

Electronic supplementary material

Bis(alkyl) rare-earth complexes supported by new tridentate amidinate ligand with a pendant diphenylphosphine oxide group.

Synthesis, structures and catalytic activity in isoprene polymerization

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Table 1S. Crystallographic data and structure refinement details for **1**, **2**, **4** and **5**.

Table 2S. Catalytic tests in isoprene polymerization initiated by systems (**3-5**)/borate/AlMe₃ (borate: [Et₃NH][BPh₄], B(C₆F₅)₃, [Ph₃C][B(C₆F₅)₄], [PhNHMe₂][B(C₆F₅)₄], [Ln]/[borate]/[AlMe₃]=1:1:10).

Figure 1S. ¹H NMR spectrum of 2-[Ph₂P(O)]C₆H₄NHC(*t*Bu)=N(2,6-Me₂C₆H₃) (**1**).

Figure 2S. ¹³C NMR spectrum of 2-[Ph₂P(O)]C₆H₄NHC(*t*Bu)=N(2,6-Me₂C₆H₃) (**1**).

Figure 3S. ³¹P NMR spectrum of 2-[Ph₂P(O)]C₆H₄NHC(*t*Bu)=N(2,6-Me₂C₆H₃) (**1**).

Figure 4S. IR spectrum of 2-[Ph₂P(O)]C₆H₄NHC(*t*Bu)=N(2,6-Me₂C₆H₃) (**1**).

Figure 5S. ¹H NMR spectrum of {2-[Ph₂P(O)]C₆H₄NC(*t*Bu)N(2,6-Me₂C₆H₃)}YCl₂(DME) (**2**).

Figure 6S. ¹³C NMR spectrum of {2-[Ph₂P(O)]C₆H₄NC(*t*Bu)N(2,6-Me₂C₆H₃)}YCl₂(DME) (**2**).

Figure 7S. ³¹P NMR spectrum of {2-[Ph₂P(O)]C₆H₄NC(*t*Bu)N(2,6-Me₂C₆H₃)}YCl₂(DME) (**2**).

Figure 8S. IR spectrum of {2-[Ph₂P(O)]C₆H₄NC(*t*Bu)N(2,6-Me₂C₆H₃)}YCl₂(DME) (**2**).

Figure 9S. ^1H NMR spectrum of $[2\text{-}(\text{P}(\text{O})\text{Ph}_2)\text{PhNC}(t\text{Bu})(2,6\text{-Me}_2\text{C}_6\text{H}_3)\text{Y}(\text{CH}_2\text{SiMe}_3)_2\text{THF}]$ (**3**).

Figure 10S. ^{13}C NMR spectrum of $[2\text{-}(\text{P}(\text{O})\text{Ph}_2)\text{PhNC}(t\text{Bu})(2,6\text{-Me}_2\text{C}_6\text{H}_3)\text{Y}(\text{CH}_2\text{SiMe}_3)_2\text{THF}]$ (**3**).

Figure 11S. ^{31}P NMR spectrum of $[2\text{-}(\text{P}(\text{O})\text{Ph}_2)\text{PhNC}(t\text{Bu})(2,6\text{-Me}_2\text{C}_6\text{H}_3)\text{Y}(\text{CH}_2\text{SiMe}_3)_2\text{THF}]$ (**3**).

Figure 12S. ^1H NMR spectrum of $[2\text{-}(\text{P}(\text{O})\text{Ph}_2)\text{PhNC}(t\text{Bu})(2,6\text{-Me}_2\text{C}_6\text{H}_3)\text{Lu}(\text{CH}_2\text{SiMe}_3)_2]$ (**5**).

Figure 13S. ^{13}C NMR spectrum of $[2\text{-}(\text{P}(\text{O})\text{Ph}_2)\text{PhNC}(t\text{Bu})(2,6\text{-Me}_2\text{C}_6\text{H}_3)\text{Lu}(\text{CH}_2\text{SiMe}_3)_2]$ (**5**).

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Table 1S. Crystallographic data and structure refinement details for **1, 2, 4** and **5**

Compound	1	2	4	5
Empirical formula	C ₃₁ H ₃₃ N ₂ OP	C ₃₇ H ₄₇ Cl ₂ N ₂ O ₄ PY	C ₄₃ H ₆₂ ErN ₂ O ₂ PSi ₂	C _{42.50} H ₅₈ LuN ₂ OPSi ₂
Formula weight	480.56	774.55	893.35	875.03
T [K]	100(2)	100(2)	100(2)	100(2)
Wavelength [Å]	0.71073	0.71073	0.71073	0.71073
Crystal system	Monoclinic	Tetragonal	Monoclinic	Triclinic
Space group	P2(1)/c	I-1	P2(1)/n	P-1
a [Å]	14.4270(2)	26.2291(7)	10.243(1)	11.6576(2)
b [Å]	18.0000(2)	26.2291(7)	19.035(2)	11.9319(2)
c [Å]	10.2803(2)	10.7386	21.678(3)	18.0960(3)
α [°]	90	90	90	74.619(1)
β [°]	102.366(1)	90	95.195(2)	72.630(1)
γ [°]	90	90	90	63.078(1)
Volume [Å ³]	2607.72(6)	7387.8(4)	4209.1(8)	2116.79(5)
Z	4	8	4	2
ρ _{calcd.} [g cm ⁻³]	1.224	1.393	1.410	1.373
Absorption coefficient [mm ⁻¹]	0.132	1.807	2.126	2.459
F(000)	1024	3224	1844	898
Crystal size [mm]	0.40×0.20×0.10	0.37×0.20×0.18	0.23×0.14×0.11	0.50×0.25×0.15
θ range for data collection [°]	3.04 to 28.00	2.05 to 26.00	2.169 to 26.00	2.83 to 27.00
Index ranges	-19≤h≤19, -23≤k≤23, -13≤l≤13	-32≤h≤32, -31≤k≤32, -13≤l≤13	-12≤h≤12, -23≤k≤23, -26≤l≤26	-14≤h≤14, 15≤k≤15, -23≤l≤23
Reflections collected	45921	31590	38140	34254
Independent reflections	6265	7213	8254	9179
R _{int}	0.0536	0.0490	0.0380	0.0406
Completeness to θ [%]	99.6	99.6	99.5	99.4
Data/restraints/parameters	6265/0/325	7213/21/432	8254/0/471	9179/0/490
Goodness-of-fit on F ²	1.036	1.001	1.066	1.037
Final R indices [I>2σ(I)]	R1 = 0.0383, wR2 = 0.0961	R1 = 0.0355, wR2 = 0.0788	R1 = 0.0309, wR2 = 0.0818	R1 = 0.0243, wR2 = 0.0504
R indices (all data)	R1 = 0.0509, wR2 = 0.1021	R1 = 0.0458, wR2 = 0.0818	R1 = 0.0367, wR2 = 0.0851	R1 = 0.0297, wR2 = 0.0516
Largest diff. peak and hole [eÅ ⁻³]	0.554/-0.260	1.157/-0.507	1.927/-0.702	1.136/-0.604

Table 2S. Catalytic tests in isoprene polymerization initiated by systems (3-5)/borate/ AlMe_3 (borate: $[\text{Et}_3\text{NH}][\text{BPh}_4]$, $\text{B}(\text{C}_6\text{F}_5)_3$, $[\text{Ph}_3\text{C}][\text{B}(\text{C}_6\text{F}_5)_4]$, $[\text{PhNHMe}_2][\text{B}(\text{C}_6\text{F}_5)_4]$, $[\text{Ln}]/[\text{borate}]/[\text{AlMe}_3]=1:1:10$).

