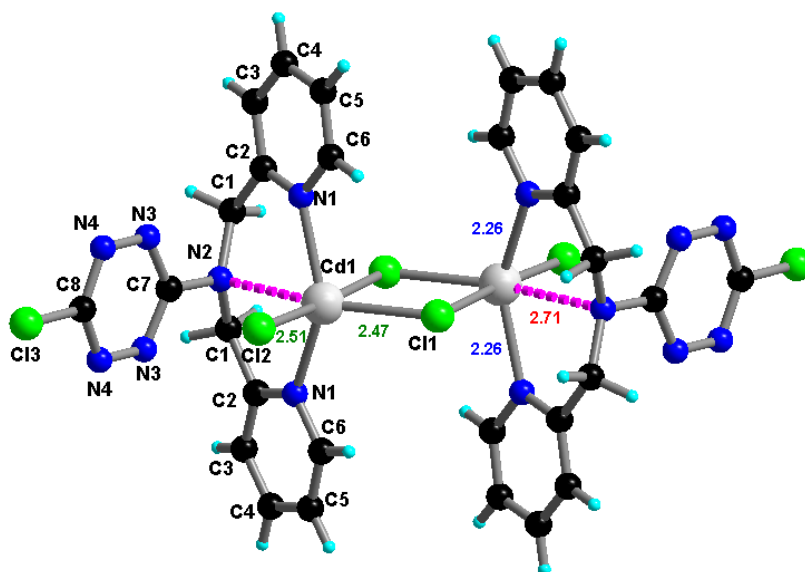
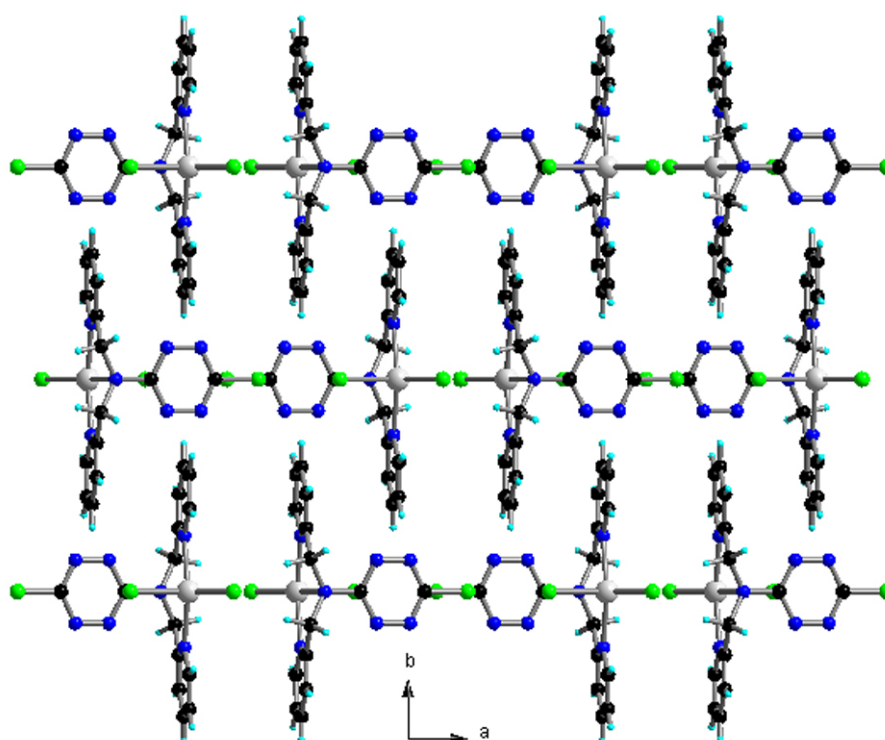
Compound **2a**

Fig. S3 Packing diagram for **2a**.

Compound **2b**Fig. S4 Molecular structure of **2b** along with the numbering scheme.Fig. S5 Packing diagram for **2b**.

