

## Electronic Supplementary Information

### CeO<sub>2</sub> nanorods/g-C<sub>3</sub>N<sub>4</sub>/N-rGO composite: Enhanced visible-light-driven photocatalytic performance and the role of N-rGO as electronic transfer media

Li Wang<sup>a</sup>, Jing Ding<sup>a</sup>, Yuanyuan Chai<sup>a</sup>, Qianqian Liu<sup>a</sup>, Jia Ren<sup>a</sup>, Xin Liu, Wei-Lin

Dai<sup>a\*</sup>

<sup>a</sup>Department of Chemistry & Shanghai Key Laboratory of Molecular Catalysis and Innovative

Materials, Fudan University, Shanghai 200433, P. R. China;

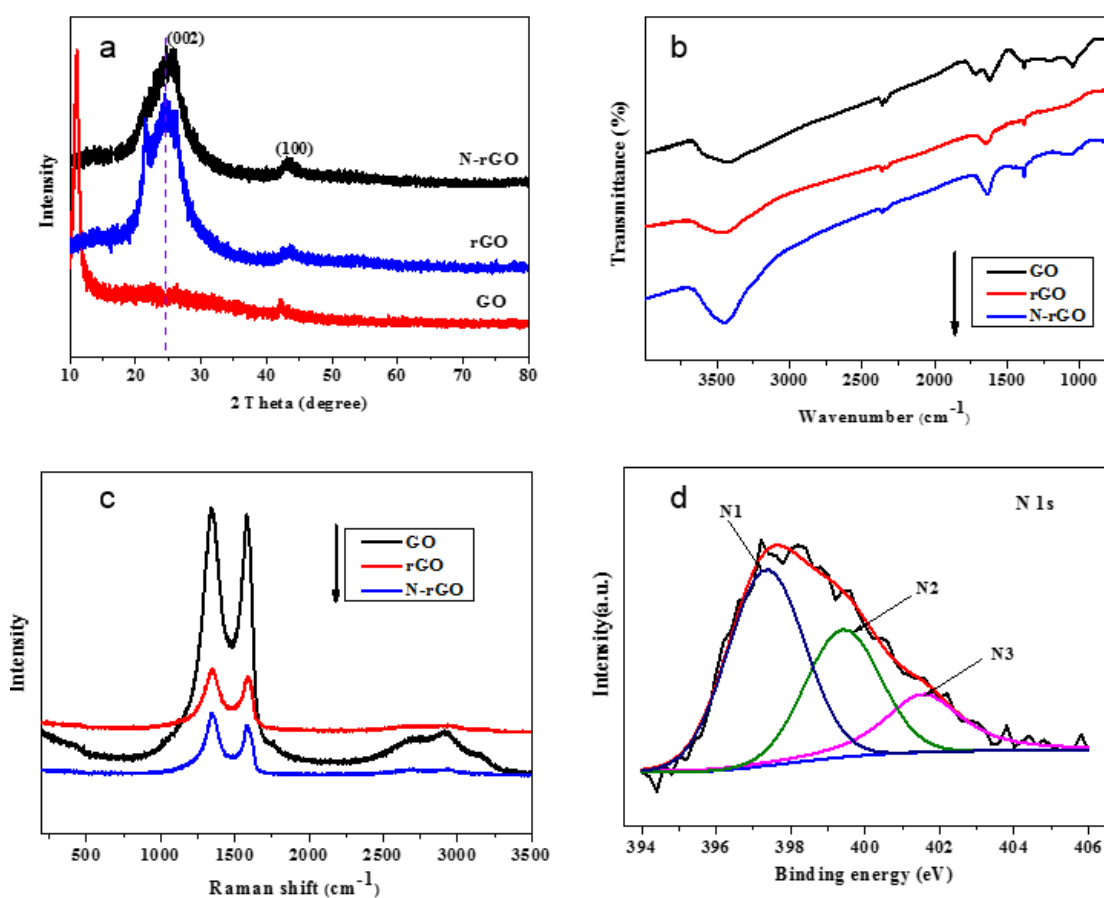
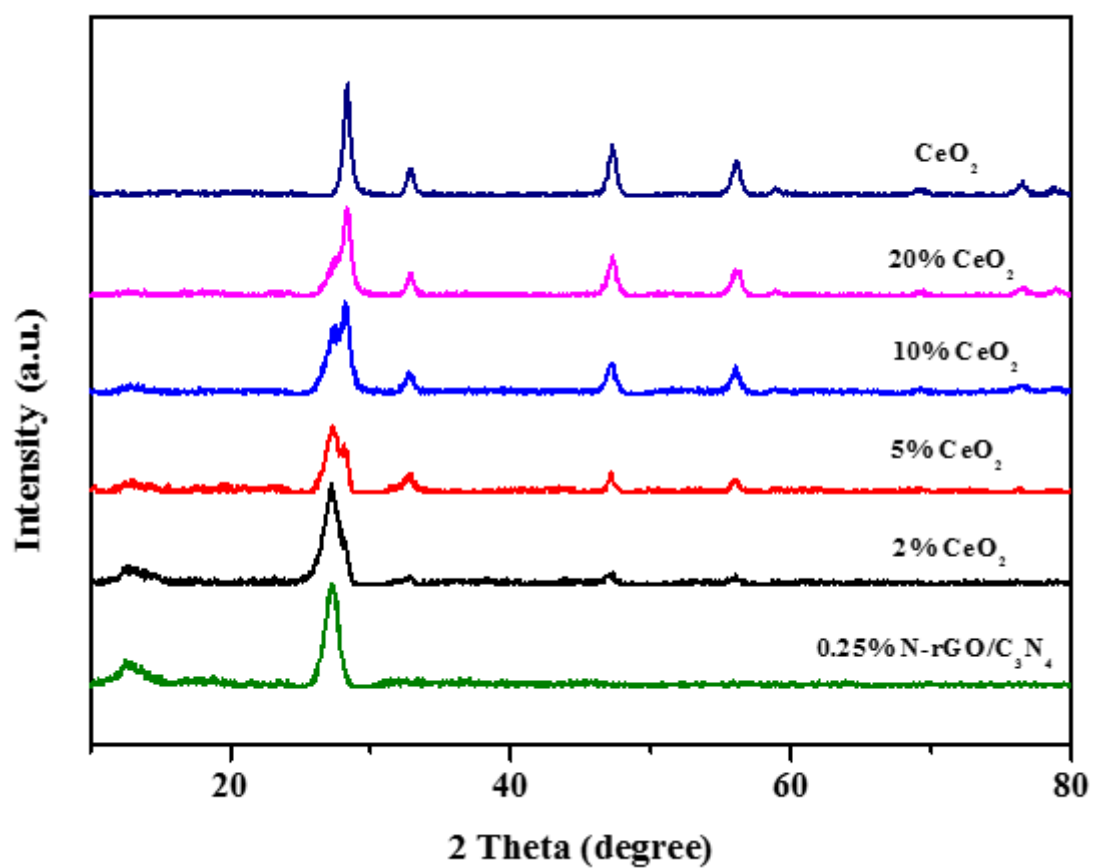
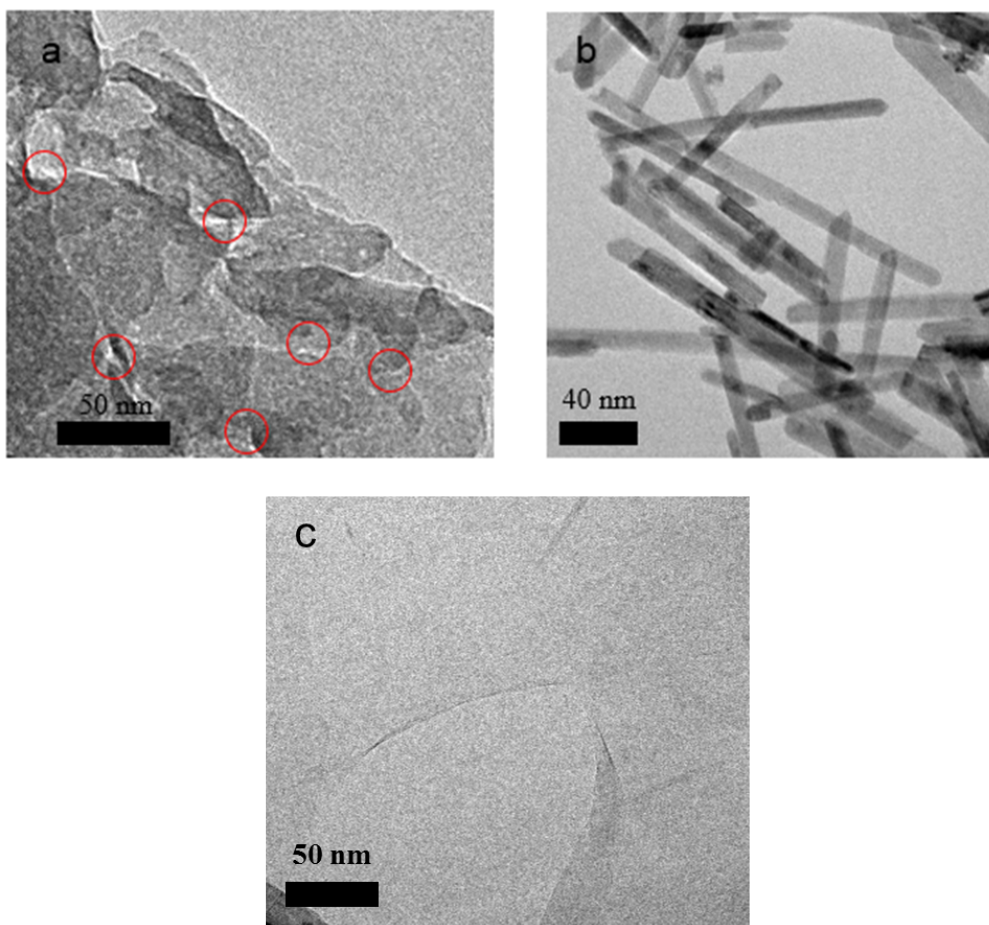