	comp.	borate	[IP]/[Ln]	t, h	Yield, %	Cis-1,4	Trans-1,4	3,4-	$M_n(\times 10^{-3})^a$	$M_n(\times 10^{-3})_{\text{calc}}^b$	M_w/M_n
1	3	Et_3NB	1000	24	0	-	-	-	-	65.0	-
2	3	$\text{B}(\text{C}_6\text{F}_5)_3$	1000	24	0	-	-	-	-	65.0	-
3	3	HNB	1000	24	0	-	-	-	-	65.0	-
4	3	TB	1000	24	0	-	-	-	-	65.0	-
5	4	Et_3NB	1000	24	0	-	-	-	-	65.0	-
6	4	$\text{B}(\text{C}_6\text{F}_5)_3$	1000	24	0	-	-	-	-	65.0	-
7	4	HNB	1000	24	0	-	-	-	-	65.0	-
8	4	TB	1000	24	0	-	-	-	-	65.0	-
9	5	Et_3NB	1000	24	0	-	-	-	-	65.0	-
10	5	$\text{B}(\text{C}_6\text{F}_5)_3$	1000	24	0	-	-	-	-	65.0	-
11	5	HNB	1000	24	0	-	-	-	-	65.0	-
12	5	TB	1000	24	0	-	-	-	-	65.0	-
13	3	-	1000	24	0	-	-	-	-	65.0	-
14	4	-	1000	24	0	-	-	-	-	65.0	-
15	5	-	1000	24	0	-	-	-	-	65.0	-

Conditions: complex (10 μmol in toluene, $[\text{AlMe}_3]:[\text{Ln}]:[\text{borate}] = 10/1/1$, T: 25 $^\circ\text{C}$.); HNB = $[\text{PhNHMe}_2][\text{B}(\text{C}_6\text{F}_5)_4]$, TB = $[\text{Ph}_3\text{C}][\text{B}(\text{C}_6\text{F}_5)_4]$, $\text{Et}_3\text{NB} = [\text{Et}_3\text{NH}][\text{BPh}_4]$; a) Determined by GPC against polystyrene standard; *The catalytic tests were performed without the addition of borate; b) $M_{\text{calc}} = ([\text{IP}]/[\text{Ln}]) \times 68.12 \times (\text{conversion})$.

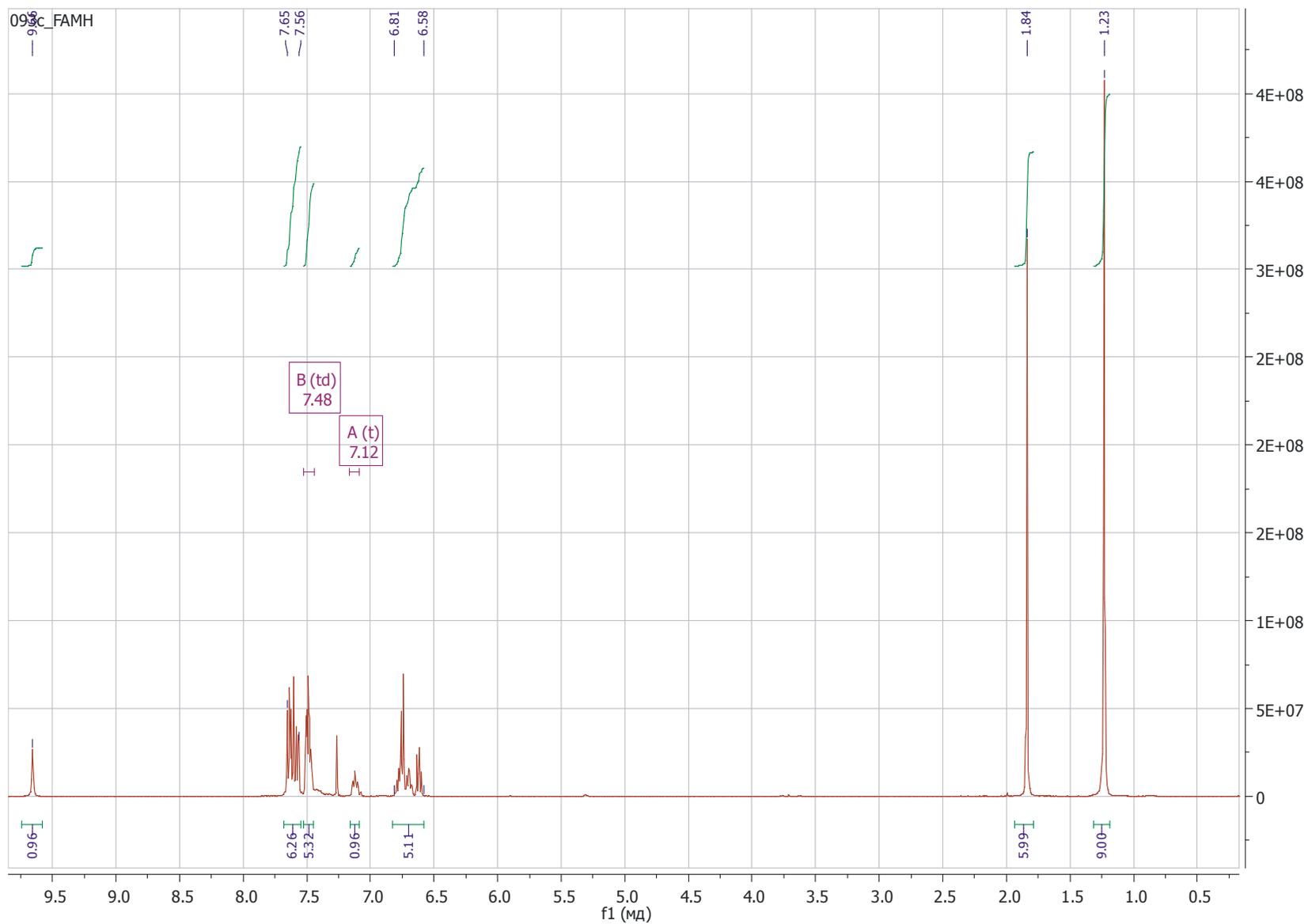


Figure 1S. ^1H NMR spectrum (400 MHz, CDCl_3 , 298 K) of 2-[$\text{Ph}_2\text{P}(\text{O})$] $\text{C}_6\text{H}_4\text{NHC}(t\text{Bu})=\text{N}(2,6\text{-Me}_2\text{C}_6\text{H}_3)$ (**1**).