Compound 2c

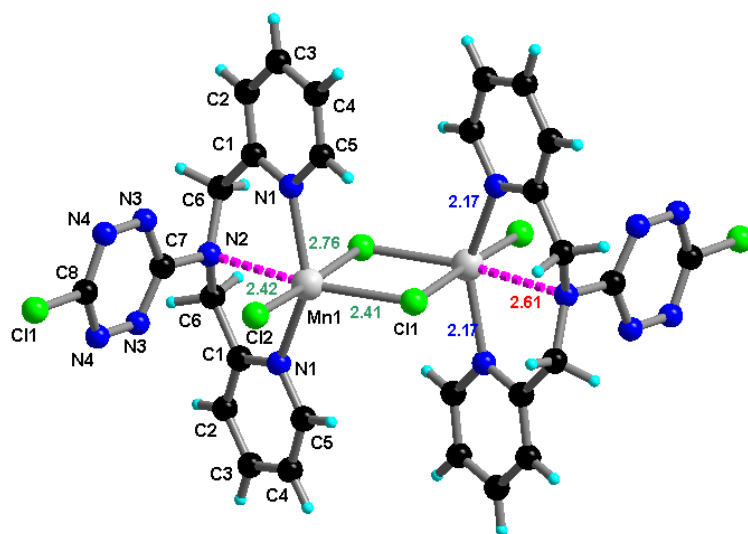


Fig. S6 Molecular structure of 2c along with the numbering scheme.

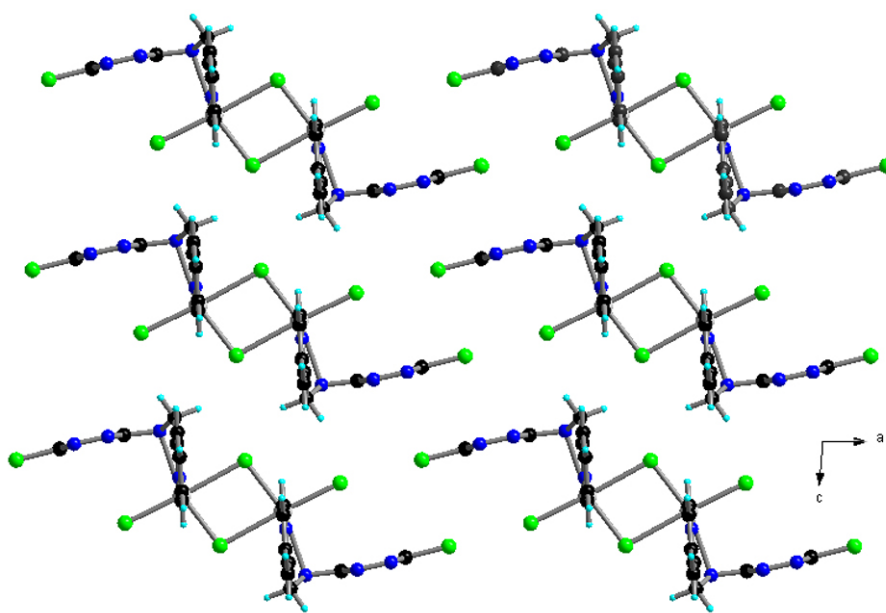


Fig. S7 Packing diagram for 2c.

Titration studies

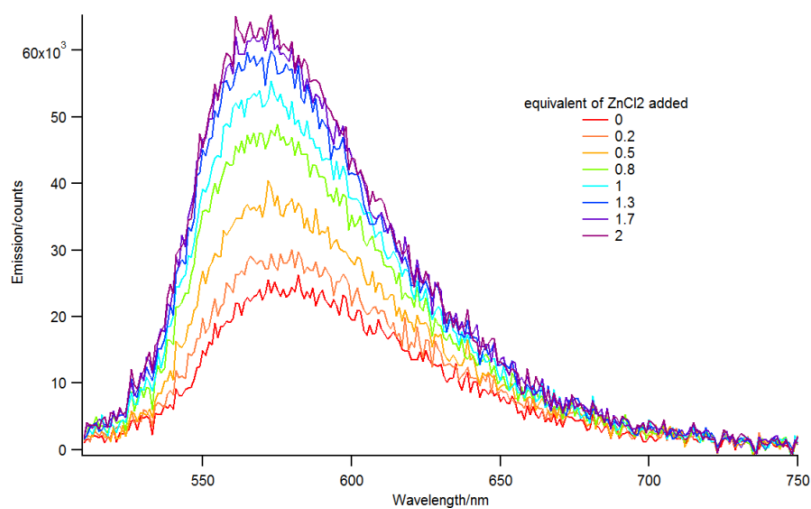


Fig. S8 Evolution of the emission spectrum on adding ZnCl_2 solution to the ligand **1** solution in acetonitrile. The molar ratio of ZnCl_2 : **1** from 0 to 2. Excitation and emission slits are all 5 nm.

