

Electron Deficient Nonplanar β -Octachlorovanadylporphyrin as Highly Efficient and Selective Epoxidation Catalyst for Olefins

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Table of Contents

	Page No
Figure S1. FT-IR spectra of (a) $H_2TPP\text{Cl}_8$ and (b) $VOTPP\text{Cl}_8$ using KBr pellets.	2
Figure S2. MALDI-TOF mass spectra of $VOTPP\text{Cl}_8$ using HABA matrix. Insets show the molecular ion peaks obtained from experiment and simulation.	3
Table S1. Selected bond lengths (Å) and bond angles ($^\circ$) for the B3LYP/6-311G(d,p) optimised geometry of $VOTPP\text{Cl}_8$ in CH_3CN . The values if the given figure represent the deviation of atoms from the porphyrin mean plane formed by 24 atoms core.	4
Figure S3. B3LYP/LANLDDZ optimized geometry showing (a) top and (b) side view of $H_2TPP\text{Cl}_8$ in CH_3CN . In side view, β -chloro and meso-phenyl groups are not shown for clarity.	5
Table S2. Selected bond lengths (Å) and bond angles ($^\circ$) for the B3LYP/6-311G(d,p) optimised geometry of $H_2TPP\text{Cl}_8$ in CH_3CN . The values if the given figure represent the deviation of atoms from the porphyrin mean plane formed by 24 atoms core.	5
Figure S4. DPV of $VOTPP\text{Cl}_8$ in CH_2Cl_2 indicating one electron redox processes.	6
Figure S5. CV and DPV of $VOTPP\text{Cl}_8$ in CH_2Cl_2 containing 0.1 M TBAPF ₆ at 298 K.	7
Figure S6. Thermogram (TG), Differential thermal analysis (DTA) and Differential thermogram (DTG) of $VOTPP\text{Cl}_8$ at the heating rate of 10 °C /minute scanned from 25 °C to 1000 °C.	8
Figure S7. MALDI-TOF mass spectra of $VOTPP\text{Cl}_8$ in presence of hydrogen peroxide and sodium bicarbonate using HABA matrix.	9
Figure S8. MALDI-TOF mass spectra of the reaction mixture after addition of the cyclohexene to oxido- peroxidovanadium(V) intermediate using HABA matrix.	9
Figure S9. UV-Visible spectra of $VOTPP\text{Cl}_8$ in CH_2Cl_2 (before catalysis) and in CH_3CN (after catalytic reaction) at 298 K.	10

