

Description of the crystal structure

The complex $[\text{Ni}(\text{pydc})(\text{pydm})]\cdot\text{H}_2\text{O}$ crystallizes in the monoclinic system with the space group $\text{P}2_1/\text{c}$. Its crystal structure consists of complex molecules of $[\text{Ni}(\text{pydc})(\text{pydm})]$ and uncoordinated water molecules. The Ni(II) central atom is hexacoordinated by two carboxylato oxygen atoms [$\text{Ni1-O1} = 2.119(4)$ Å, $\text{Ni1-O3} = 2.151(4)$ Å] and one pyridine nitrogen atom [$\text{Ni1-N1} = 1.962(5)$ Å] from the tridentate dipicolinato ligand, and two hydroxyl oxygen atoms [$\text{Ni1-O5} = 2.110(4)$ Å, $\text{Ni1-O6} = 2.138(4)$ Å] and one pyridine nitrogen atom [$\text{Ni1-N2} = 1.992(5)$ Å] from the tridentate pyridine-2,6-dimethanol. The complex molecules and uncoordinated water molecules are linked to 2-D supramolecular frameworks through hydrogen bonds between hydroxyl oxygen atoms of the pyridine-2,6-dimethanol and uncoordinated carboxylato oxygen atoms of dipicolinato ligands [$\text{O5-H5O}\cdots\text{O4}$ with $\text{O5}\cdots\text{O4}$ distance of $2.622(6)$ Å], between hydroxyl oxygen atoms of pyridine-2,6-dimethanol and uncoordinated water molecules [$\text{O6-H6O}\cdots\text{O1W}$ with $\text{O6}\cdots\text{O1W}$ distance of $2.656(6)$ Å], between uncoordinated water molecules and carboxylato oxygen atoms of dipicolinato ligands [$\text{O1W-H1W}\cdots\text{O1}$, $\text{O1W-H2W}\cdots\text{O5}$ with $\text{O}\cdots\text{O}$ distances of $2.700(6)$ and $2.985(6)$ Å, respectively] and form double layers. In the crystal structure of **1** the $\pi\cdots\pi$ stacking interactions are observed between pyridine rings of pyridine-2,6-dimethanol with $c_{\text{g}}\cdots c_{\text{g}}$ distances of 3.58 and 3.40 Å and shift-distances of 1.77 and 1.23 Å (for detail see ref. [1]).