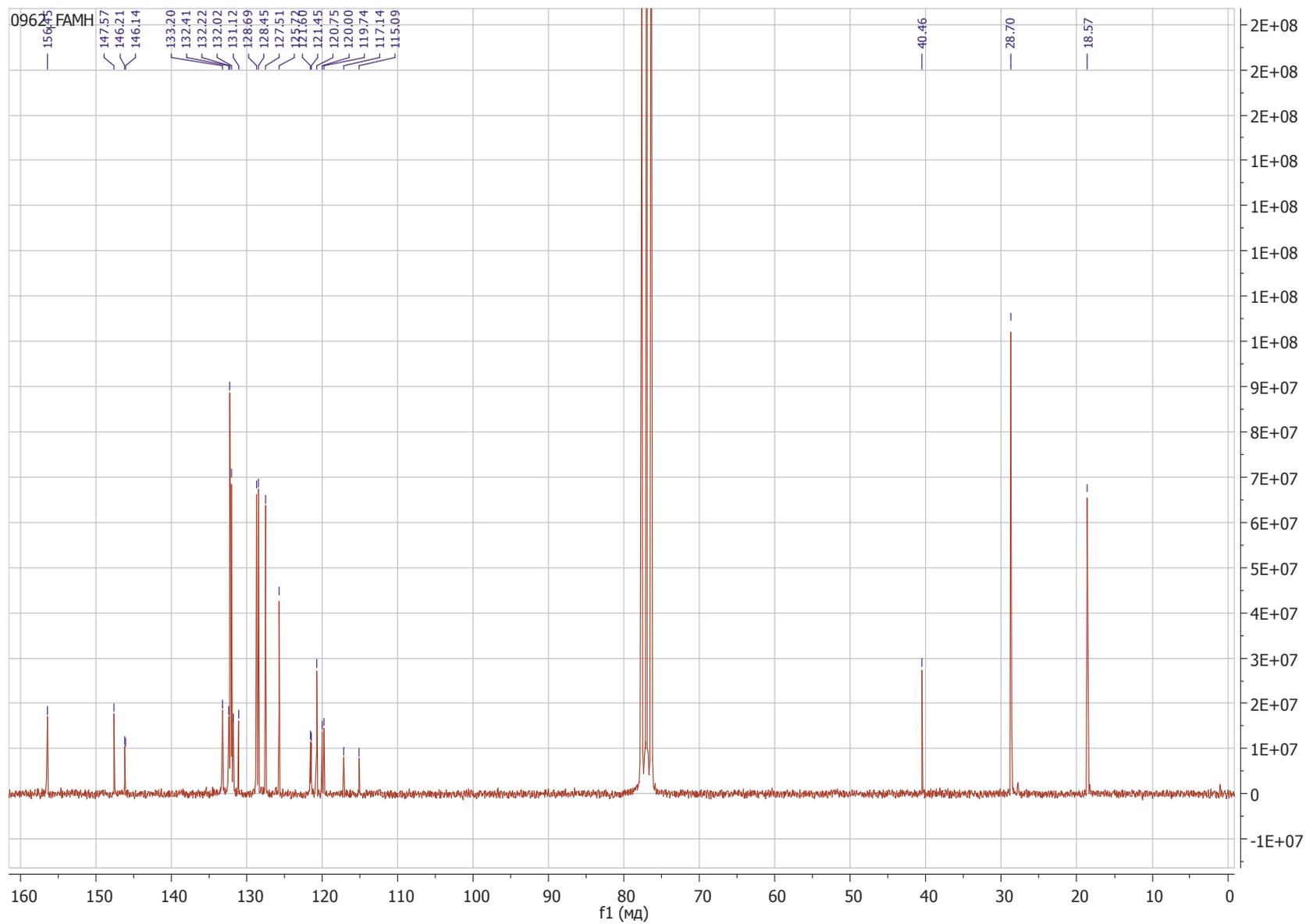


Figure 2S. ^{13}C NMR spectrum (50 MHz, CDCl_3 , 298 K) of 2-[$\text{Ph}_2\text{P}(\text{O})$] $\text{C}_6\text{H}_4\text{NHC}(\text{tBu})=\text{N}(2,6\text{-Me}_2\text{C}_6\text{H}_3)$ (**1**).

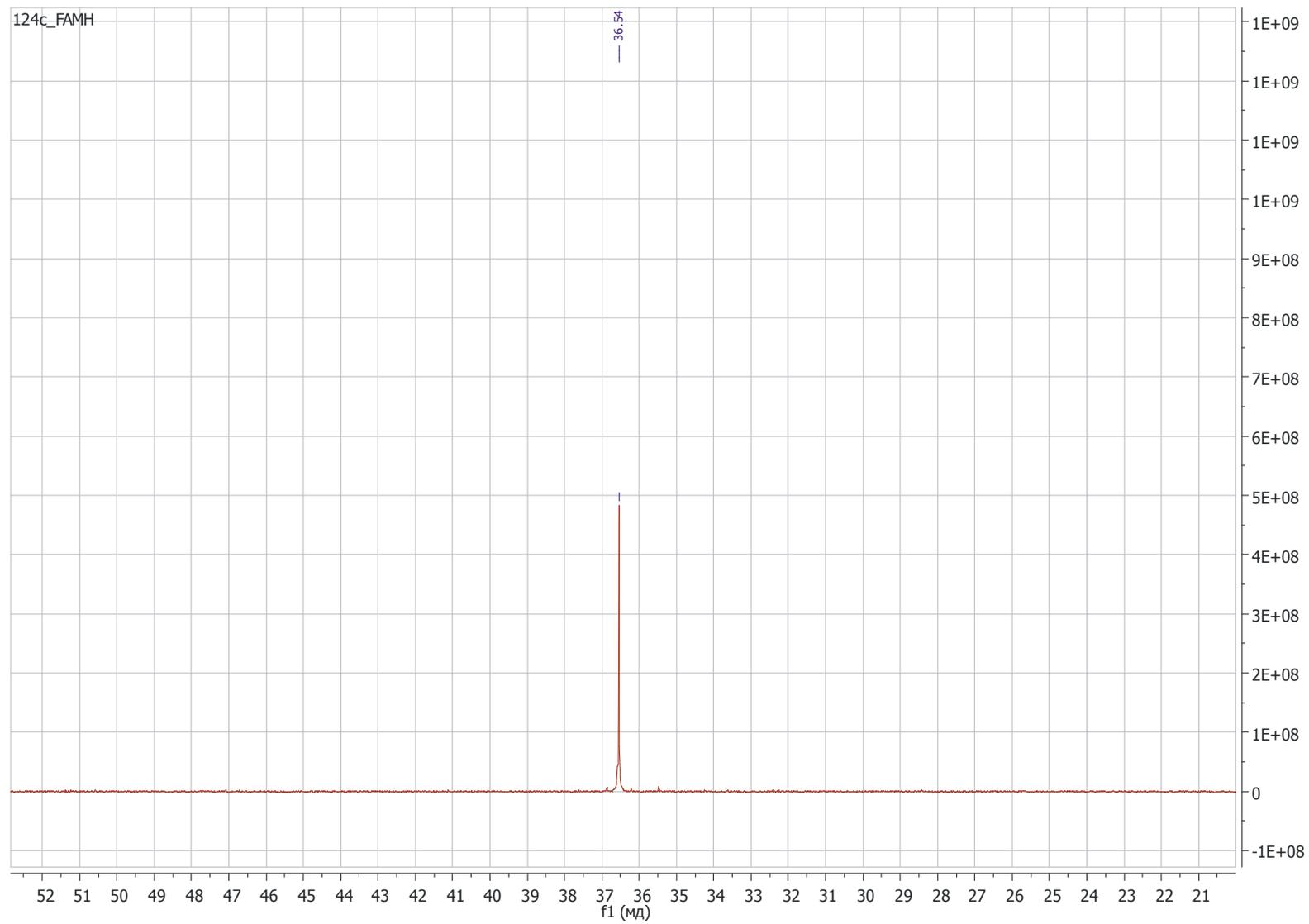


Figure 3S. ^{31}P NMR spectrum (81 MHz, CDCl_3 , 298 K) of 2-[$\text{Ph}_2\text{P}(\text{O})$] $\text{C}_6\text{H}_4\text{NHC}(t\text{Bu})=\text{N}(2,6\text{-Me}_2\text{C}_6\text{H}_3)$ (**1**).