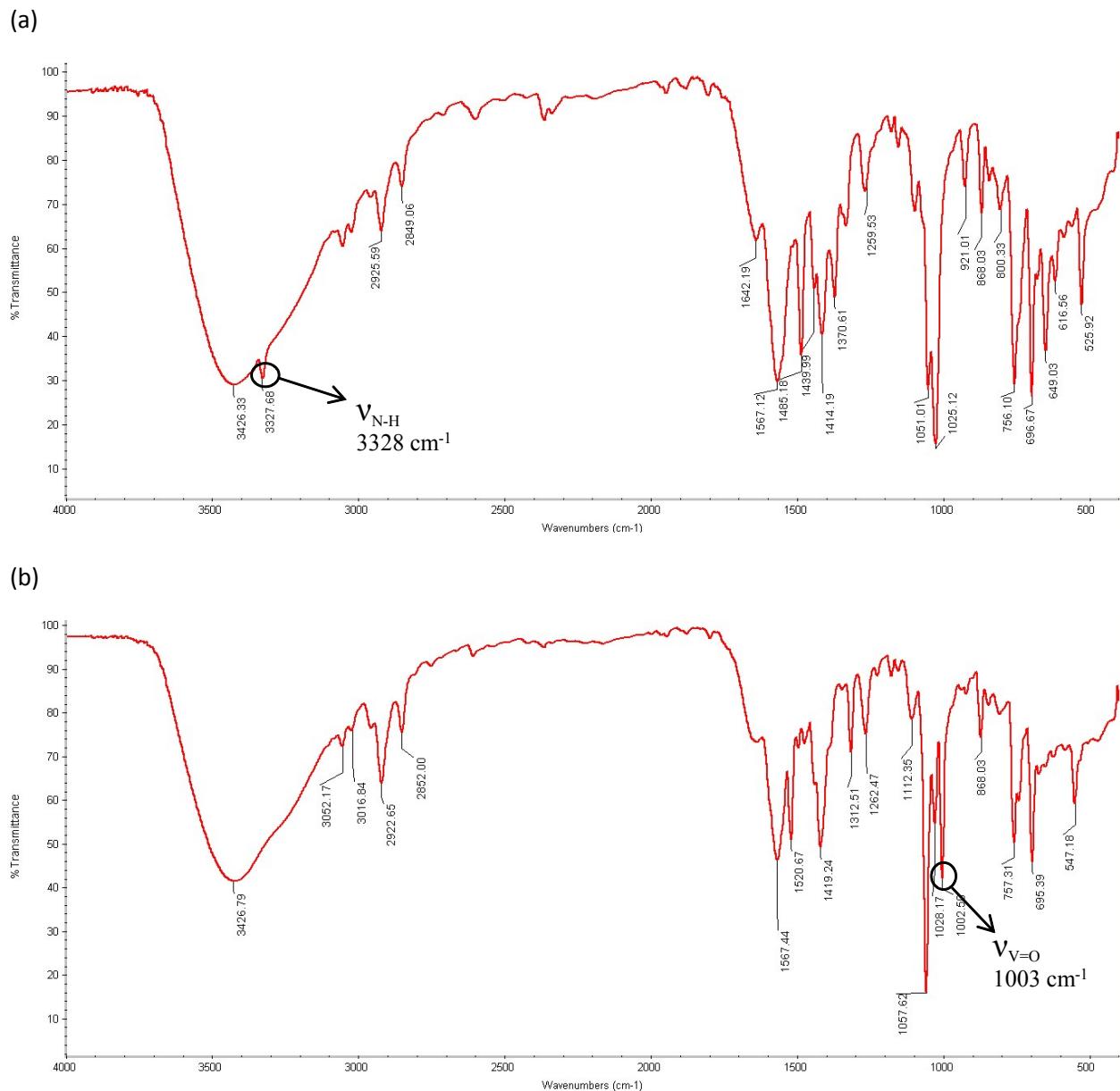
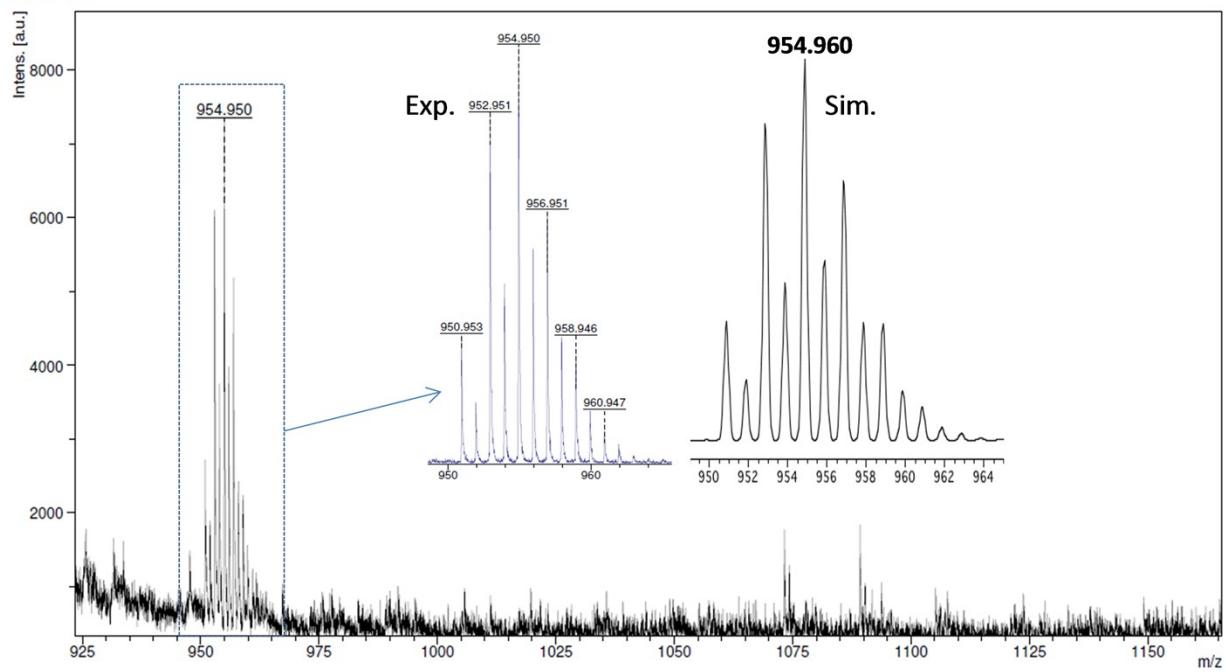


