

Supplementary Information

A novel metal-organic framework based on bismuth and trimesinic acid: synthesis, structure and properties

Guanzhi Wang,^a Yuanyuan Liu,^{*a} Baibiao Huang,^{*a} Xiaoyan Qin,^a Xiaoyang Zhang^a and Ying Dai^b

^a State Key Laboratory of Crystal Materials, Shandong University, Jinan 250100, P. R. China.

^b School of Physics, Shandong University, Jinan 250100, P. R. China.

Email: yyliu@sdu.edu.cn (Yuanyuan Liu);

bbhuang@sdu.edu.cn (Baibiao Huang)

1 Experimental Section

All chemicals used were of analytical grade, and were used without further purification.

1.1 Synthesis of Bi-BTC.

The compound 1 was synthesized by solvothermal method. In the experiment, $\text{Bi}(\text{NO}_3)_3 \cdot 5\text{H}_2\text{O}$ (0.2360g, 0.49 mmol) and H_3BTC (0.1930g, 0.92 mmol) were added in a mixture of DMF/MeOH (5mL, 1:3). The solution was then sealed in a 100 mL Teflon-lined autoclave, heated to 120°C for 45h, kept at this temperature for 12h and then cooled down slowly at the rate of 2°C/min to room temperature. Colorless crystals were washed by DMF, methanol and then dried in air at 60°C.

1.2 Characterization

The X-ray power diffraction (XRD) data of the as-prepared sample was obtained by an X-ray powder diffraction (Bruker AXS D8). Thermogravimetry-differential thermal analysis (TG/DTA) was performed using a Diamond TG/DTA analyzer at a heating rate of 5°C/min in air. Fourier transform infrared (FT-IR) spectra was performed on a Bruker ALPHA-T spectrometer using KBr pellets. The UV-vis diffuse reflectance spectra was determined by a Shimadzu UV 2550 recording spectrophotometer with BaSO_4 as a reference. Steady state fluorescence was measured on a high sensitivity fluorescence spectrometer (Edinburgh FS920). The pump source is an Opolette HE 355 II tunable laser system with a laser pulse width of 5 ns and repetition rate of 20Hz.

1.3 Single- Crystal Structure Determination

The data was collected using a Bruker SMART APEXII CCD area detector on a D8 goniometer and with graphite-monochromated and 0.5 mm-Mono Cap-collimated Mo-K α radiation ($\lambda = 0.71073 \text{ \AA}$) at room temperature in ω scan method. The INTEGRATE program in APEX2 software was used for reduction and cell refinement. Multi-scan absorption corrections were applied by the SCALE program for area detector. The structure was solved by the direct method and refined using the

full-matrix least-squares method on F^2 by program SHELX-97. All the atoms were refined anisotropically except hydrogen atoms and hydrogen atoms were placed in idealized positions.

The structure was refined as a twin. The unit cell includes large regions of disordered solvent molecules and could not be modeled properly. PLATON/SQUEEZE was employed to calculate the diffraction contribution from the solvent molecules and, thus produced a set of solvent-free diffraction intensities, and the results were appended to the bottom of the CIF file. According to the SQUEEZE calculations combined with the TG/DTA data, the solvent region are corresponding to nearly 1 DMF and 2 methanol molecules per unit cell. So the tentative formula for this compound is $[\text{Bi}(\text{BTC})(\text{DMF})]\cdot\text{DMF}\cdot 2 (\text{CH}_3\text{OH})$. The contents of the solvent region are included in the unit cell contents but not included in the refinement model.

1.3 Photocatalytic reaction

Photocatalytic oxygen production reaction was performed in a top-irradiation vessel which is connected to a glass-enclosed gas circulation system. In the procedure, 50 mg of the sample and 100mg AgNO_3 were mixed in 50mL aqueous solution with constant stirring. The reaction temperature was kept at 5°C . The amount of O_2 evolution was determined by a gas chromatograph (Techcomp GC7890 II).

2 Characterization of Bi-BTC

2.1 Crystal data and structure refinement

Table S1. Crystal data and structure refinement for Bi-BTC

| | |
|---|---|
| Empirical formula | C ₁₇ H ₂₅ BiN ₂ O ₁₀ |
| Formula weight | 626.19 |
| Temperature | 293 K |
| Wavelength | 0.71073 Å |
| Crystal system | Monoclinic |
| Space group | P2 ₁ /n |
| Unit cell dimensions | $a = 10.137(2)$ Å $\alpha = 90^\circ$ $b = 15.017(3)$ Å $\beta = 99.441(2)^\circ$ $c = 11.422(3)$ Å $\gamma = 90^\circ$ |
| Volume | 1715.2(7) Å ³ |
| Z | 4 |
| Density (calculated) | 2.425 g·cm ⁻³ |
| Radiation type | Mo K α |
| Absorption coefficient (μ) | 10.306 mm ⁻¹ |
| F(000) | 912 |
| Crystal size | 0.143 × 0.125 × 0.040 mm ³ |
| ω range for data collection | 2.260 to 27.627° |
| Index ranges | -13 ≤ h ≤ 13, -19 ≤ k ≤ 19, -14 ≤ l ≤ 14 |
| Reflections collected | 19035 |
| Independent reflections | 19035 [R _{int} = 0.0675] |
| Final R ₁ values [$I > 2\sigma(I)$] | 0.0485 |
| Final wR(F ²) values ($I > 2\sigma(I)$) | 0.1470 |
| Final R ₁ values (all data) | 0.0618 |
| Final wR(F ₂) values (all data) | 0.1592 |
| Goodness of fit on F ² | 0.871 |

