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## **Supporting Information for**

## The Selectivity of Water-based Pyrophosphate Recognition is Tuned by Metal Substitution in Dimetallic Receptors

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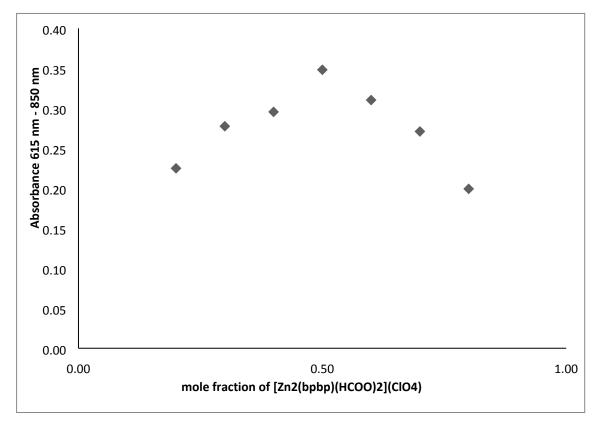


Figure S1. Job plot of 50  $\mu$ M [Zn<sub>2</sub>(bpbp)(HCOO)<sub>2</sub>](ClO<sub>4</sub>) with pyrocatechol violet.

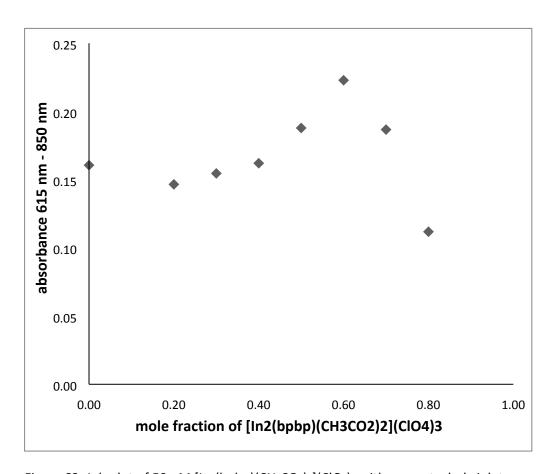


Figure S2. Job plot of 50  $\mu$ M [In<sub>2</sub>(bpbp)(CH<sub>3</sub>CO<sub>2</sub>)<sub>2</sub>](ClO<sub>4</sub>)<sub>3</sub> with pyrocatechol violet.

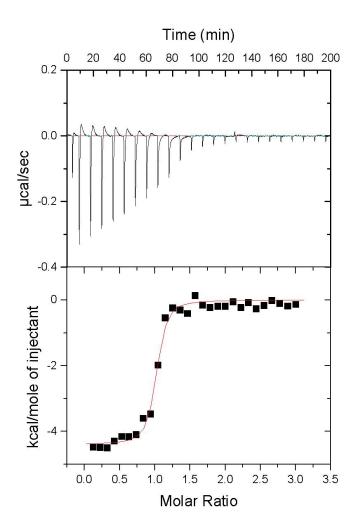


Figure S3. Binding isotherm for titration of 0.32 mM  $Na_4P_2O_7$  into 0.017 mM  $[Ga_2(bpbp)(OH)_2(OH_2)_2](CIO_4)_3$ 

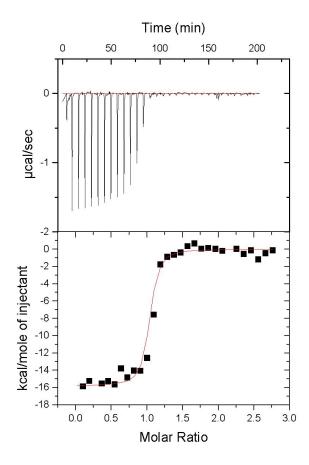


Figure S4. Binding isotherm for titration of 0.339 mM  $Na_4P_2O_7$  into 0.025 mM  $[Zn_2(bpbp)(HCO_2)_2](CIO_4)$ 

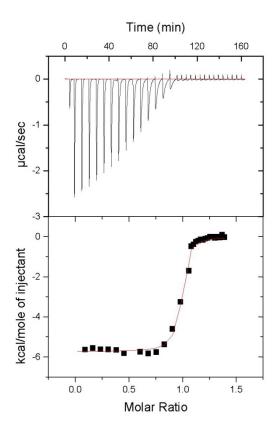


Figure S5. ITC binding isotherm for titration of 0.21 mM  $[Ga_2(bpbp)(OH)_2(H_2O)_2](ClO_4)_3$  with 2.47 mM  $Na_2HPO_4$  in 10 mM HEPES.