Figure S1. FT-IR spectra of (a) H_2TPPCL_8 and (b) VOTPPCL_8 using KBr pellets.

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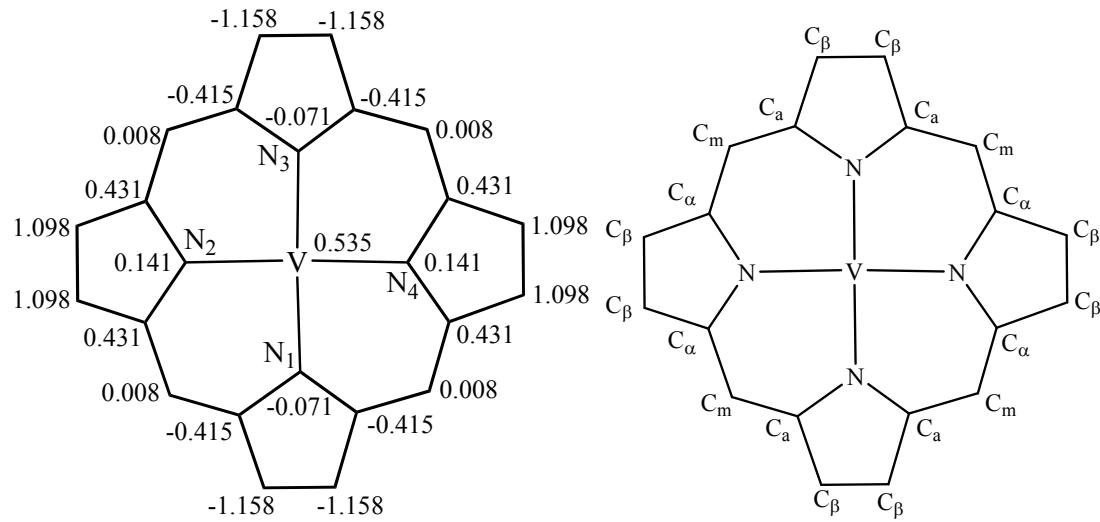


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Figure S2. MALDI-TOF mass spectra of VOTPPCl₈ using HABA matrix. Insets show the molecular ion peaks obtained from experiment and simulation.

Table S1. Selected bond lengths (\AA) and bond angles ($^\circ$) for the B3LYP/6-311G(d,p) optimised geometry of VOTPPCl₈ in CH₃CN. The values if the given figure represent the deviation of atoms from the porphyrin mean plane formed by 24 atoms core.



Bond Length (\AA)		Bond Angle ($^\circ$)	
V-N	2.065	M-N-C_a	123.00
V-O	1.602	N-M-N	153.06
N-C _a	1.402	N-C_a-C_m	123.80
C _a -C _b	1.452	N-C _a -C _b	107.82
C_b-C_b	1.375	C_b-C_a-C_m	128.16
C _a -C _m	1.413	C _a -C _b -C _b	107.89
Delta C_b (Angstrom)^a	1.128	C _a -N-C _a	107.90
Delta 24 (Angstrom)^b	0.535		
Delta Metal (Angstrom)^c	0.515		

^aDeviation of β -pyrrole carbons from porphyrin mean plane formed by 24-atom core

