Supporting Information

Safe disposal of radioactive iodide by Ag₂O grafted Nb₂Na₂O₆·H₂O nanofibers

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Fig.S1. Removal of Γ anions from aqueous solution by $Ag_2O-Na_2Nb_2O_6 \cdot H_2O$ with different Na:Ag ratios.



Fig.S2. The effect of temperature on I⁻ anions absorption $(Ag_2O-Na_2Nb_2O_6 \cdot H_2O)$



Fig.S3. The TEM images of $Na_2Nb_2O_6 \cdot H_2O$ nanofibers. (a) TEM at low magnification. (b) HRTEM image of a single nanoparticle, and the fast flourier transformation (FFT) shows inset. (c) SAED pattern.



Fig.S4. Formation of Ag₂O nanoparticles on the hydroxyl group rich $Na_2Nb_2O_6 \cdot H_2O$ nanofibers surface and subsequent removal of I⁻ ions



Fig.S5. Raman spectra of $Na_2Nb_2O_6 \cdot H_2O$ nanofibers, Ag_2O nanocrystals anchored $Na_2Nb_2O_6 \cdot H_2O$ nanofibers and AgI nanocrystals coated $Na_2Nb_2O_6 \cdot H_2O$ nanofibers.



Fig.S6. SEM images and EDS spectra of the $Na_2Nb_2O_6 \cdot H_2O$ nanofibers, Ag_2O nanocrystals anchored $Na_2Nb_2O_6 \cdot H_2O$ nanofibers and AgI nanocrystals coated $Na_2Nb_2O_6 \cdot H_2O$ nanofibers.



Fig. S7. TEM images of Ag₂O nanocrystals anchored on the surface of Na₂Nb₂O₆·H₂O nanofibers. (a) Na₂Nb₂O₆·H₂O nanofibers with small amount of Ag₂O nanocrystals. (b) Na₂Nb₂O₆·H₂O nanofibers with large amount of Ag₂O nanocrystals.



Fig.S8. XPS spectra of $Na_2Nb_2O_6 \cdot H_2O$ nanofibers, Ag_2O nanocrystals anchored $Na_2Nb_2O_6 \cdot H_2O$ nanofibers and AgI nanocrystals coated $Na_2Nb_2O_6 \cdot H_2O$ nanofibers.



Fig.S9. The leaching concentration of Γ anions as a function of temperature