Supporting information

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for

AIE-Active Organoboron Complexes with Highly Efficient

Solid-State Luminescence and Their Application as Gas

Sensitive Materials

Shuwen Gong,^{†,a} Qingsong Liu,^{†,a} Xiaoqing Wang,^{*,a,b} Bo Xia,^a Zhipeng Liu,^{*,a} and Weijiang He^{*,c}

^{*a*} Institute of Functional Organic Molecules and Materials, School of Chemistry and Chemical Engineering, Liaocheng University, No.1 Hunan Road, Liaocheng, 252000, People's Republic of China; E-mail: <u>chliuzp@163.com</u>; wangxiaoqing@lcu.edu.cn.

^b School of Material Science and Engineering, Liaocheng University, No.1 Hunan Road, Liaocheng, 252000, People's Republic of China;

^c State Key Laboratory of Coordination Chemistry, Coordination Chemistry Institute, School of Chemistry and Chemical Engineering, Nanjing University, Nanjing, P.R. China, 210093. E-mail: heweij69@nju.edu.cn.

⁺Both authors contribute equally to this work.



Figure S1. Absorption spectra of compound 3 (a), compound 4 (b), compound 5 (c) and compound 6 (d) in various solvents.



Figure S2. Absorption and emission of compound 3 (a), compound 4 (b), compound 5 (c) and compound 6 (d) in the solid state.



Figure S3. Changes in the quantum yields of compound **3** (a), compound **4** (b), compound **5** (c) and compound **6** (d) in THF/water mixtures (10 μ M) with varied volumetric fractions of water (f_w).



Figure S4. Normalized emission spectra of compounds 3 (a), 4 (b), 5 (c) and 6 (d) in THF/water

mixtures (10 μ M) with f_w of 90%.



Figure S5. Crystal packing structures of compound **3**. The red dotted lines show intermolecular $C-H\cdots F$ and $C-H\cdots \pi$ interactions within **3**.



Figure S6. Crystal packing structures of compound **4**. The red dotted lines show intermolecular $C-H\cdots F$ and $C-H\cdots \pi$ interactions within **4**.



Figure S7. Crystal packing structures of compound **5**. The red dotted lines show intermolecular $C-H\cdots S$ and $C-H\cdots \pi$ interactions within **5**.



Figure S8. Crystal packing structures of compound **6**. The red dotted lines show intermolecular $C-H\cdots\pi$ interactions within **6**.



Figure S9. Calculated absorption spectra of compound 3 (a), compound 4 (b), compound 5 (c) and compound 6 (d)



Figure S10. Moldecular orbital amplitude plots and energy levels of HOMOs and LUMOs of compound **6+H**⁺ calculated by using B3LYP/6-31G(d, p) basis set with G03 Program.







Figure S13. ¹H NMR spectrum of compound 5 in CDCl₃.







Figure S17. ¹³C NMR spectrum of compound 5 in CDCl₃.