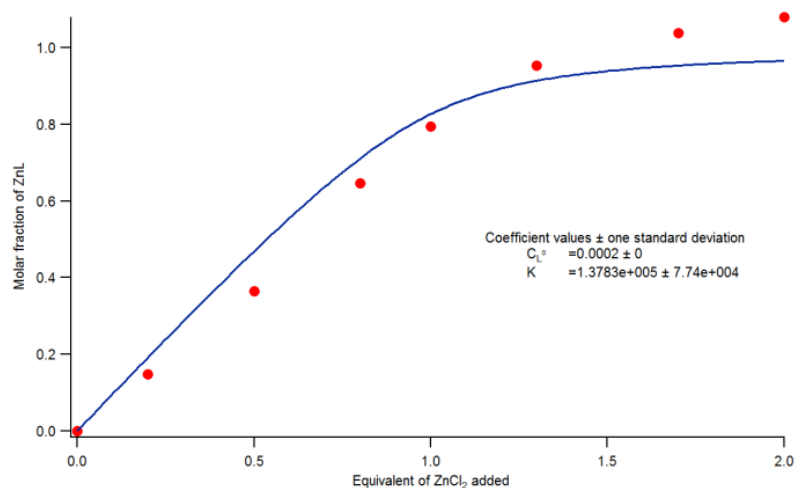
