

Supporting information

IR- spectra.

Table S1 IR-spectra (cm⁻¹ in KBr) of starting compounds and complexes **1-4**.

	{CpMo (CO) ₃ } ₂	{Cp*Mo (CO) ₂ } ₂	PPN-Cl	C ₆₀	C ₇₀	(PPN ⁺) {CpMo(CO) ₂ (η^2 -C ₆₀)} ⁻ (1)	(PPN ⁺) {CpMo(CO) ₂ (η^2 -C ₇₀)} ⁻ ·C ₆ H ₁₄ (2)	(PPN ⁺) {Cp*Mo(CO) ₂ (η^2 -C ₆₀)} ⁻ ·C ₆ H ₅ CN ·C ₆ H ₄ Cl ₂ (3)	(PPN ⁺) {CpMo(CO) ₃ } ⁻ (4)
Cp(*) Mo(CO) _x , x = 2, 3	417w 451m 477w 501m 547m 587m 824s 1016w 1264w 1419w 1427w C=O 1886s 1899s 1953s C-H 3095w 3115w	448w 482w 495m 531s 1027m 1384s 1416w 1450w 1478w C=O 1825s 1870s C-H 2861w 2908w 2970w				- 437w - 495m* 546m* - - 1026w* 1282s* - - C=O 1798s 1819s 1888s C-H 3088w -	- 438w 454w* 497m* 540m 580w* - - 1283s* 1414m* 1433m C=O - 1812s 1898s C-H 3091w -	- 438w 484w 503s* 532s* 1025w* - - 1281s* 1454m* 1481w* C=O 1787s 1878s C-H 2853w 2924w 2961w	- 447w 461w 502s* 542m* 616m - - 1272s* - - C=O 1768s 1786s 1895s C-H 3088w
PPN ⁺			500s 531s 550s 694m 724m 746m 754m 997m 1024w 1075w 1115s 1262s - - 1439s 1483m 1587m 3046w			495m* 531s* 546m* 690s 722s 752w - 998m 1026w 1074w 1113s 1282s* 1300m 1319m 1436s 1479w 1587w 3057w	497m* 533s 545m 691s 721s* 743w* - 998w 1026w 1089w* 1114s 1283s* 1298m 1318m 1438m 1482w 1588w 3056w	503s* 532s* 546m 689s 722s 744w 753w 998m 1025w* 1070w 1114s 1261s 1281s* 1300m 1436s 1481w* 1587w 3054w	502s* 535s 542m* 693s 723s 752m 758m 998m 1027w 1072w 1117s 1272s* 1289s 1303s 1438s 1482m 1586w 3055w
Fullerene				C ₆₀ 526s - 576m - 1182m 1429s	C ₇₀ 457w 534s 564m 576s - 642m 673m 794m 1132w - 1413w 1429s	Coord. C ₆₀ 531s* 566w 576w 582w 1182m 1402w 1416s	Coord. C ₇₀ 454w* 530s 567w 580w* 607w 639w 672w 791m - 1406m 1414m* 1426s*	Coord. C ₆₀ ⁻ 532m* 576m 1183m 1392s 1412w	
Solvent							C ₆ H ₁₄ 721s* 743w* 1089w* - 1462w 2926w 2951w	C ₆ H ₄ Cl ₂ - 740w 1025w* 1454m*	

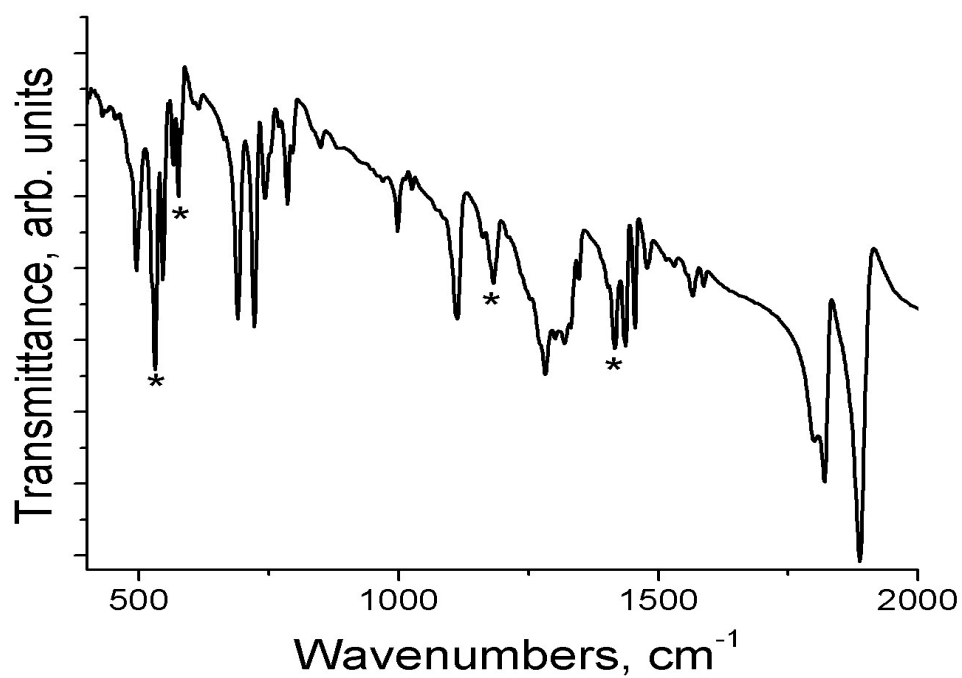


Fig. S1 IR-spectrum of $(\text{PPN}^+)\{\text{CpMo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ (**1**) in KBr pellet prepared in anaerobic conditions. Absorption bands of C_{60} are marked by asterisks.