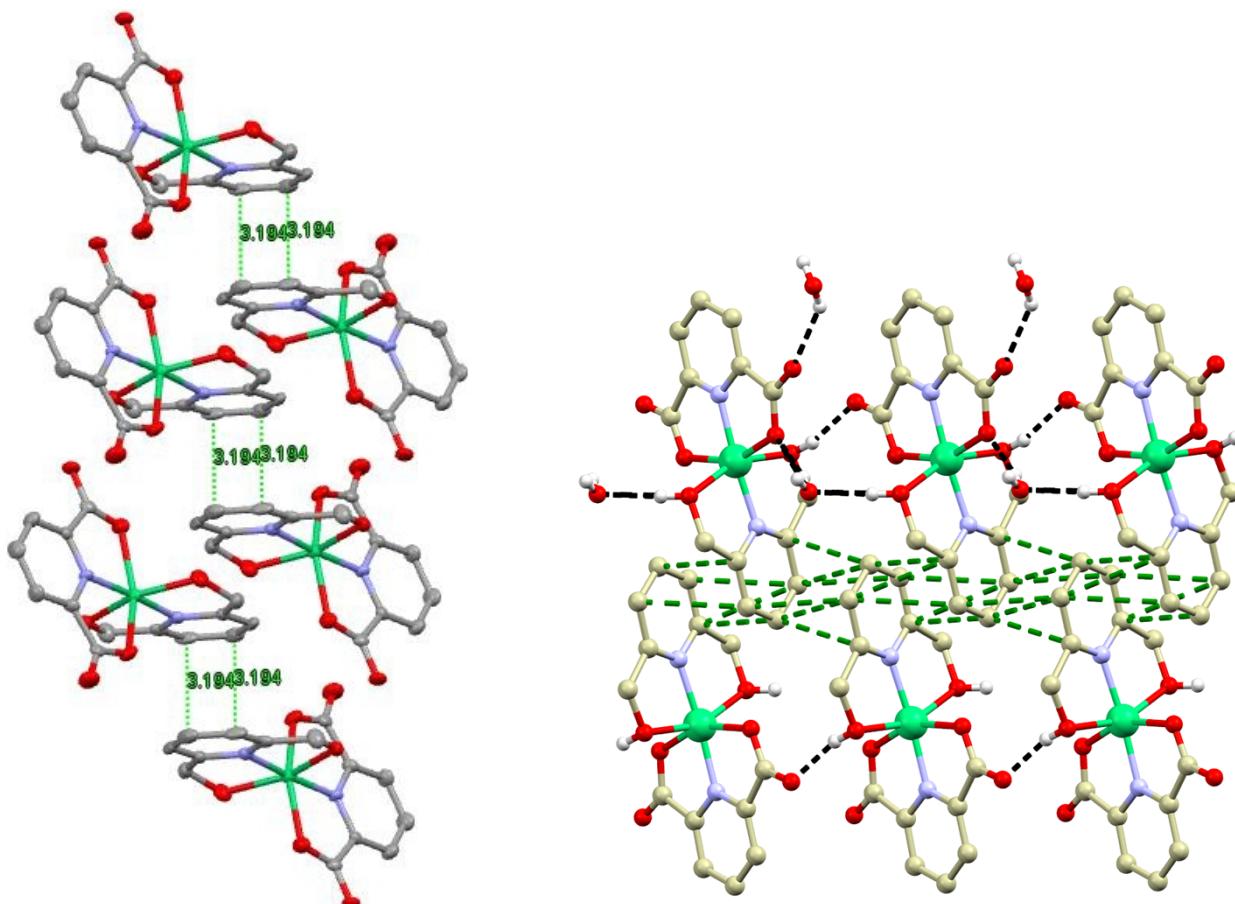


Fig. S1. Zoomed view to the crystal structure of **1**.

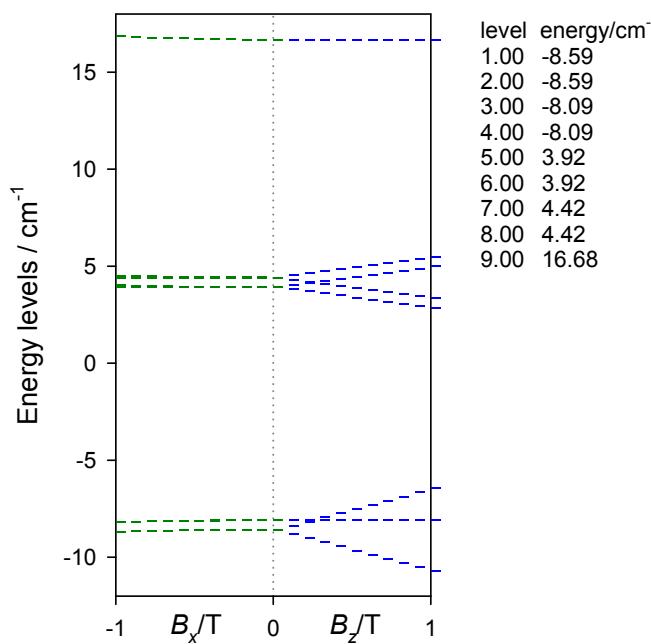


Fig. S2. Calculated energy levels for a dimeric **[1]₂** complex in the weak exchange limit using the spin Hamiltonian formalism. Magnetic parameters resulting from the magnetic data fitting: $J/hc = +0.25\text{ cm}^{-1}$, $g_z = 2.289$, $g_x = 2.277$, $D/hc = -12.5\text{ cm}^{-1}$, $\chi_{\text{TIM}} = 0.99 \times 10^{-9}\text{ m}^3\text{ mol}^{-1}$, $zj/hc = -0.039\text{ cm}^{-1}$; $R(\chi) = 0.010$, $R(M) = 0.030$.

