

Electronic Supplementary Information (ESI)

Water-Soluble Tb³⁺ and Eu³⁺ Complexes Based on Task-Specific Ionic Liquid Ligand and Their Application in luminescent poly(vinyl alcohol) Films

Qing-Feng Li,^{*a} Dan Yue,^a Gen-wu Ge,^a Xiaodi Du,^a Yanchuan Gong,^a Zhenling Wang^{*a} and Jianhua Hao^{*b}

^a The Key Laboratory of Rare Earth Functional Materials and Applications, Zhoukou Normal University, Zhoukou 466001, P. R. China

^b Department of Applied Physics, The Hong Kong Polytechnic University, Hong Kong, P. R. China

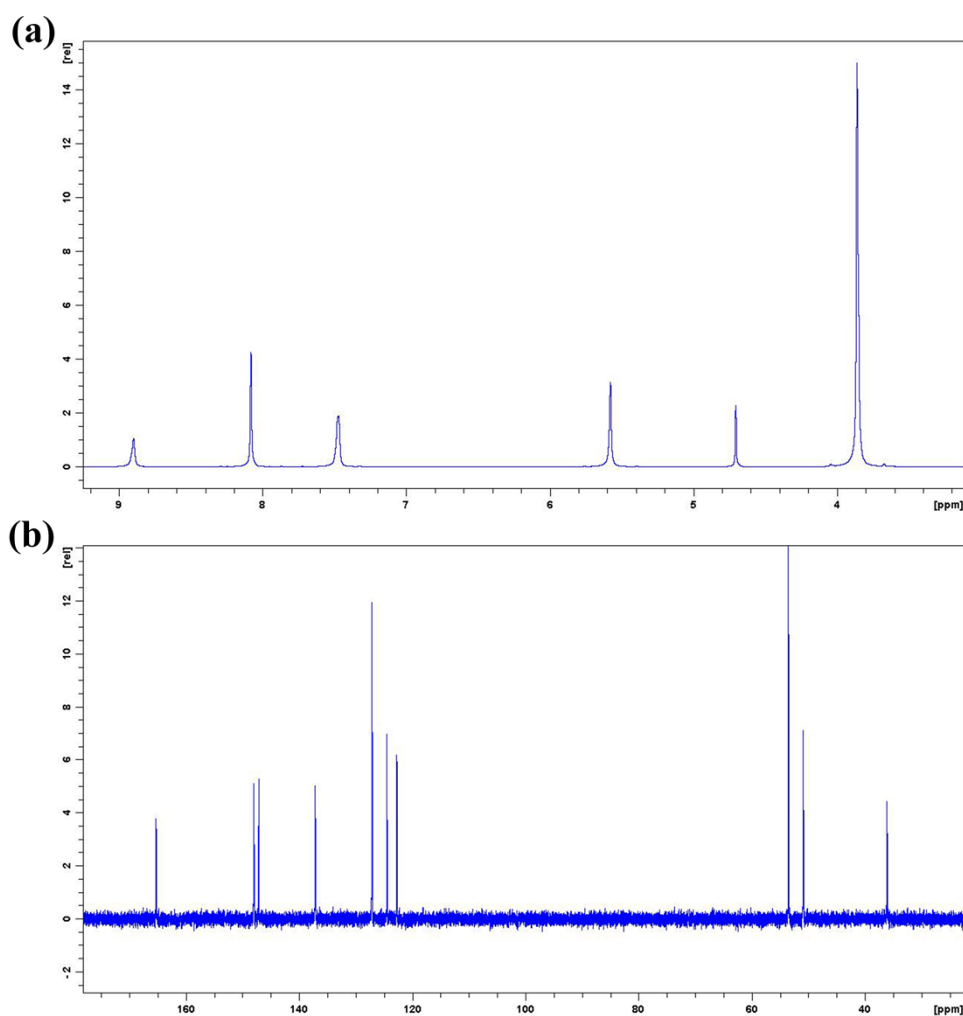


Fig. S1 ¹H-NMR (a) and ¹³C-NMR (b) of compound 4, Solvent: D₂O, ¹H-NMR Solvent peak: 4.7 ppm