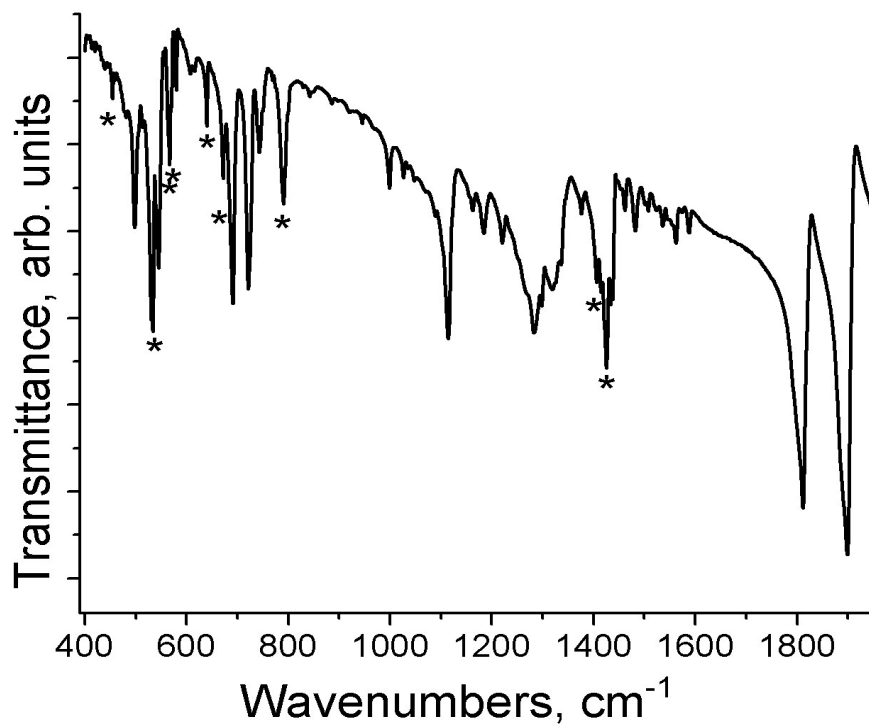


Fig. S2 IR-spectrum of $(\text{PPN}^+)\{\text{CpMo}(\text{CO})_2(\eta^2\text{-C}_{70})\}^- \cdot \text{C}_6\text{H}_{14}$ (**2**) in KBr pellet prepared in anaerobic conditions. Absorption bands of C_{70} are marked by asterisks.

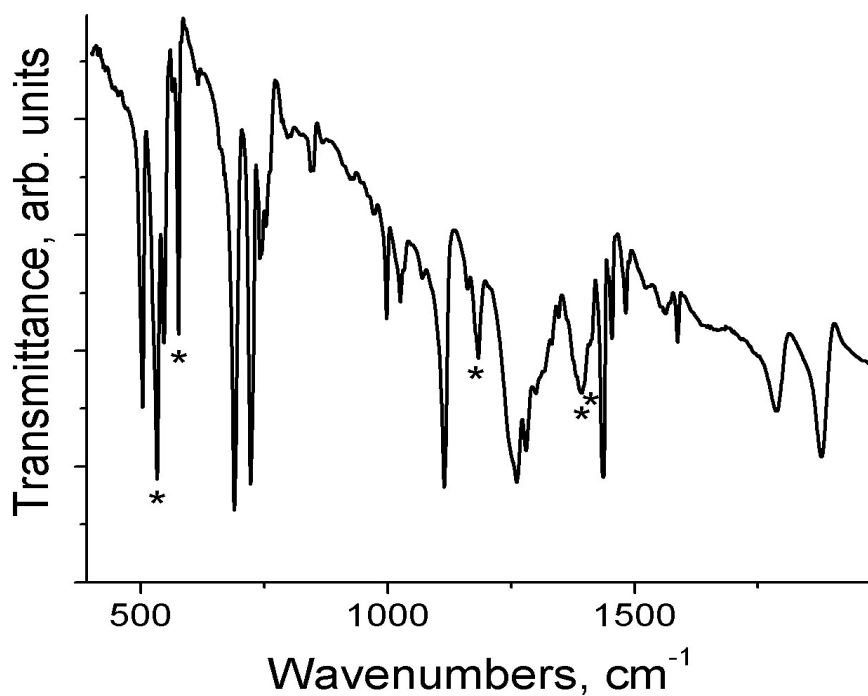


Fig. S3 IR-spectrum of $(\text{PPN}^+)\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-\cdot\text{C}_6\text{H}_4\text{Cl}_2\cdot\text{C}_6\text{H}_5\text{CN}$ (**3**) in KBr pellet prepared in anaerobic conditions. Absorption bands of C_{60} are marked by asterisks.

