

Figure S1. TG and DSC curves of complexes 1 (top) and 2 (bottom).

Complexes investigated show similar thermal behaviour. Their noticeable weight loss starts at about 150 °C. The first decomposition step for 1 and 2 is characterized by the endothermic peak at 196 and 193 °C, correspondingly. Further heating is accompanied by exothermic peak at 318 (1) and 308 °C (2).



**Figure S2.** A hydrogen-bonded polymeric chain extending along the *a* axis in the crystal structure of 1-*tert*-Butyltetrazole (on the left), and crystall packing of 1-*tert*-Butyltetrazole viewed along the *a* axis, with dashed lines showing  $\pi \cdots \pi$  interactions between the tetrazole rings (on the right).



Figure S3. Field dependence of the magnetization of complex 2 as measured (square) and as calculated (doted).



**Figure S4.** B3LYP/6-311+G(d,p)//B3LYP/6-31G(d) calculated MESP contour map for 1methyltetrazole. One hydrogen atom is not shown for clarity.



**Figure S5.** B3LYP/6-311+G(d,p)//B3LYP/6-31G(d) calculated MESP contour map for 1-phenyltetrazole.





Figure S6. X-ray powder diffraction patterns of complexes 1 (top) and 2 (bottom) registered at 296 K using CuK $\alpha$  radiation.