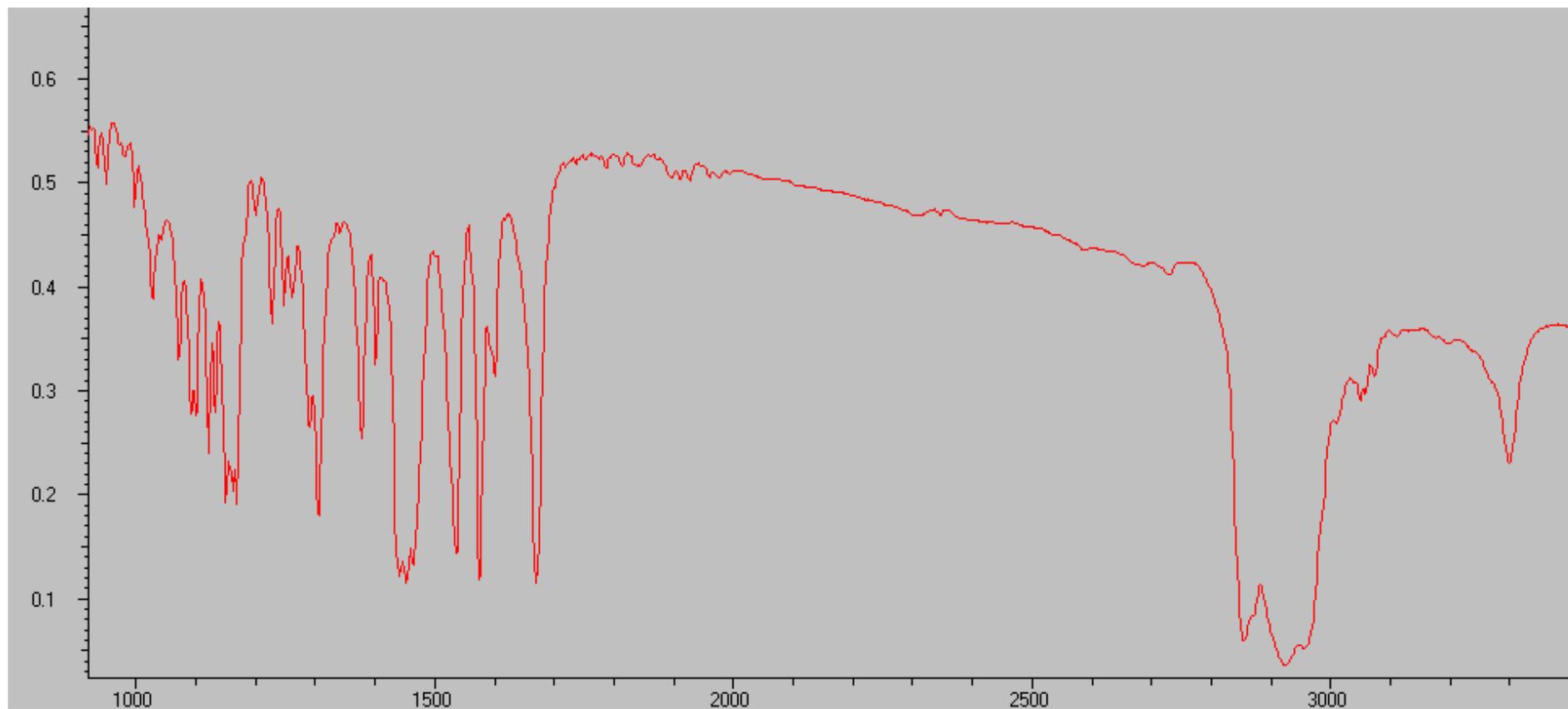


Figure 4S. IR spectrum of 2-[Ph₂P(O)]C₆H₄NHC(*t*Bu)=N(2,6-Me₂C₆H₃) (**1**).

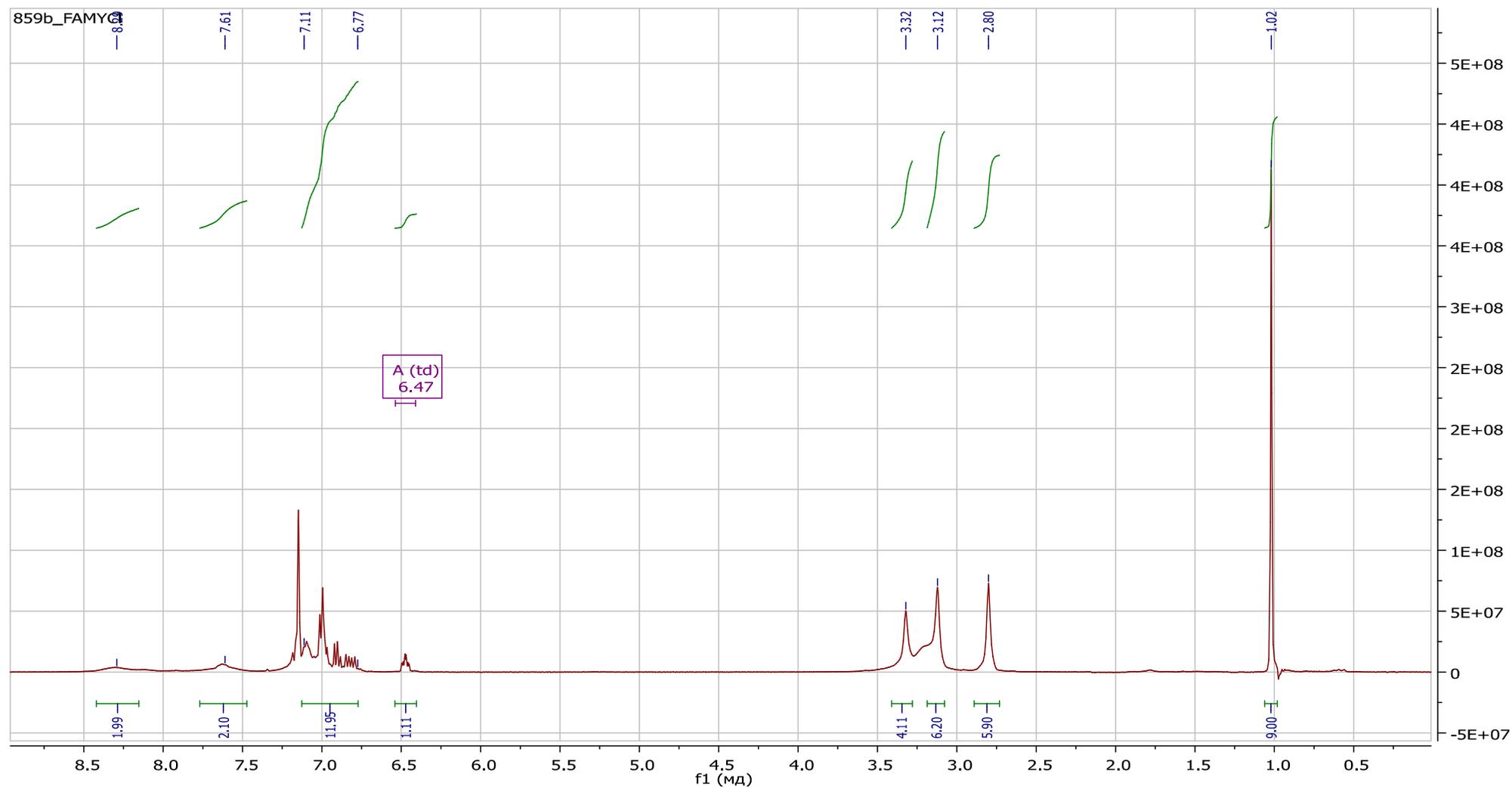


Figure 5S. ^1H NMR spectrum (400 MHz, C_6D_6 , 298 K) of $\{2\text{-}[\text{Ph}_2\text{P}(\text{O})]\text{C}_6\text{H}_4\text{NC}(t\text{Bu})\text{N}(2,6\text{-Me}_2\text{C}_6\text{H}_3)\}\text{YCl}_2(\text{DME})$ (**2**).