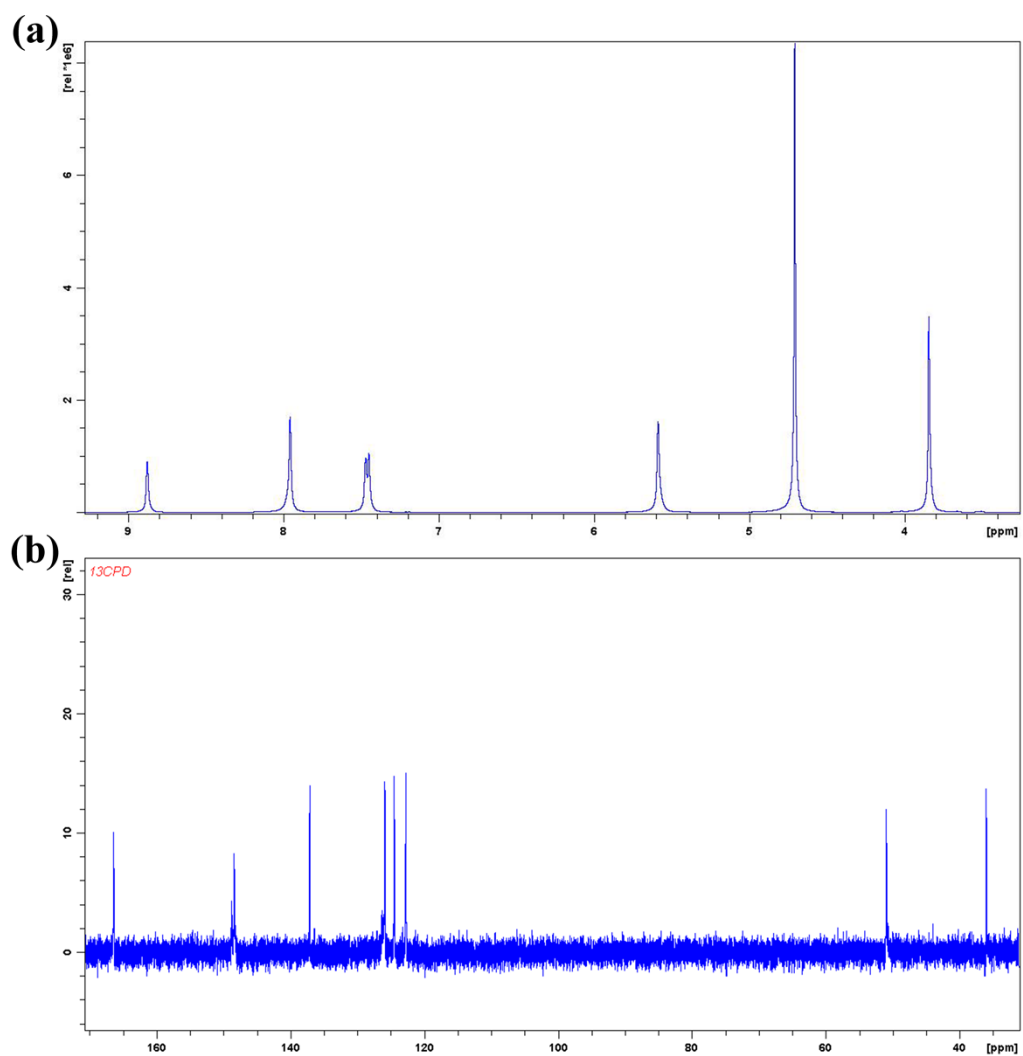


Fig. S2 ^1H -NMR (a) and ^{13}C -NMR (b) of IMI-DPA, Solvent: D_2O , ^1H -NMR Solvent peak: 4.7 ppm

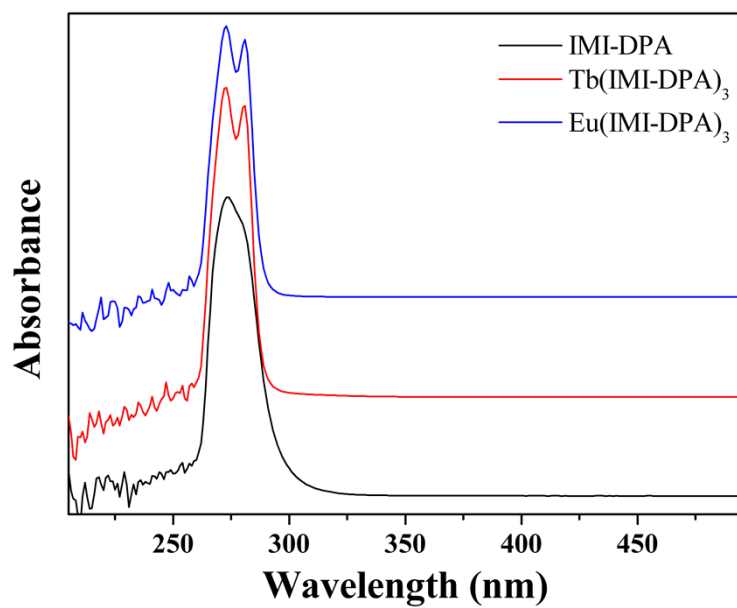


Fig. S3 UV-vis absorption spectra of the IMI-DPA (0.1 mmol/L) and complexes (0.05 mmol/L) in aqueous solution