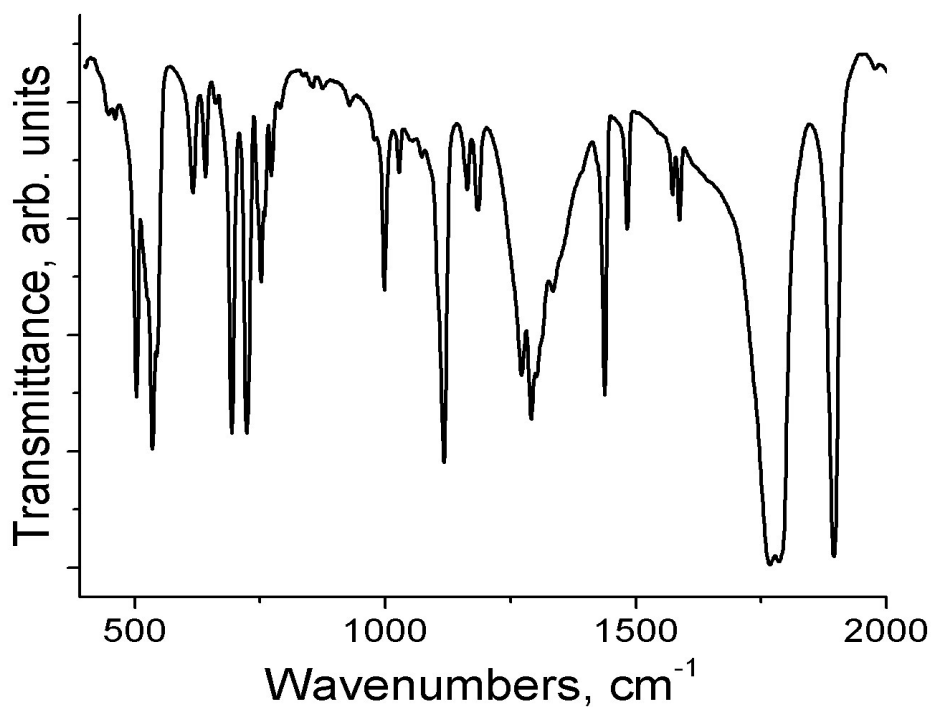


Fig. S4 IR-spectrum of $(\text{PPN}^+)\{\text{CpMo}(\text{CO})_3\}^-$ (**4**) in KBr pellet prepared in anaerobic conditions.

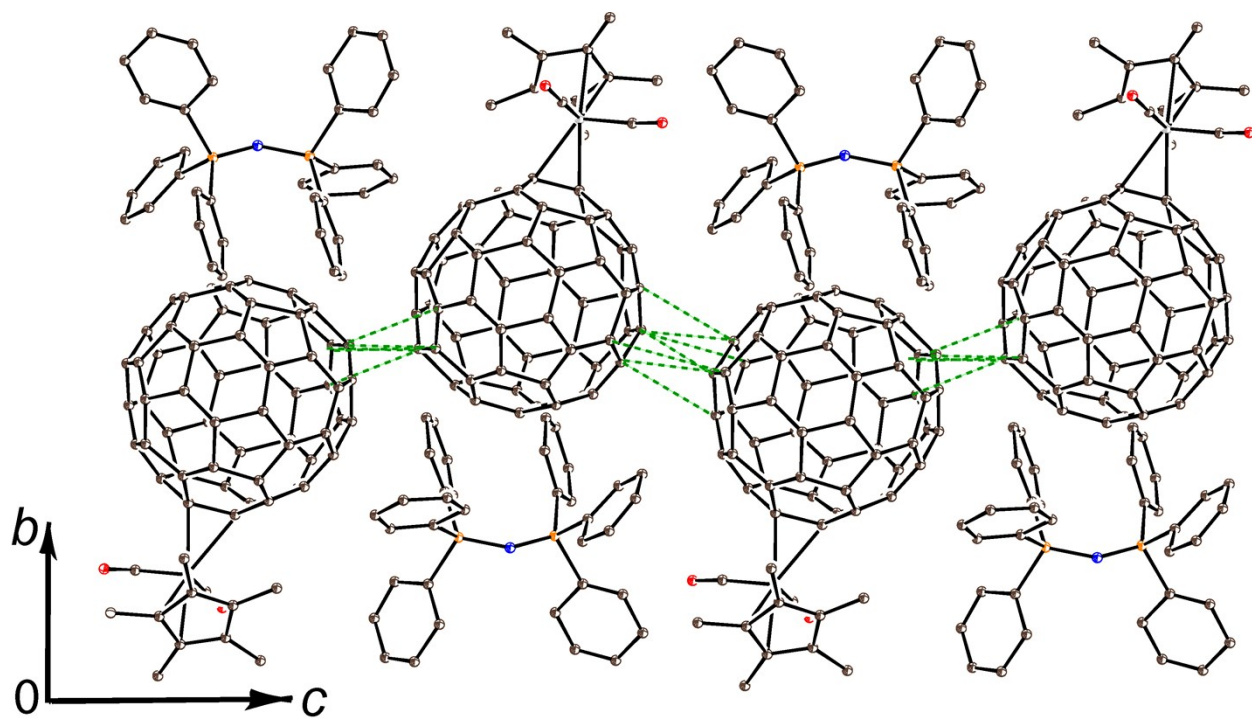


Fig. S5 View on zigzag fullerene chains in complex **3** arranged along the *c* axis and formed by closely packed $\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ anions.

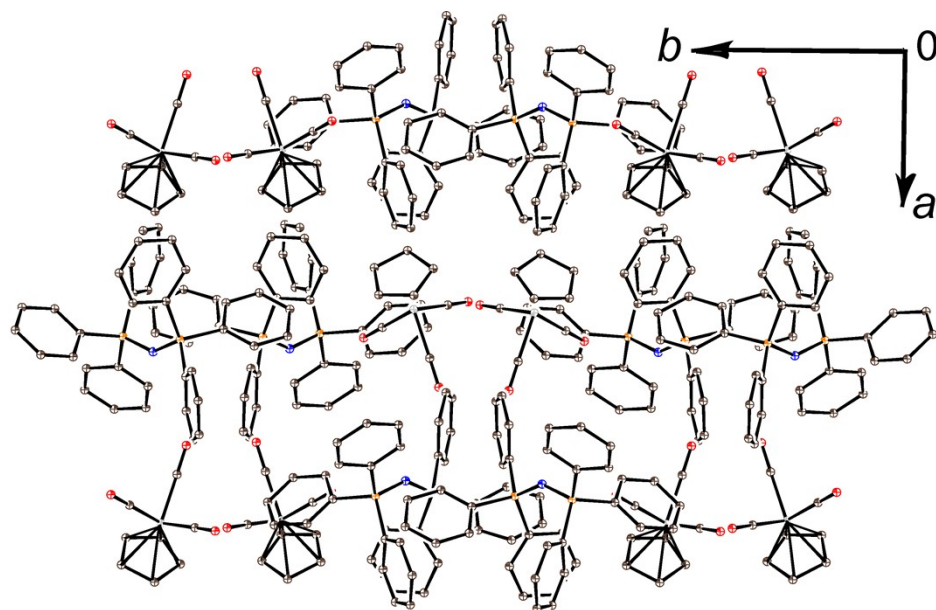


Fig. S6 View on the crystal structure of $(\text{PPN}^+)\{\text{CpMo}(\text{CO})_3\}^-$ (**4**) along the *c* axis.