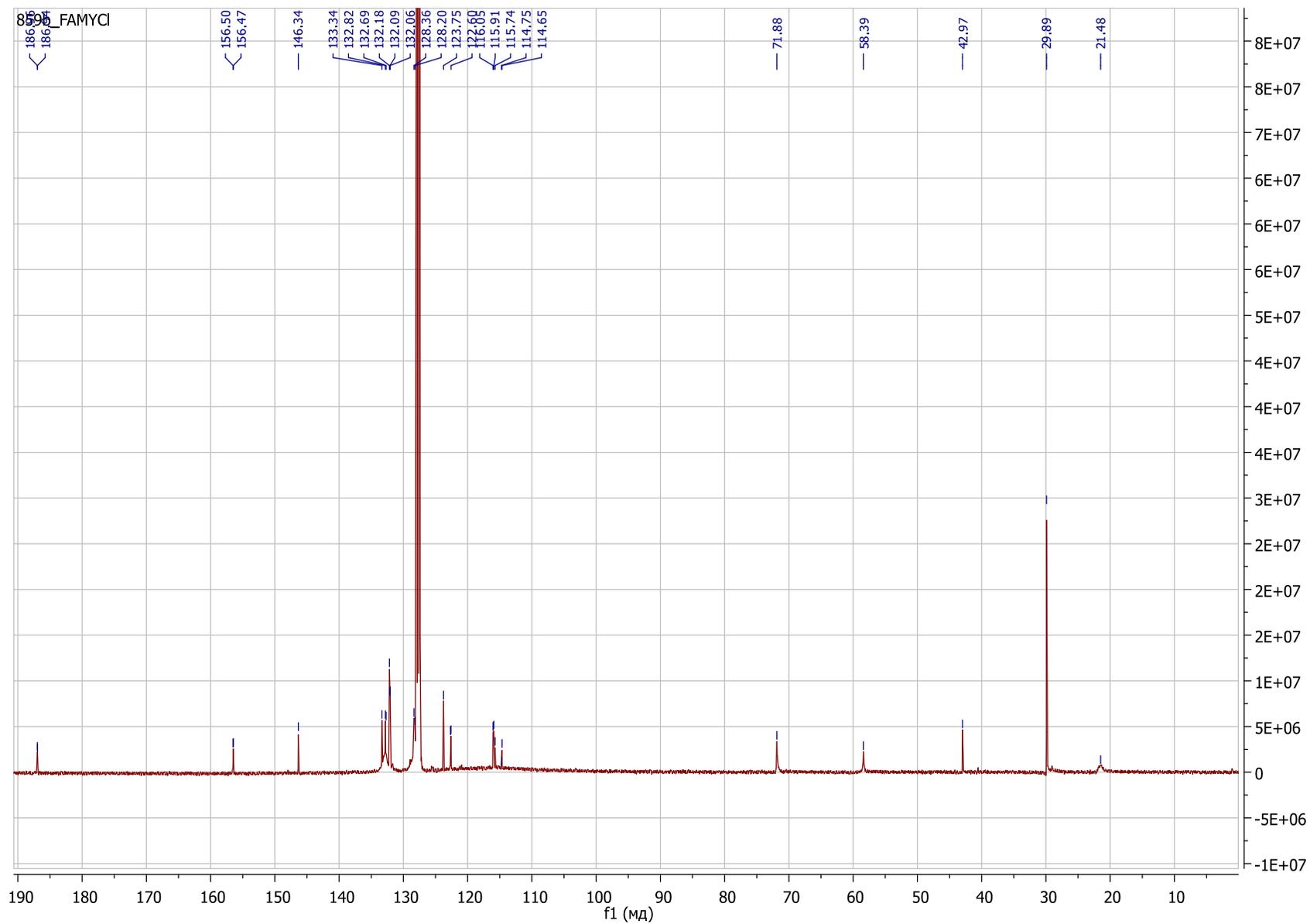


Figure 6S. ^{13}C NMR spectrum (100 MHz, C_6D_6 , 298 K) of $\{2\text{-}[\text{Ph}_2\text{P}(\text{O})]\text{C}_6\text{H}_4\text{NC}(t\text{Bu})\text{N}(2,6\text{-Me}_2\text{C}_6\text{H}_3)\} \text{YCl}_2(\text{DME})$ (**2**).

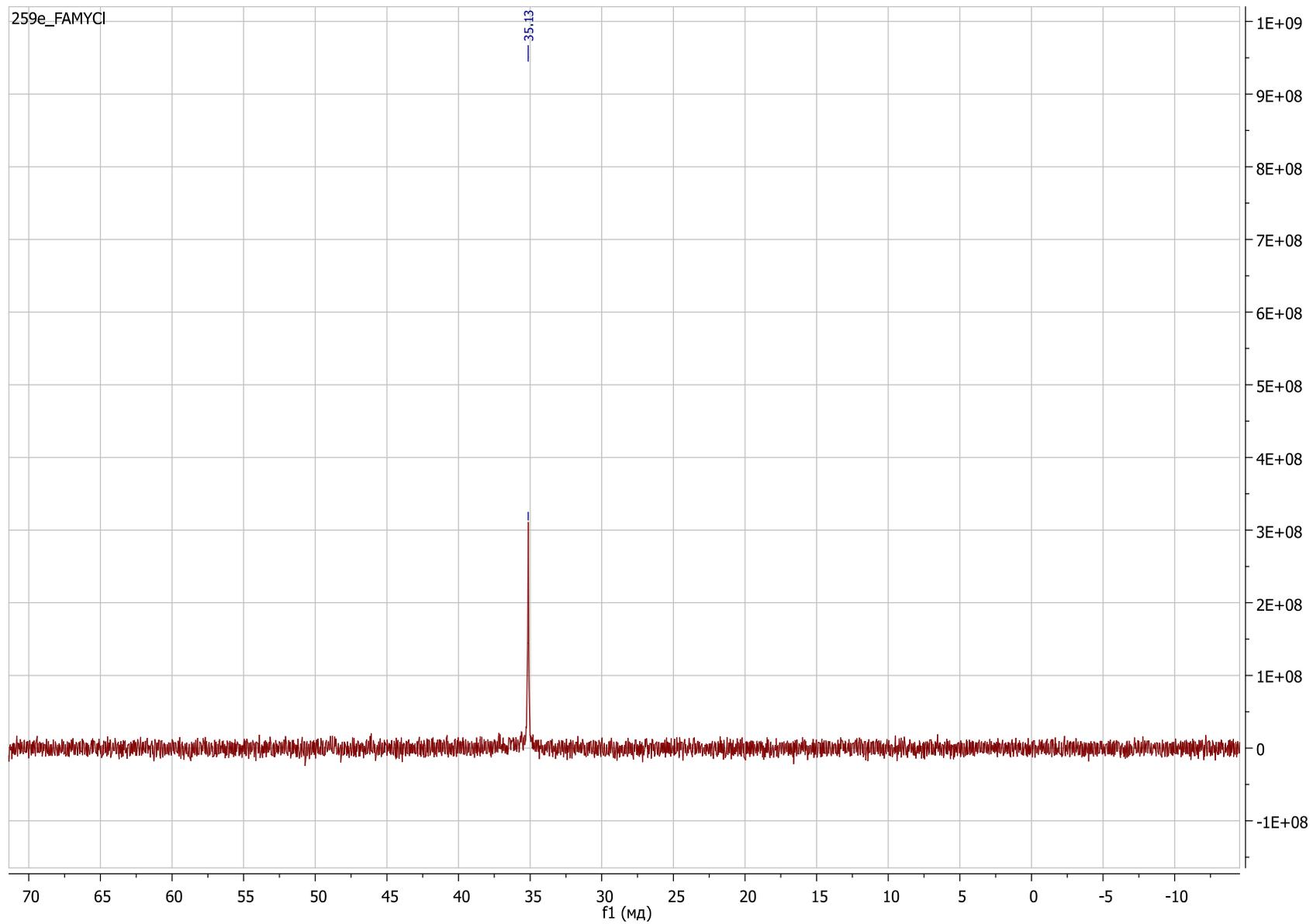


Figure 7S. ^{31}P NMR spectrum (81 MHz, C_6D_6 , 298 K) of $\{2\text{-}[\text{Ph}_2\text{P}(\text{O})\text{C}_6\text{H}_4\text{NC}(t\text{Bu})\text{N}(2,6\text{-Me}_2\text{C}_6\text{H}_3)]\text{YCl}_2(\text{DME})\}$ (**2**).