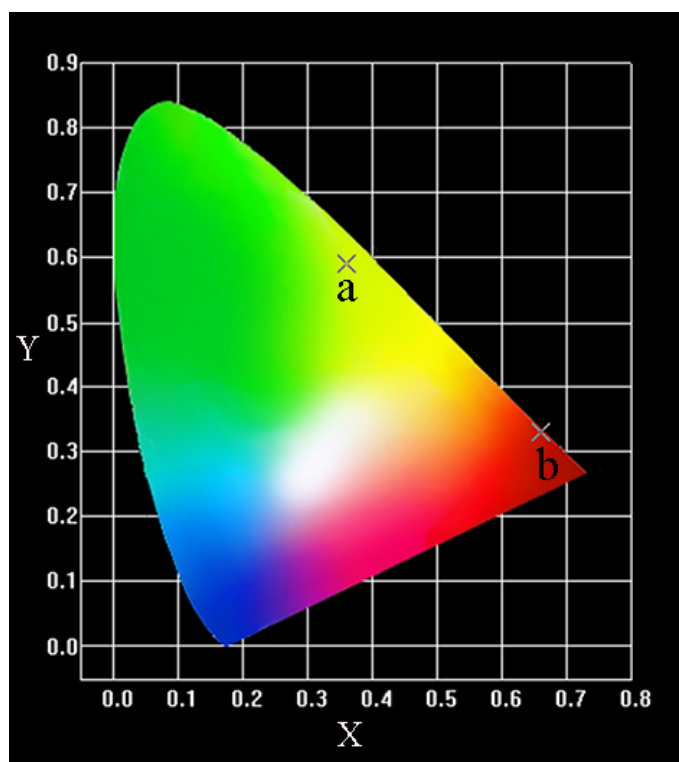


Fig. S4 Chromaticity diagram (CIE) of the Tb(IMI-DPA)₃ (a) and Eu(IMI-DPA)₃ (b) under 290 nm UV illumination

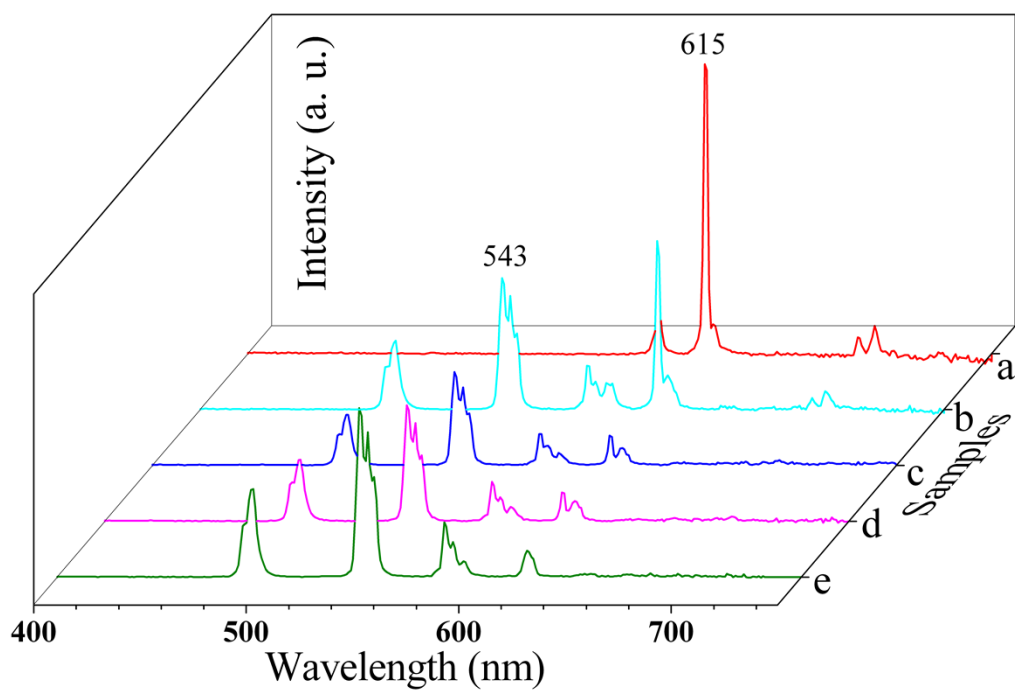


Fig. S5 The emission spectra of the luminescence PVA films upon excitation at 290 nm: (a) PVA-Eu; (b) PVA-Eu₁Tb₁; (c) PVA-Eu₁Tb₃; (d) PVA-Eu₁Tb₅; (e) PVA-Tb