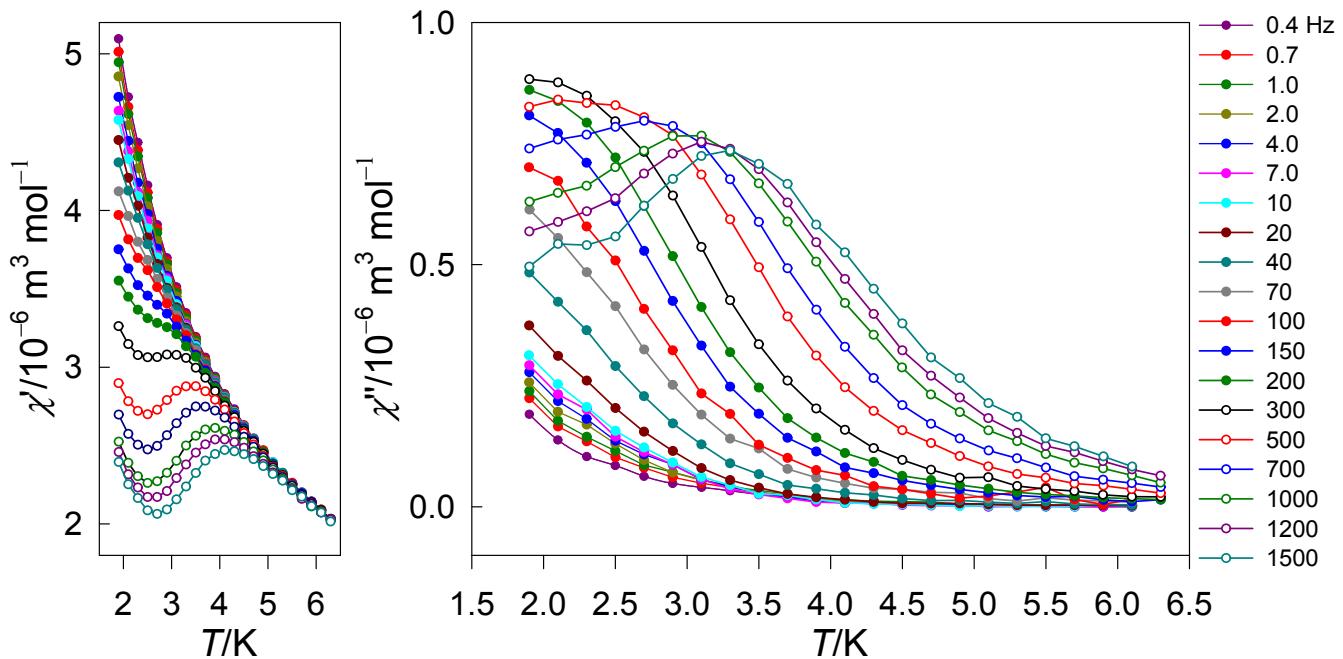