^bDeviation of 24 atoms from the porphyrin mean plane

^cdeviation of metal atom (V^{IV}) from the porphyrin mean plane

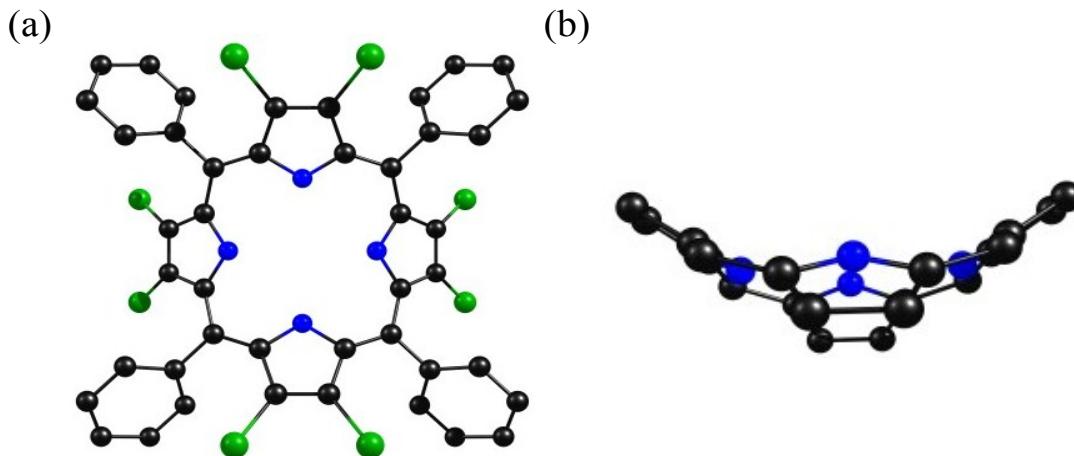
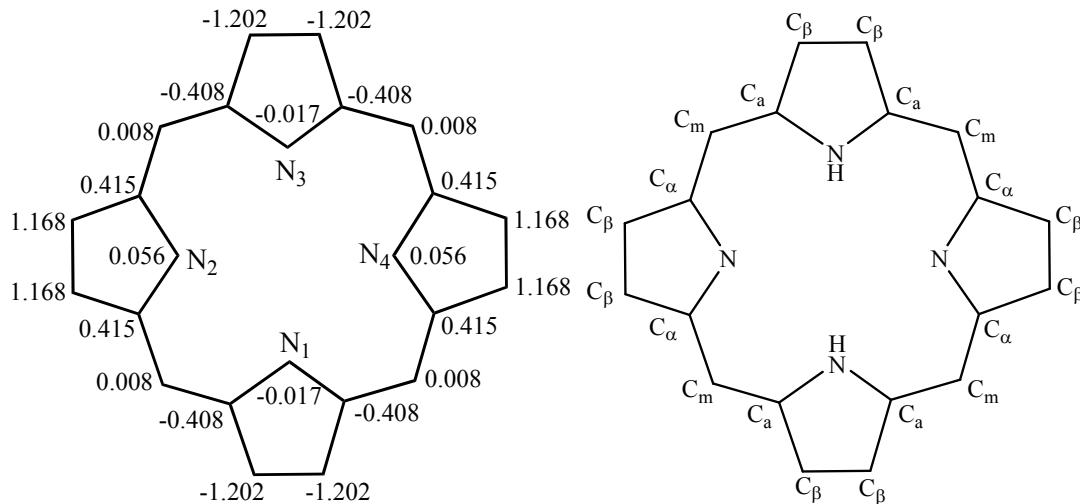


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^aDeviation of β -pyrrole carbons from porphyrin mean plane formed by 24-atom core.

^bDeviation of 24 atoms from the porphyrin mean plane.