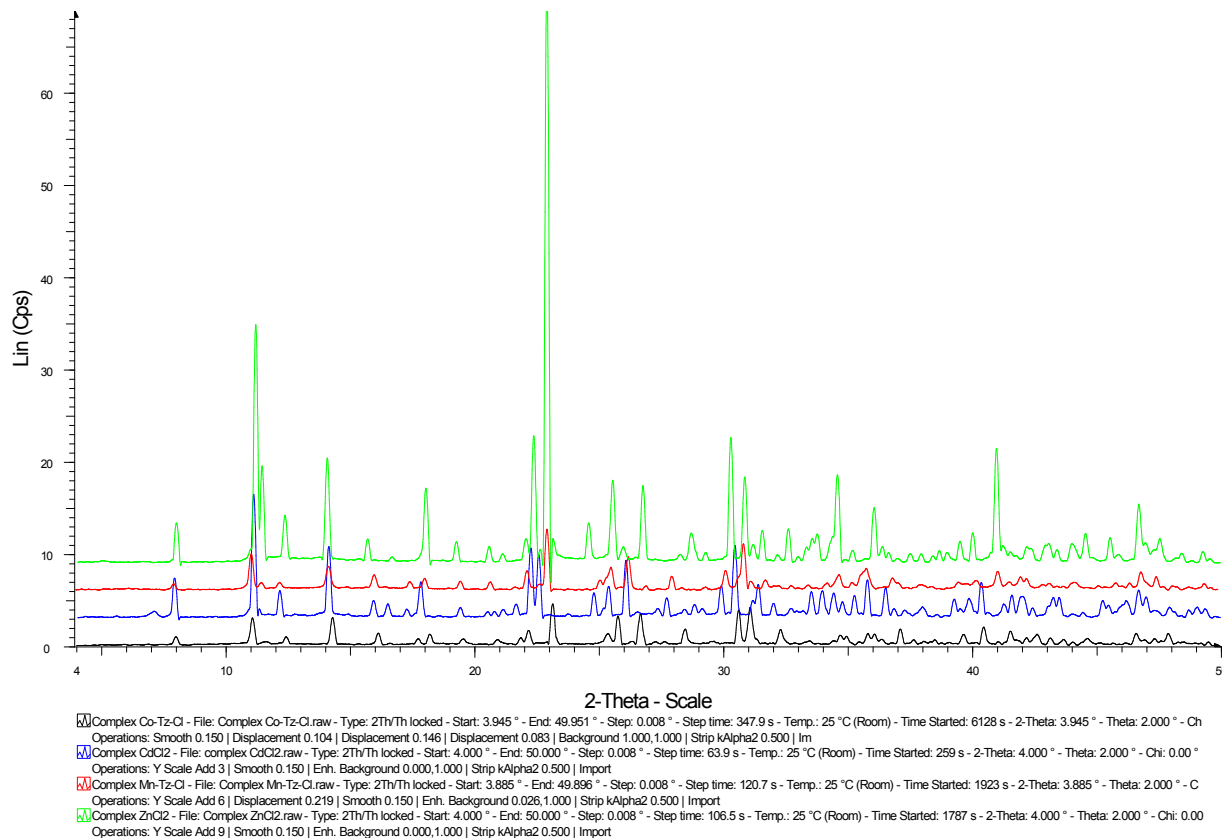
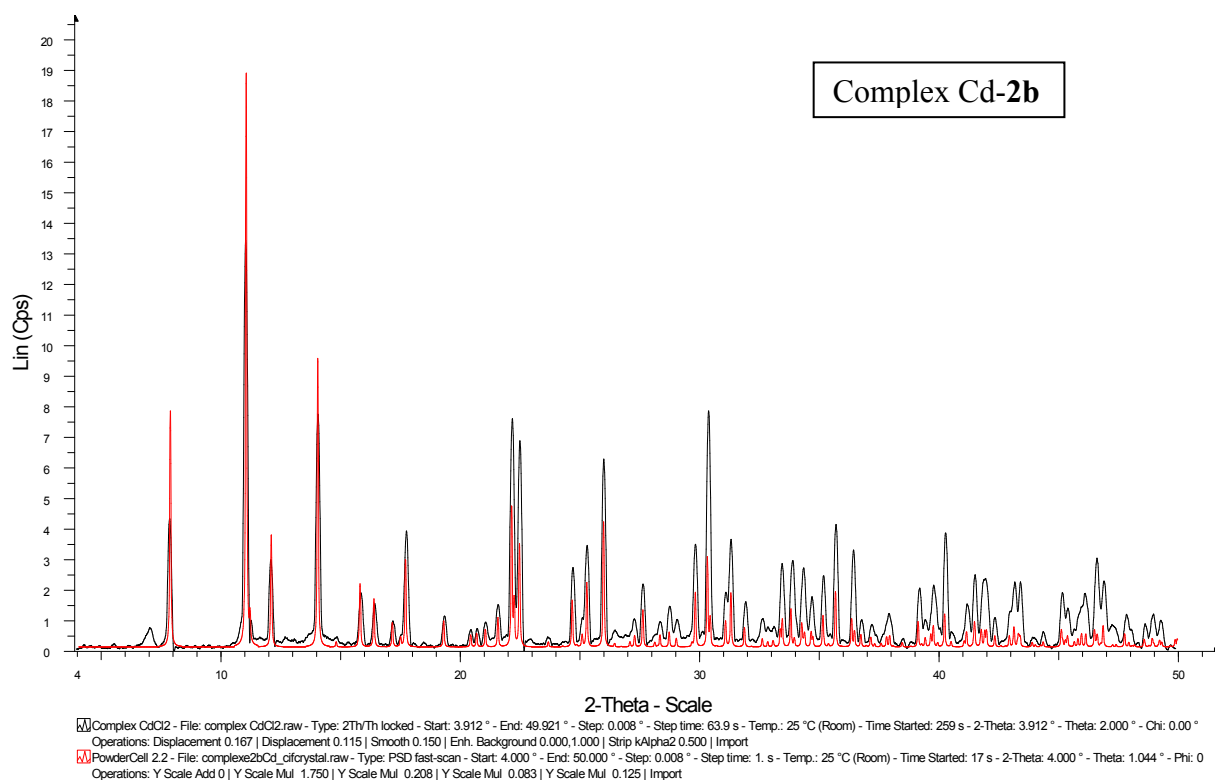