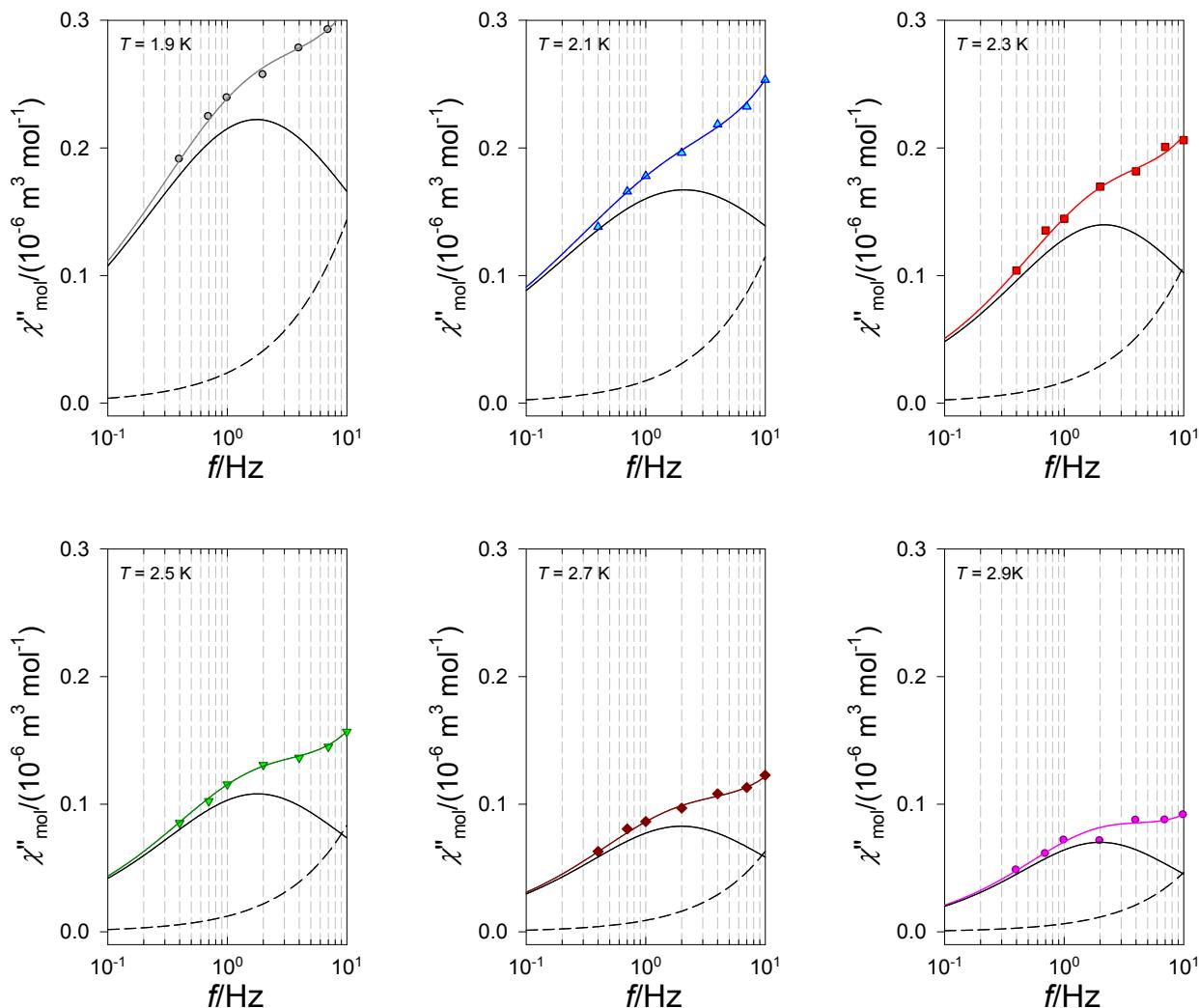