Data of EPR spectra for complex 3.

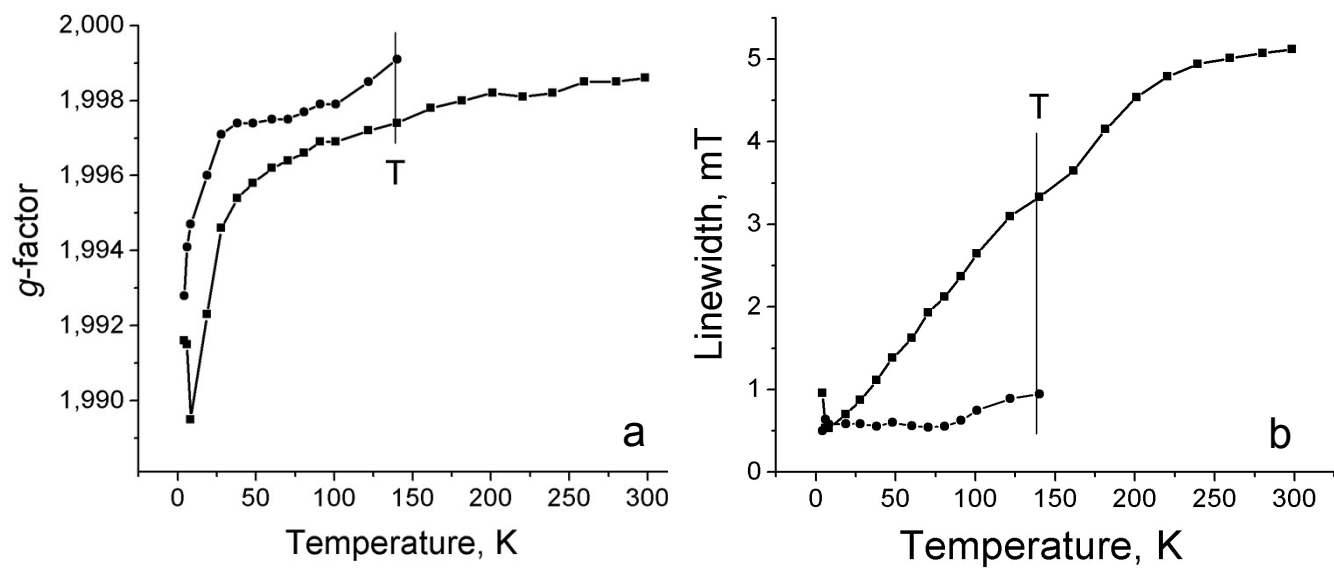


Fig. S7 Temperature dependence of g-factor (a) and linewidth (b) of EPR signal from polycrystalline complex **3**. T marks the temperature of splitting of EPR signal.

Theoretical calculations

Table S2. Calculated state, total energies (E), relative energies (ΔE), $\langle S^2 \rangle$ values, and number of imaginary frequencies (NImag.) of the singlet and triplet states in $\{\text{CpMo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ and $\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ at the CAM-B3LYP-D3 and M11/cc-pVTZ-PP/cc-pVDZ levels of theory

Method ^a	State	E / hartree	ΔE / eV	$\langle S^2 \rangle$	NImag.
$\{\text{CpMo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$					
RCAM-B3LYP-D3	1A	-2773.3329	—	0	0
RM11	1A	-2773.1358	—	0	0
RCAM-B3LYP-D3	1A	-2773.2421	0	0	— ^b
UCAM-B3LYP-D3	3A	-2773.1923	1.35	2.025	— ^b
RM11	1A	-2773.0376	0.00	0	— ^b
UM11	3A	-2772.9910	1.27	2.031	— ^b
$\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$					
RCAM-B3LYP-D3	1A	-2969.8076	0	0	0
UCAM-B3LYP-D3	3A	-2969.7809	0.73	2.024	0
RM11	1A	-2969.5867	0	0	0
UM11	3A	-2969.5633	0.63	2.030	0
RCAM-B3LYP-D3	1A	-2969.1740	0	0	— ^b
UCAM-B3LYP-D3	3A	-2969.1367	1.02	2.045	— ^b
RM11	1A	-2968.9446	0	0	— ^b
UM11	3A	-2968.9095	0.95	2.040	— ^b

^a The cc-pVTZ-PP and cc-pVDZ basis sets were used for molybdenum and the other atoms, respectively.

^b X-ray structure was used without geometry optimization.

Table S3. Observed and calculated bond lengths and Wiberg bond indices of the 1A state in $\{\text{CpMo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ at the RCAM-B3LYP-D3 and RM11/cc-pVTZ-PP/cc-pVDZ levels of theory

	Bond length, Å			Wiberg bond index	
	Obs.	RCAM-B3LYP-D3	RM11	RCAM-B3LYP-D3	RM11
Mo1–C61	2.368(2)	2.414	2.434	0.281	0.273
Mo1–C62	2.354(3)	2.356	2.371	0.334	0.326
Mo1–C63	2.351(3)	2.330	2.345	0.355	0.342
Mo1–C64	2.369(3)	2.356	2.371	0.334	0.326
Mo1–C65	2.392(3)	2.414	2.434	0.281	0.273
Mo1–C66	1.943(3)	1.946	1.942	1.325	1.310
Mo1–C67	1.947(3)	1.946	1.942	1.325	1.310
C66–O1	1.157(3)	1.163	1.159	2.013	2.033
C67–O2	1.164(3)	1.163	1.159	2.013	2.033
Mo1–C9	2.252(2)	2.234	2.220	0.572	0.586
Mo1–C1	2.241(2)	2.234	2.220	0.572	0.586
C9–C1	1.497(3)	1.509	1.516	0.976	0.961
C9–C10	1.478(3)	1.484	1.486	1.062	1.058
C9–C8	1.480(3)	1.482	1.484	1.057	1.052
C1–C2	1.475(3)	1.484	1.486	1.062	1.058
C1–C5	1.483(3)	1.482	1.484	1.057	1.052
Total of Mo				6.035	5.963