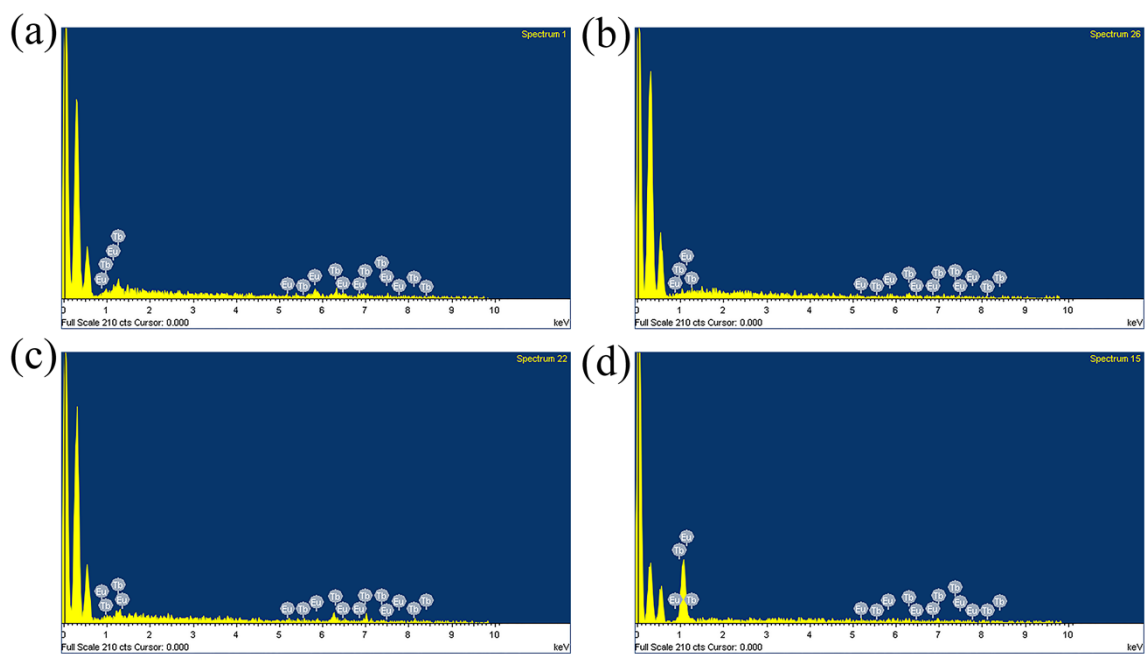


Fig. S6 EDX graphs of PVA-Eu₁Tb₁ (a), PVA-Eu₁Tb₃ (b), PVA-Eu₁Tb₅ (c), PVA-Na-BTC-Eu₁Tb₁ (d)

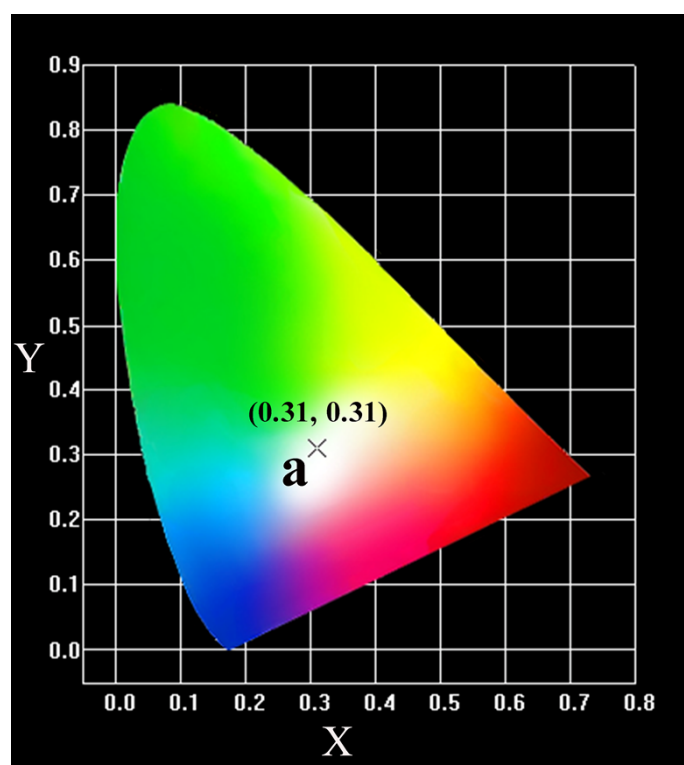


Fig. S7 Chromaticity diagram (CIE) of the PVA-Na-BTC-Eu₁Tb₁ (a) under 305 nm UV illumination