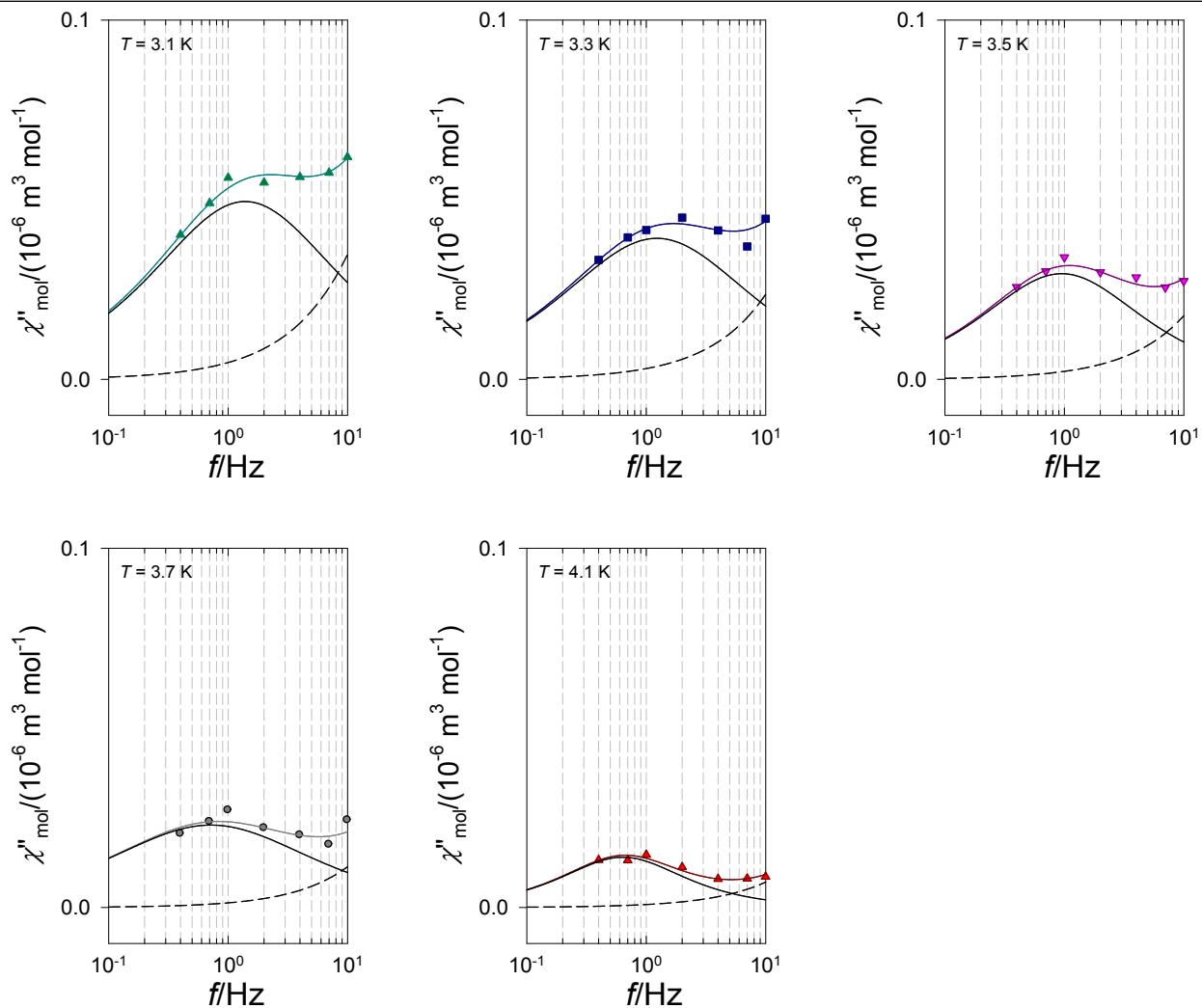
Figure S3. The in-phase χ' and out-of-phase χ'' molar susceptibility (SI units) for **1** at the applied external field $B_{\text{DC}} = 0.2\text{ T}$. Lines serve as a guide for eyes.