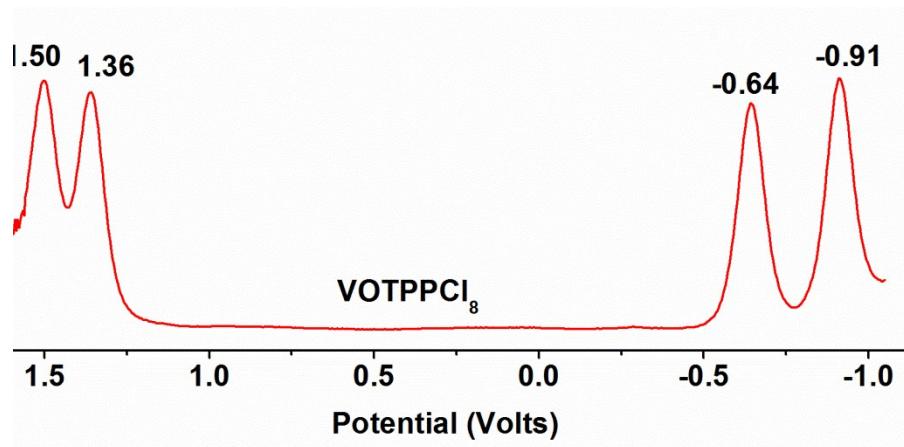


Figure S4. DPV of VOTPPCl₈ in CH₂Cl₂ indicating one electron redox processes.

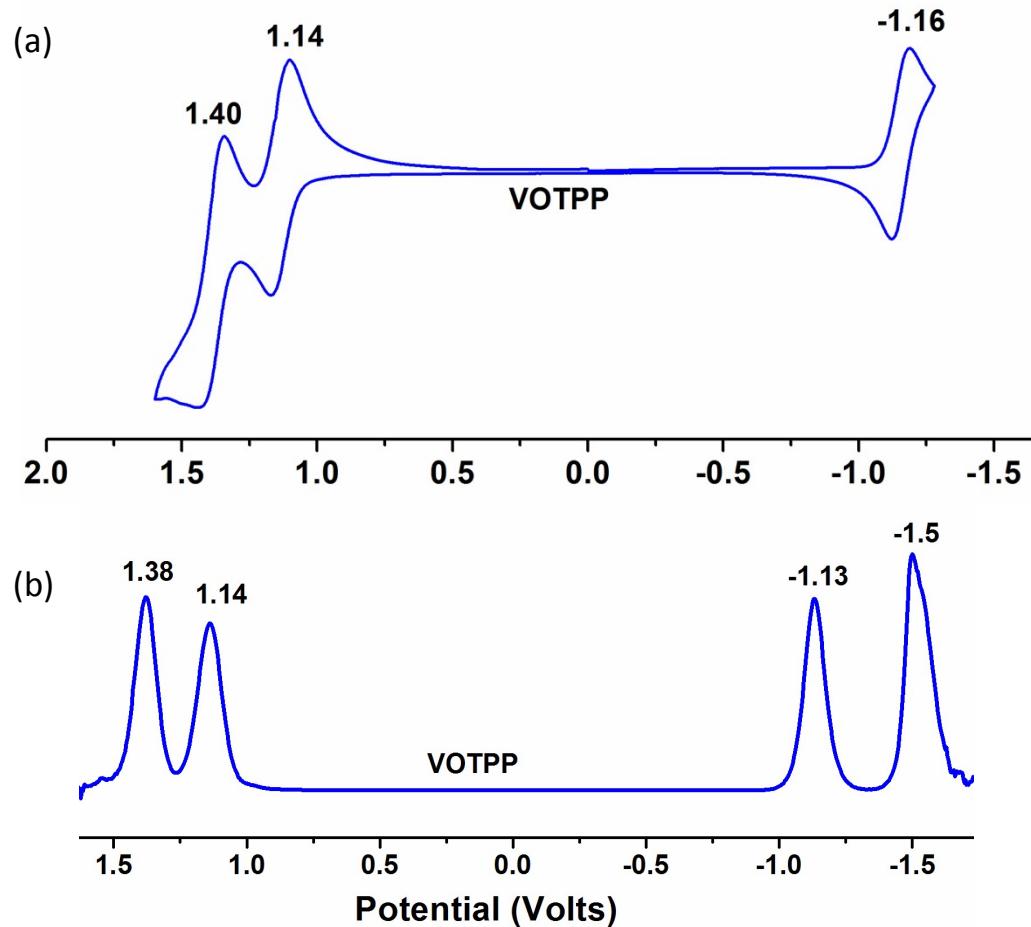


Figure S5. CV and DPV of VOTPPCl_8 in CH_2Cl_2 containing 0.1 M TBAPF_6 at 298 K.

