Electronic Supplementary Information

Nano-sized Mn oxide on halloysite or high surface area montmorillonite as an efficient catalyst for water oxidation with cerium(IV) ammonium nitrate: Supports from natural sources

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Scheme S1 Set up for water-oxidation experiments.



Fig. S1 FTIR spectrum of halloysite



Fig. S2 FTIR spectrum of Mn oxide on halloysite (calcined temperature: 200 °C).



Fig. S3 FTIR spectrum of Mn oxide on halloysite (calcined temperature: 400 °C).



Fig. S4 FTIR spectrum of montmorillonite.



Fig. S5 FTIR spectrum of Mn oxide on montmorillonite (calcined temperature: 100 °C).



Fig. S6 FTIR spectrum of Mn oxide on montmorillonite (calcined temperature: 400 °C).



Fig. S7 XRD pattern of halloysite.



Fig. S8 XRD pattern of Mn oxide on halloysite (calcined temperature: 100 °C).



Fig. S9 XRD pattern of Mn oxide on halloysite (calcined temperature: 400 °C).



Fig. S10 XRD pattern of Mn oxide on halloysite (calcined temperature: 500 °C).



Fig. S11 XRD pattern of Mn oxide on montmorillonite.



Fig. S12 XRD pattern of Mn oxide on montmorillonite (calcined temperature: 100 °C).



Fig. S13 XRD pattern of Mn oxide on montmorillonite (calcined temperature: 500 °C).









Fig. S14 SEM images from halloysite (a-d).



a







d Fig. S15 SEM images from Mn oxide on halloysite (calcined temperature: 60 °C) (a-d).







 $f_{\text{figure}} = 15.00 \text{ kV WD} = 6 \text{ mm}}$

d

Fig. S16 SEM images from montmorillonite (a-d).







Fig. S17 SEM images of Mn oxide on montmorillonite (calcined temperature: 60 °C) (ad).







c



d

Fig. S18 TEM images of Mn oxide on halloysite (calcined temperature: 60 °C) (a-d).





b



c



d

Fig. S19 TEM images of Mn oxide on montmorillonite (calcined temperature: $60 \,^{\circ}$ C) (a-d).



Fig. S20 The rate of oxygen evolution in the presence of 50 mg halloysite (a) and montmorillonite (b) (50 mg) ([Ce(IV)]: 0.11 M).