Table S1. Parameters of the extended Debye model for **1** (two relaxation processes) at $B_{DC} = 0.2$ T.

T/K	χ_s	χ_{T1}	α_1	$\tau_1 / 10^{-3} s$	χ_{T2}	α_2	$\tau_2 / 10^{-3} s$	$R(\chi')/\%$	$R(\chi'')/\%$
1.9	2.08(1)	4.42(3)	0.20(1)	0.542(5)	5.33(3)	0.42(3)	91(6)	0.18	0.82
2.1	1.93(1)	4.16(5)	0.17(1)	0.480(6)	4.94(5)	0.48(5)	77(11)	0.29	1.21
2.3	1.80(1)	4.04(3)	0.17(1)	0.432(6)	4.53(3)	0.34(6)	74(9)	0.40	0.93
2.5	1.73(1)	3.85(1)	0.16(1)	0.363(3)	4.23(2)	0.34(5)	90(19)	0.26	0.95
2.7	1.64(1)	3.68(1)	0.15(1)	0.286(2)	3.96(2)	0.34(6)	81(10)	0.29	0.50
2.9	1.58(1)	3.53(1)	0.13(1)	0.225(2)	3.74(1)	0.27(6)	80(10)	0.30	0.74
3.1	1.51(2)	3.39(1)	0.13(1)	0.171(2)	3.54(2)	0.28(8)	116(22)	0.26	0.91
3.3	1.53(2)	3.26(1)	0.10(1)	0.136(2)	3.38(2)	0.29(11)	131(40)	0.30	1.00
3.5	1.48(3)	3.13(1)	0.10(1)	0.106(3)	3.22(2)	0.20(16)	169(66)	0.29	1.49
3.7	1.56(3)	3.01(1)	0.06(1)	0.093(3)	3.09(3)	0.33(20)	224(180)	0.31	0.84
4.1	1.39(6)	2.80(1)	0.07(1)	0.053(3)	2.83(1)	0.05(30)	252(162)	0.27	1.40
5.1	1.82(9)	2.39(1)	0.01(4)	0.050(10)	-	-	-	0.47	3.7
6.1	1.85(5)	2.08(1)	0	0.049(11)	-	-	-	0.71	8.0

^a SI unit for the molar magnetic susceptibility [$10^{-6} \text{ m}^3 \text{ mol}^{-1}$]. Standard deviations in parentheses (last digit). R – discrepancy factor of the fit for dispersion χ' and absorption χ'' , respectively.



Fig. S4. Deconvolution lines of the fitted χ'' vs f curves in the low-frequency window.

