Supporting Information

Heteroleptic naphthalo-phthalocyaninates of lutetium: synthesis and spectral and conductivity properties

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Table S1. High-resolution mass spectrometry MALDI TOF/TOF data.

Compound	Mass found	Monoisotopic
(Molecular formula)		mass calculated
$3a (C_{128}H_{72}LuN_{16})$	2007.5519	2007.5534
3b (C ₁₂₈ H ₇₅ Cl ₈ LuN ₁₆)	2290.1597	2290.3277
3c (C ₁₇₆ H ₁₀₅ LuN ₁₆)	2616.8997	2616.8116
3d (C ₁₂₈ H ₇₃ LuN ₁₆ O ₈)	2136.4387	2136.5205
3e (C ₁₂₈ H ₇₆ Cl ₈ LuN ₁₆ O ₈)	2419.2322	2419.2943
$3f(C_{176}H_{106}LuN_{16}O_{16})$	2873.7937	2873.7375

Table S2. ¹H NMR data.

Compound	H _{Ph} (or H _{PhO})	β -H _{Pc}	β -H _{Nc}	α -H _{Pc}	α -H _{Nc}	Solvent
NcLuPc ¹⁰	-	8.13	8.72	8.72	9.32	[D ₆]DMSO
^{Ph} PcLu ^{Ph} Pc ⁴	7.35-7.43	-	-	9.02	-	[D ₆]DMSO–CDCl ₃ (3:1, V/V)
39	7.40-7.56	8.17	8.72	8.81	9.42	[D ₆]DMSO
Ja	7.38-7.54	8.16	8.67	8.83	9.40	[D ₆]DMSO:CDCl ₃ (3:1, V:V)
3b	7.40-7.54	-	8.72	8.77	9.45	[D ₆]DMSO:CDCl ₃ (3:1, V:V)
3c	7.72-7.75 and 7.87-7.90	-	8.54	9.26	9.56	[D ₆]DMSO
3d	7.18-7.28 and 7.44-7.50	7.94	8.08	8.81	9.15	[D ₈]THF
3e	7.55-7.74	-	8.14	8.61-8.73	9.16	[D ₆]DMSO
3f	7.51-7.59 and 7.67-7.75	-	8.15	8.17	9.27	[D ₆]DMSO



Figure S1. MALDI-TOF mass spectrum of a reaction mixture for synthesis of complex 3c.



Figure S2. High-resolution MALDI-TOF/TOF mass spectrum of **3b**, isotopic patterns for the molecular ion (inset A) and simulated MS patterns of the molecular ion (inset B).



Figure S3. High-resolution MALDI-TOF/TOF mass spectrum of **3c**, isotopic patterns for the molecular ion (inset A) and simulated MS patterns of the molecular ion (inset B).



Figure S4. High-resolution MALDI-TOF/TOF mass spectrum of **3d**, isotopic patterns for the molecular ion (inset A) and simulated MS patterns of the molecular ion (inset B).



Figure S5. High-resolution MALDI-TOF/TOF mass spectrum of **3e**, isotopic patterns for the molecular ion (inset A) and simulated MS patterns of the molecular ion (inset B).



Figure S6. High-resolution MALDI-TOF/TOF mass spectrum of **3f**, isotopic patterns for the molecular ion (inset A) and simulated MS patterns of the molecular ion (inset B).



Figure S7. ¹H NMR spectra of complexes **3b** and **3e**.



Figure S8. ¹H NMR spectra of complexes **3c** and **3f**.



Figure S9. UV-Vis spectra of reduced by N2H4 •H2O forms of heteroleptic complexes 3 in DMSO.



Figure S10. I-V curves for thin films of heteroleptic complexes 3.



Figure S11. UV-Vis spectra of thin films and solutions in CCl_4 for heteroleptic complexes **3**. The Q-band: B-band intensity ratios for thin films are given in brackets.



Figure S12. Conductivity of a thin film of complex 3a as a function of temperature (ln(σ) vs. l/T).



Figure S13. Conductivity of a thin film of complex **3b** as a function of temperature $(ln(\sigma) \text{ vs. } l/T)$.



Figure S14. Conductivity of a thin film of complex 3c as a function of temperature (ln(σ) vs. l/T).



Figure S15. Conductivity of a thin film of complex 3d as a function of temperature (ln(σ) vs. l/T).



Figure S16. Conductivity of a thin film of complex 3e as a function of temperature (ln(σ) vs. l/T).



Figure S17. Conductivity of a thin film of complex **3f** as a function of temperature $(ln(\sigma) \text{ vs. } l/T)$.



Figure S18. Tauc plot for the Q-band of the thin film of phthalocyanine $\mathbf{3b}$.



Figure S19. Tauc plot for the RV-band of phthalocyanine 3b.