Fig. S1 XRD (a), FT-IR (b), Raman spectra (c) and XPS spectra (d) of N-rGO sample.



**Fig. S2** XRD patterns of the ternary  $\text{CeO}_2$  nanorods/(g- $\text{C}_3\text{N}_4$ /0.25%N-rGO) composites with different content of  $\text{CeO}_2$  nanorods.



**Fig. S3** TEM images of pure g-C<sub>3</sub>N<sub>4</sub> (a), pure CeO<sub>2</sub> nanorods (b) and **grapheme (c)**

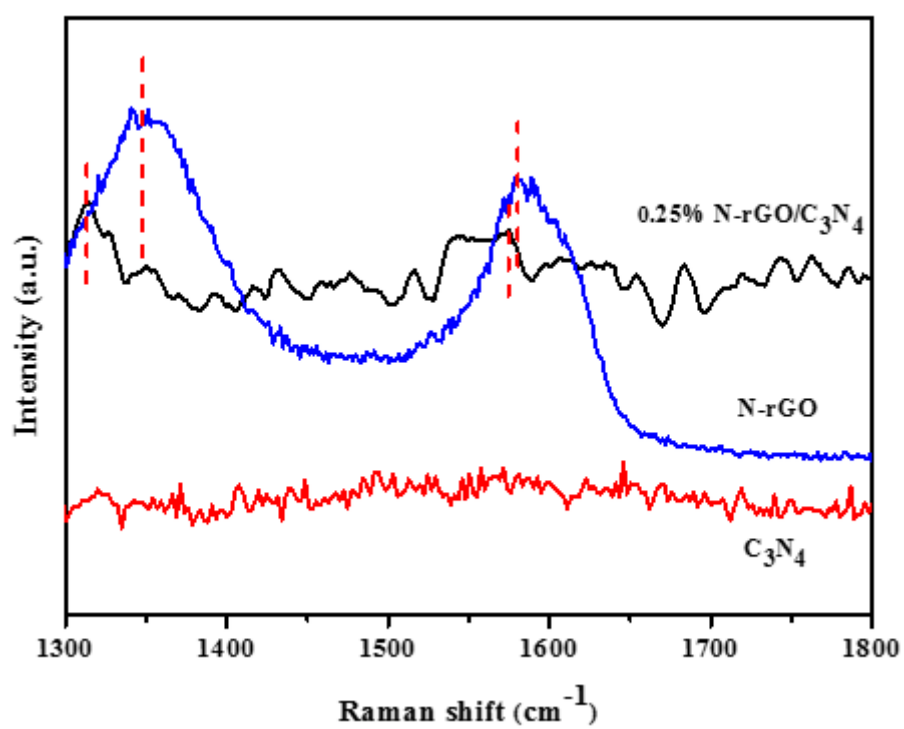


Fig. S4 Raman spectra of N-rGO, g-C<sub>3</sub>N<sub>4</sub> and g-C<sub>3</sub>N<sub>4</sub>/0.25%N-rGO.

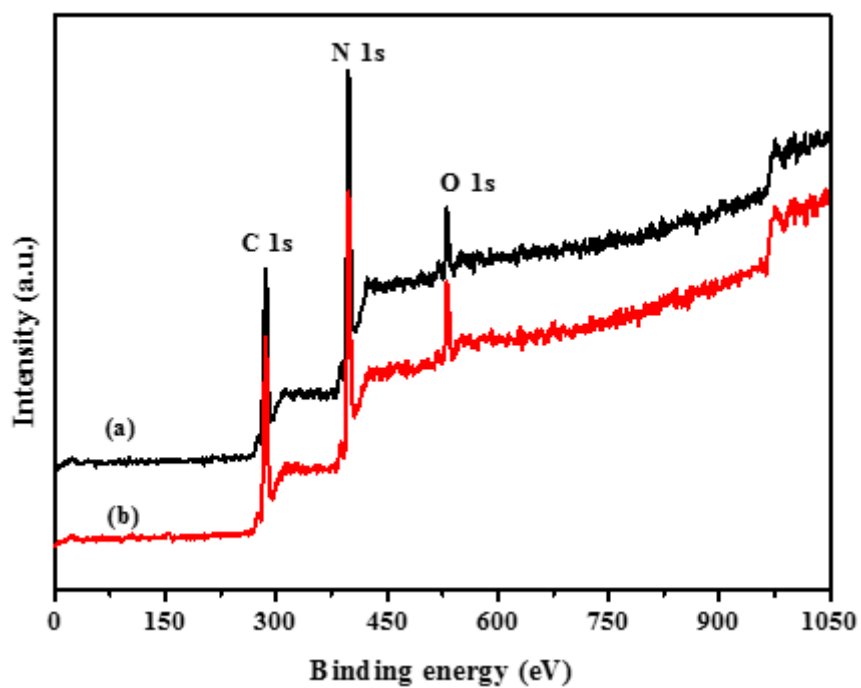
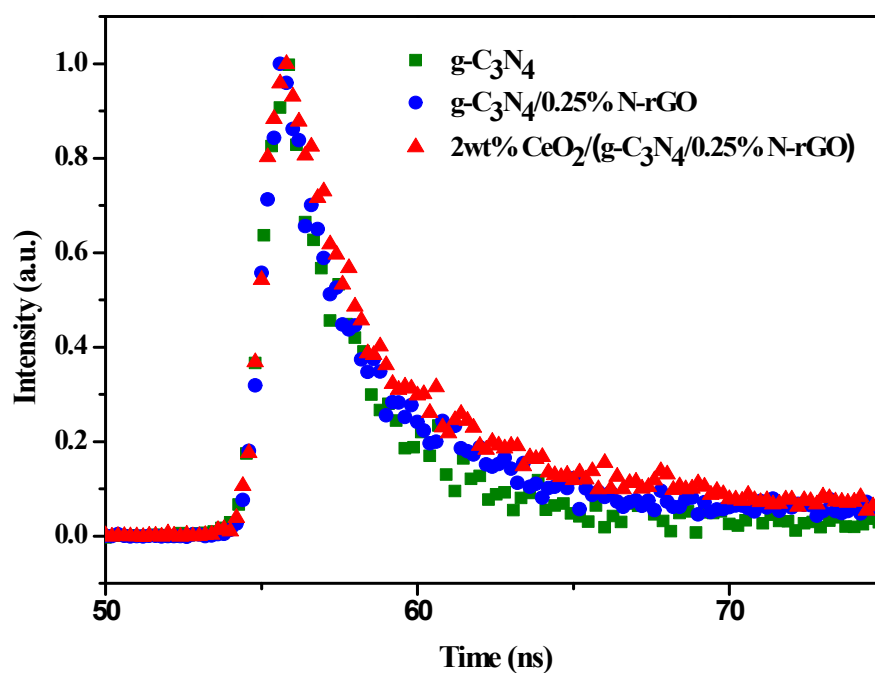