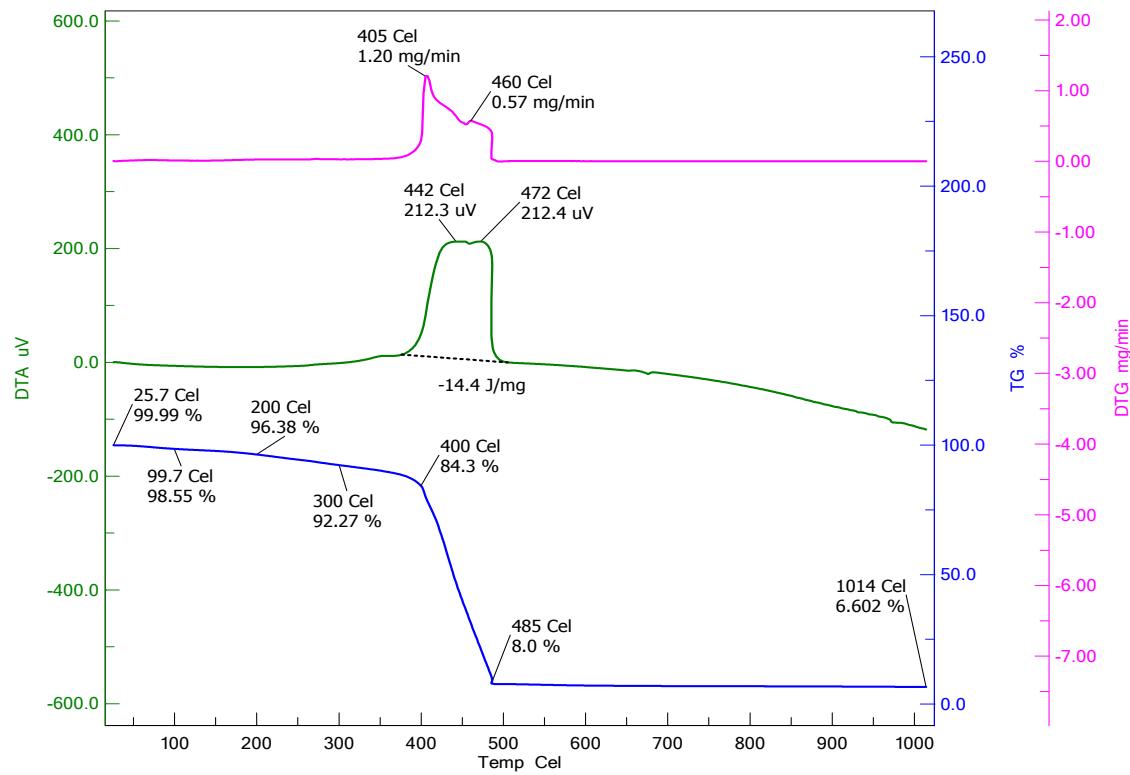


Figure S6. Thermogram (TG), Differential thermal analysis (DTA) and Differential thermogram (DTG) of VOTPPCl₈ at the heating rate of 10 °C /minute scanned from 25 °C to 1000 °C.