Table S4. Observed and calculated bond lengths and Wiberg bond indices of the 1A state in $\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ at the RCAM-B3LYP-D3 and RM11/cc-pVTZ-PP/cc-pVDZ levels of theory

	Bond length, Å			Wiberg bond index	
	Obs.	RCAM-B3LYP-D3	RM11	RCAM-B3LYP-D3	RM11
Mo1B–C1B	2.306(7)	2.344	2.359	0.329	0.315
Mo1B–C2B	2.450(6)	2.447	2.464	0.258	0.248
Mo1B–C3B	2.513(7)	2.447	2.462	0.258	0.252
Mo1B–C4B	2.415(7)	2.344	2.351	0.329	0.324
Mo1B–C5B	2.284(7)	2.305	2.317	0.346	0.332
Mo1B–C11B	1.815(16)	1.940	1.931	1.329	1.332
Mo1B–C12B	1.943(14)	1.940	1.942	1.329	1.296
C11B–O1B	1.23(2)	1.164	1.162	1.998	2.011
C12B–O2B	1.228(16)	1.164	1.160	1.998	2.028
Mo1B–C302	2.385(13)	2.247	2.236	0.549	0.560
Mo1B–C307	2.205(11)	2.247	2.229	0.549	0.568
C302–C307	1.525(19)	1.511	1.518	0.974	0.959
C302–C301	1.476(17)	1.481	1.482	1.068	1.064
C302–C303	1.421(17)	1.482	1.484	1.059	1.054
C307–C314	1.466(17)	1.481	1.484	1.068	1.062
C307–C315	1.528(17)	1.482	1.483	1.059	1.055
Total of Mo				6.016	5.942

Table S5. Observed and calculated bond lengths and Wiberg bond indices of the 3A state in $\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ at the UCAM-B3LYP-D3 and UM11/cc-pVTZ-PP/cc-pVDZ levels of theory

	Bond length, Å			Wiberg bond index	
	Obs.	UCAM-B3LYP-D3	UM11	UCAM-B3LYP-D3	UM11
Mo1B–C1B	2.306(7)	2.355	2.359	0.314	0.307
Mo1B–C2B	2.450(6)	2.408	2.410	0.269	0.265
Mo1B–C3B	2.513(7)	2.397	2.397	0.275	0.273
Mo1B–C4B	2.415(7)	2.338	2.340	0.340	0.334
Mo1B–C5B	2.284(7)	2.317	2.323	0.349	0.340
Mo1B–C11B	1.815(16)	1.985	1.998	1.172	1.125
Mo1B–C12B	1.943(14)	1.970	1.973	1.189	1.160
C11B–O1B	1.23(2)	1.153	1.148	2.087	2.128
C12B–O2B	1.228(16)	1.154	1.151	2.078	2.105
Mo1B–C302	2.385(13)	2.306	2.267	0.455	0.485
Mo1B–C307	2.205(11)	2.234	2.209	0.525	0.553
C302–C307	1.525(19)	1.465	1.472	1.103	1.083
C302–C301	1.476(17)	1.486	1.489	1.044	1.032
C302–C303	1.421(17)	1.485	1.489	1.043	1.032
C307–C314	1.466(17)	1.487	1.490	1.039	1.031
C307–C315	1.528(17)	1.489	1.492	1.035	1.028
Total of Mo				5.453	5.381

Table S6. Sum of charge densities for the 1A and 3A states in $\{\text{CpMo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ and $\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ by Mulliken and natural population analyses at the CAM-B3LYP-D3 and M11/cc-pVTZ-PP/cc-pVDZ levels of theory^a.

	1A state		3A state	
	Mulliken	NPA ^b	Mulliken	NPA ^b
$\{\text{CpMo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$				
CAM-B3LYP-D3				
Mo1	-0.589	-0.892		
C66-O1	0.121	0.250		
C67-O2	0.121	0.250		
Cp	0.084	0.029		
C ₆₀	-0.737	-0.638		
M11				
Mo1	-0.469	-0.916		
C66-O1	0.112	0.284		
C67-O2	0.112	0.284		
Cp	0.044	-0.005		
C ₆₀	-0.799	-0.648		
$\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$				
CAM-B3LYP-D3				
Mo1B	-0.453	-0.815	-0.411	-0.533
C11B-O1B	0.113	0.241	0.180	0.300
C12B-O2B	0.113	0.241	0.182	0.289
Cp*	0.048	0.018	0.216	0.107
C ₆₀	-0.822	-0.686	-1.166	-1.163
M11				
Mo1B	-0.373	-0.842	-0.360	-0.542
C11B-O1B	0.096	0.268	0.178	0.346
C12B-O2B	0.106	0.286	0.186	0.326
Cp*	0.042	-0.020	0.223	0.076
C ₆₀	-0.869	-0.692	-1.227	-1.206

^a Geometry optimization was carried out.

^b Natural population analysis.

Table S7. Observed and calculated frequencies (cm^{-1}) of CO stretching mode of the 1A states in $\{\text{CpMo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ and $\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ at the RCAM-B3LYP-D3 and RM11/cc-pVTZ-PP/cc-pVDZ levels of theory.