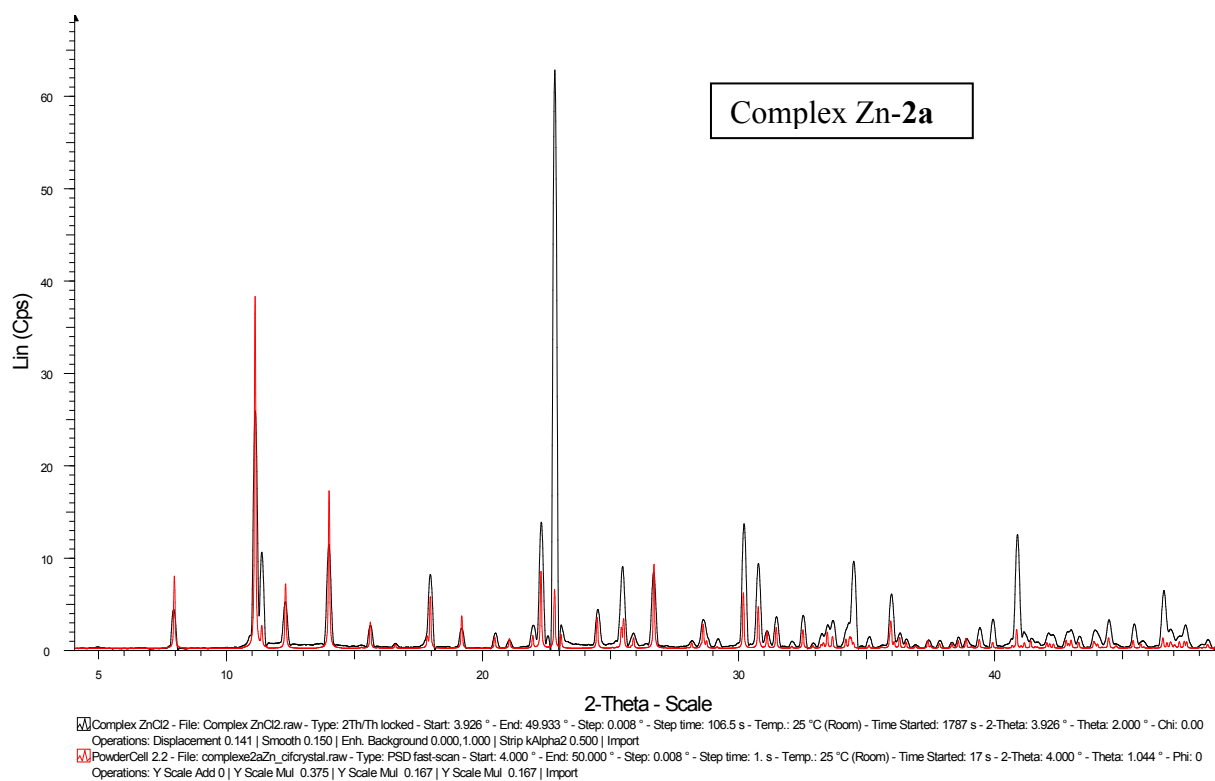