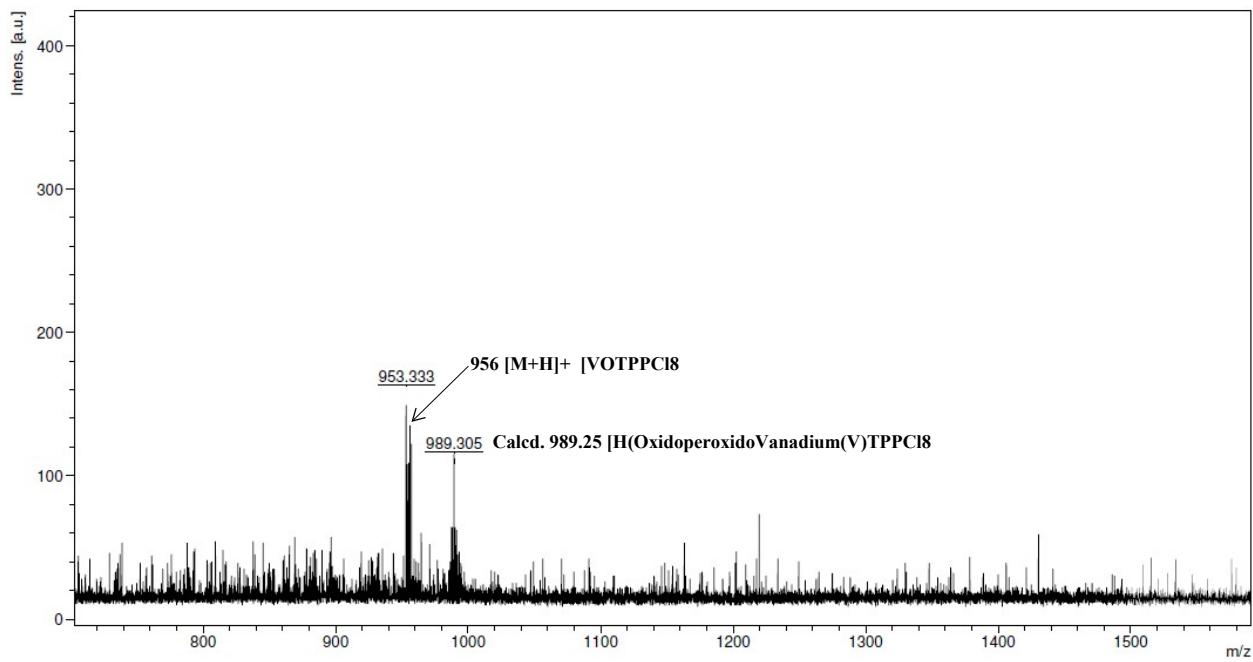


Figure S7. MALDI-TOF mass spectra of VOTPPCl₈ in presence of hydrogen peroxide and sodium bicarbonate using HABA matrix.