Obs.	Calc. ^a	
	RCAM-B3LYP-D3	RM11
$\{\text{CpMo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$		
1798		
1819	1820	1822
1888	1879	1877
$\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$		
1787	1807	1808
1878	1868	1866

^a The calculated frequencies at RCAM-B3LYP-D3 and RM11/cc-pVTZ-PP/cc-pVDZ levels are scaled by 0.9303 and 0.9244, respectively.

Table S8. Sum of spin densities for the 3A state^a in $\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ by Mulliken and natural population analyses at the UCAM-B3LYP-D3 and UM11/cc-pVTZ-PP/cc-pVDZ levels of theory.

	UCAM-B3LYP-D3		UM11	
	Mulliken	NPA ^b	Mulliken	NPA ^b
Mo1B	0.813	0.783	0.830	0.805
C11B–O1B	0.118	0.109	0.099	0.093
C12B–O2B	–0.010	–0.004	–0.011	–0.006
Cp*	0.068	0.082	0.067	0.079
C ₆₀	1.011	1.030	1.015	1.029

^a Geometry optimization was carried out. ^b Natural population analysis.

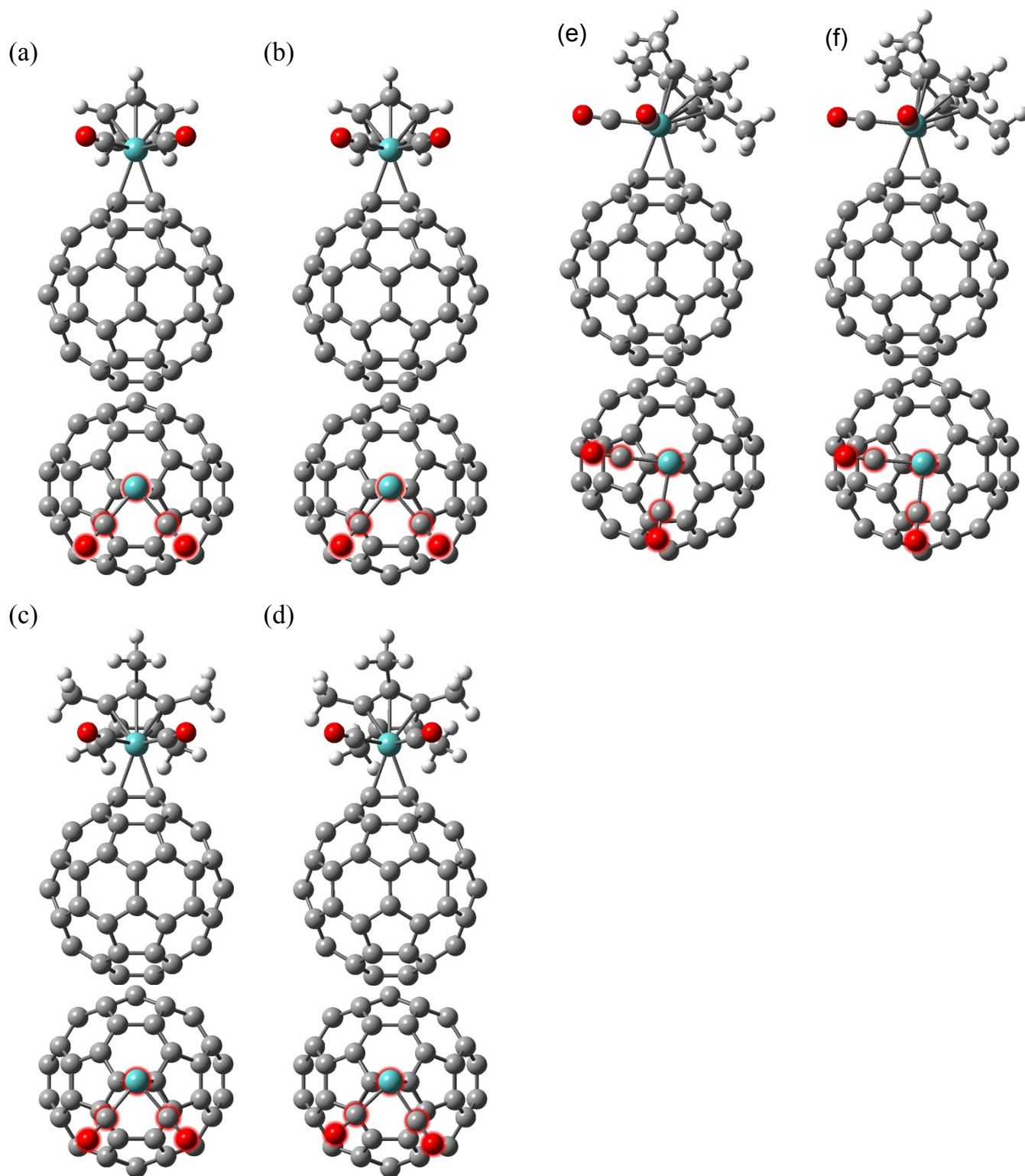


Fig. S8 Optimized structures: $\{\text{CpMo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ in the 1A states at the (a) RCAM-B3LYP-D3 and (b) RM11; $\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ in the 1A states at the (c) RCAM-B3LYP-D3 and (d) RM11, and the 3A states at the (e) UCAM-B3LYP-D3 and (f) UM11/cc-pVTZ-PP/cc-pVDZ levels of theory. In the top view, cyclopentadienyl and pentamethylcyclopentadienyl ligands are omitted, and $\text{Mo}(\text{CO})_2$ moiety is highlighted for clarity.