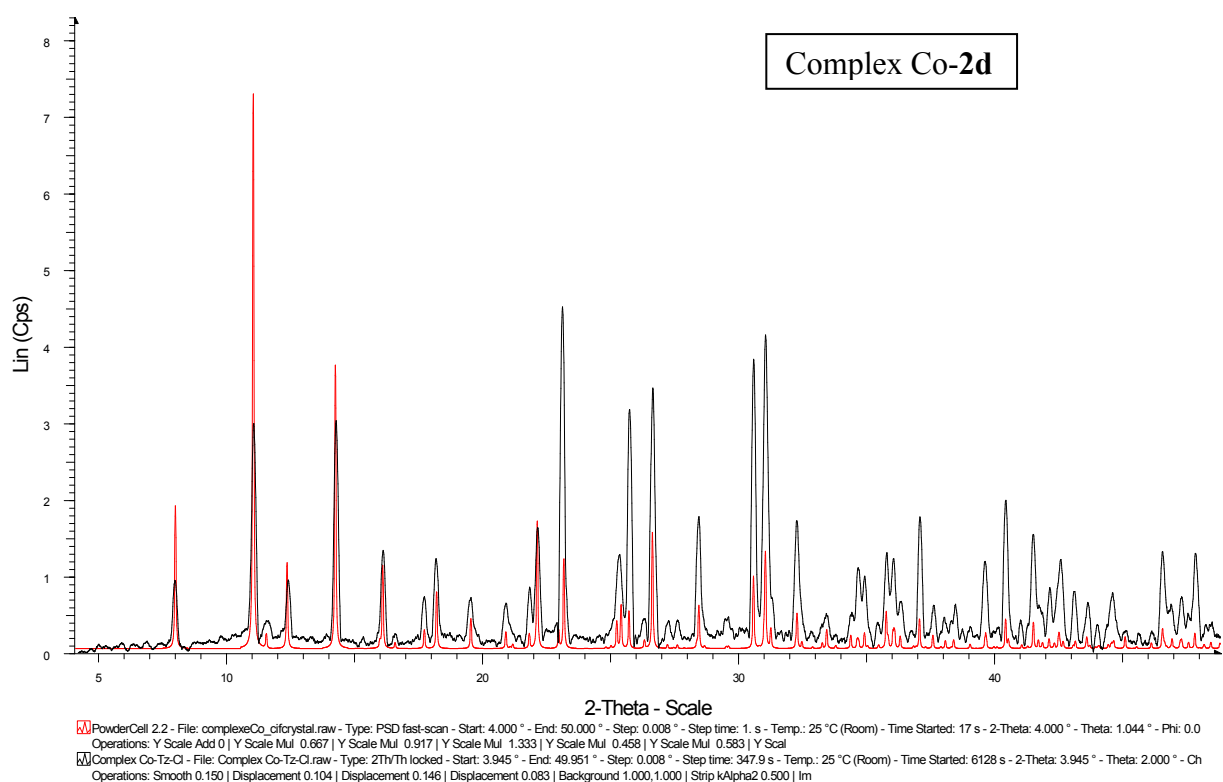
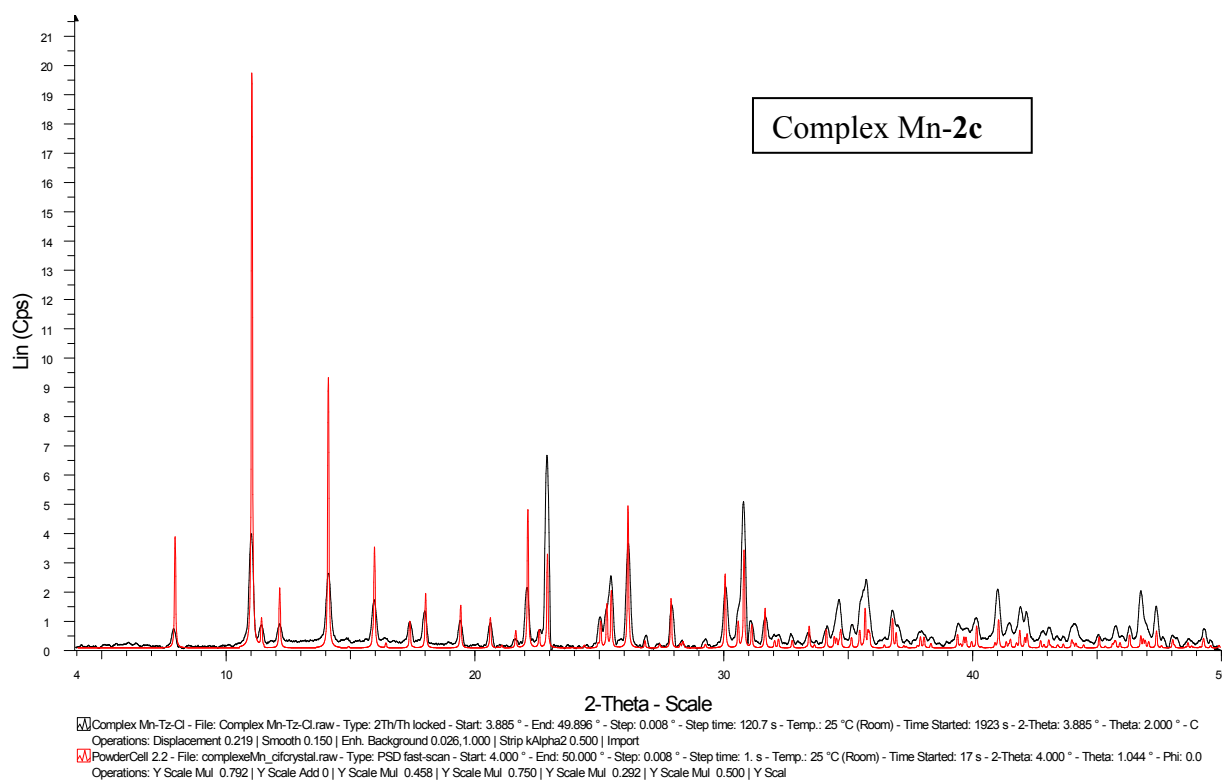
Fig. S9 Normalized evolution of the emission spectrum on adding ZnCl_2 solution to the ligand **1** solution in acetonitrile at an initial concentration of 2×10^{-4} M (●), least squares fit with $K = 1.4 \times 10^5$ (—).