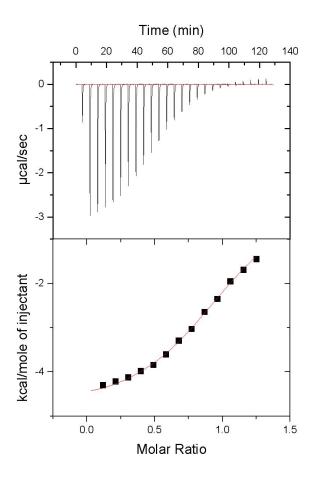


Figure S6. ITC binding isotherm for titration of 0.12 mM  $[Zn_2(bpbp)(HCO_2)_2](ClO_4)$  with 2.47 mM  $Na_2HPO_4$  in 10 mM HEPES.

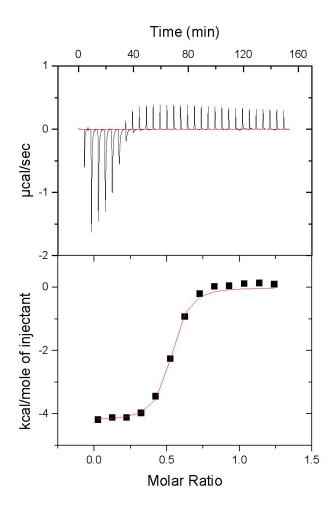


Figure S7. ITC binding isotherm for titration of 0.13 mM  $[In_2(bpbp)(CH_3CO_2)_2](CIO_4)_3$  with 2.47 mM  $Na_2HPO_4$  in 10 mM HEPES.

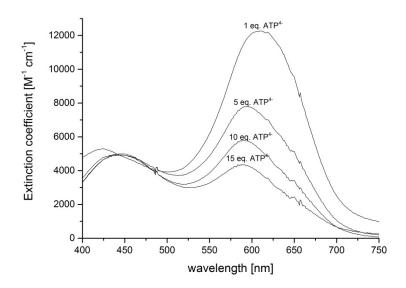
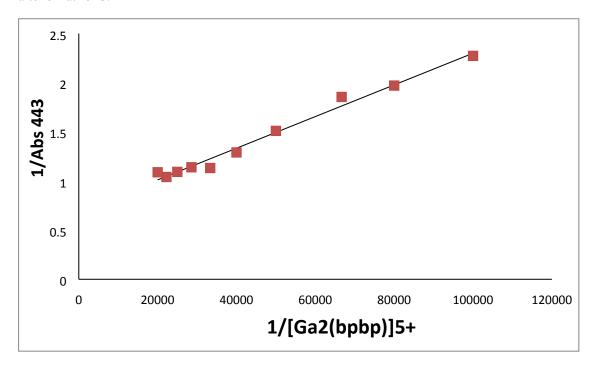


Figure S8. Uv-vis of 25  $\mu$ M  $\{Ga_2(bpbp)(H_nPV)\}^{1+n+}$  in 10 mM HEPES, pH 7.2 with 1 eq., 5 eq., 10 eq. or 15 eq. ATP<sup>4-</sup> after 8 h at 25°C.



**Figure S9**. Benesi-Hildebrand plot of  $[Ga_2(bpbp)(OH)_2(H_2O)_2](ClO_4)_3$  titrated into 50  $\mu$ M PV and linear fit for calculation  $K_a$ . Equation and  $R^2$  value is displayed in plot.

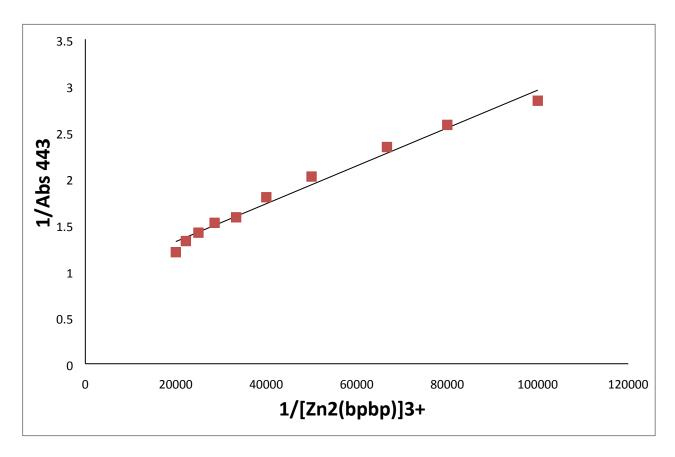


Figure S10. Benesi-Hildebrand plot of  $[Zn_2(bpbp)(HCO_2)_2](ClO_4)$  titrated into 50  $\mu$ M PV and linear fit for calculation  $K_a$ . Equation and  $R^2$  value is displayed in plot.