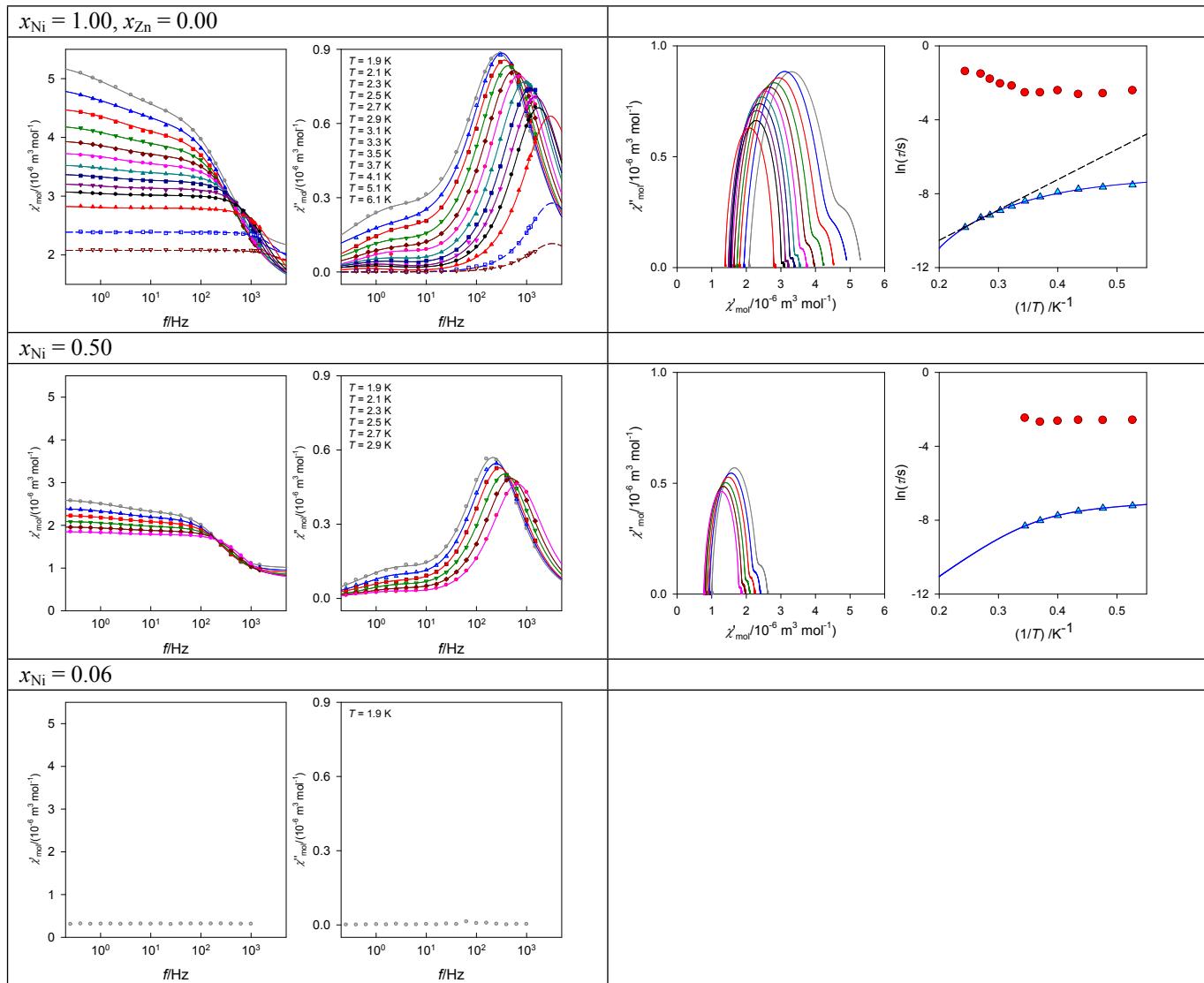
Doping experiments

Fig. S5. Comparison of the AC susceptibility measurements for various degrees of Ni:Zn doping.

Table S2. Parameters of the extended Debye model for **2** (two relaxation processes) at $B_{DC} = 0.2$ T.

T/K	χ_s	χ_{T1}	α_1	$\tau_1 / 10^{-3}$ s	χ_{T2}	α_2	$\tau_2 / 10^{-3}$ s	$R(\chi')/\%$	$R(\chi'')/\%$
1.9	1.00(1)	2.29(1)	0.09(1)	0.73(1)	2.61(1)	0.27(2)	76(4)	0.26	1.8
2.1	0.93(1)	2.17(1)	0.09(1)	0.64(1)	2.40(1)	0.24(3)	75(5)	0.28	1.7
2.3	0.89(1)	2.06(1)	0.07(1)	0.55(1)	2.24(1)	0.27(3)	75(6)	0.25	1.5
2.5	0.84(1)	1.96(1)	0.07(1)	0.43(1)	2.10(1)	0.23(4)	72(6)	0.31	1.2
2.7	0.80(1)	1.86(1)	0.05(1)	0.33(1)	1.97(1)	0.28(4)	68(7)	0.24	1.3
2.9	0.78(1)	1.78(1)	0.05(1)	0.24(1)	1.86(1)	0.24(5)	85(9)	0.25	0.73

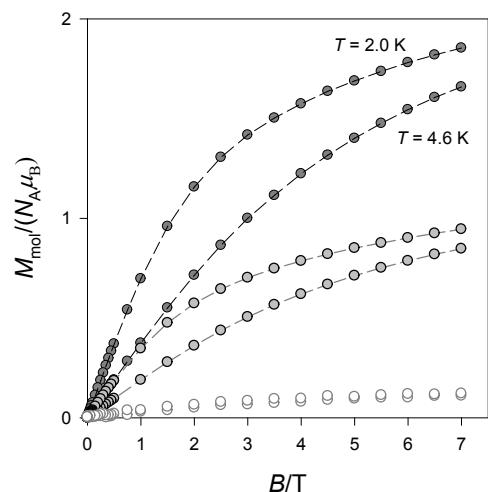


Fig. S6. DC magnetization per formula unit for Ni:Zn doped materials with the mole fraction $x_{\text{Ni}} = 1.00$ (dark gray), 0.50 (gray), and 0.06 (white).