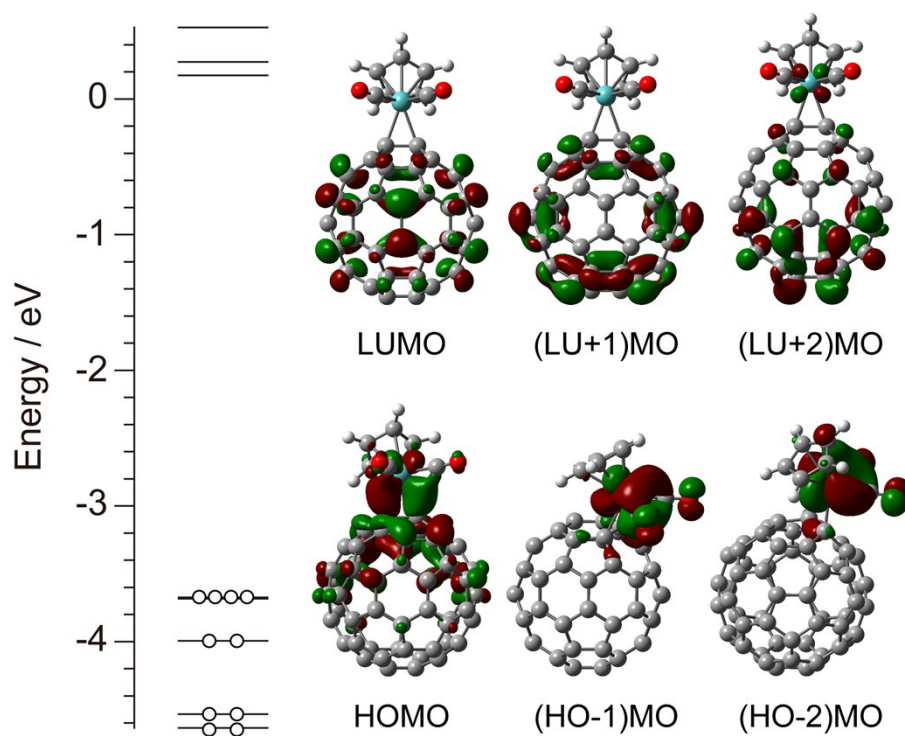


Fig. S9 Energy diagrams for the frontier Kohn-Sham orbitals of the 1A state in $\{\text{CpMo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ calculated at the RCAM-B3LYP-D3 levels of theory.

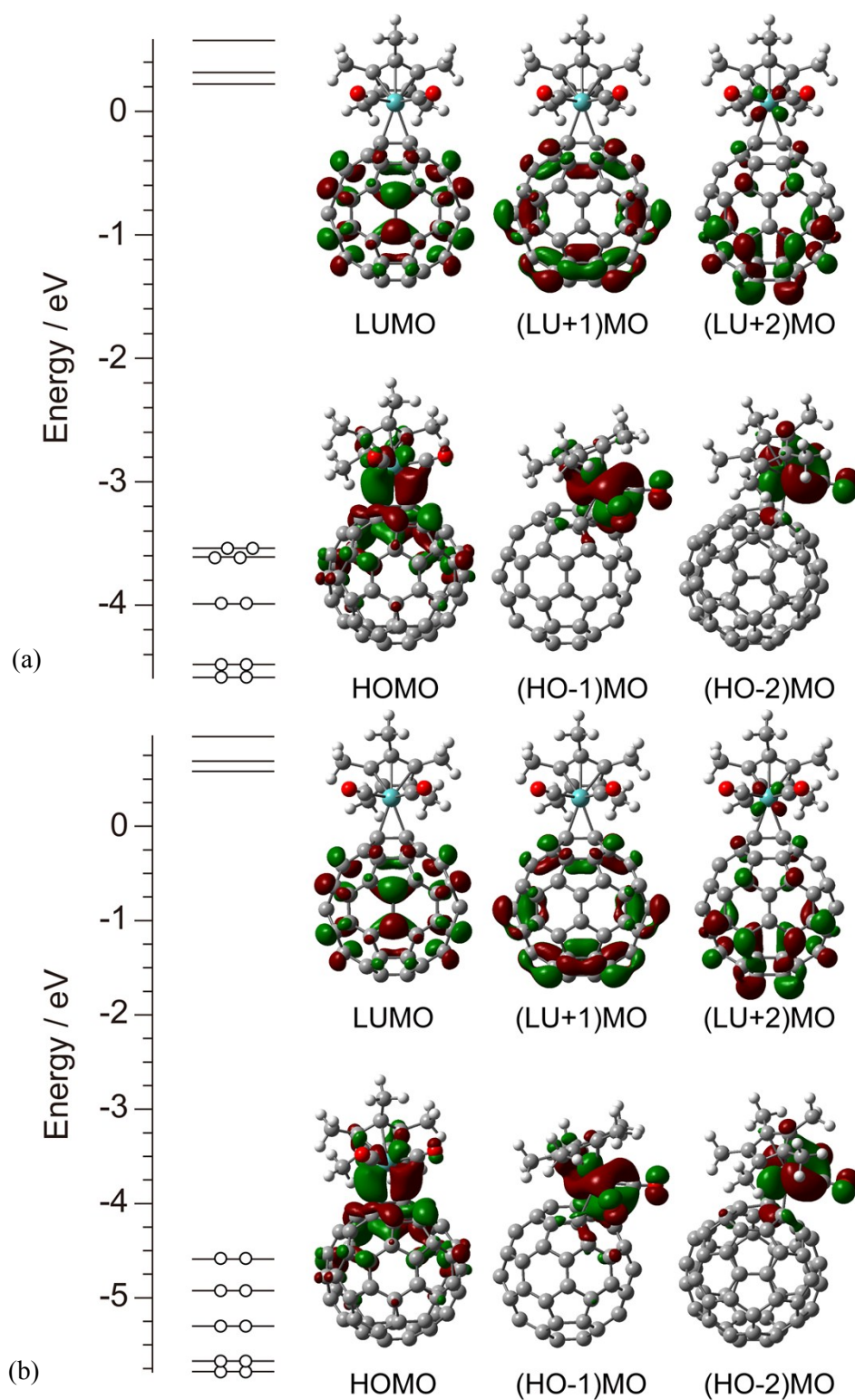


Fig. S10 Energy diagrams for the frontier Kohn-Sham orbitals of the 1A state in $\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ calculated at the (a) RCAM-B3LYP-D3 and (b) RM11/cc-pVTZ-PP/cc-pVDZ levels of theory.