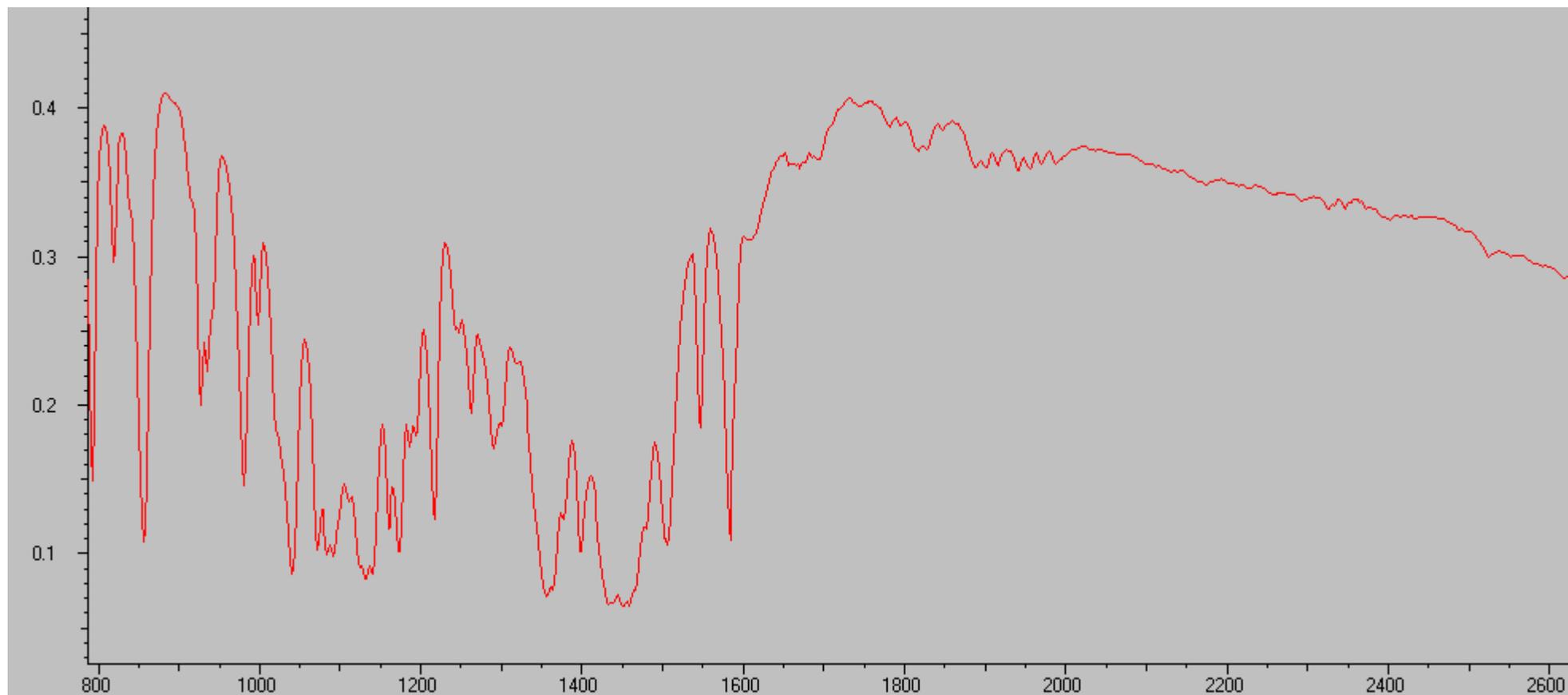


Figure 8S. IR spectrum of {2-[Ph₂P(O)]C₆H₄NC(*t*Bu)N(2,6-Me₂C₆H₃)}YCl₂(DME) (**2**).

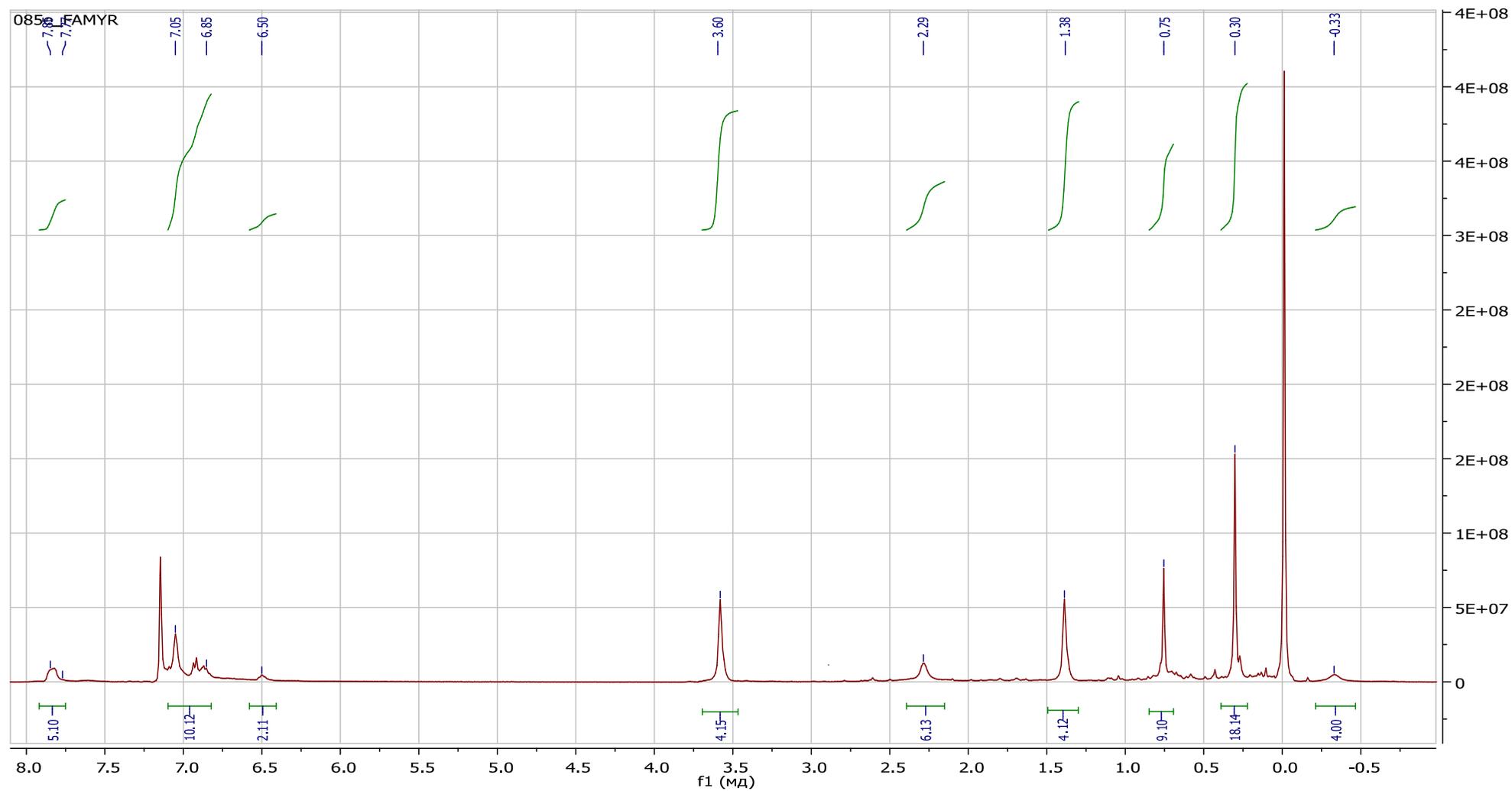


Figure 9S. ^1H NMR spectrum (400 MHz, C_6D_6 , 298 K) of $[2\text{-}(\text{P}(\text{O})\text{Ph}_2)\text{PhNC}(t\text{Bu})(2,6\text{-Me}_2\text{C}_6\text{H}_3)\text{Y}(\text{CH}_2\text{SiMe}_3)_2\text{THF}]$ (**3**).