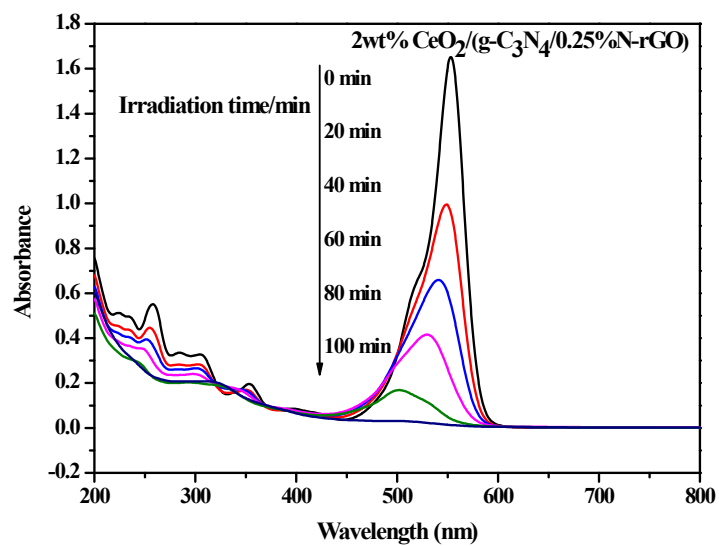
Fig. S5 XPS survey spectra of g-C<sub>3</sub>N<sub>4</sub> (a) and g-C<sub>3</sub>N<sub>4</sub>/0.25%N-rGO (b).

**Table S1** BET surface areas of different samples

Sample	$S_{\text{BET}}$ ( $\text{m}^2/\text{g}$ )
$\text{C}_3\text{N}_4$	16.2
$\text{g-C}_3\text{N}_4/0.25\%\text{rGO}$	23.9
$\text{g-C}_3\text{N}_4/0.25\%\text{N-rGO}$	25.9
2wt% $\text{CeO}_2/(\text{g-C}_3\text{N}_4/0.25\%\text{N-rGO})$	27.0



**Fig. S6** Time-resolved fluorescence decay spectra of the  $\text{g-C}_3\text{N}_4$ ,  $\text{g-C}_3\text{N}_4/0.25\%\text{N-rGO}$  and 2wt%  $\text{CeO}_2$  nanorods/ $(\text{g-C}_3\text{N}_4/0.25\%\text{N-rGO})$  samples



**Fig. S7** The temporal absorption spectrum changes of RhB aqueous solution in the presence of 2wt% CeO<sub>2</sub>/(g-C<sub>3</sub>N<sub>4</sub>/0.25%N-rGO) under visible light irradiation.