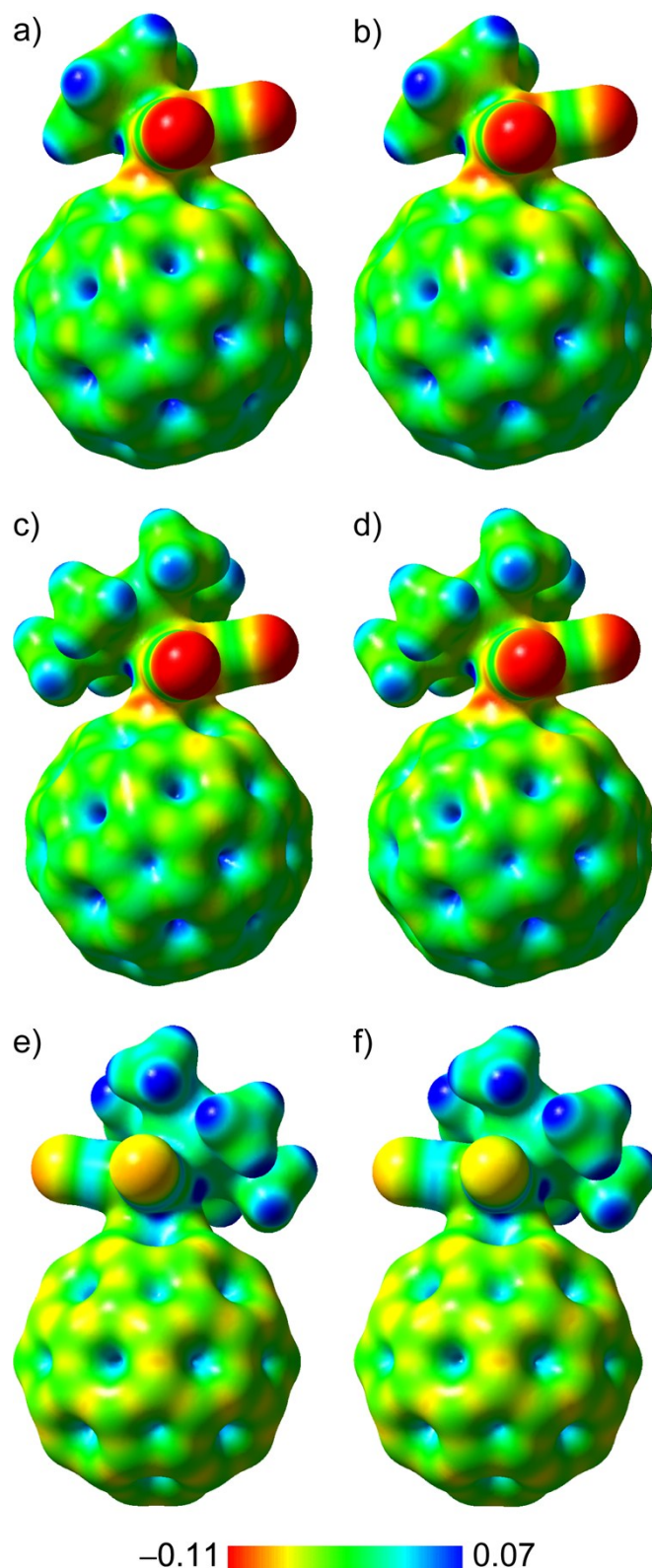


Fig. S11 Electrostatic potential maps on the 0.02 electron/au³ of electron density surface in {CpMo(CO)₂(η^2 -C₆₀)}⁻ for the ¹A states at the (a) RCAM-B3LYP-D3 and (b) RM11, and {Cp*Mo(CO)₂(η^2 -C₆₀)}⁻ for the ¹A states at the (c) RCAM-B3LYP-D3 and (d) RM11 and the ³A states at the (e) UCAM-B3LYP-D3 and (f) UM11/cc-pVTZ-PP/cc-pVDZ levels of theory.

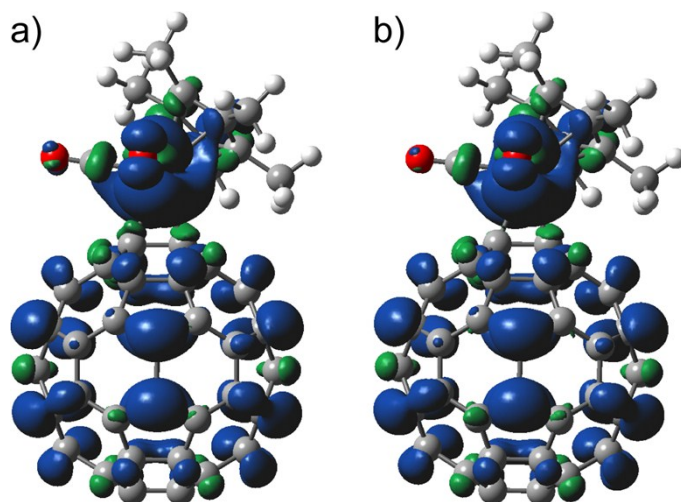


Fig. S12 The isosurface plots on spin density distribution of the 3A state in $\{\text{Cp}^*\text{Mo}(\text{CO})_2(\eta^2\text{-C}_{60})\}^-$ calculated at the (a) UCAM-B3LYP-D3 and (b) UM11/cc-pVTZ-PP/cc-pVDZ levels of theory. The isosurface value is 0.0009 electron/ au^3 . The isosurfaces in blue and green denote the positive and negative spin density, respectively.