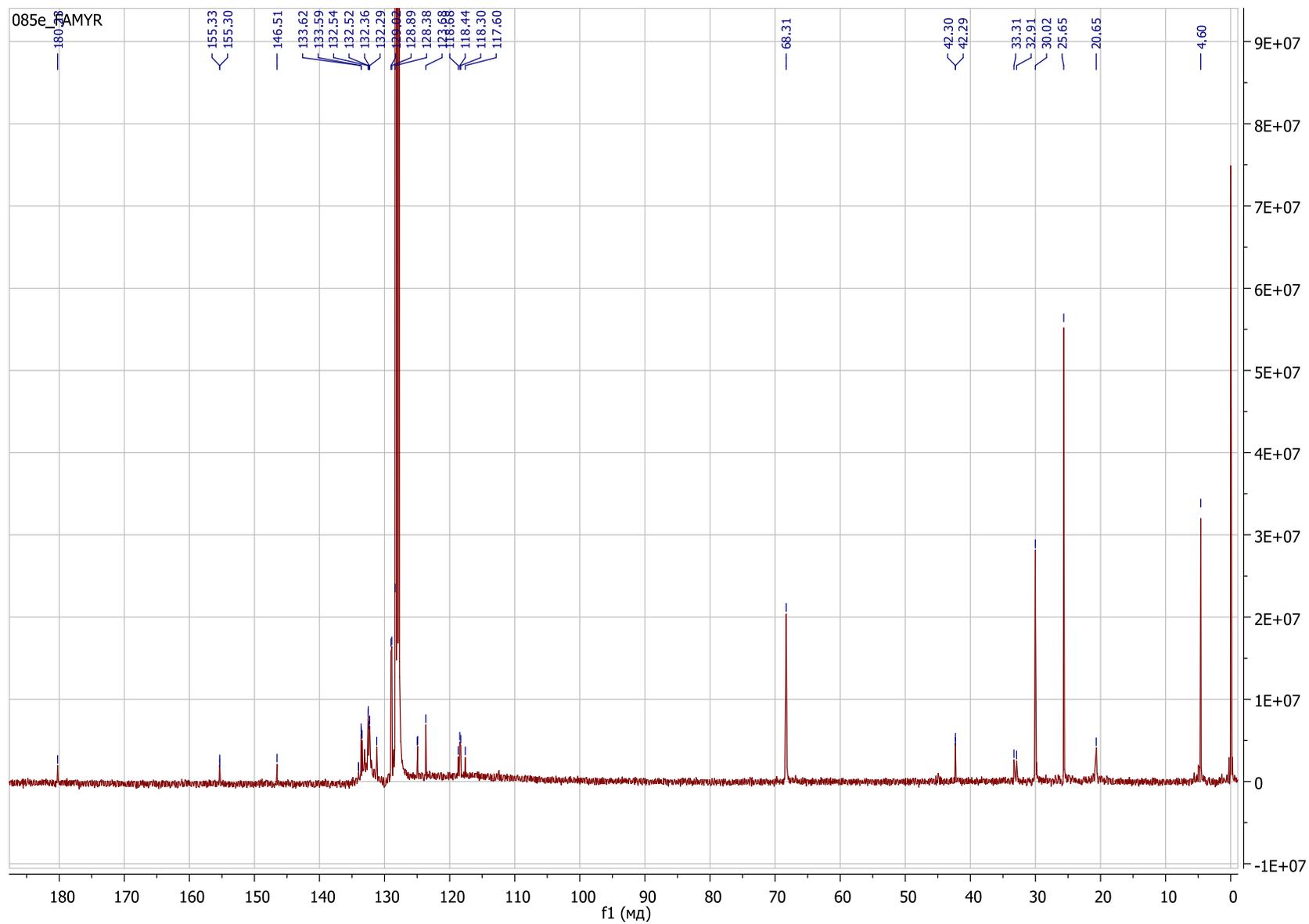


Figure 10S. ^{13}C NMR spectrum (100 MHz, C_6D_6 , 298 K) of $[2\text{-}(\text{P}(\text{O})\text{Ph}_2)\text{PhNC}(t\text{Bu})(2,6\text{-Me}_2\text{C}_6\text{H}_3)\text{Y}(\text{CH}_2\text{SiMe}_3)_2\text{THF}]$ (**3**).

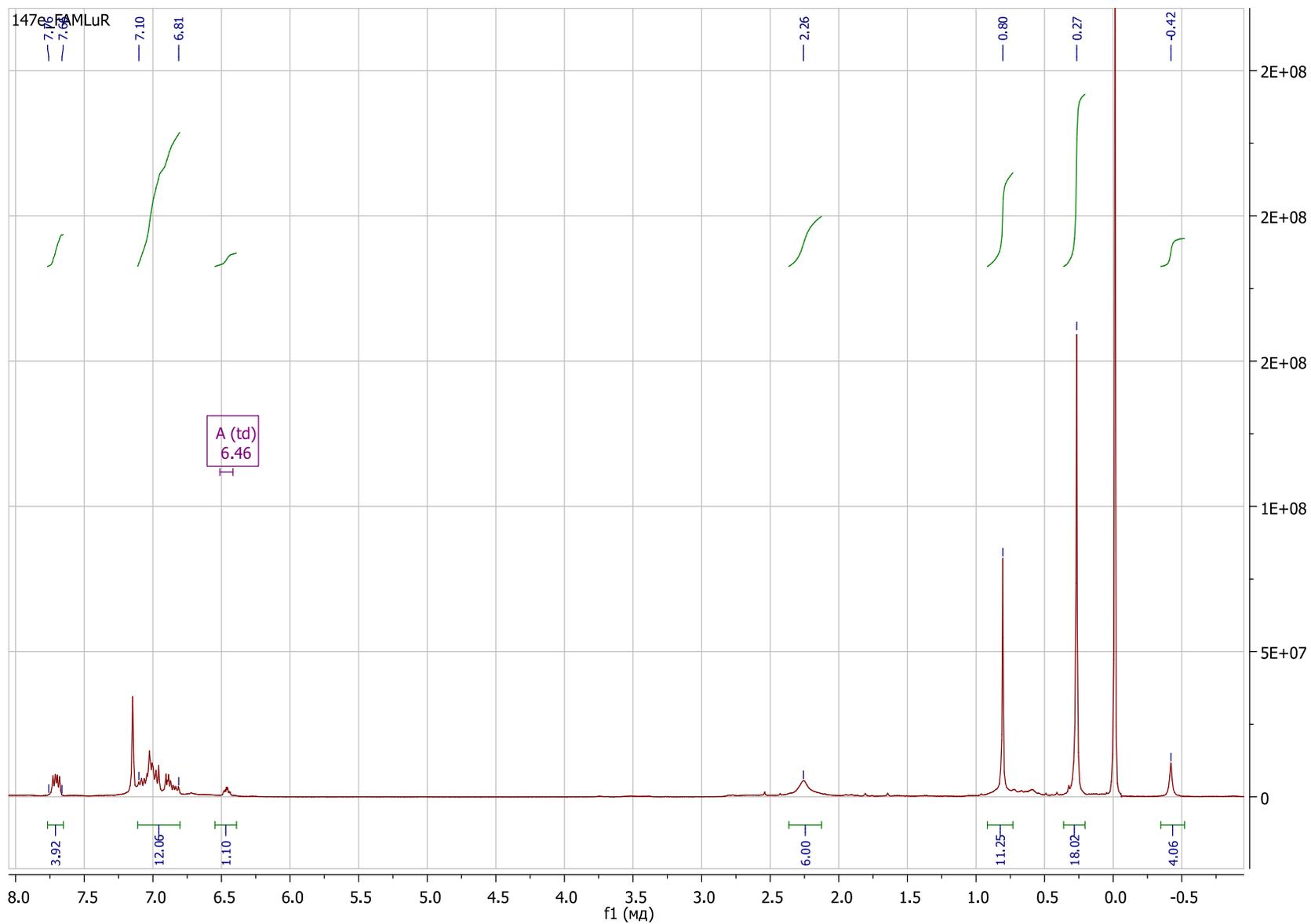


Figure 12S. ^1H NMR spectrum of (400 MHz, C_6D_6 , 298 K) $[2\text{-}(\text{P}(\text{O})\text{Ph}_2)\text{PhNC}(t\text{Bu})(2,6\text{-Me}_2\text{C}_6\text{H}_3)\text{Lu}(\text{CH}_2\text{SiMe}_3)_2]$ (**5**).