2.2 XRD pattern

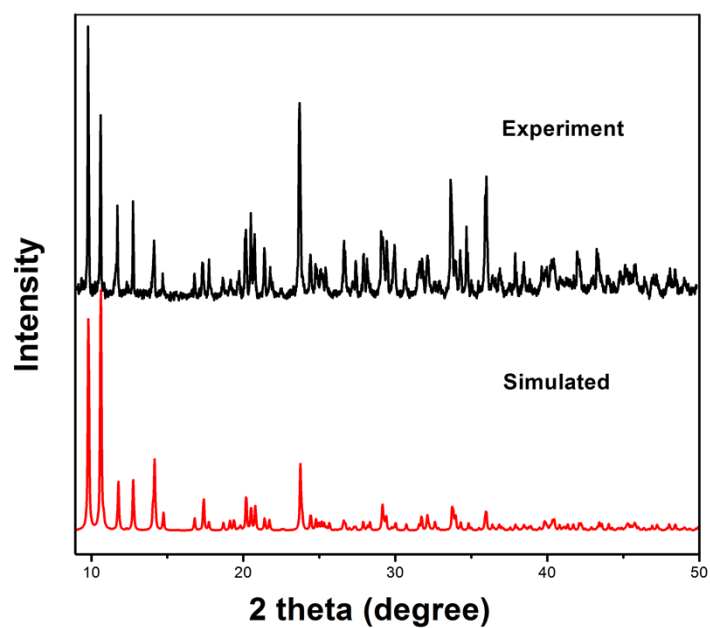


Figure S1. XRD patterns of the as-prepared Bi-BTC.

2.3 The FT-IR spectra of Bi-BTC

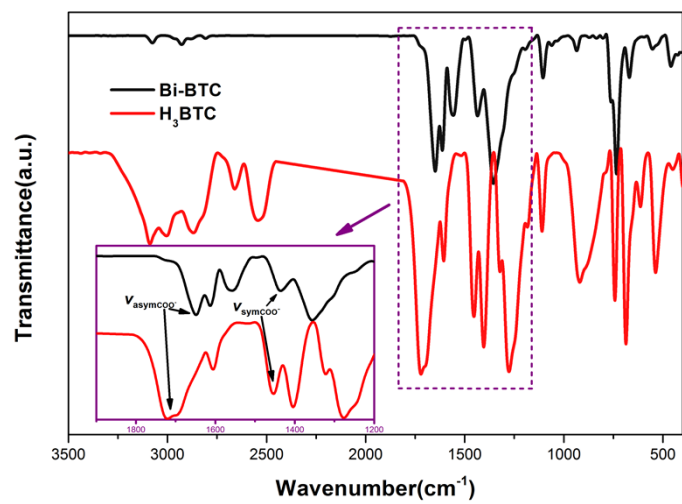


Figure S2. The FT-IR spectra of Bi-BTC

2.4 TG/DTA of Bi-BTC

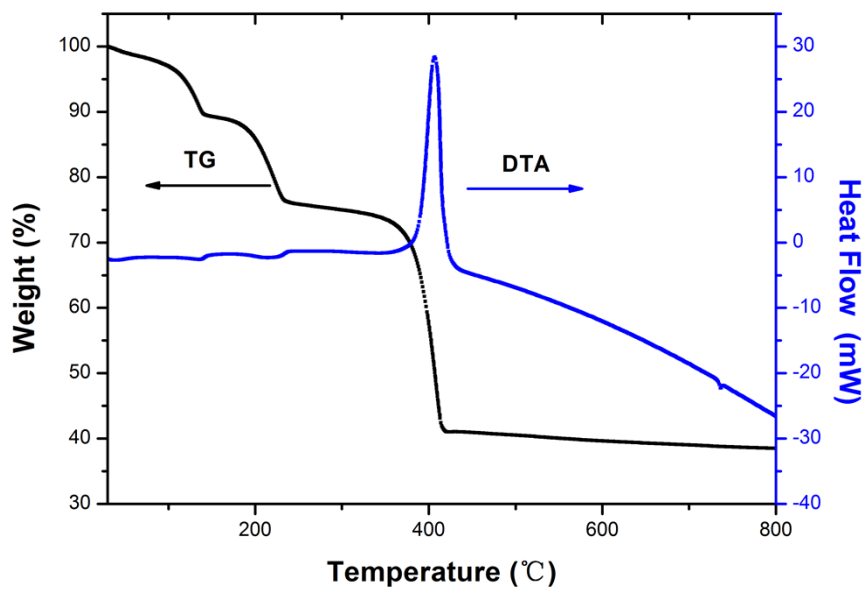


Figure S3. The TG/DTA curve of Bi-BTC.