***Ab Initio* Calculations**

Quantum-chemical calculations of the ZFS based on X-ray diffraction-determined structure were performed with the ORCA program [2]. For this purpose, an *ab initio* wave function based method has been chosen, such as complete active space self-consistent field (CASSCF) improved with the second-order *N*-electron valence perturbation theory (NEVPT2). An active space in which eight electrons are distributed into the five nickel d-orbitals (CAS(8,5)) was employed along with the TZVP basis set for all elements. In CASSCF procedure, the orbitals were optimized for the average of 10 triplet (3F and 3P terms of the free Ni(II)) and 15 singlet (1G , 1D and 1S terms) roots. The state-averaged (SA) CASSCF calculations were performed on top of quasi-restricted DFT/BP86/TZVP orbitals. The *D*-values were calculated through quasi-degenerate perturbation theory (QDPT) in which the spin-orbit coupling (SOC) operator (in SOMF approximation) is diagonalized in the basis of the non-relativistic SA-CASSCF/NEVPT2 eigenfunctions [3]. Diagonalization of the resulting matrix yields the energy levels and eigenvectors of the coupled states that were used for constructing the *D*-tensor through the formalism based on effective Hamiltonian approach.

Table S3. Calculated CAS(8,5)/NEVPT2 transition energies and SOC splitting of the ground state for the complex **1**.

Root	Multiplicity	$\Delta E/\text{cm}^{-1}$	$\Delta E^{\text{SOC}}/\text{cm}^{-1}$
0	3	0	0, 9.0, 19.9
1	3	6956	
2	3	9306	
3	3	11029	
0	1	13596	
4	3	15006	
1	1	16483	
5	3	17026	
6	3	17725	
2	1	22969	
3	1	24741	
7	3	25315	
4	1	26082	
8	3	28218	
5	1	29212	
6	1	29792	
9	3	30228	
7	1	33306	
8	1	35426	
9	1	37534	
10	1	37761	
11	1	38760	
12	1	39909	
13	1	40462	
14	1	69013	

Table S4. Individual contributions to D and E of the excited triplet and singlet states computed at the CAS(8,5)/NEVPT2 level for the complex **1**.

Root	Multiplicity	D/cm^{-1}	E/cm^{-1}
1	3	-60.33	-0.02
2	3	21.42	-16.81
3	3	16.50	11.46
4	3	1.22	-0.34
5	3	0.79	-0.03
6	3	0	0.13
7	3	0.08	0.07
8	3	0.05	-0.04
9	3	-0.02	0
0	1	0.30	0
1	1	0.03	0
2	1	16.29	-0.01
3	1	-6.63	5.31
4	1	-5.44	-4.42
5	1	-1.12	-0.60
6	1	0.10	0
7	1	-0.28	0.28
8	1	-0.02	-0.02
9	1	0.80	0
10	1	0.02	0.01
11	1	-1.06	-0.44
12	1	0.40	0.01
13	1	-0.79	0.31
14	1	0.01	0

References

- [1] J. Miklovič, A. Packová, P. Segľa, J. Titiš, M. Koman, J. Moncol', R. Boča, V. Jorík, H. Krekuska, D. Valigura, Inorg. Chim. Acta, 2015, 429, 73.
- [2] (a) Neese, F. ORCA – An Ab Initio, Density Functional and Semi-empirical Program Package, Version 3.0.3. (b) Atanasov, M.; Ganyushin, D.; Pantazis, D. A.; Sivalingam, K.; Neese, F. Inorg. Chem. 2011, 50, 7460. (c) Angeli, C.; Borini, S.; Cestari, M.; Cimiraglia, R. J. Chem. Phys. 2004, 121, 4043. (d) Angeli, C.; Cimiraglia, R.; Evangelisti, S.; Leininger, T.; Malrieu, J.-P. J. Chem. Phys. 2001, 114, 10252. (e) Angeli, C.; Cimiraglia, R.; Malrieu, J.-P. J. Chem. Phys. 2002, 117, 9138.
- [3] (a) Neese, F. J. Chem. Phys. 2005, 122, 34107. (b) Ganyushin, D.; Neese, F. J. Chem. Phys. 2006, 125, 24103. (c) Neese, F. J. Chem. Phys. 2007, 127, 164112.