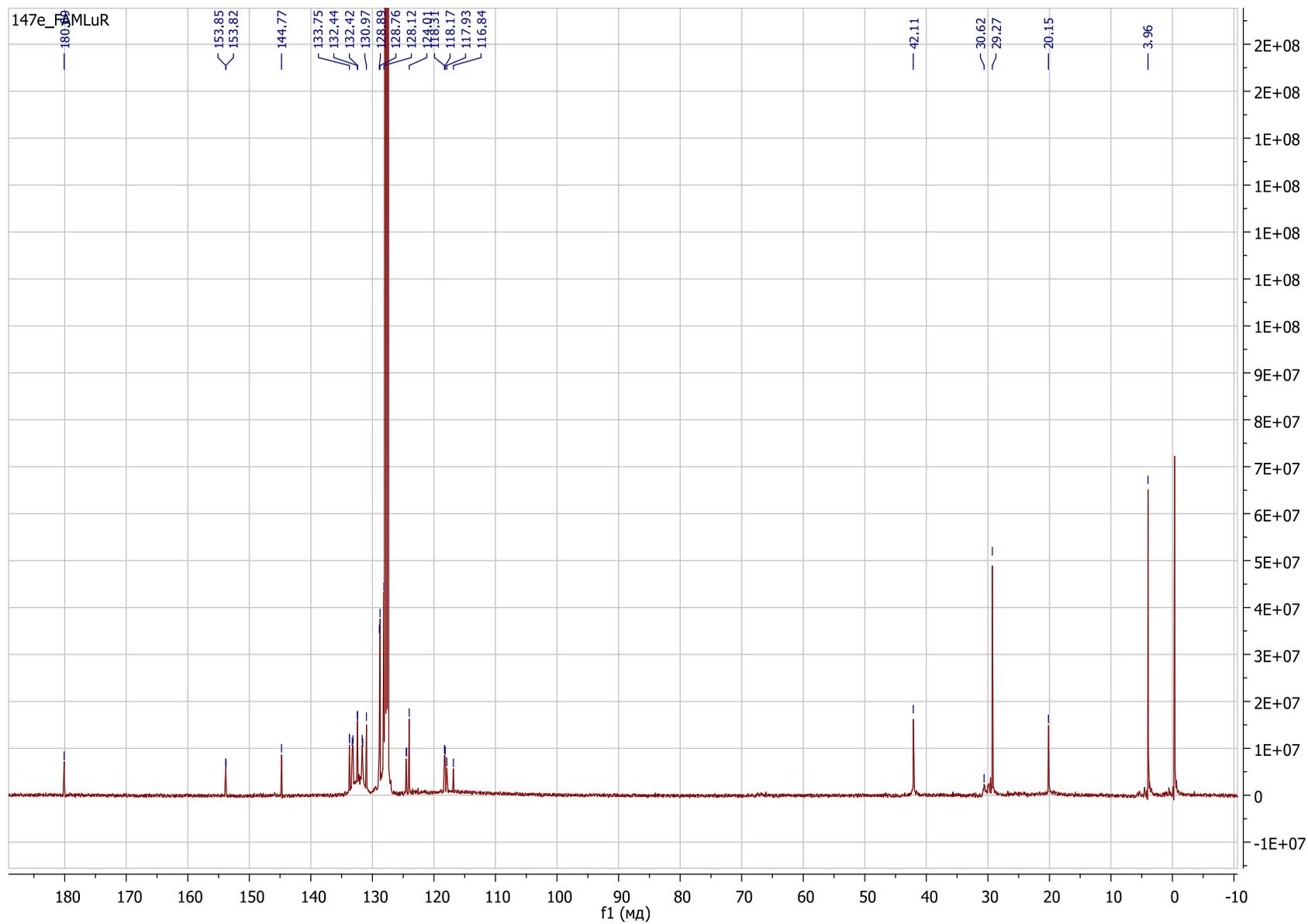


Figure 13S. ^{13}C NMR spectrum (100 MHz, C_6D_6 , 298 K) of $[2\text{-}(\text{P}(\text{O})\text{Ph}_2)\text{PhNC}(t\text{Bu})(2,6\text{-Me}_2\text{C}_6\text{H}_3)\text{Lu}(\text{CH}_2\text{SiMe}_3)_2]$ (**5**).

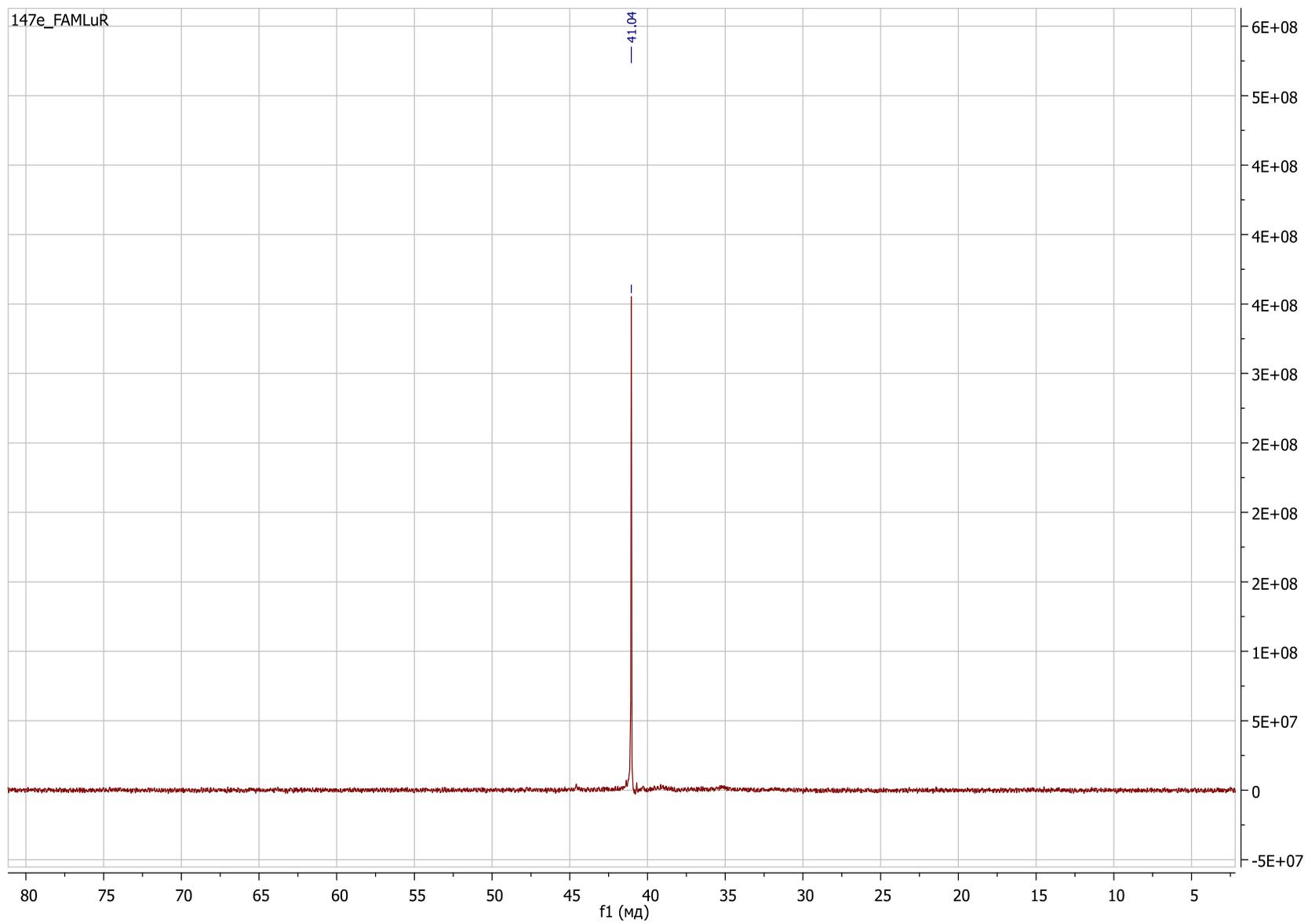


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