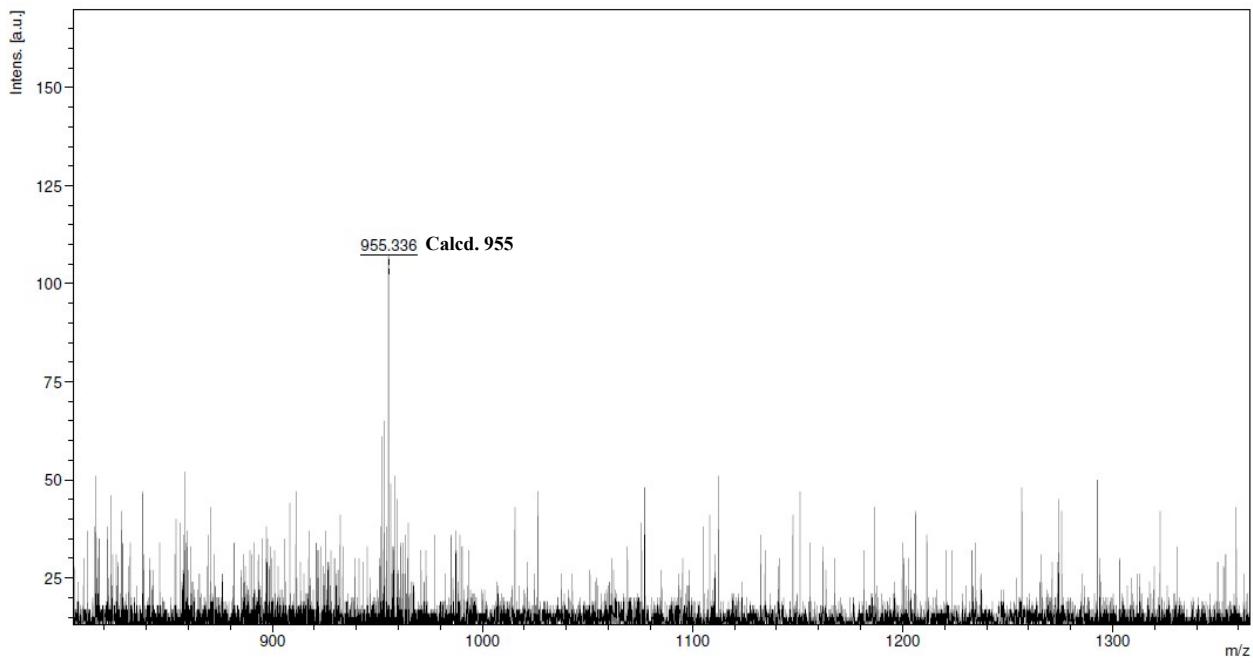


Figure S8. MALDI-TOF mass spectra of the reaction mixture after addition of the cyclohexene to oxido-peroxidovanadium(V) intermediate using HABA matrix.

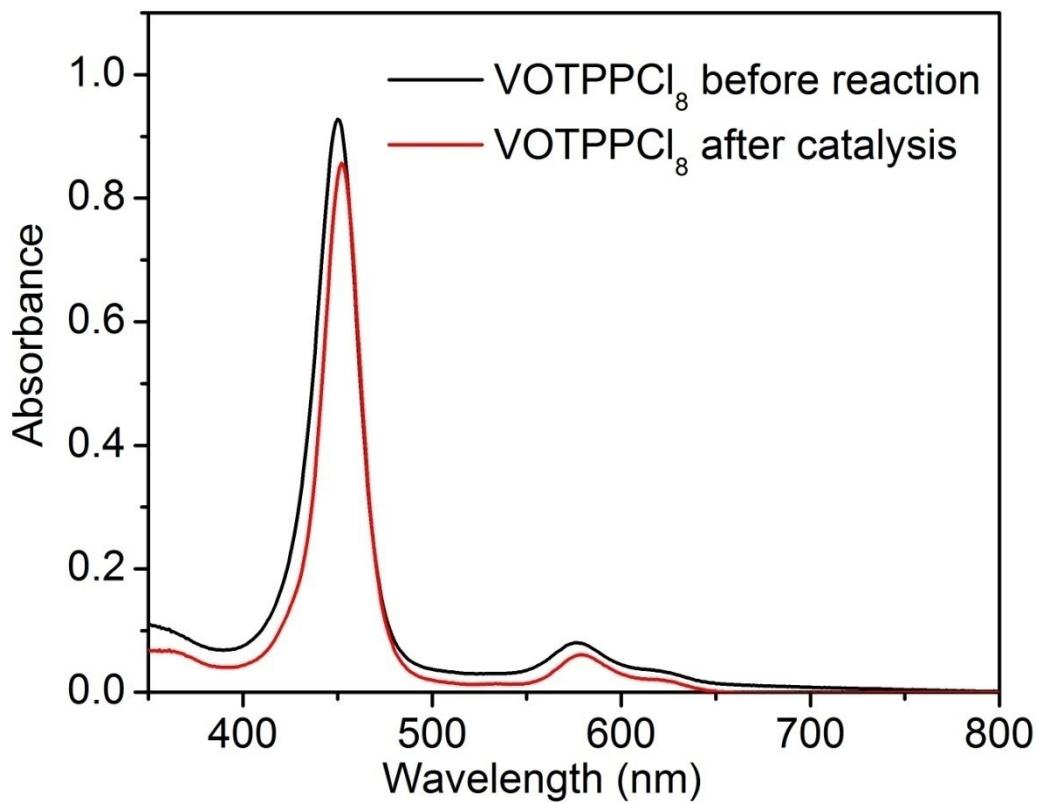


Figure S9. UV-Visible spectra of VOTPPCl₈ in CH₂Cl₂ (before catalysis) and in CH₃CN (after catalytic reaction) at 298 K.