PXRD measurements

X-ray powder diffraction measurements were carried out on a D8 BRUKER diffractometer using Cu-K α 1,2 radiation, equipped with a linear Vantec super speed detector. The X-ray powder diffraction (XRPD) patterns of all complex (**2a**, **2b**, **2c**, **2d**) show that all the observed reflections (black color) can be indexed in the unit cell parameters obtained from single crystal X-ray diffraction experiments (red color).







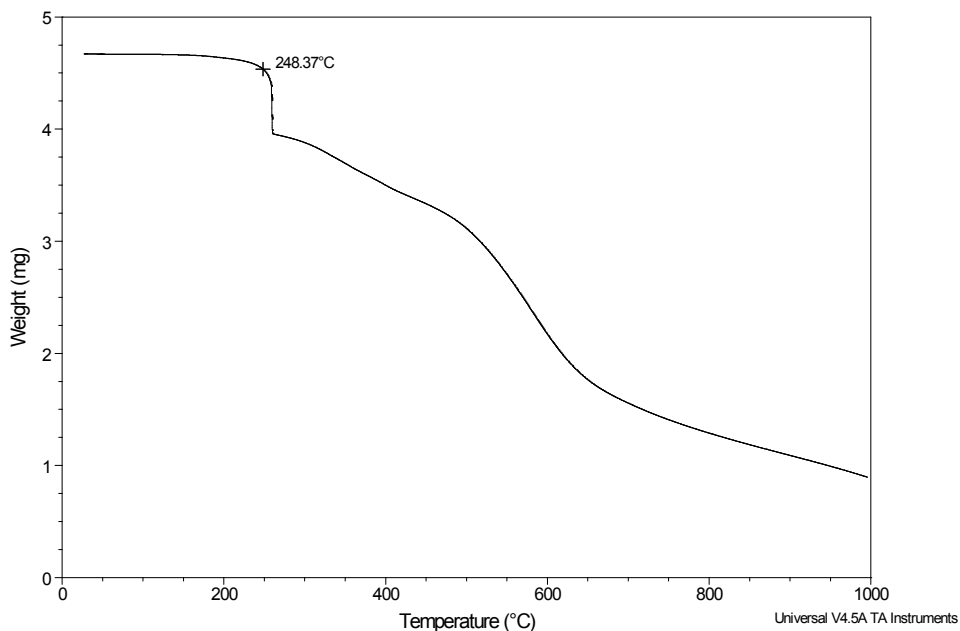
TGA measurements

Thermogravimetric studies (TGA) were performed on a Q500 TA INSTRUMENTS analyzer under nitrogen flow. The two complexes **2b** (up) and **2d** (bottom) are stable till about 250 °C and then start to decompose.

Sample: Complex CdCl₂
Size: 4.6700 mg
Method: Ramp

TGA

File: C:\...\Narcis_complexCdCl₂_18032015.001
Operator: MA
Run Date: 18-Mar-2015 12:16
Instrument: TGA Q500 V20.10 Build 36



Sample: CoTzCl
Size: 6.9030 mg
Method: Ramp

TGA

File: C:\...\Narcis_CoTzCl_16032015.001
Operator: MA
Run Date: 16-Mar-2015 15:42
Instrument: TGA Q500 V20.10 Build 36

